

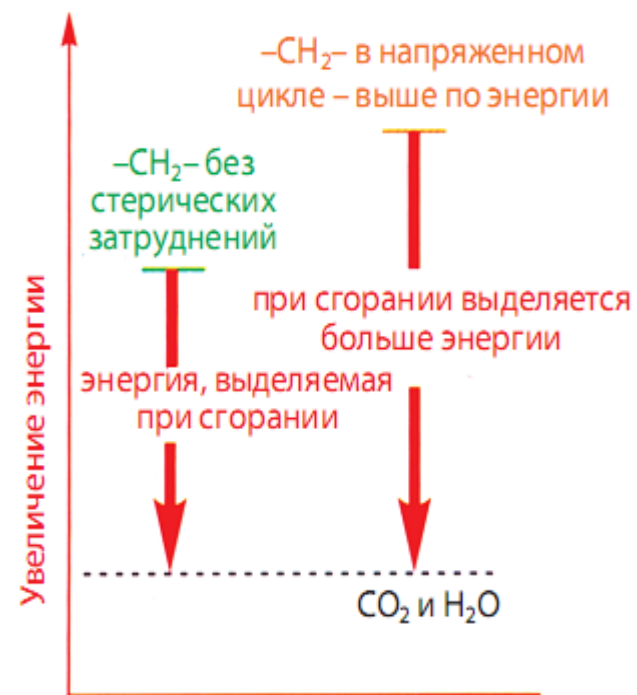
7. Образование циклов

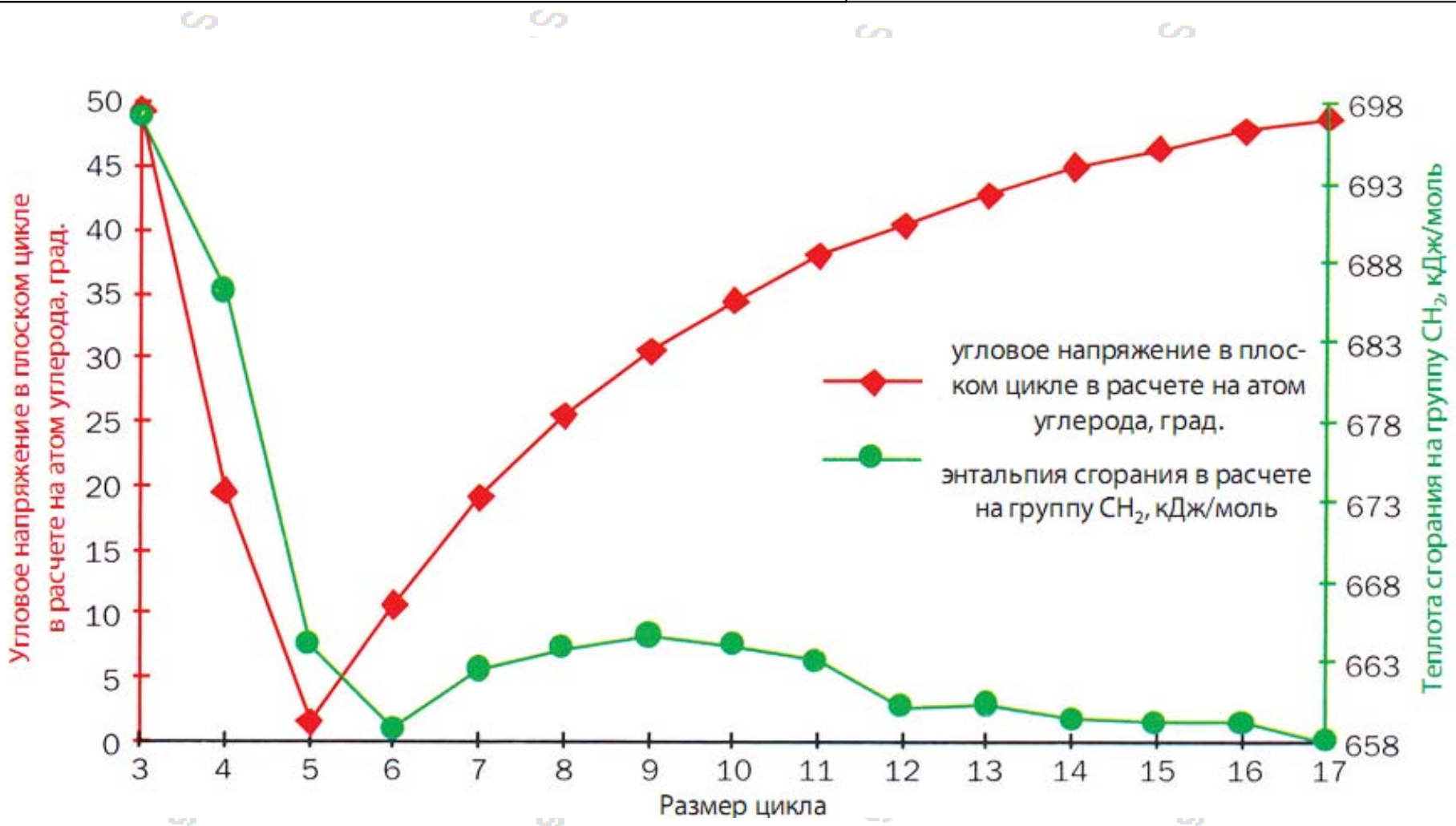
- Общие подходы к синтезу циклических молекул, понятие о циклизации и циклоприсоединении, способы сборки циклов
- Особенности реакций циклизации: правила Болдуина
- Трехчленные циклы: циклопропаны, азиридины, оксираны
- Четырехчленные циклы: циклобутаны, оксетаны
- Циклизации «анион-алкен/алкин», «радикал-алкен/алкин», «катион-алкен/алкин»
- Пятичленные циклы
- Шестичленные циклы

Общая энергия напряжения является суммой трех основных составляющих: углового напряжения, напряжения заслоненных, противостоящих С-Н связей и так называемого трансаннулярного напряжения. Угловое напряжение (синонимы напряжение углерод-углеродных связей или напряжение Байера) вызвано растяжением или сжатием валентных углов и отклонением их от тетраэдрического $109^{\circ}28'$. Напряжение заслоненных С-Н связей (синонимы торсионное напряжение или напряжение Питцера) по своей природе аналогично отталкиванию атомов водорода в заслоненной конформации этана и других предельных углеводородов для двух соседних заслоненных С-Н связей; энергии этого взаимодействия оценивается приблизительно в 1 ккал/моль. Трансаннулярное напряжение или напряжение Прелога обусловлено взаимодействием в пространстве двух или большего числа атомов водорода при атомах углерода на противоположных концах цикла. Его следует принимать во внимание, главным образом, для средних циклов C_8-C_{11} .

Таблица 18.2. Теплоты сгорания некоторых линейных алканов

Линейный алкан	n в $\text{CH}_3(\text{CH}_2)_n\text{CH}_3$	$-\Delta H_{\text{сгорания}}$, кДж/моль
Этан	0	1560
Пропан	1	2220
Бутан	2	2877
Пентан	3	3536
Гексан	4	4194
Гептан	5	4853
Октан	6	5511
Нонан	7	6171
Декан	8	6829
Ундекан	9	7487
Додекан	10	8148

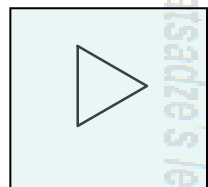




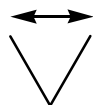
Напряжение по Питцеру и Прелогу

- Общие принципы образования циклов

Образование циклов

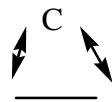


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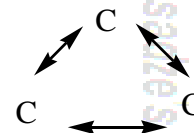


Циклизация

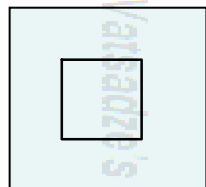
Циклоприсоединение



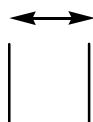
[1+2]



[1+1+1]

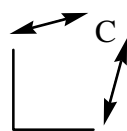


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Циклизация

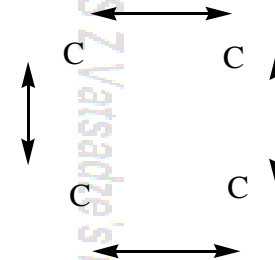
Циклоприсоединение



[1+3]



[2+2]



[1+1+1+1]

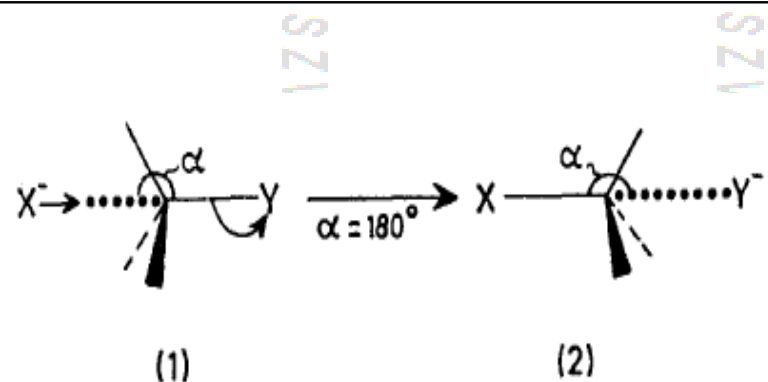


FIGURE 1: *Tet.*

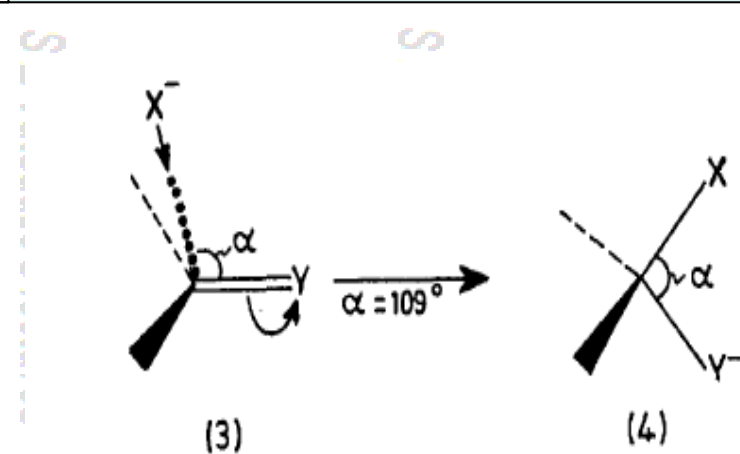
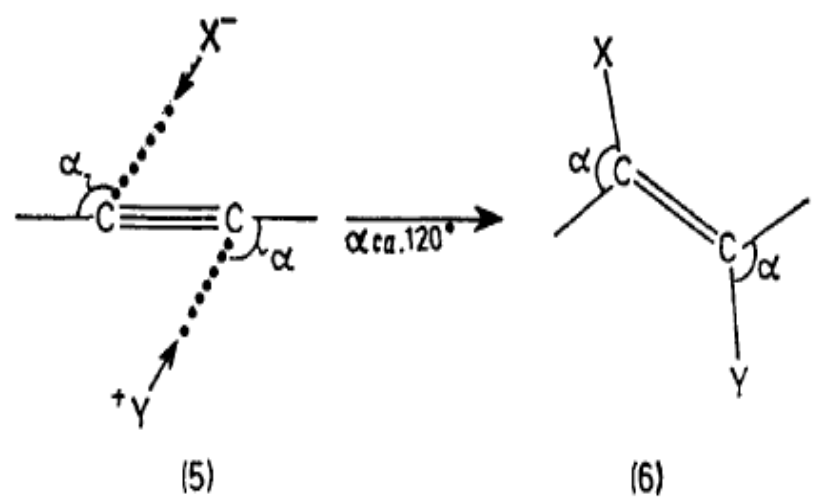


FIGURE 2: *Trig.*

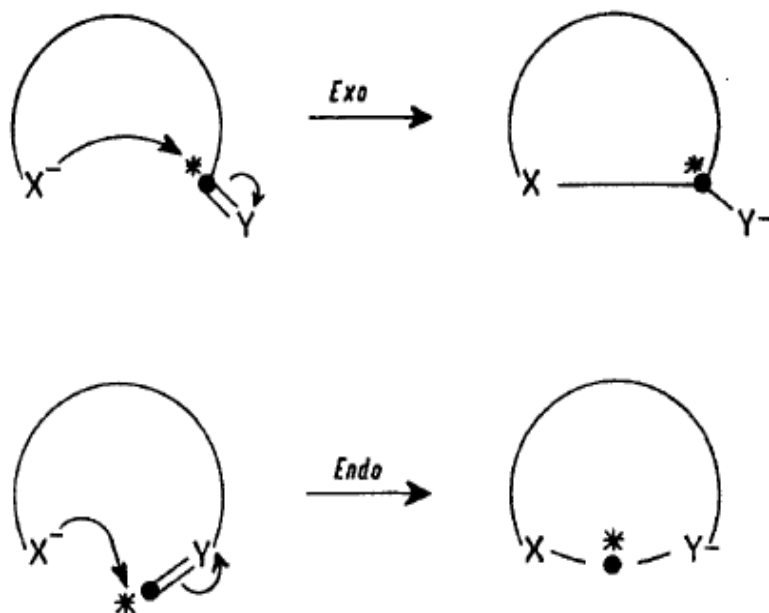
SZ Varsadze's lectures



SZ Varsadze's lectures

FIGURE 3: *Dig.*

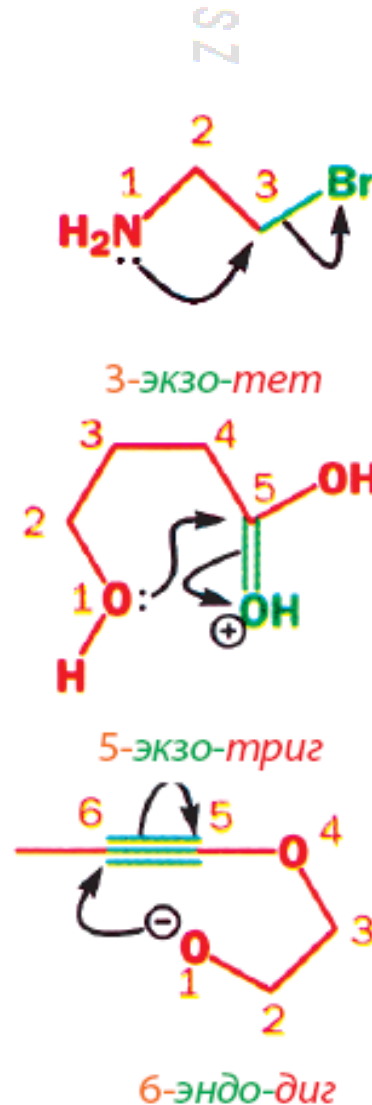
RING-FORMING reactions are important and common processes in organic chemistry. I now adumbrate a set of simple rules which I have found useful in predicting the relative facility of different ring closures. I believe these



will be useful to organic chemists, especially in planning syntheses. Also these rules indicate certain experiments which may be helpful to define more precisely their limits. The rules are of a stereochemical nature and it is likely that unambiguous cases of all the possibilities I will discuss are as yet unknown.

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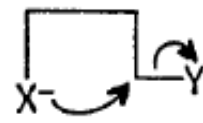


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3-Exo-Tet

S Z Vatsadze's



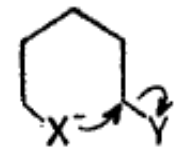
4-Exo-Tet

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5-Exo-Tet

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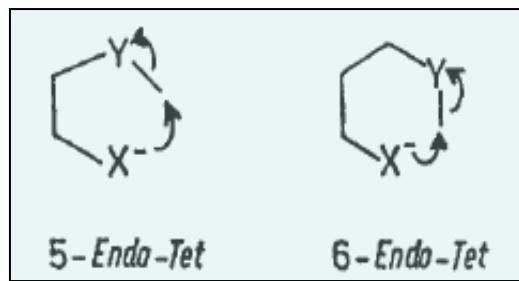


6-Exo-Tet

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7-Exo-Tet



5-Endo-Tet

6-Endo-Tet

SCHEME 2: Tetrahedral

tures

tures

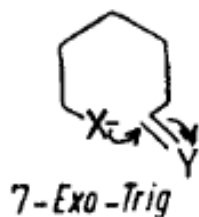
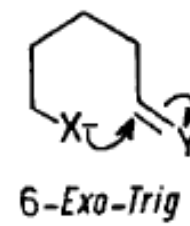
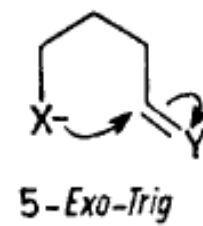
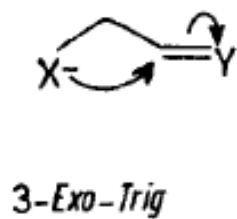
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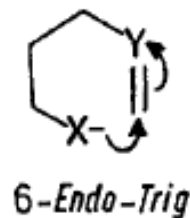
SZ1

SZ

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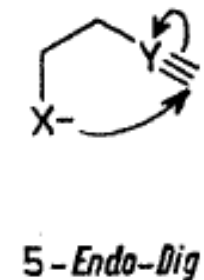
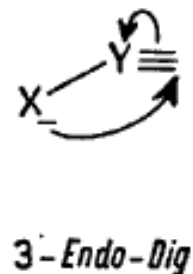
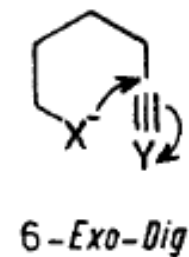
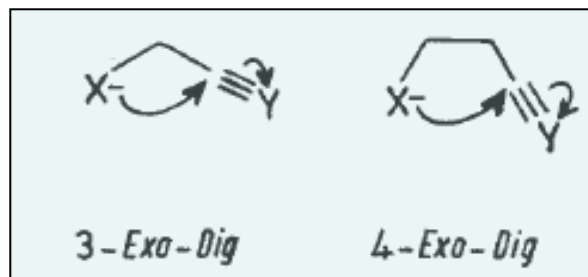
SCHEME 3: Trigonal

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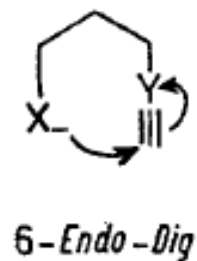
SZVa

SZV

SZV



S Z Vatsadze's lectures



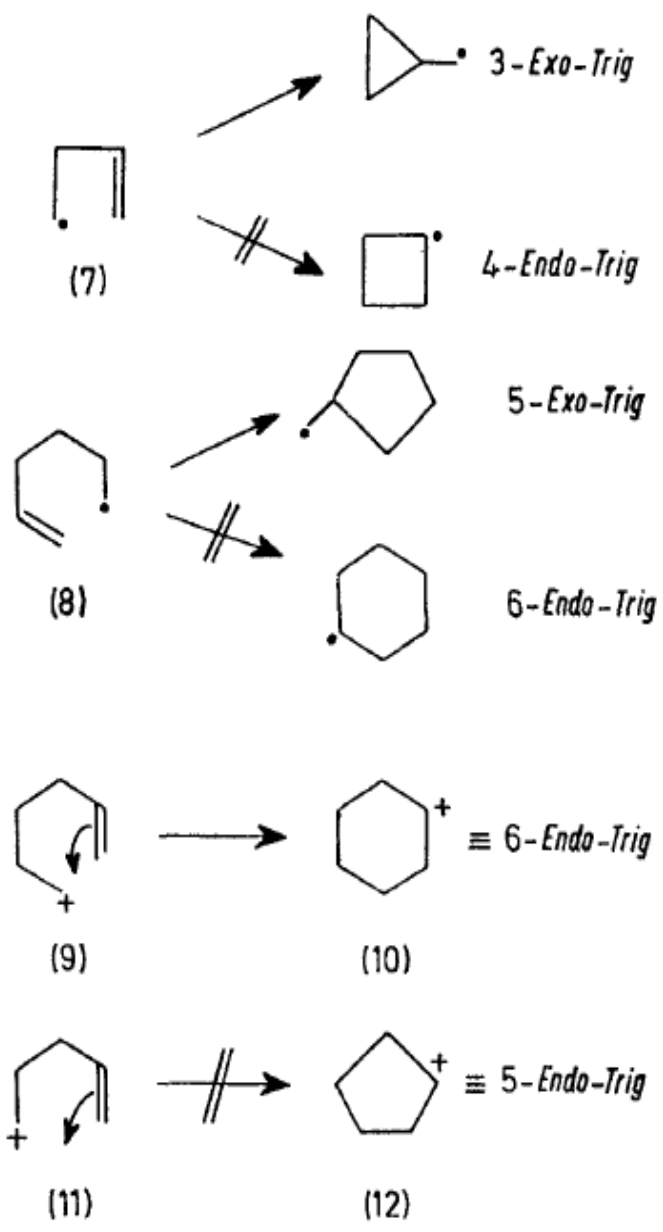
SCHEME 4: Digonal

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S Z Vatsadze's lectures

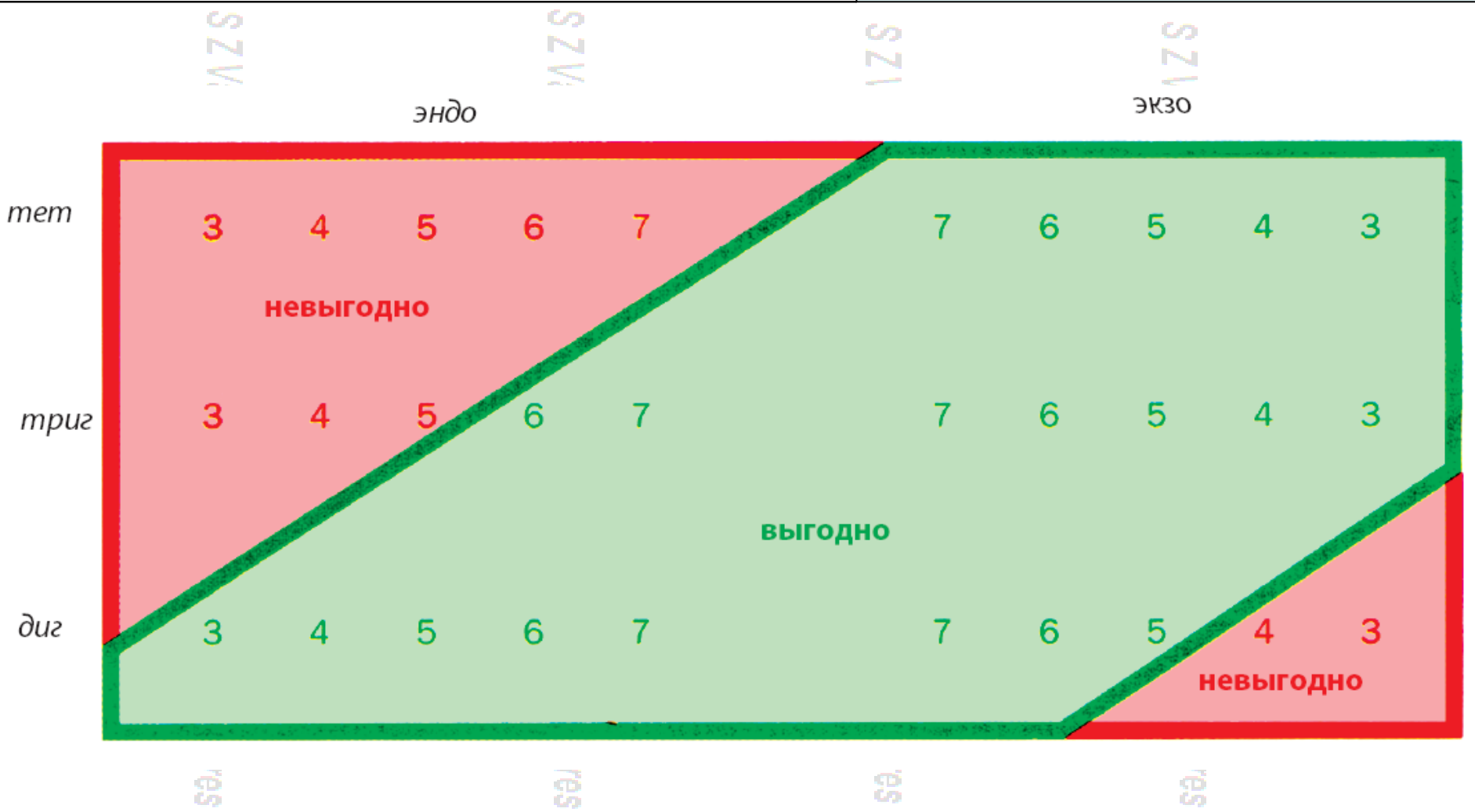
S Z Vatsadze's lectures

S Z Vatsadze's lectures



• Правила Болдуина

Образование циклов





1. Циклопропаны:

- реакция Симмонса-Смита
- diaзосоединения
- илиды серы
- S_N2 -реакции

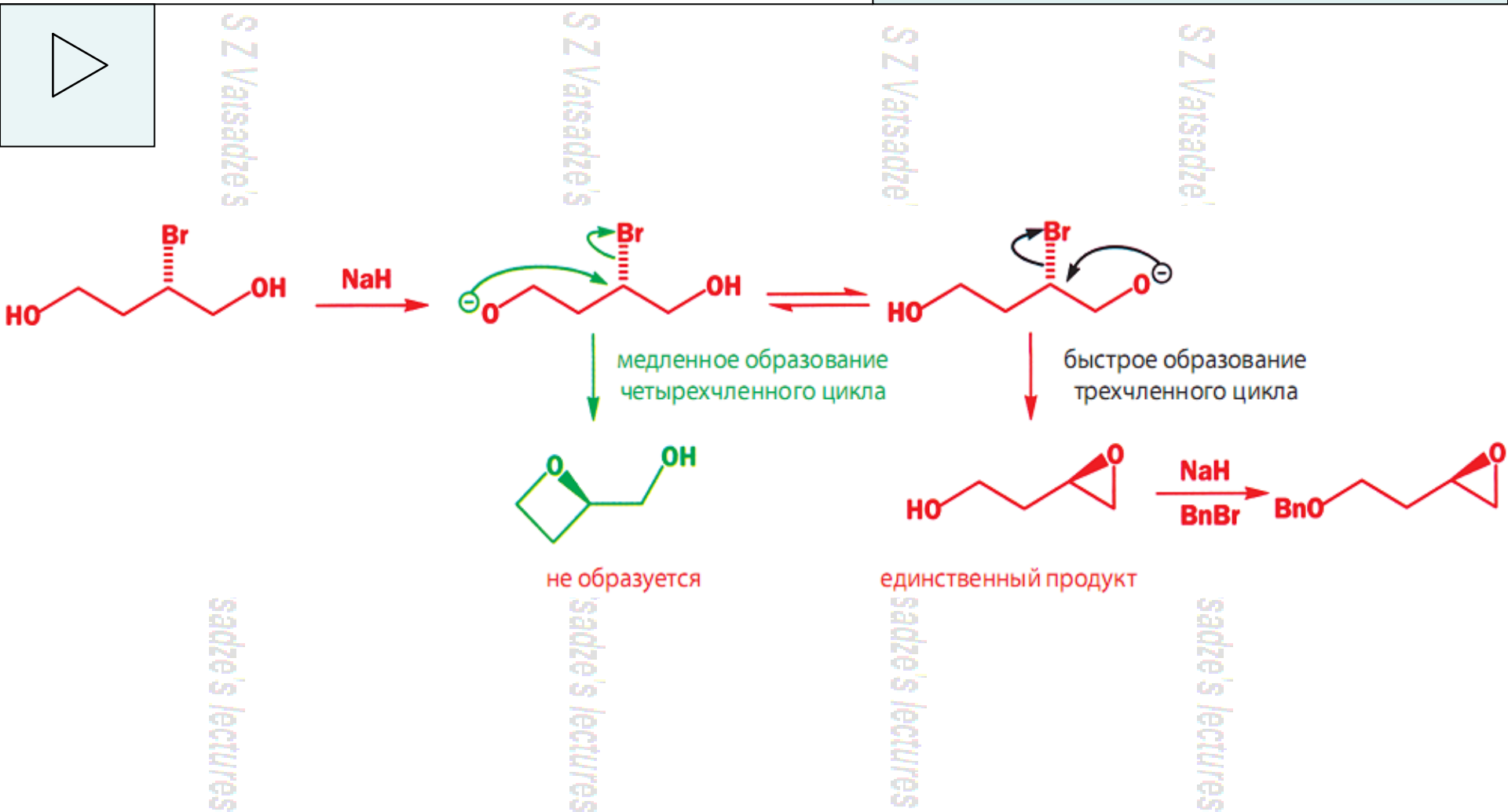
2. Эпоксиды:

- надкислоты, гидропероксиды, диоксираны
- галогенгидрины
- конденсация Дарзана
- илиды серы
- эпоксидирование, катализ. переходными металлами

3. Азиридины:

- нитрены
- S_N2 -реакции

S_N2 замыкание цикла

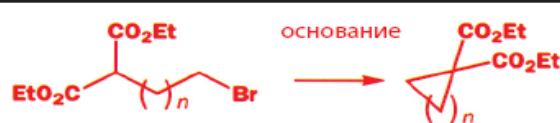


• S_N2 замыкание цикла



Таблица 42.3. Скорости реакций замыкания циклов

Размер цикла	Продукт	Относительная скорость ^a	Продукт реакции ^b	Относительная скорость ^a	Оценка реакции
3		0,07			С умеренной скоростью
4		0,001		0,58	Медленная
5		100		833	Очень быстрая
6		1		1	Быстрая
7		0,002		0,0087	Медленная
8				0,00015	Очень медленная



• Циклопропанирование



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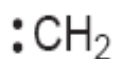
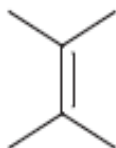
Parman, Schweizer *Org. React.* **1963**, 13, 55.

Moss *Acc. Chem. Res.* **1989**, 22, 15.

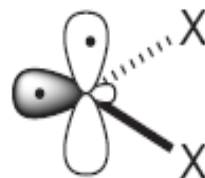
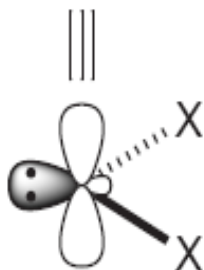
Acc. Chem. Res. **1980**, 13, 58.

Kostikov, Molchanov, Khlebnikov *Russ. Chem. Rev.* **1989**, 58, 654.

$2\pi^s + 2\omega^a$
cycloaddition



Addition of a singlet carbene proceeds by a concerted process in a *syn* fashion.



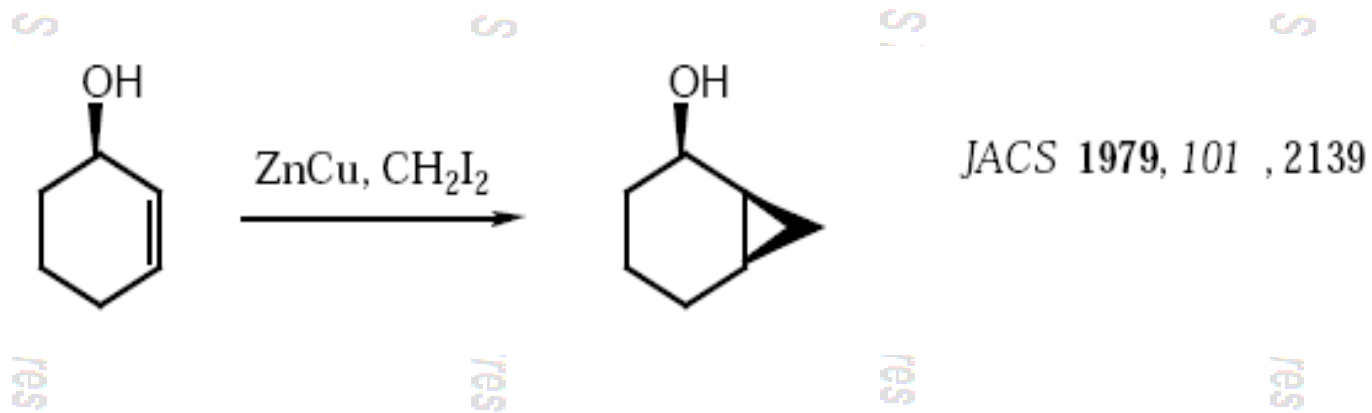
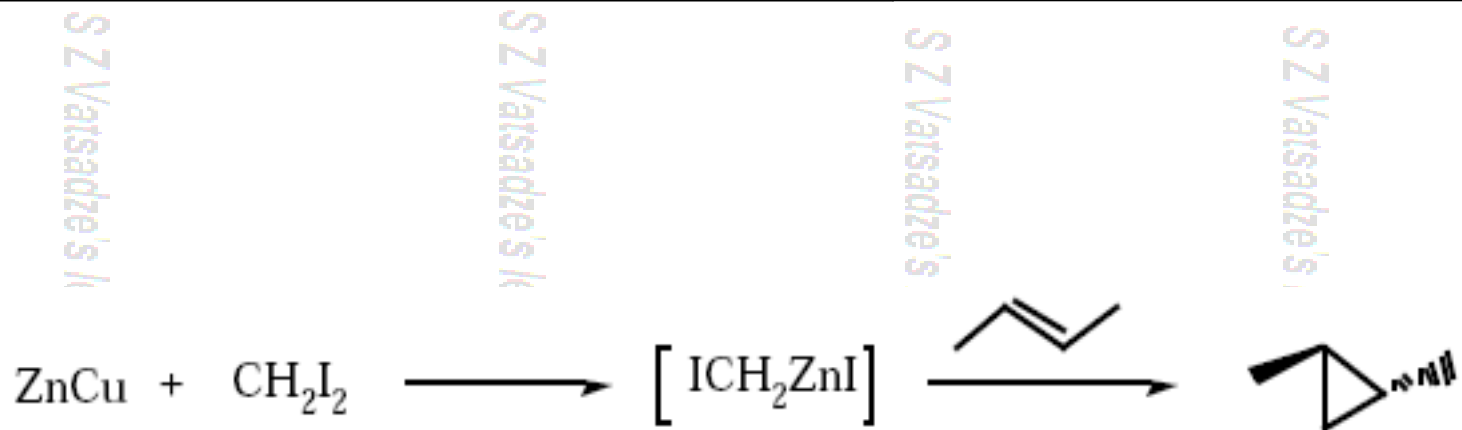
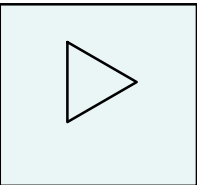
Triplet carbene behaves as a diradical.

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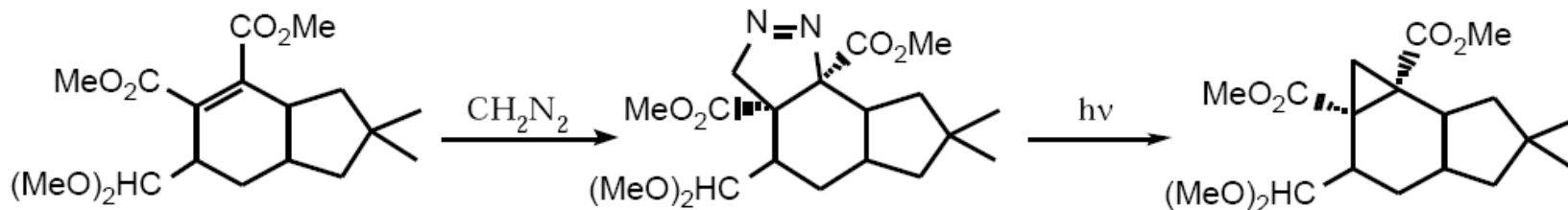


S Z Vaisa

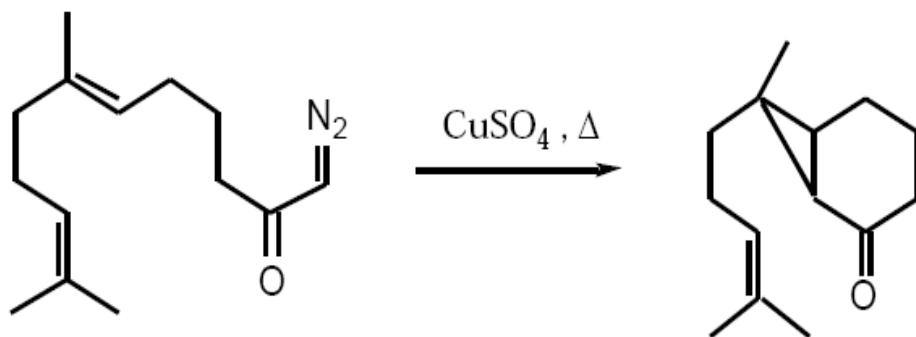
S Z Vaisa

S Z Vaisa

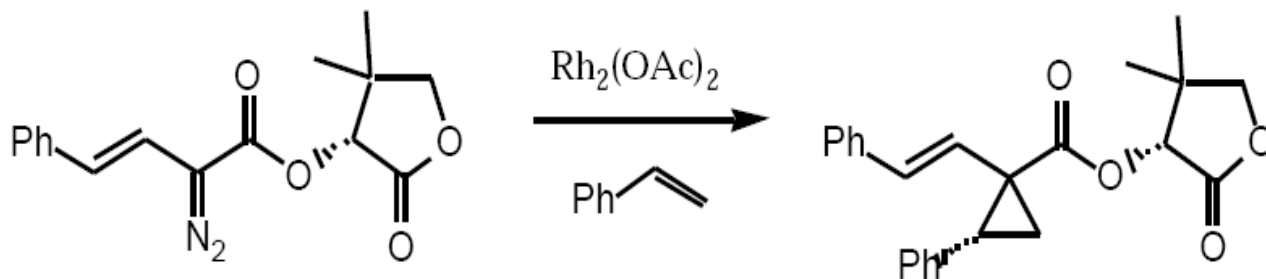
S Z Vaisa



JACS 1975,
97, 6075

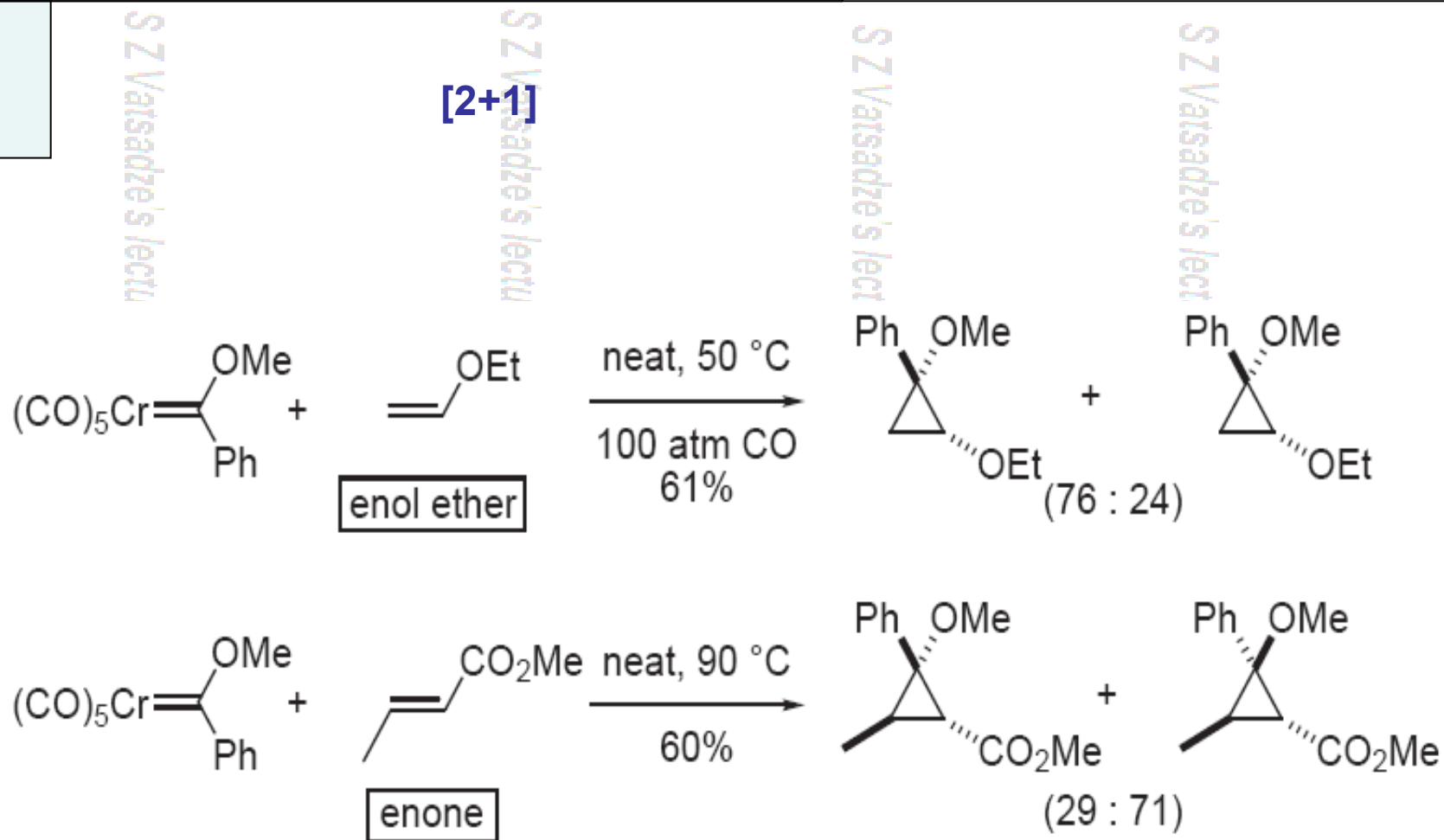


JACS 1970, 92, 3429
JACS 1969, 91, 4318



JACS 1993, 115, 9468

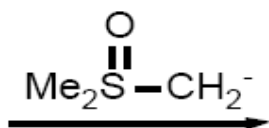
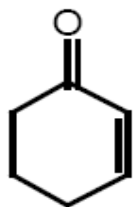
91 % yield
89 % de



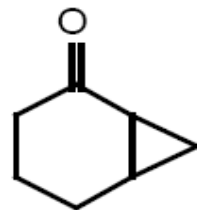
Fischer, Dötz *Chem. Ber.* **1972**, *105*, 3966.
Chem. Ber. **1972**, *105*, 1356.



S Z Vatsadze



S Z Vatsadze

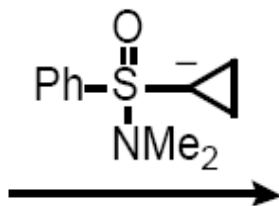
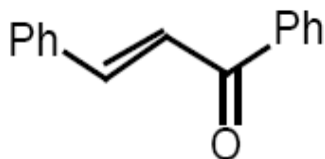


S Z Vatsadz

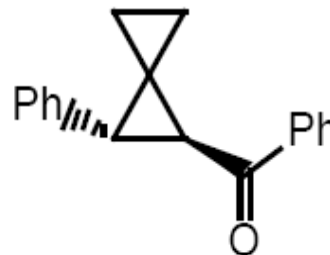
Tetrahedron
1987, 43, 2609

S Z Vatsadze's lectures

S Z Va



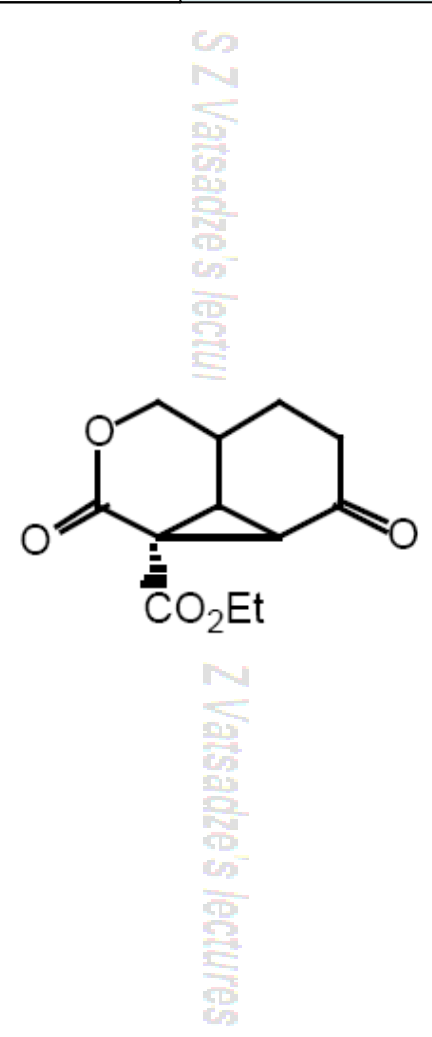
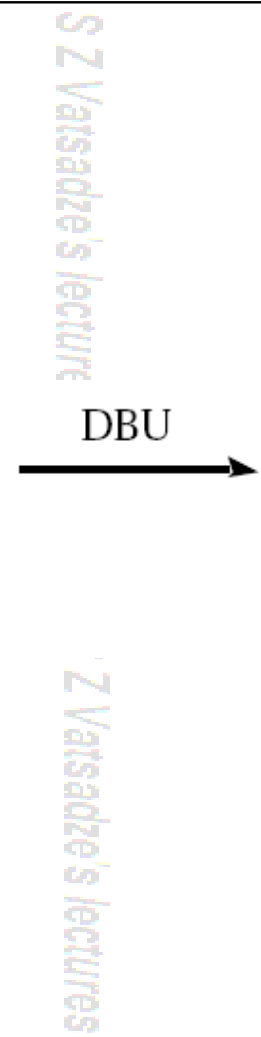
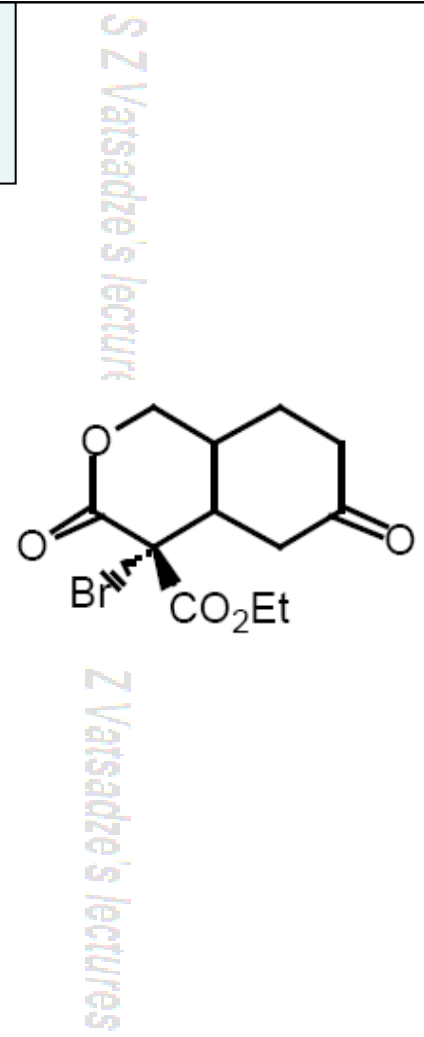
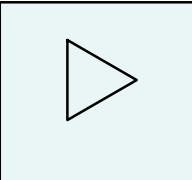
S Z Va



S Z Va

ACR 1973,
6, 341

S Z Va



S Z Vatsadze's lectu

JACS 1978,
99, 1940

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Циклизации типа «радикал-радикал»

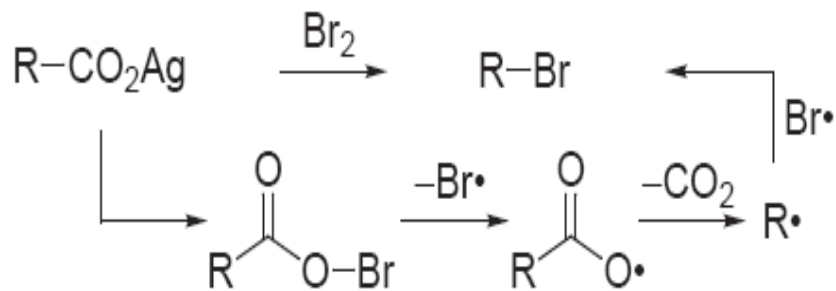


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Hunsdiecker, H.; Hunsdiecker, C. *Ber.* 1942, 75, 291.



Wiberg *Acc. Chem. Res.* 1984, 17, 379.

lectures

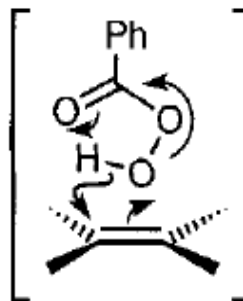
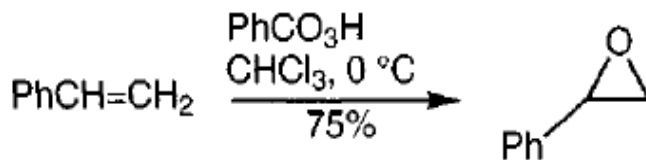
lectures

lectures

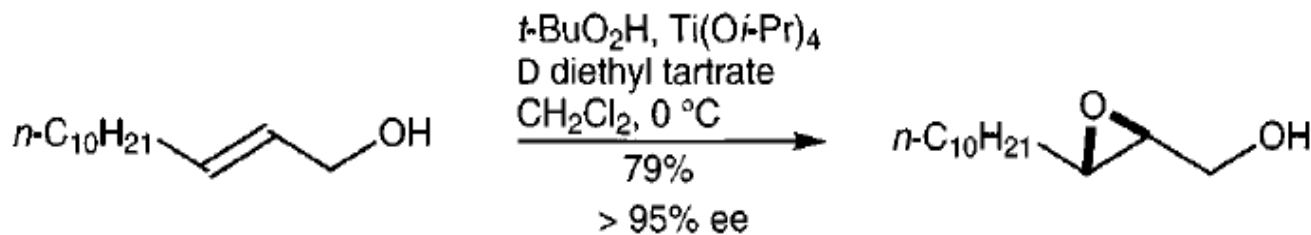
lectures



-надкислоты, гидропероксиды, диоксираны

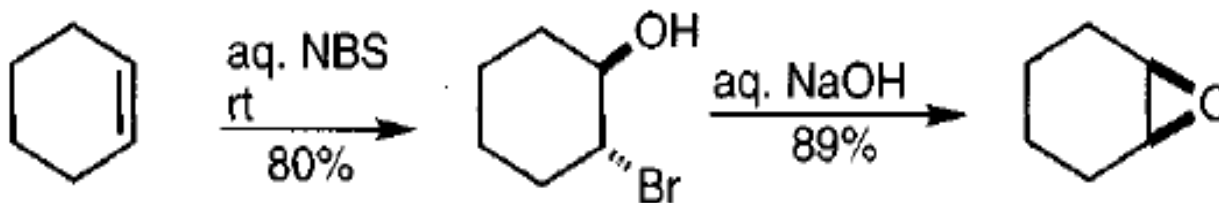


-эпокси́дирование, катализируемое переходными металлами
(Шарплесс, Якобсен)

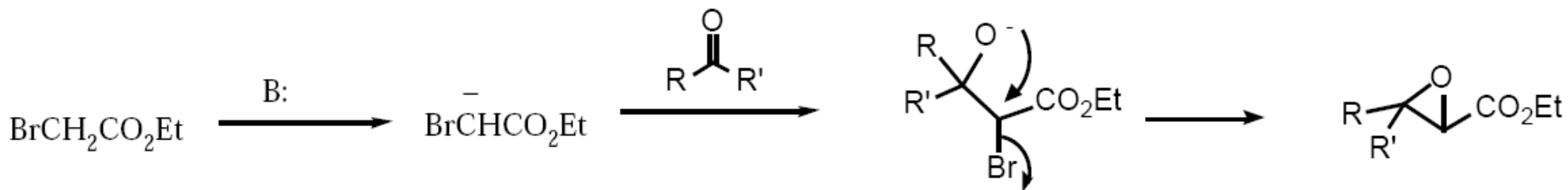




-галогенгидрины:



-конденсация Дарзана





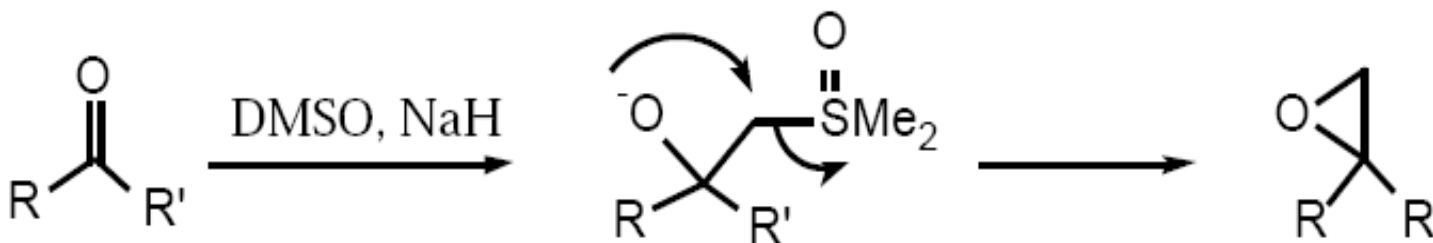
S Z Vatsadze's lectures

-илиды серы

S Z Vatsadze's lectures

S Z Vatsadze's lecture

S Z Vatsadze's lecture



sadze's lectures

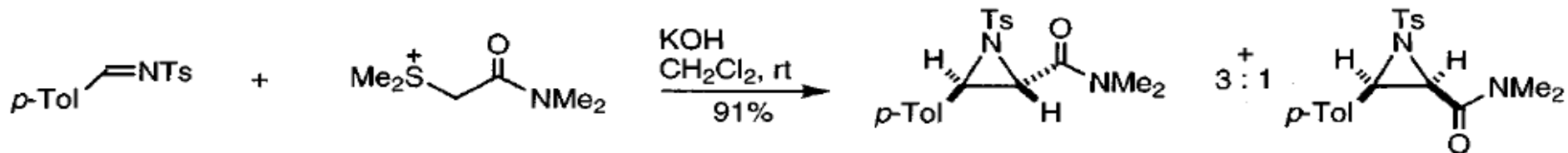
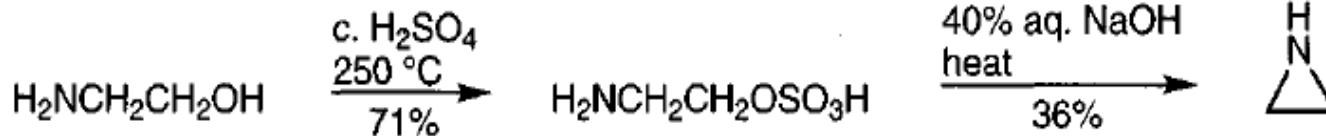
sadze's lectures

sadze's lectures

sadze's lectures

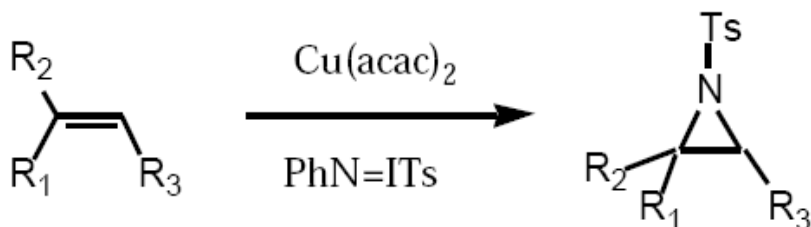


- S_N2-реакции



(не напоминает Кори-Чайковского?)

-нитрены:



J. Org. Chem. .1991, 56, 6744



1. Циклобутаны и циклобутены:

- [2+2]-циклоприсоединение кетены + алкены
- S_N2 -реакции
- ацилоиновая конденсация

- илиды серы

2. Оксетаны:

- [2+2]-циклоприсоединение (реакция Патерно-Бухи)
- S_N2 -реакции
- илиды серы
- β -лактоны

• [2+2]-циклоприсоединение

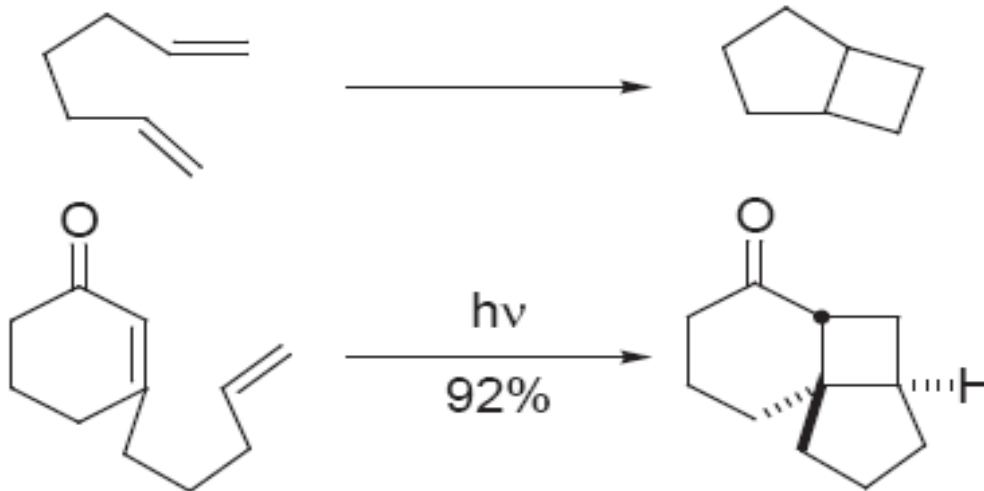


S Z Vatsadze's lectures

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S Z Vatsadze's lectures



S Z Vatsadze's lectures

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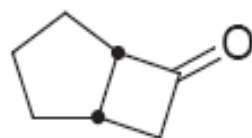
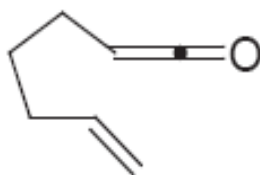
S Z Vatsadze's lectures

Cargill *Tetrahedron Lett.* 1978, 4465.



S Z Vaisadze's //

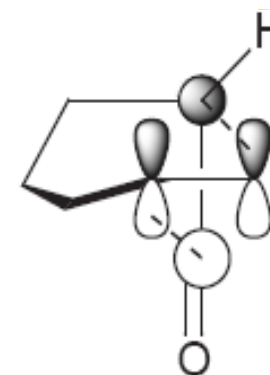
Org. React. **1995**, 45, 159.



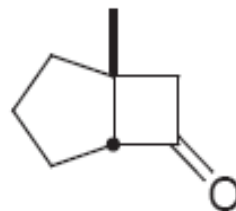
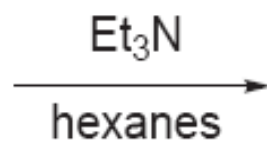
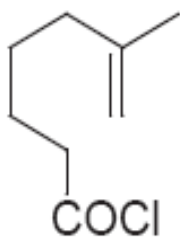
S Z Vaisadze's //

S Z Vaisadze's

S Z Vaisadze's



$2\pi^s + 2\pi^a$
Cycloaddition



Baldwin J. Chem. Soc., Chem. Commun. **1972**, 1337.

• Фишеровские карбены



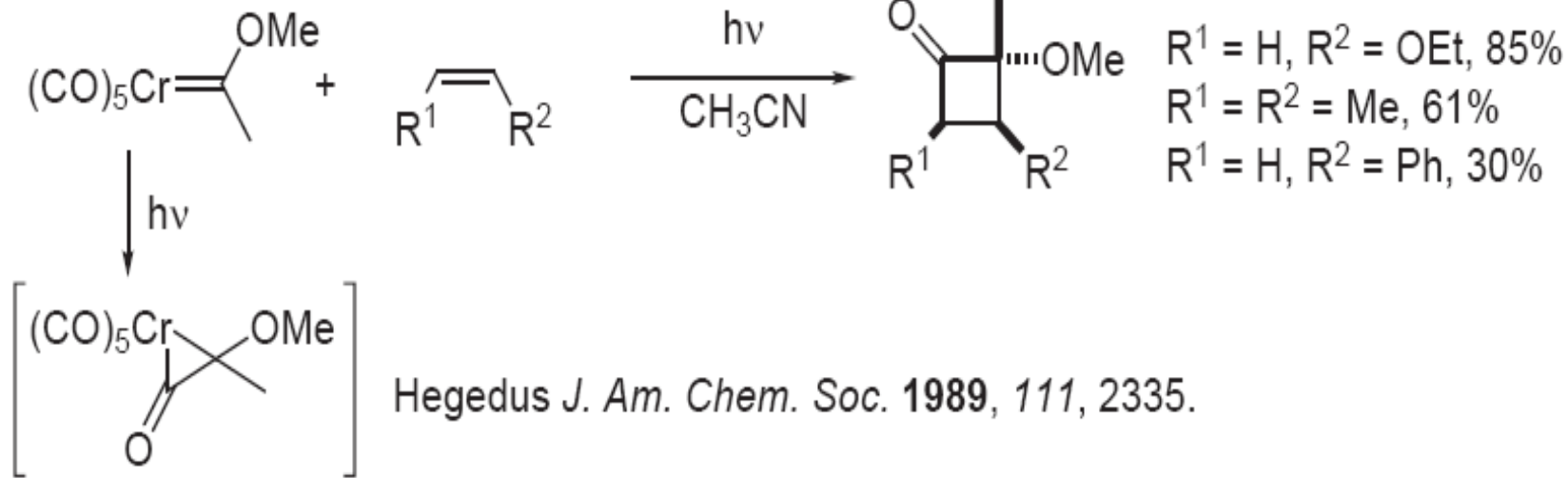
S Z Vatsadze's lectures

S Z Vatsadze's lectures

S Z Vatsadze's lectures

S Z Vatsadze's lectures

[2+1+1]



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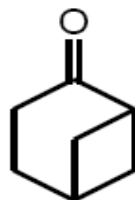
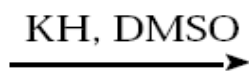
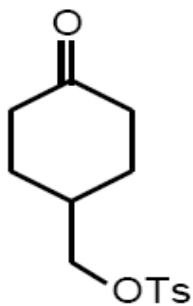
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SZ Vaisadze's lectures



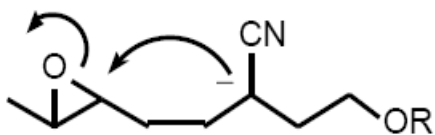
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SZ Vaisadze's

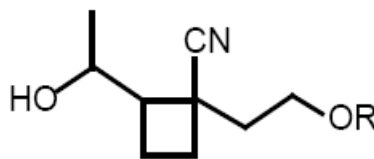
JACS 1980,
102, 1404

SZ Vaisadze's lectures

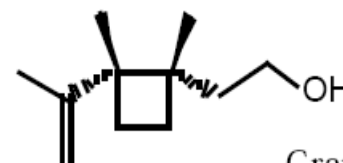
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SZV



SZV



JACS 1974, 96,
5268, 5272

SZV

Grandisol

is

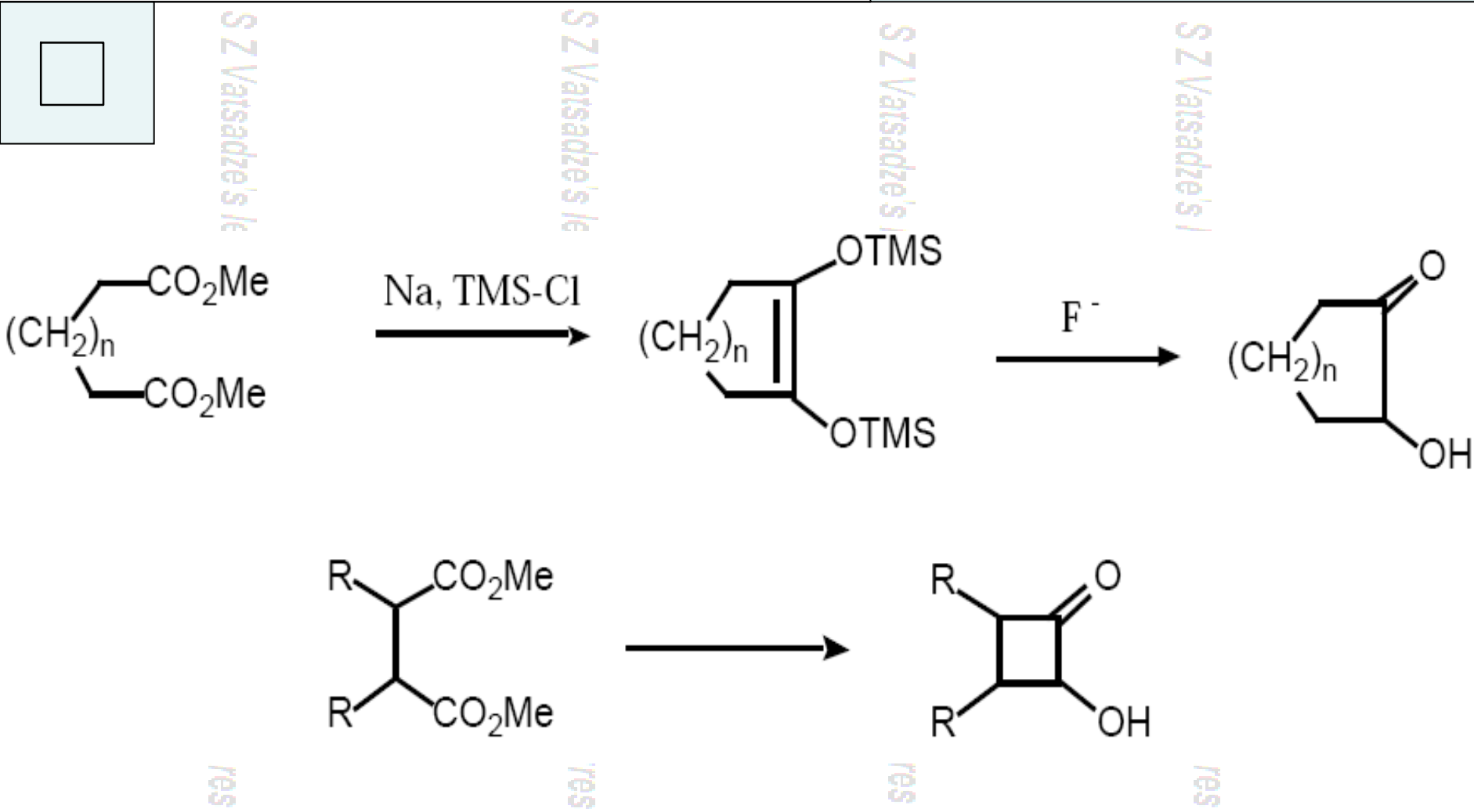
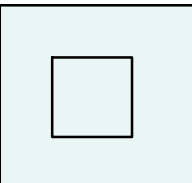
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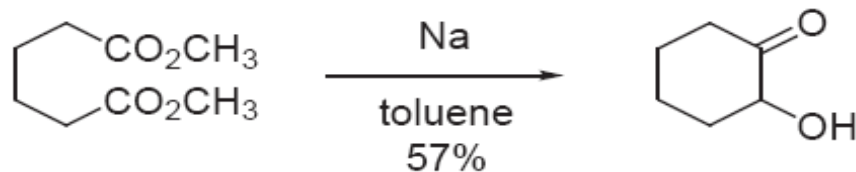
• Ацилоиновая конденсация

Образование циклов



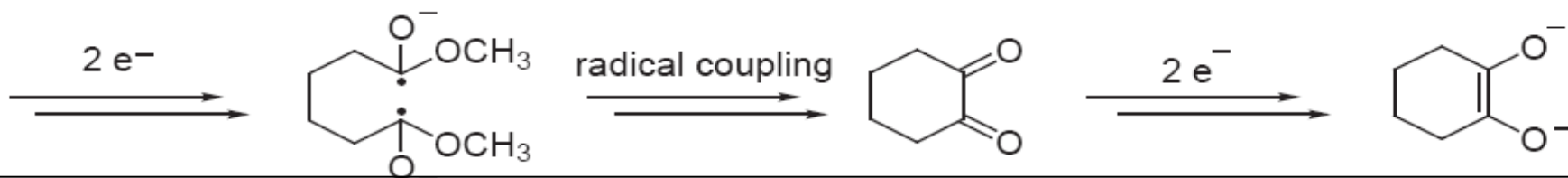
• Восстановительная димеризация карбонил

a. Acyloin Condensation

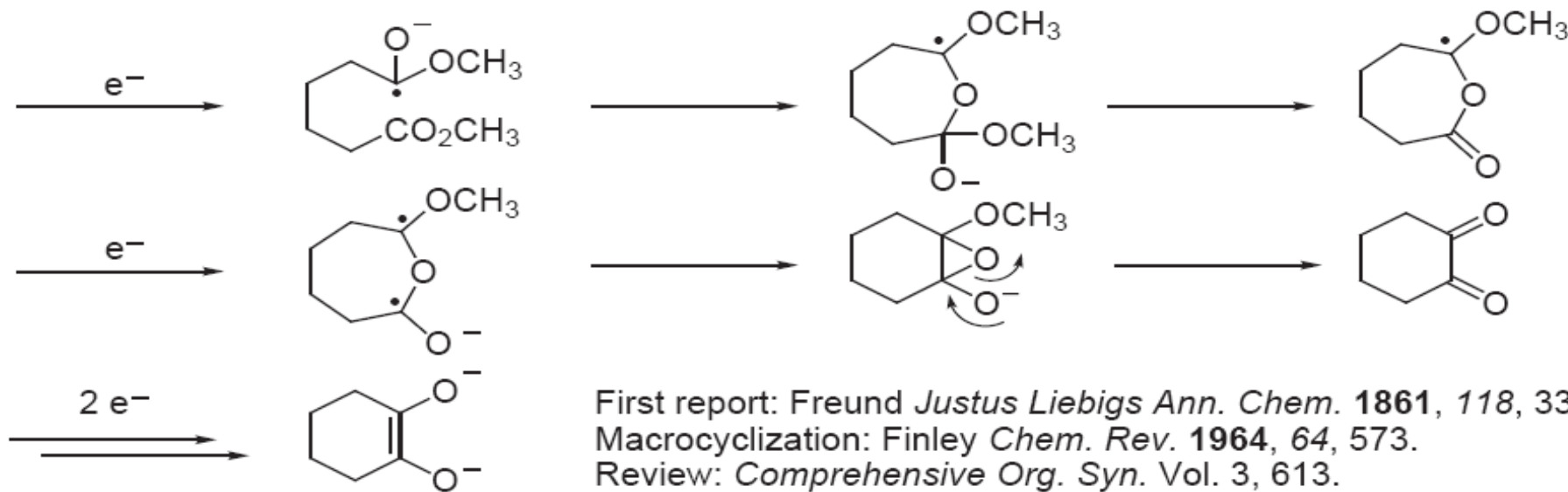


Sheehan *J. Am. Chem. Soc.* **1950**, *72*, 3376.

- Mechanism



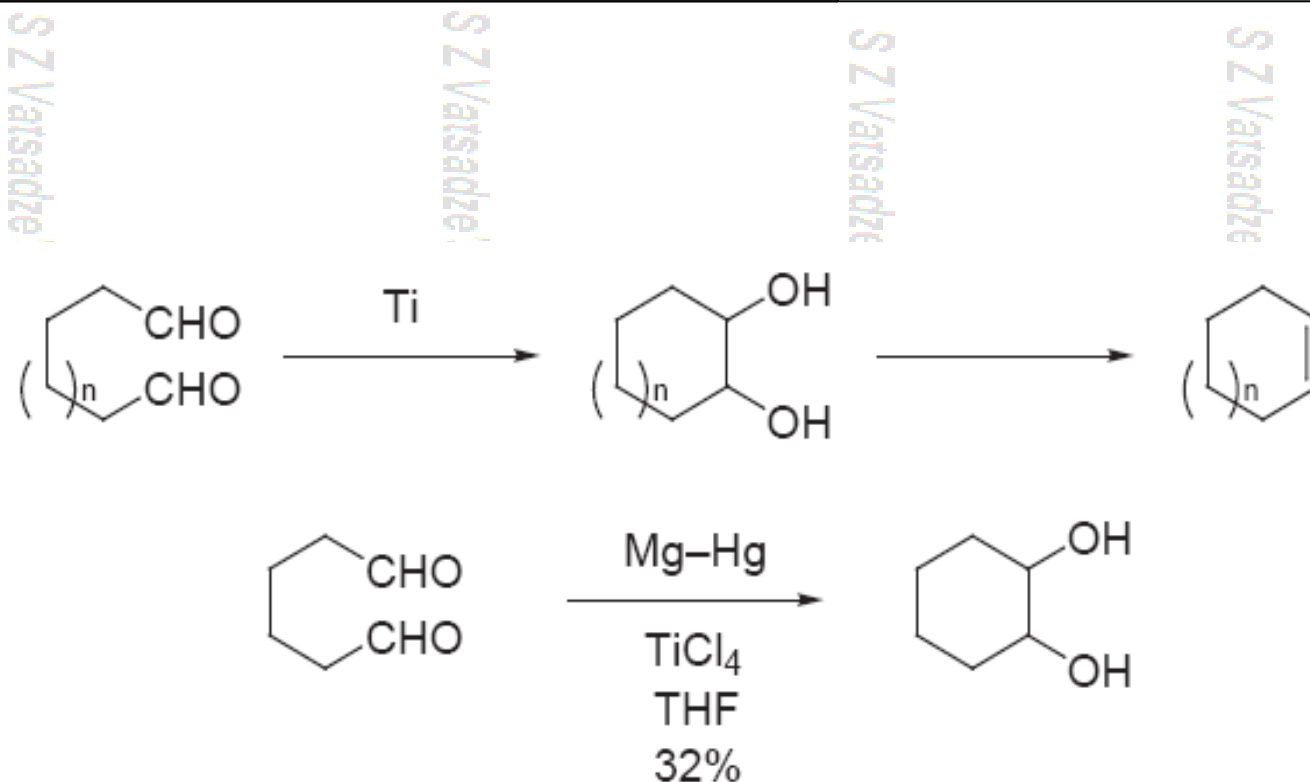
- Alternative



First report: Freund *Justus Liebigs Ann. Chem.* **1861**, *118*, 33.
 Macrocyclization: Finley *Chem. Rev.* **1964**, *64*, 573.
 Review: *Comprehensive Org. Syn.* Vol. 3, 613.

• Восстановительная димеризация карбонил

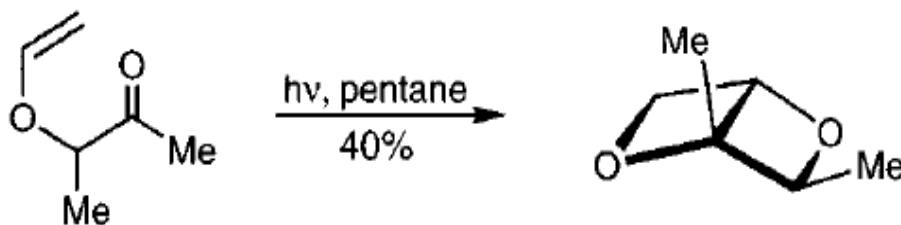
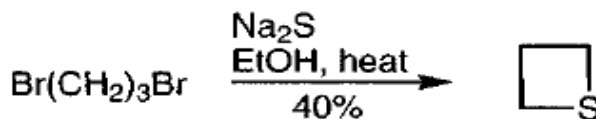
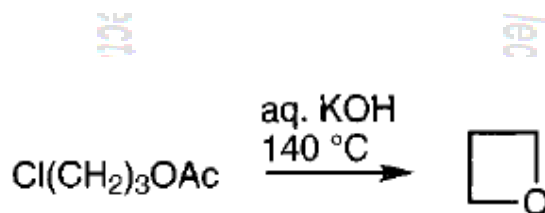
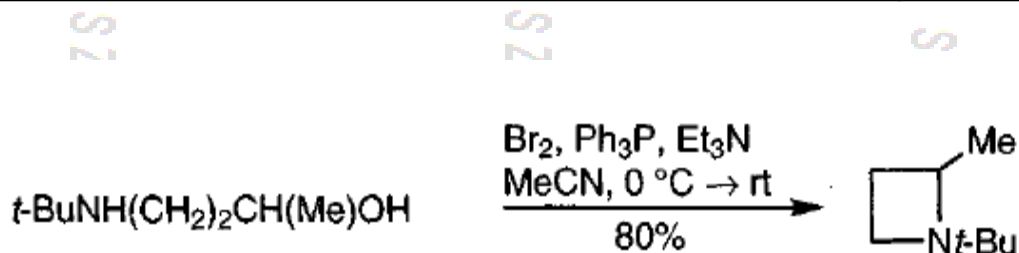
Образование циклов



Corey, Danheiser *J. Org. Chem.* 1976, 41, 260.

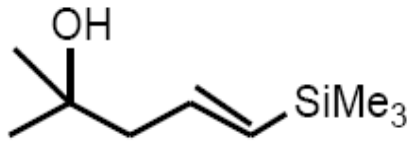
Вспомните реакцию МакМурри!

Синтез 4-членных гетероциклов

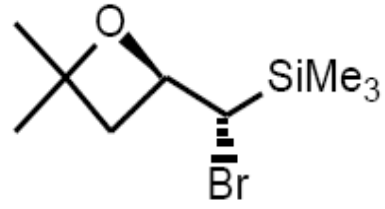




-S_N2-реакции



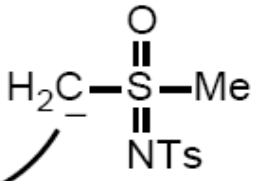
NBS



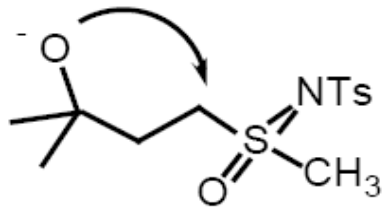
-илиды серы (вспомните Кори-Чайковский!)



SZ Vats



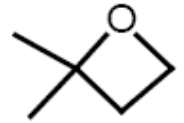
SZ Vats



SZ Vats



SZ Vats





SZVar

SZVar

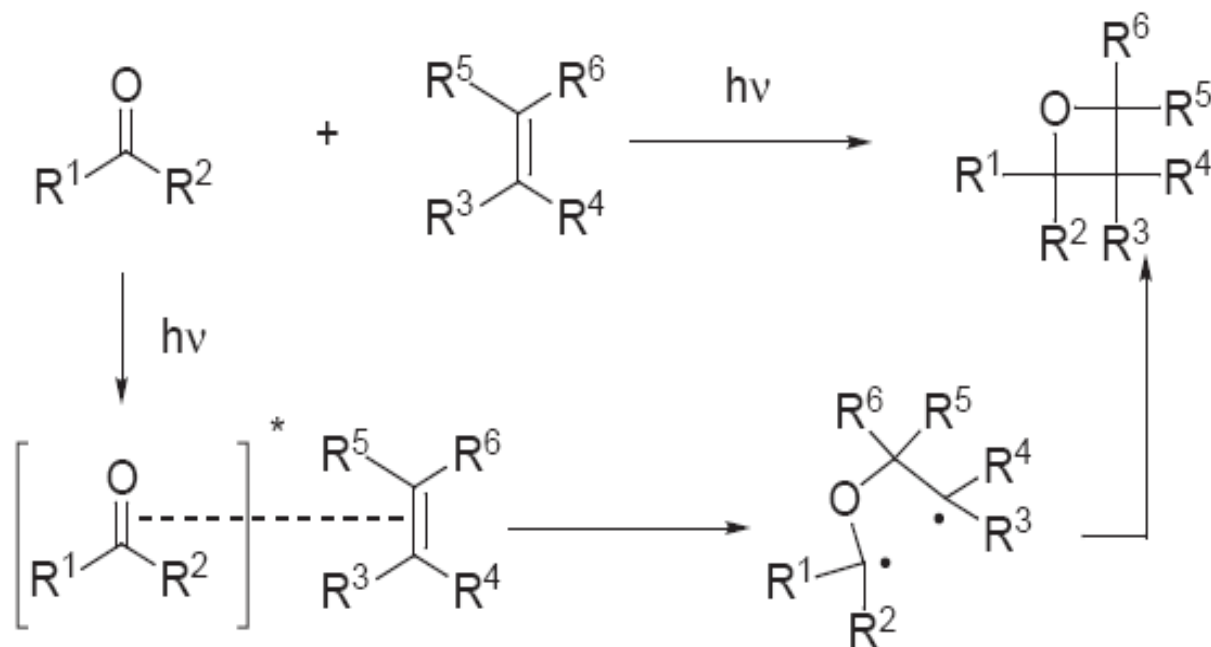
SZVa

SZVa

Comprehensive Org. Syn., Vol. 5, 151.

Dermuth Synthesis 1989, 152.

First studied in detail by Buchi *J. Am. Chem. Soc.* 1954, 76, 4327.



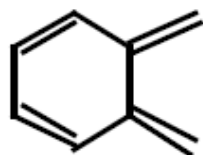
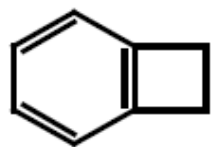
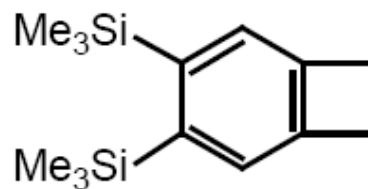
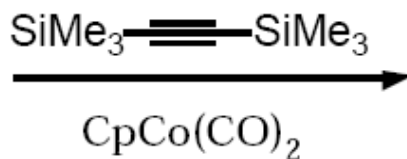
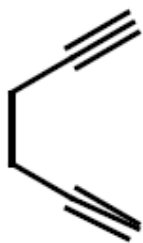
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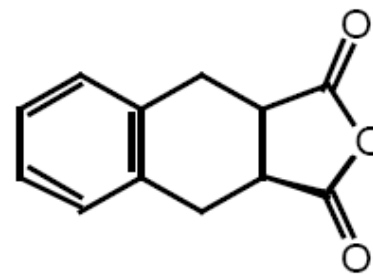
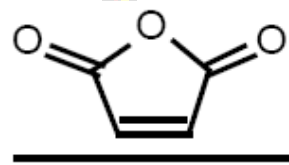


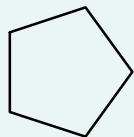
1. Циклобутаны и циклобутены:
- разное



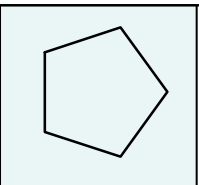
benzocyclobutane

o-quinodimethane

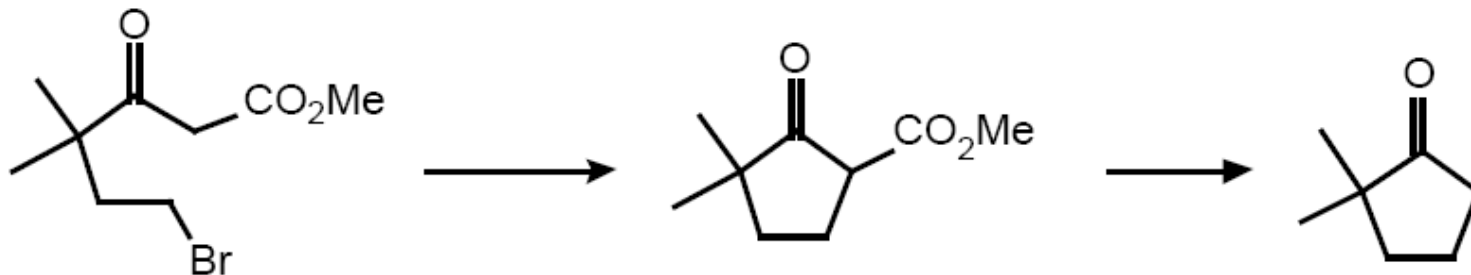




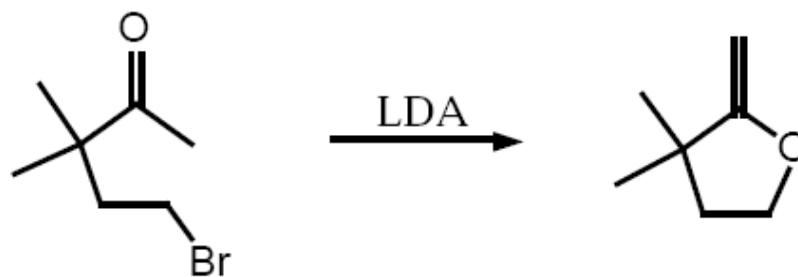
- **S_N2 -реакции**
- **ацилоиновая конденсация**
- **присоединение по Михаэлю**
- **альдольная конденсация**
- **внутримолекулярное олефинирование по Виттигу**
- **метатезис с замыканием цикла**
- **диазосоединения**
- **радикальные циклизации**
- **реакция Посона-Хэнда**
- 1,3-диполярное присоединение ([3+2]-циклоприсоединение)
- реакции расширения и сужения циклов:
 - a. 3 @ 5
 - b. 4 @ 5
 - c. 6 @ 5
- циклизация по Назарову
- циклизация по Нойори
- фотоциклизации арен+алкен



5-ЭКЗО-ТЕТ

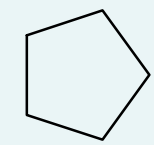


Но:



Почему:

• Внутримолекулярное алкилирование енолятов



S Z Vaisa

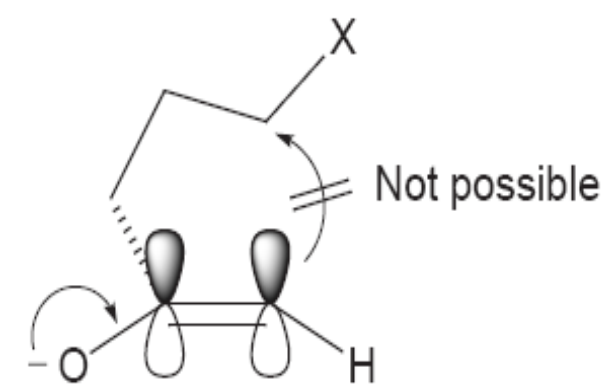
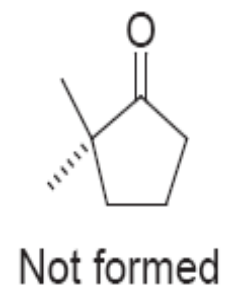
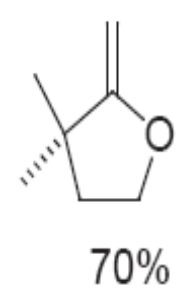
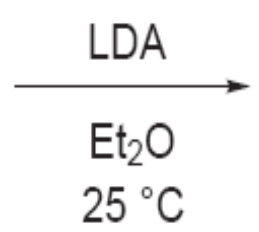
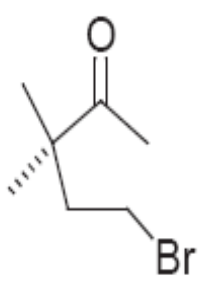
S Z Vaisa

S Z Vaisa

S Z Vaisa



- Note Baldwin's Rules
 Preceded by Eschenmoser
Helv. Chim. Acta 1970, 53, 2059.



tu/res

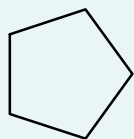
tu/res

tu/res

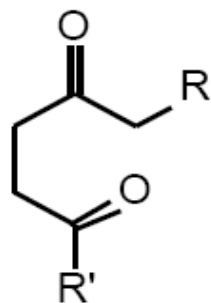
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• Альдольно-кратоновая конденсация

Образование циклов



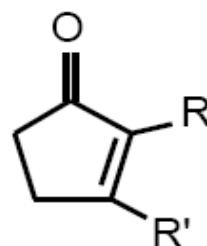
S Z Vaisadze's lectures



S Z Vaisadze's

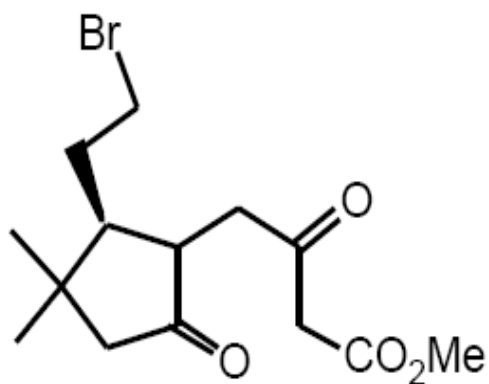


S Z Vaisadze's

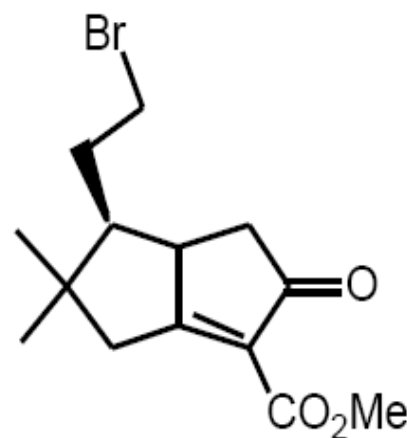


S Z Vaisadze's lectures

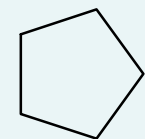
5-ЭКЗО-ТРИГ



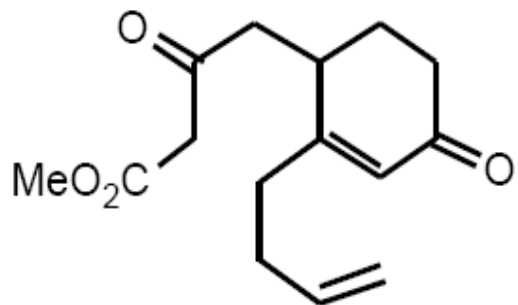
NaOMe, MeOH



5-ЭКЗО-ТРИГ



S Z Vaisadze's lecture



dze's lectures

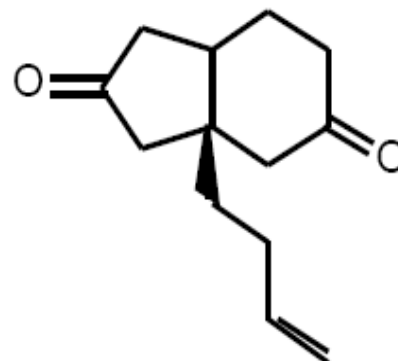
S Z Vaisadze's lecture

dze's lectures



S Z Vaisadze's lecture

dze's lectures

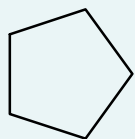


S Z Vaisadze's lecture

dze's lectures

• Внутримолекулярное олефинирование

Образование циклов



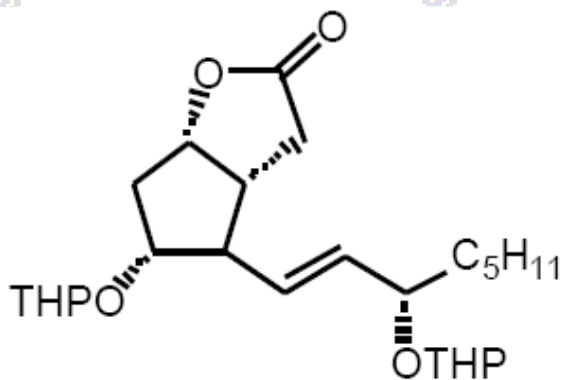
SZ Vats

SZ Vats

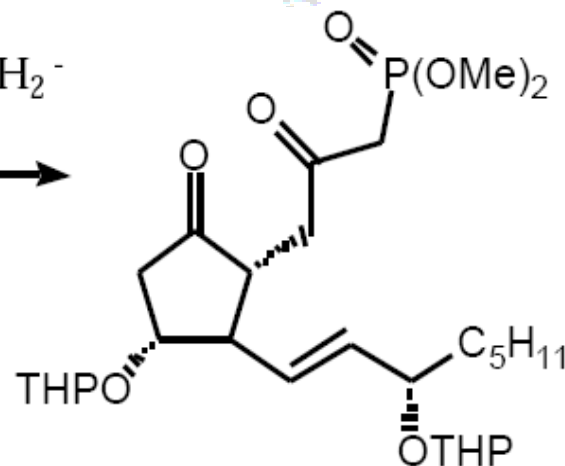
SZ Var

SZ Var

5-ЭКЗО-ТРИГ



1) $(\text{MeO})_2\text{P}(\text{O})\text{CH}_2^-$
2) Collins

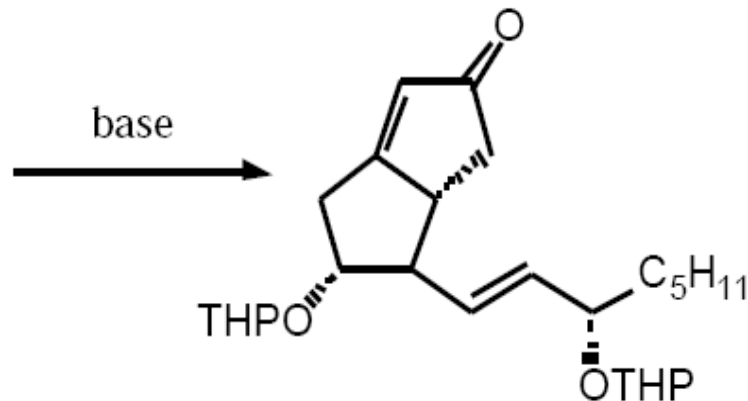


SZ Vatsadze's lectures

SZ I

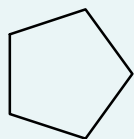
SZ I

SZ Vatsadze's lectures

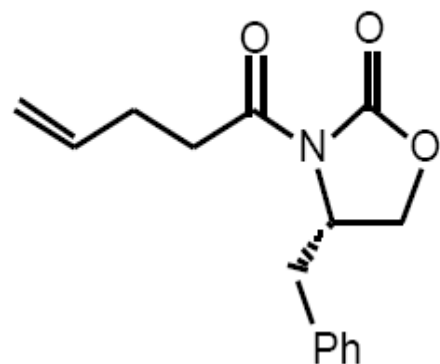


• Метатезис с замыканием цикла

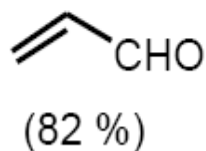
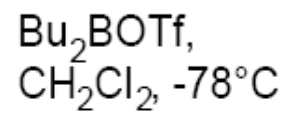
Образование циклов



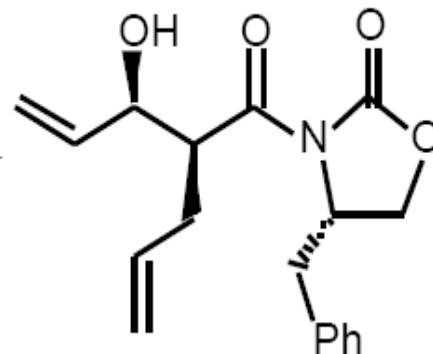
S Z Vatsa



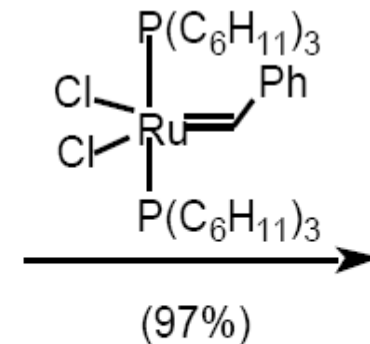
S Z Vatsa



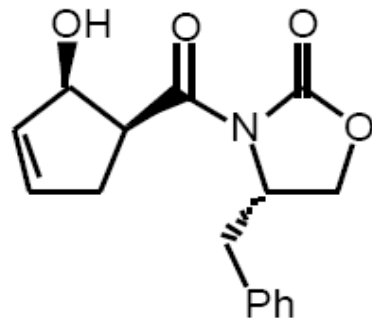
S Z Vatsa



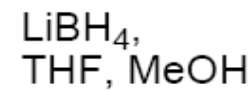
S Z Vatsa



Vatsa's lectures

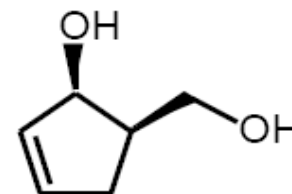


Vatsa

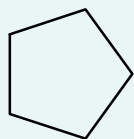


(78%)

Vatsa



Vatsa

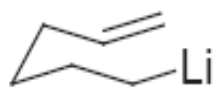


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S Z Vatsadze's lectu

S Z Vatsadze's lectu

S Z Vatsadze's lectu



stable at $-78\text{ }^{\circ}\text{C}$
 $t_{1/2} = 5.5\text{ min at }25\text{ }^{\circ}\text{C}$

Bailey *J. Am. Chem. Soc.* **1992**, *114*, 8053.
J. Am. Chem. Soc. **1991**, *113*, 5720.
J. Am. Chem. Soc. **1987**, *109*, 2442.

Intramolecular carbometalation, review:
Comprehensive Org. Syn., Vol. 4, 871.

e's lectures

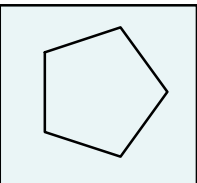
e's lectures

e's lectures

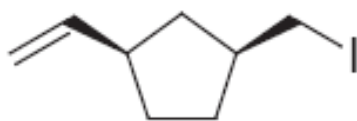
e's lectures

• Анионные циклизации: примеры

Образование циклов



1ZV

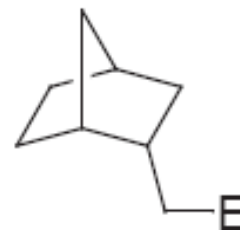


1ZS

1. ^tBuLi

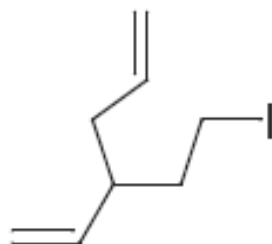
2. E⁺

1Z1



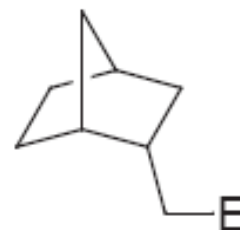
63–91%

1Z1

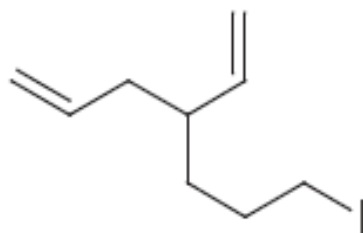


tandem

cyclizations

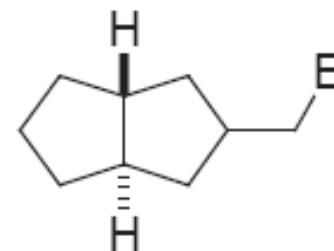


65–90%



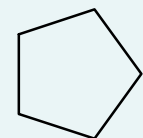
tandem

cyclizations



65–87%

• Анионные циклизации: примеры



SZVa

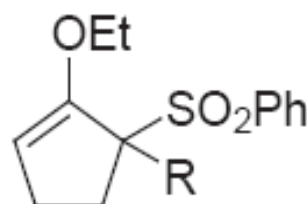
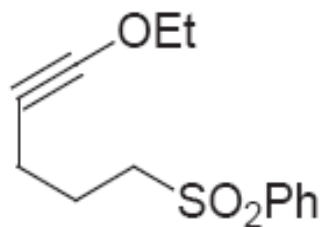
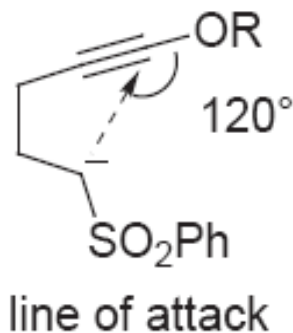
SZVa

SZVa

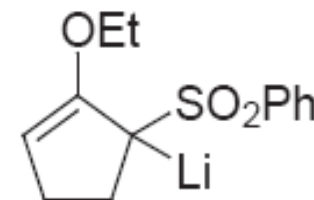
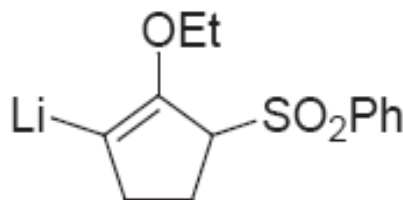
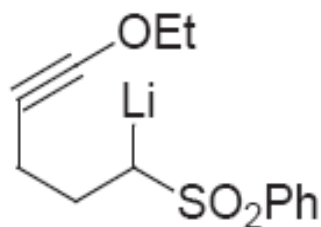
SZVa

Funk J. Am. Chem. Soc. 1993, 115, 7023.

5-endo-dig cyclization



RX



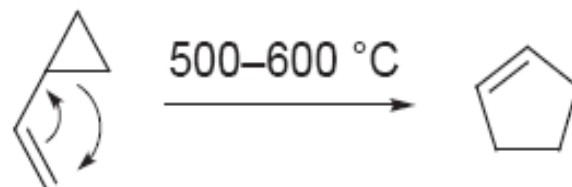
• 1,3-Сигматропные перегруппировки:
винилциклопропан

Образование циклов



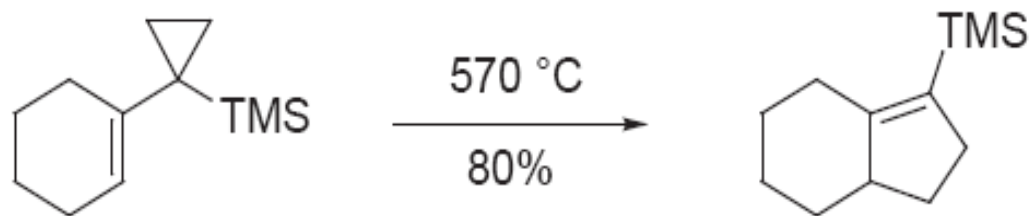
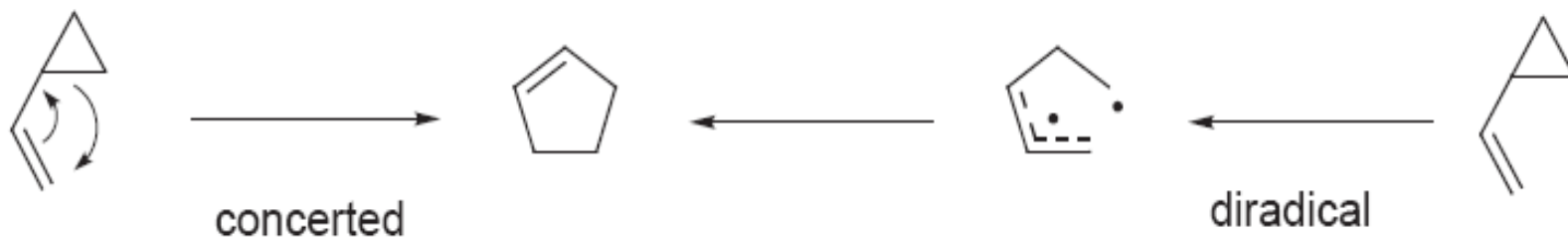
First report: Neureiter *J. Org. Chem.* **1959**, 24, 2044.

Review: Hudlicky *Chem. Rev.* **1989**, 89, 165.

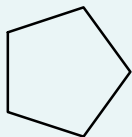


Org. React. **1985** 33, 247.

Mechanism:



Paquette Tetrahedron Lett. **1982**, 23, 263.



Что еще по циклопентанам:

• **радикальные циклизации**

-реакции расширения и сужения циклов:

a. 3 ® 5

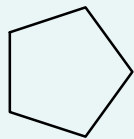
b. 4 ® 5

c. 6 ® 5

-ацетиловая конденсация

-1,3-диполярное присоединение

-фотоциклизации арен+алкен



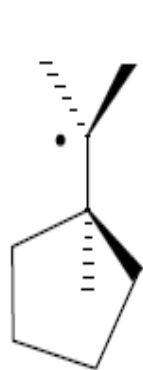
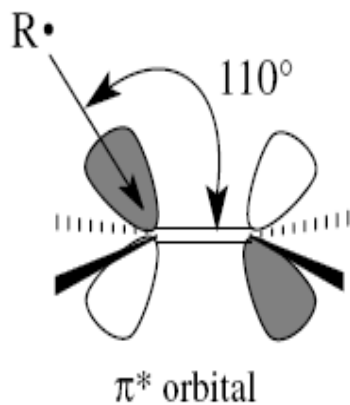
S Z Vats

Вспоминаем правила Болдуина!!!

S Z Vats

S Z Vats

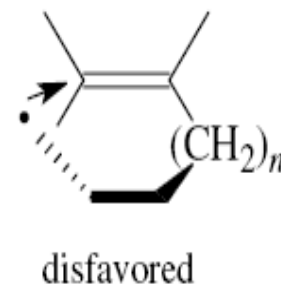
S Z Vats



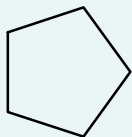
and



but



- Циклизации типа «радикал-олефин»



Радикалы бывают **электрофильные** и **нуклеофильные**.

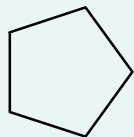
Алкильные радикалы относятся к **нуклеофильным** радикалам:

- Relative rates of addition to $\text{CH}_2=\text{CHPO}(\text{OEt})_2$: typical electron-deficient olefin.

$k_{\text{rel}} =$	CH_3^\bullet	$\text{CH}_3\text{CH}_2^\bullet$	$\text{CH}_3\text{OCH}_2^\bullet$	$(\text{CH}_3)_2\text{CH}^\bullet$	$(\text{CH}_3)_3\text{C}^\bullet$
	1	1	2.7	4.8	24

• Циклизации типа «радикал-олефин»

Образование циклов

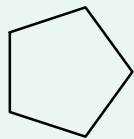


Стереoeлектронные требования

olefin	% addition to:		k_{rel}
	Ca	Cb	
	>95	<5	1.16
	>95	<5	18.4
	>95	<5	2×136
	50	50	2×0.50
	50	50	2×0.63
	>95	5	15
	<5	>95	13.9

• Циклизации типа «радикал-олефин»

Образование циклов

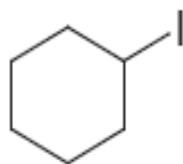


SZ Vais?

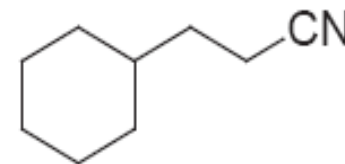
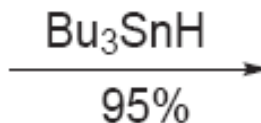
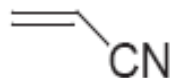
SZ Vais?

SZ Vais

SZ Vais

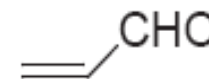
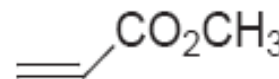
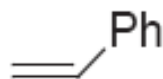
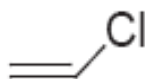
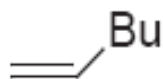
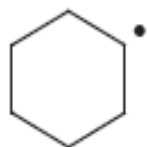


+



Nucleophilic radical

Electrophilic acceptor alkene



k_{rel}

1.0

8.4

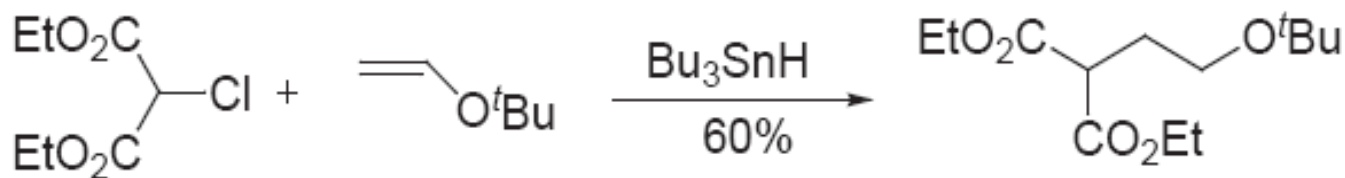
84

3000

8500

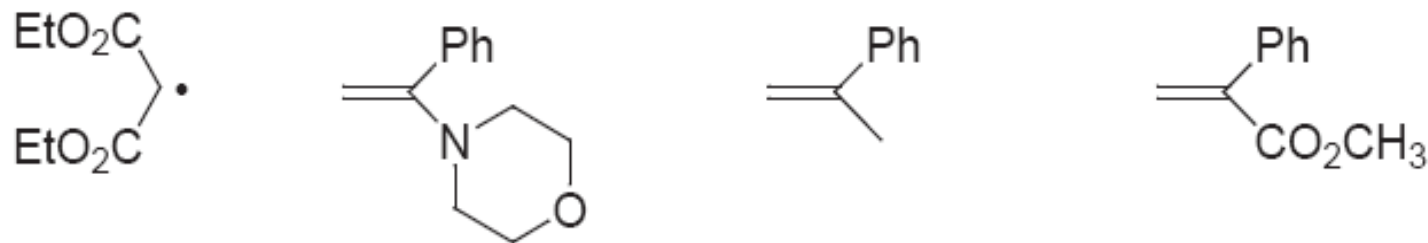
• Циклизации типа «радикал-олефин»

Электрофильные радикалы – стабилизированы электроноакцепторными группами:



Electrophilic radical

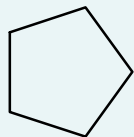
Nucleophilic acceptor alkene



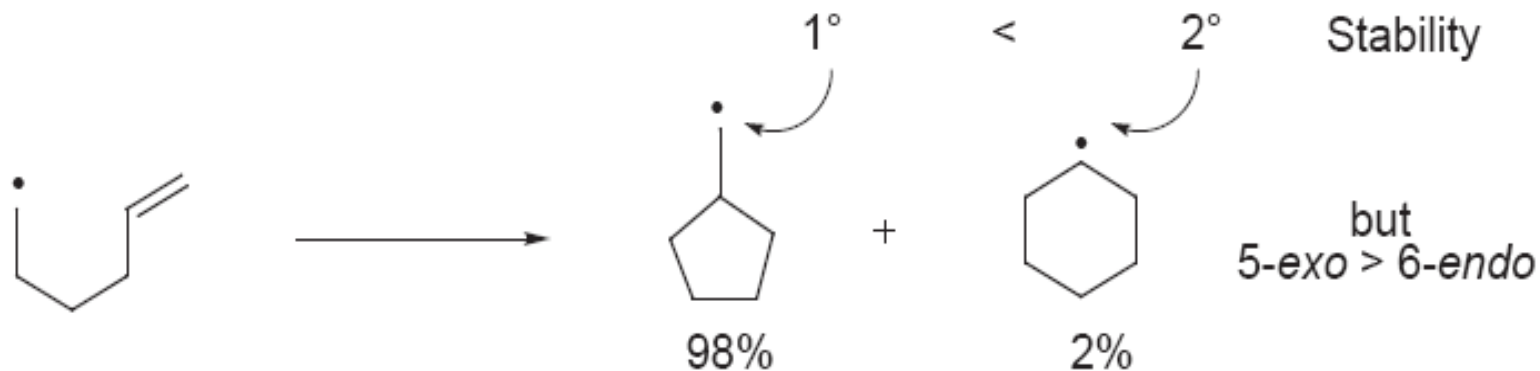
k_{rel}	23	3.5	1
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• Циклизации типа «радикал-олефин»

Образование циклов

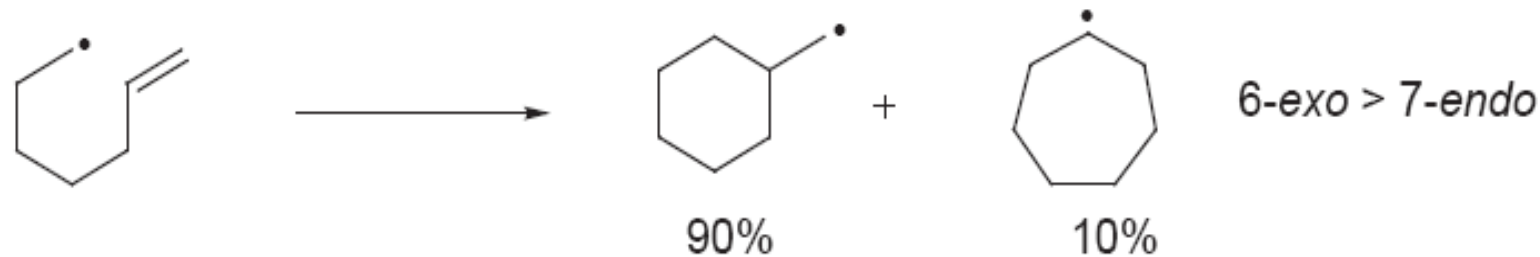


Скорости, регио- и диастереоселективность



Beckwith *J. Chem. Soc., Chem. Commun.* **1974**, 472.

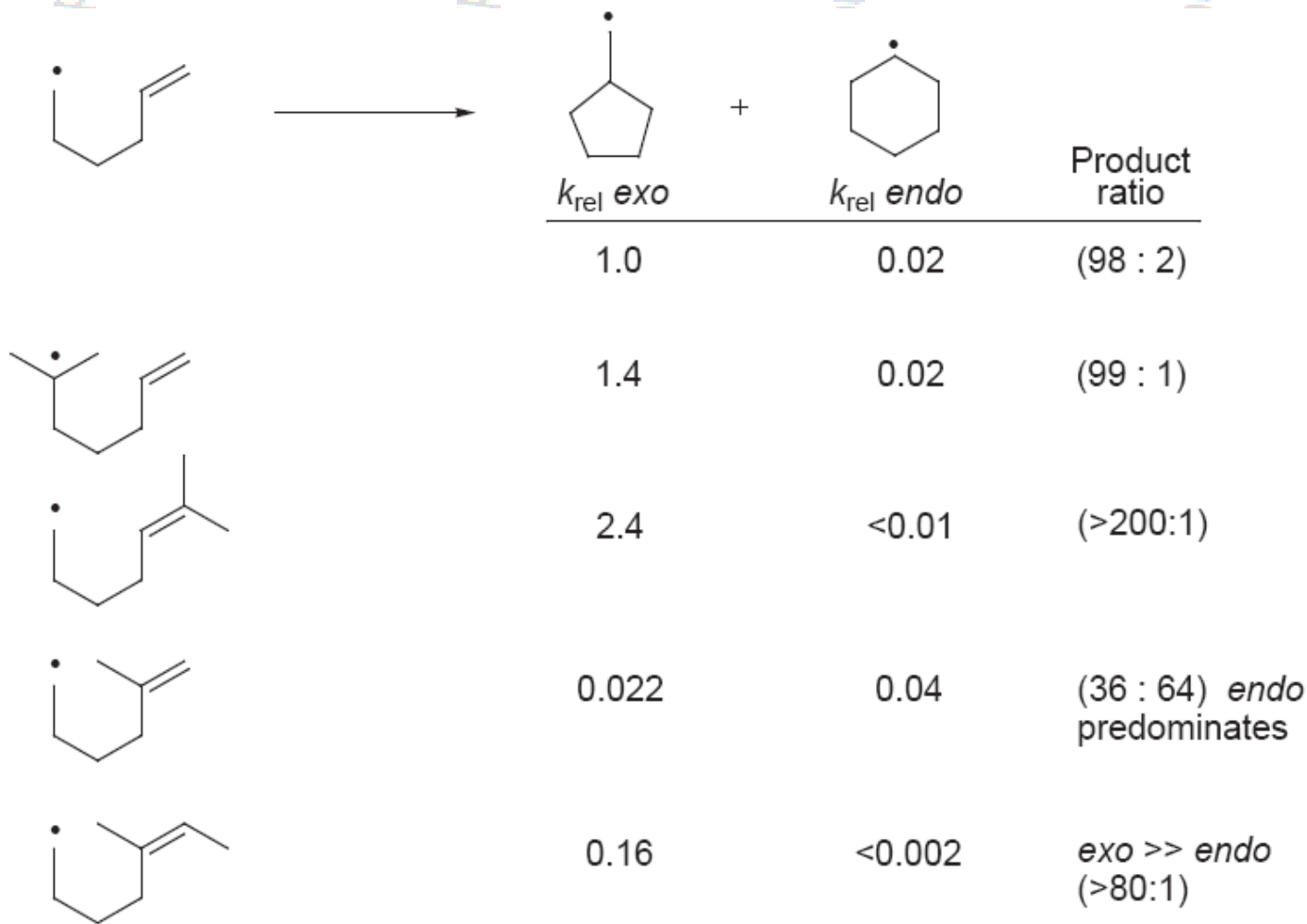
Beckwith *J. Chem. Soc., Chem. Commun.* **1980**, 484.



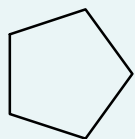
- Циклизации типа «радикал-олефин»

Образование циклов

Скорости, регио- и диастереоселективность



Циклизации типа «радикал-олефин»



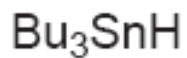
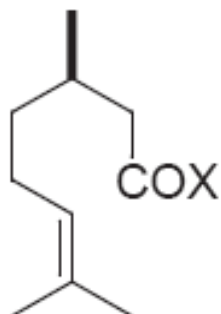
S Z Vatsa

Карбонилрадикалы

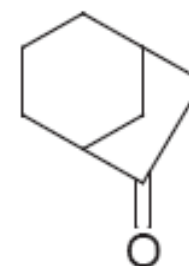
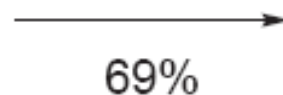
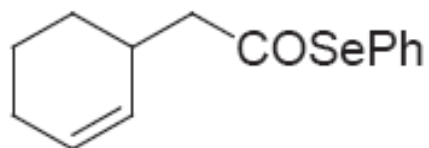
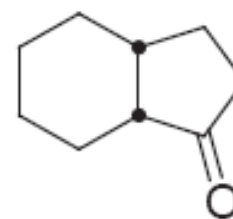
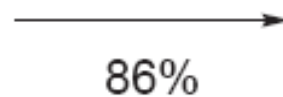
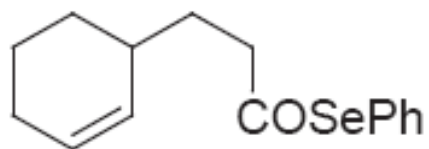
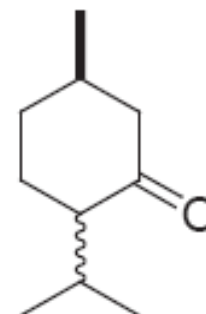
S Z Vatsa

S Z Vatsa

S Z Vatsa



X = SePh, 84%
X = SPh, 0%
X = Cl, 59%

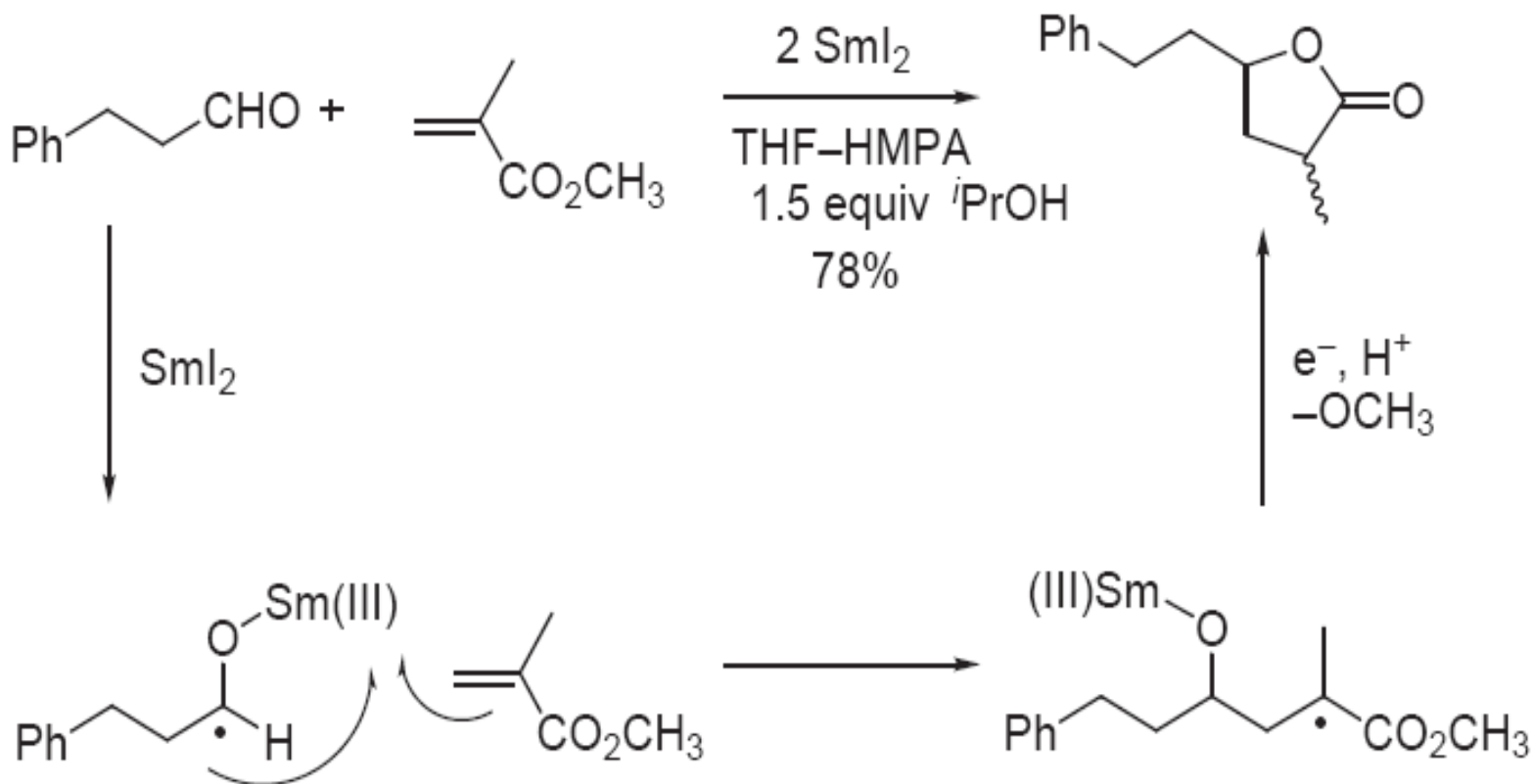


• Восстановительное сочетание с участием SmI_2

Molander *Chem. Rev.* **1992**, *92*, 29.

Molander in *Chemistry of the Carbon Metal Bond*, Hartley, F. R.; Patai, S., Eds.; Wiley: NY, 1989, Vol. 5

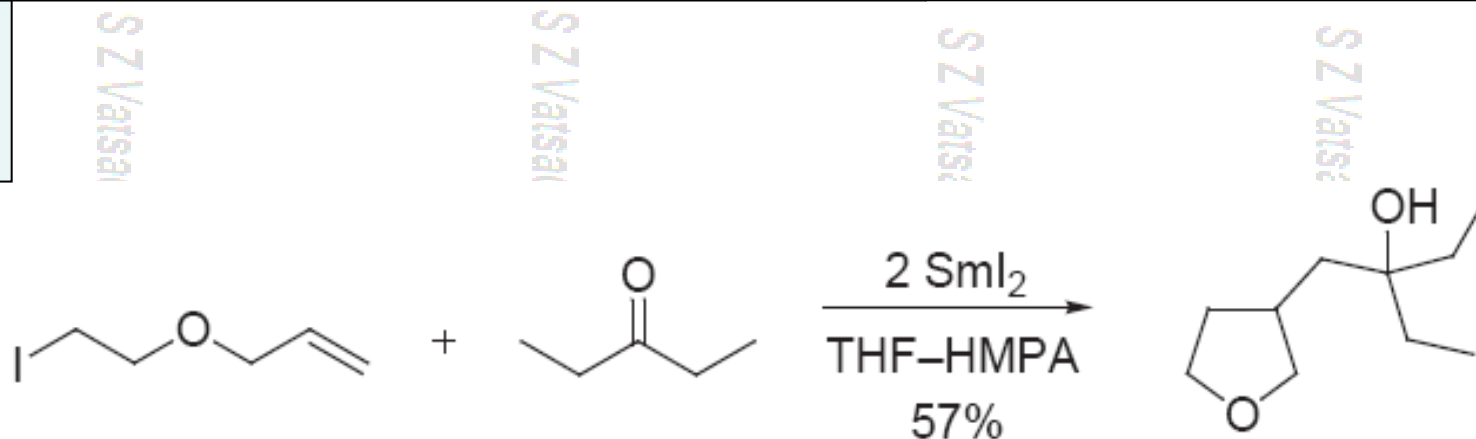
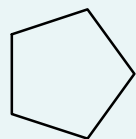
Molander in *Comprehensive Org. Syn.*, Vol. 1, 262.



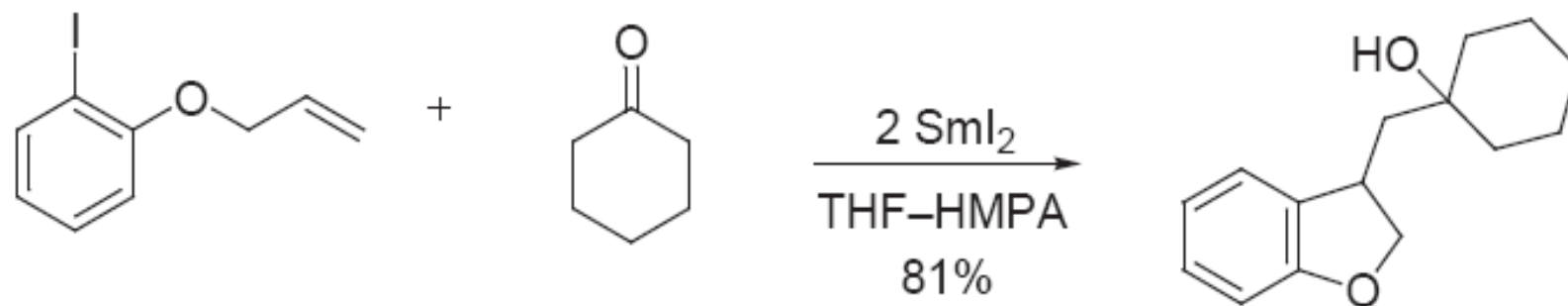
Inanaga *Tetrahedron Lett.* **1986**, *27*, 5763.

Tetrahedron Lett. **1989**, *30*, 2837.

- Восстановительное сочетание с участием SmI_2



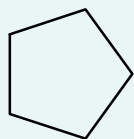
Molander *J. Org. Chem.* **1990**, *55*, 6171.



Curran *Synlett* **1990**, 773.

• Реакция Посона-Хэнда

Образование циклов



[2 + 2 + 1]

Comprehensive Org. Syn., Vol. 5, pp 1037–1064.

Org. React. **1991**, 40, 1.

Pauson *Tetrahedron* **1978**, 41, 5855.

Schore *Chem. Rev.* **1988**, 88, 1081.

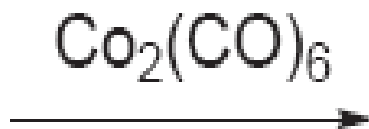
Brummond *Tetrahedron* **2000**, 56, 3263.

First detailed study: Khand *J. Chem. Soc., Perkin Trans. 1* **1973**, 977.

SZ Vatsadze's lectures

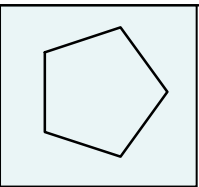


CO



• Реакция Посона-Хэнда

Образование циклов



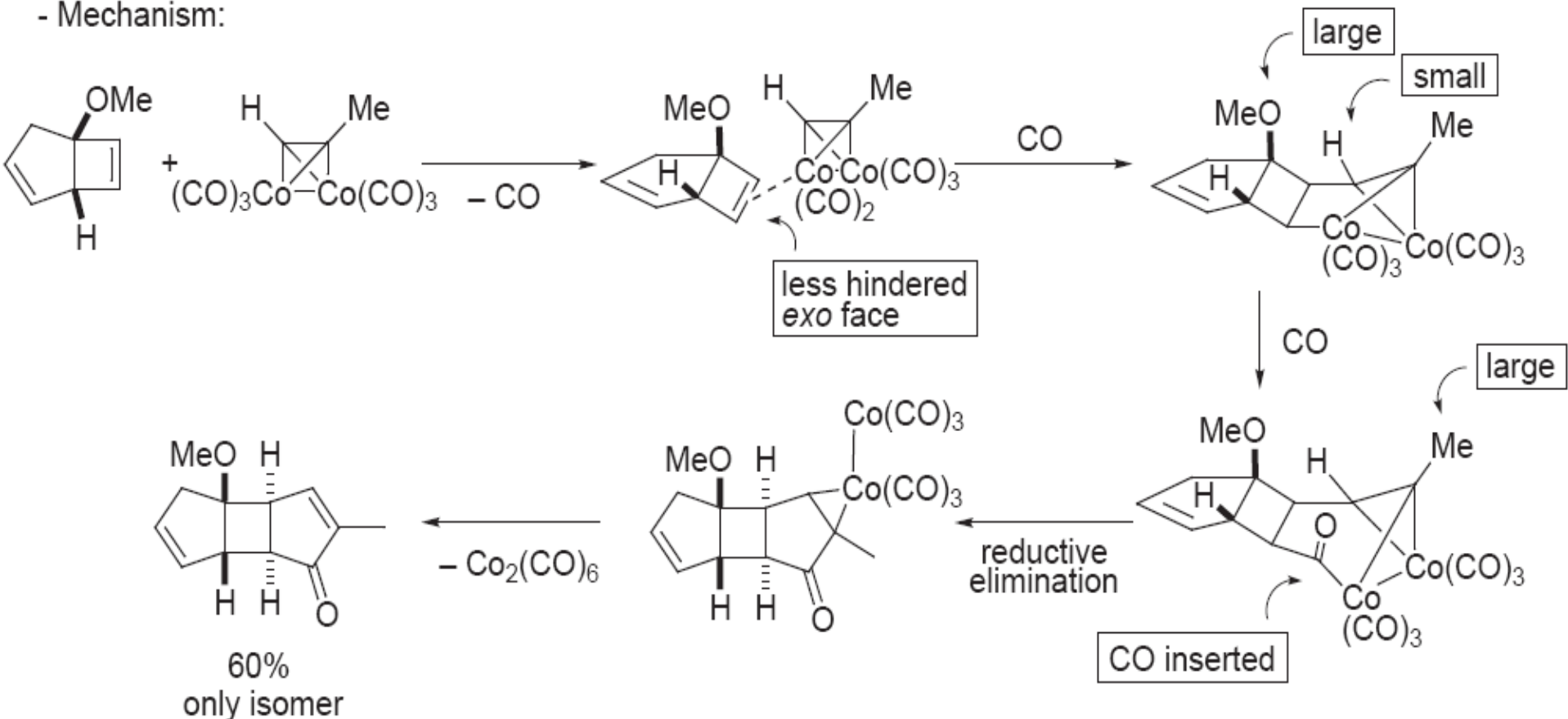
S Z Varsadze

S Z Varsadze

S Z Varsadze

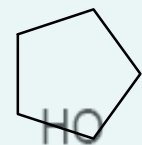
S Z Varsadze

- Mechanism:

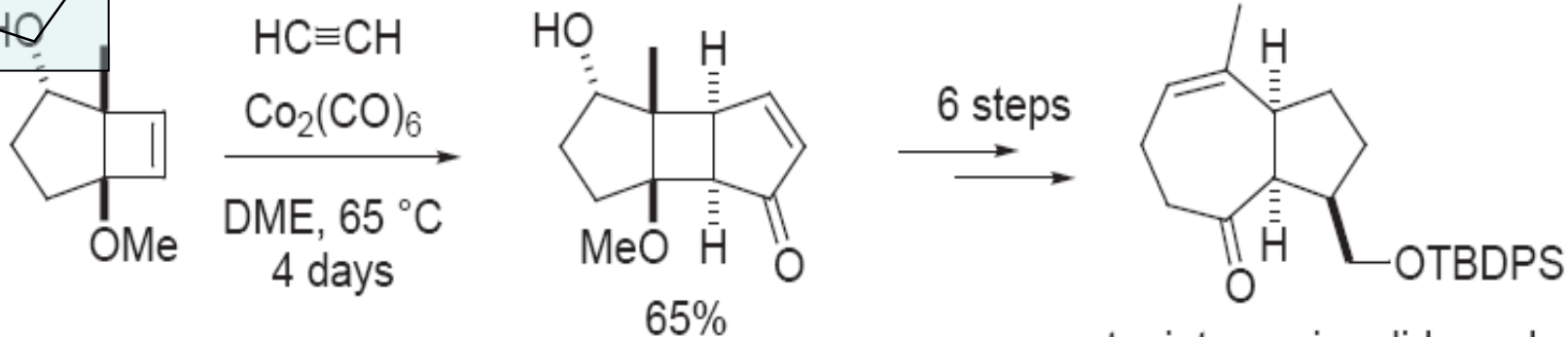


• Реакция Посона-Хэнда: примеры

Образование циклов



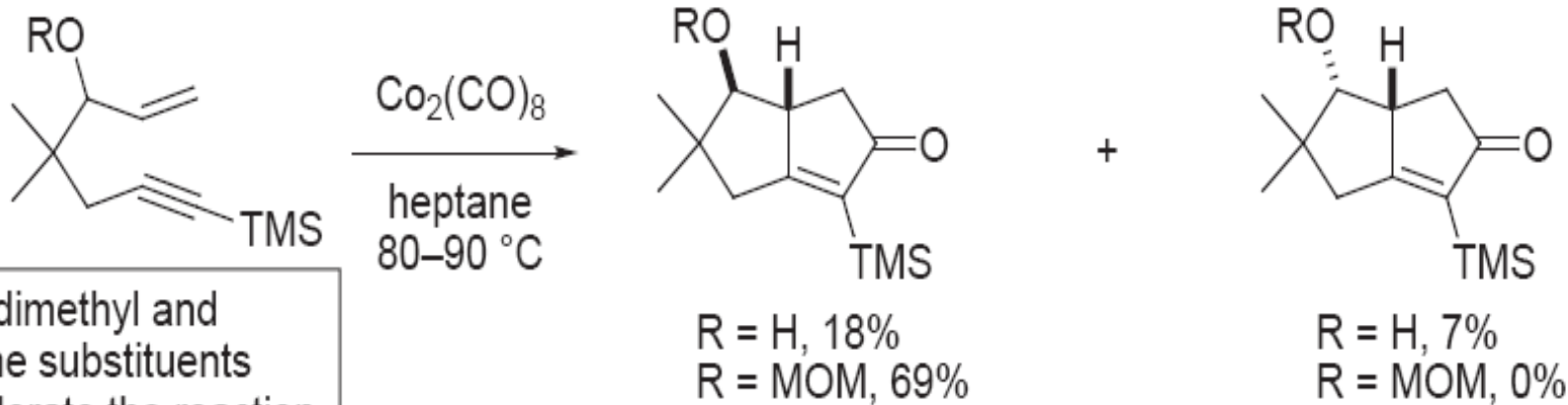
Межмолекулярная



Schore *J. Org. Chem.* **1987**, 52, 3595.

entry into guaianolide and pseudo-guaianolide natural products

Внутримолекулярная

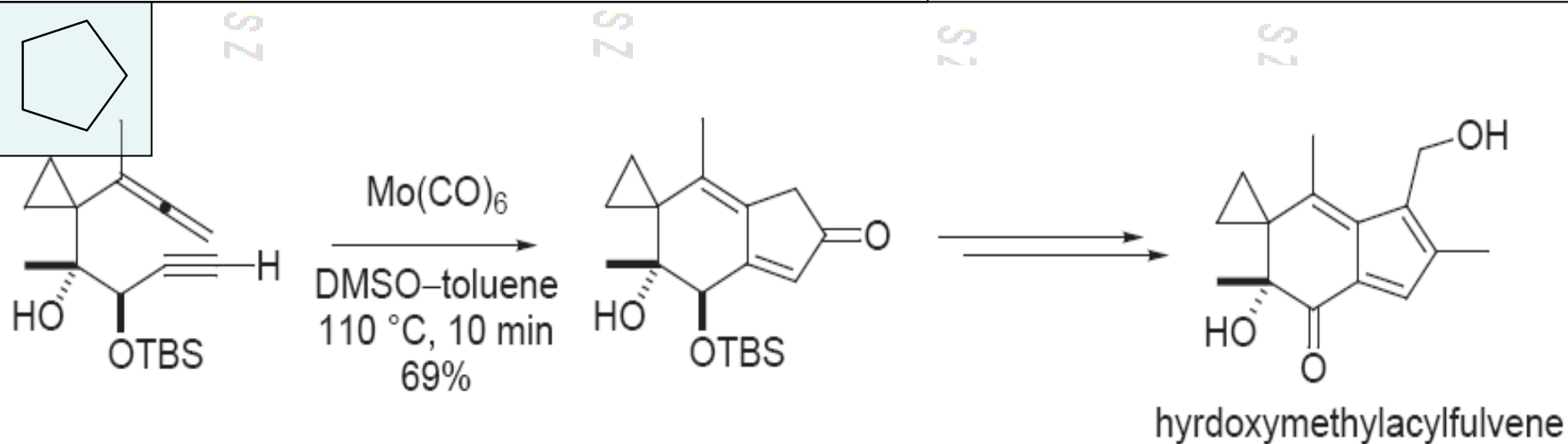


The dimethyl and alkyne substituents accelerate the reaction.

Magnus *J. Am. Chem. Soc.* **1983**, 105, 2477.

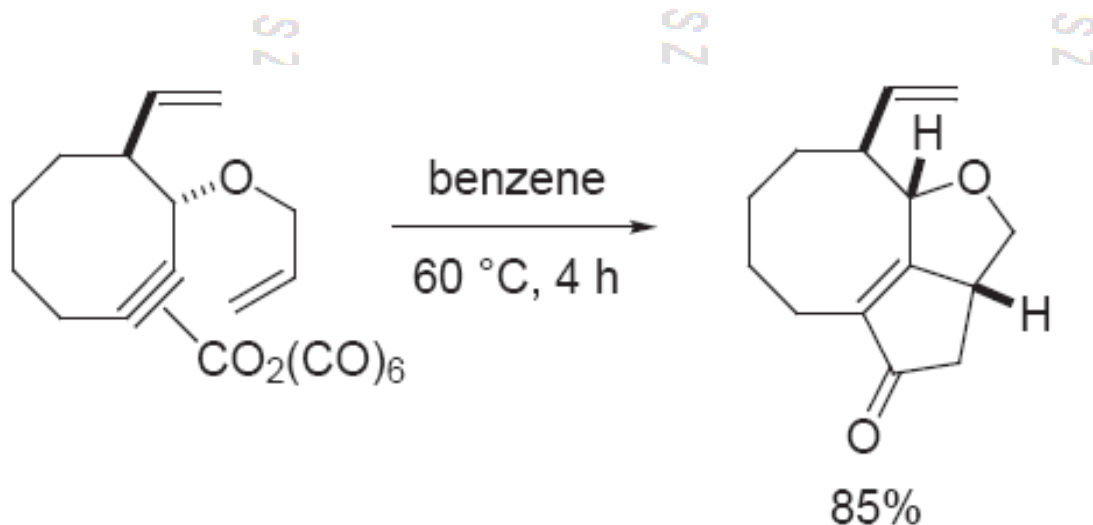
• Реакция Посона-Хэнда: примеры

Образование циклов

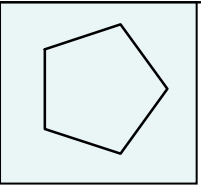


Brummond *J. Am. Chem. Soc.* 2000, 122, 4915.

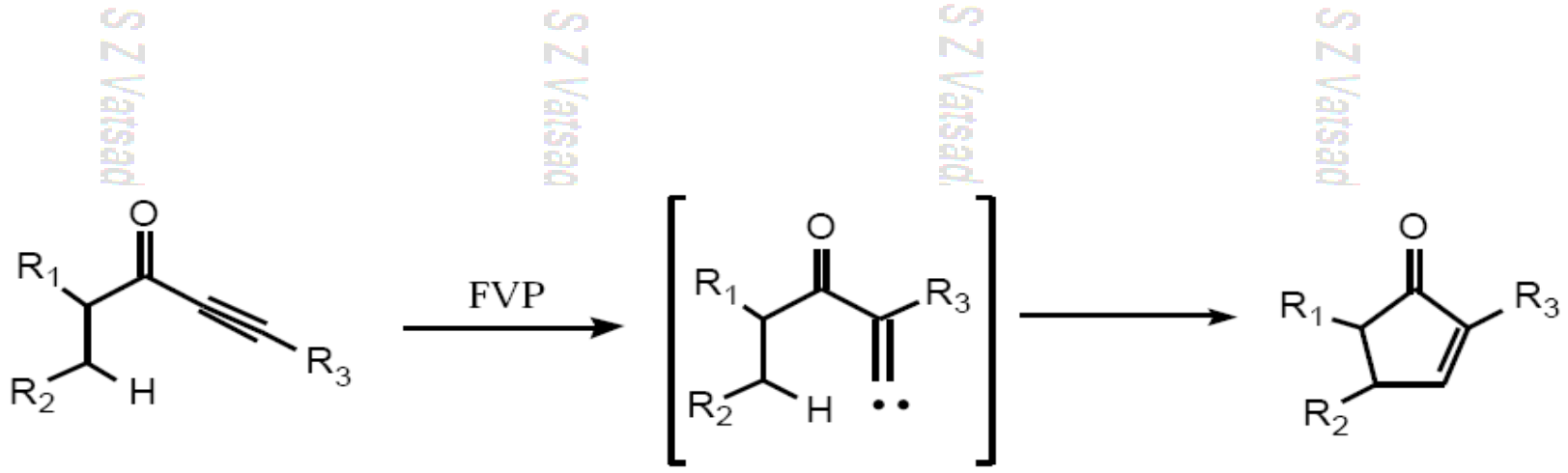
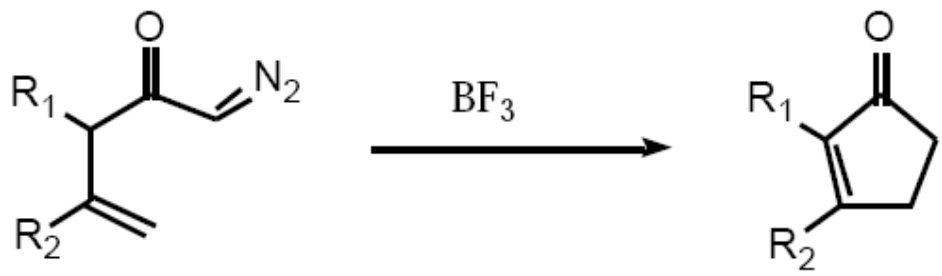
SZ Varsadze's lectures

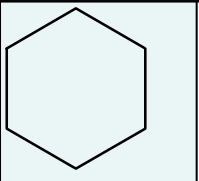


Schreiber *J. Am. Chem. Soc.* 1986, 108, 3128.



-дiazosоединения



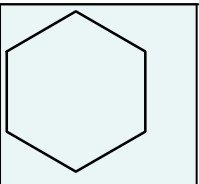


- сложнэфирная и альдольная конденсация
- присоединение по Михаэлю
- аннелирование по Робинсону
- метатезис с замыканием цикла
- циклизации катион+алкен
- Фишеровские карбены (р-я Дётса)
- тримеризация ацетиленов
- циклизация по Бергману и родственные процессы

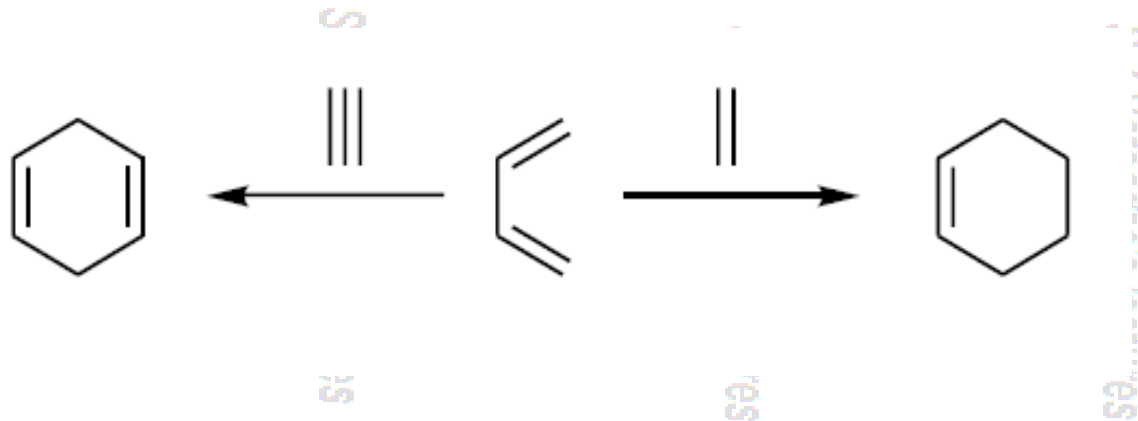
- внутримолекулярное олефинирование по Виттигу
- S_N2 -реакции
- ацилоиновая конденсация
- реакция Дильса-Альдера, *o*-хинодиметаны
- расширение и сужение циклов
- ароматическое замещение с послед. гидрированием или восстановлением по Берчу
- радикальные циклизации
- сигматропные перегруппировки
- внутримолекулярная еновая реакция

• [4+2]-циклоприсоединение

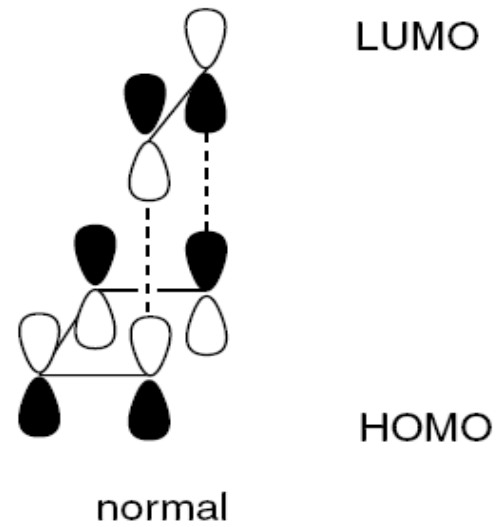
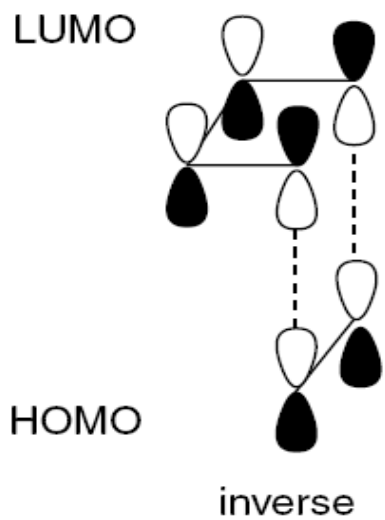
Образование циклов



S Z Vatsadze's lectures

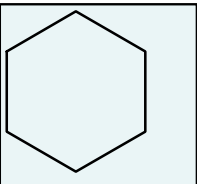


S Z Vatsadze's lectures



• Реакция Робинсона

Образование циклов

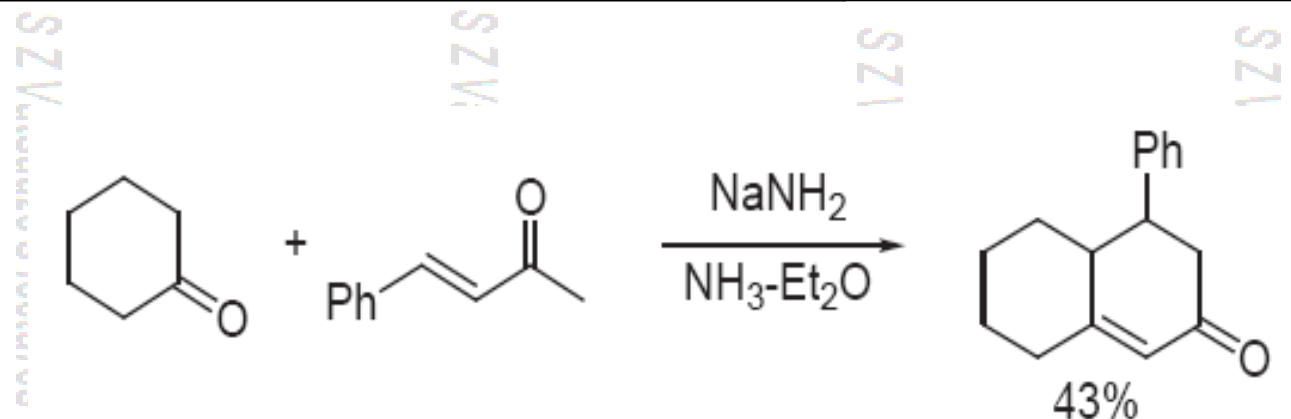
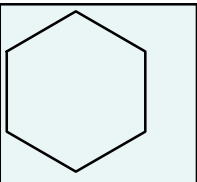


1947

SIR ROBERT ROBINSON for his investigations on plant products of biological importance, especially the alkaloids.



1886-1975



Robinson *J. Chem. Soc.* 1935, 1285.

Reviews

M. Jung, *Tetrahedron* 1976, 32, 3.
Org. React. 1959, 10, 179.
Org. React. 1968, 16, 3.
Synthesis 1976, 777.
Synthesis 1969, 49.

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se.

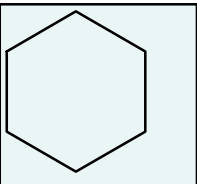
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Robinson *J. Chem. Soc.* 1917, 762. (tropinone)

• Реакция Робинсона

Образование циклов



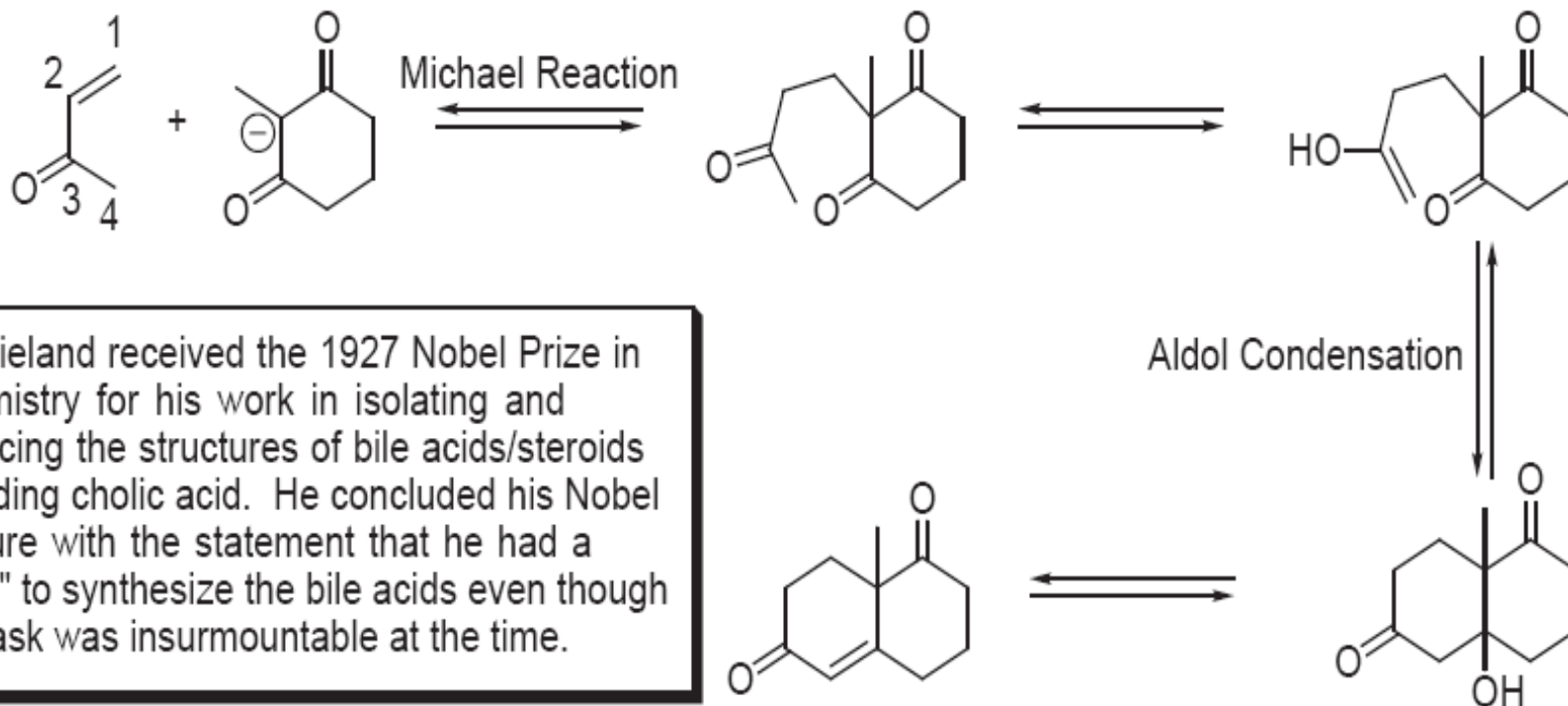
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S Z Vatsai

- Formally, a [4 + 2] condensation approach



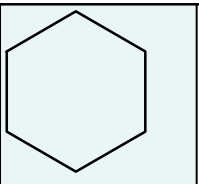
H. Wieland received the 1927 Nobel Prize in Chemistry for his work in isolating and deducing the structures of bile acids/steroids including cholic acid. He concluded his Nobel Lecture with the statement that he had a "duty" to synthesize the bile acids even though the task was insurmountable at the time.

Wieland–Miescher ketone

Wieland and Miescher *Helv. Chim. Acta* 1950, 33, 2215.

Реакция Робинсона

Образование циклов



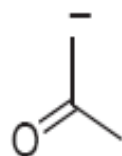
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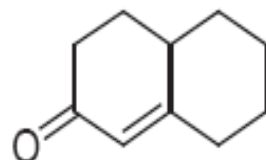
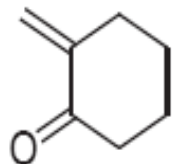
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- Alternative "[3 + 3] Robinson Annulation"



+



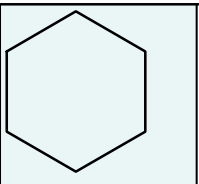
Both the [4 + 2] and [3 + 3] approaches were first generalized by Robinson *J. Chem. Soc.* 1937, 53.

sadze's lectures

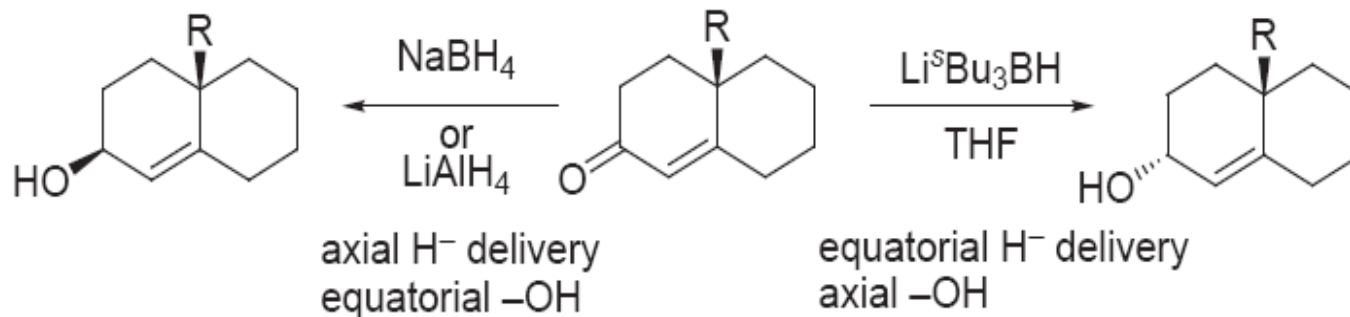
sadze's lectures

sadze's lectures

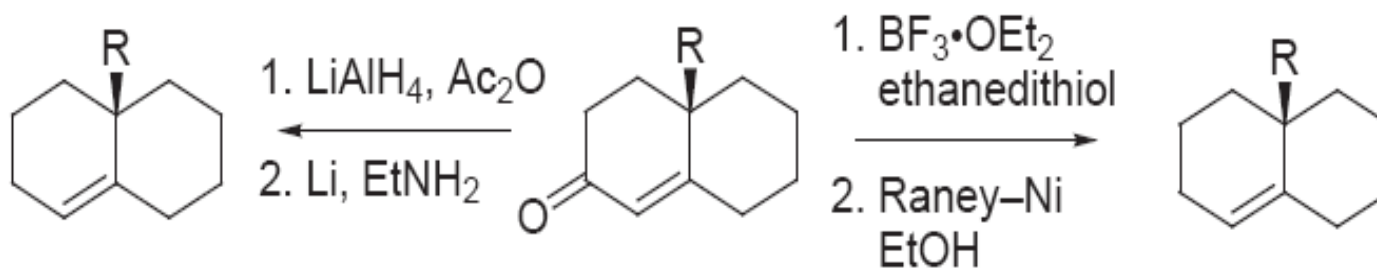
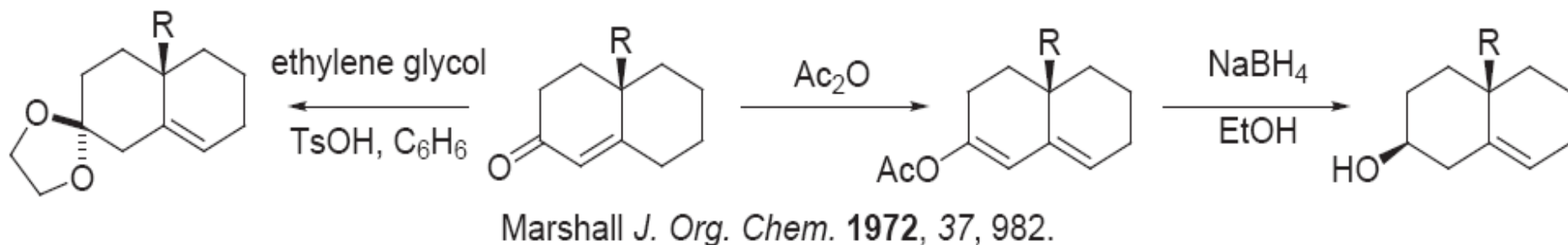
sadze's lectures



Некоторые синтетические превращения Робинсоновских продуктов

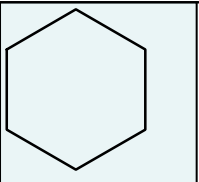


- Deconjugation with ketalization or reduction

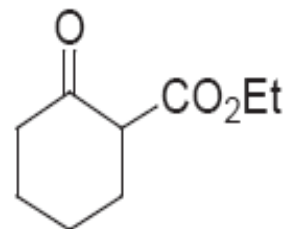
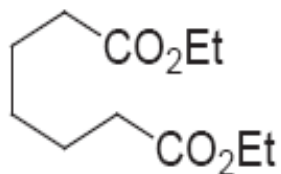


- Внутримолекулярное ацилирование енолятов

Образование циклов



Конденсация Дикмана



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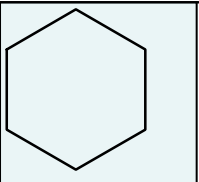
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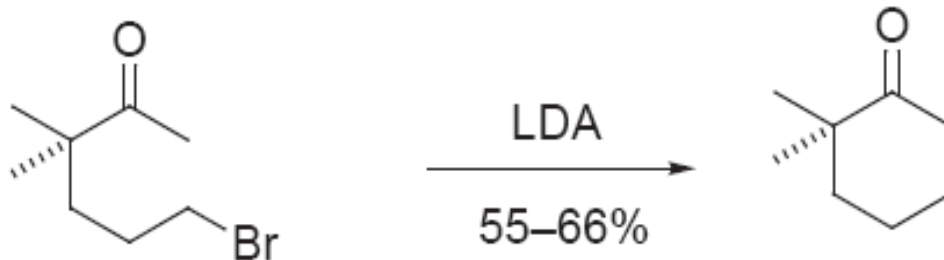
S Z Vatsadze's lectures

• Внутримолекулярное алкилирование енолятов

Образование циклов



S Z Vatsadze's lectures



House *J. Org. Chem.* 1978, 43, 700.

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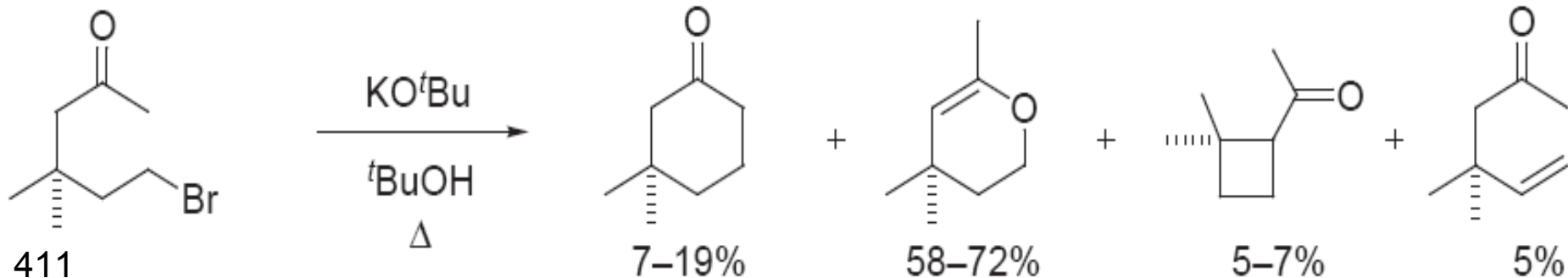


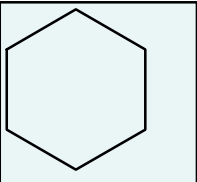
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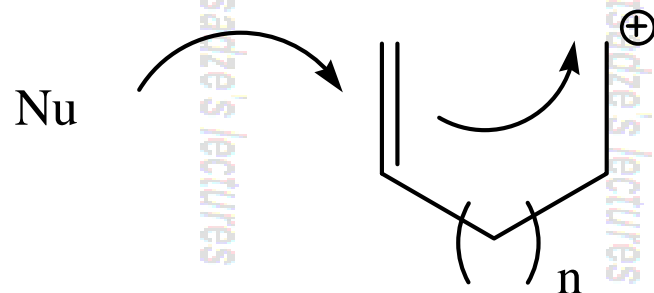
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Сравните с т/д условиями:





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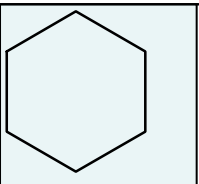
S Z Vaisadze's lectures

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- | | |
|-------------|--|
| Johnson | <i>Acc. Chem. Res.</i> 1968 , 1, 1.
<i>Angew. Chem., Int. Ed. Eng.</i> 1976 , 15, 9.
<i>Bioorg. Chem.</i> 1976 , 5, 51. |
| van Tamelen | <i>Acc. Chem. Res.</i> 1968 , 1, 111. |
| Harding | <i>Bioorg. Chem.</i> 1973 , 2, 248. |
| Goldsmith | <i>Fortschr. Chem. Org. Nat.</i> 1972 , 29, 363. |
| Lansbury | <i>Acc. Chem. Res.</i> 1972 , 5, 311. |
| Speckamp | <i>Recl. Trav. Chim. Pays-Bas.</i> 1981 , 100, 345. |
| Sutherland | <i>Comprehensive Org. Syn.</i> Vol 3, pp 341–377. |



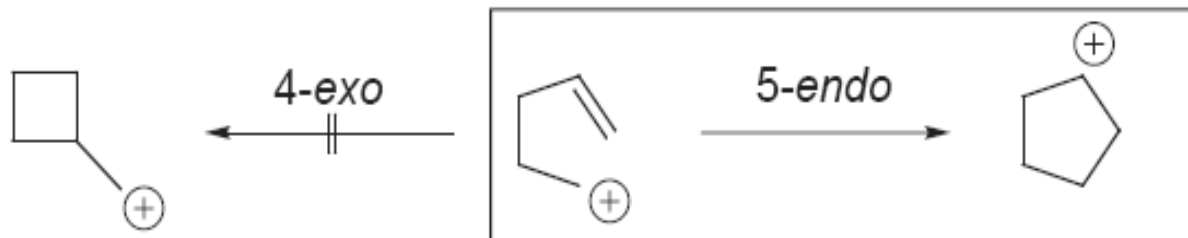
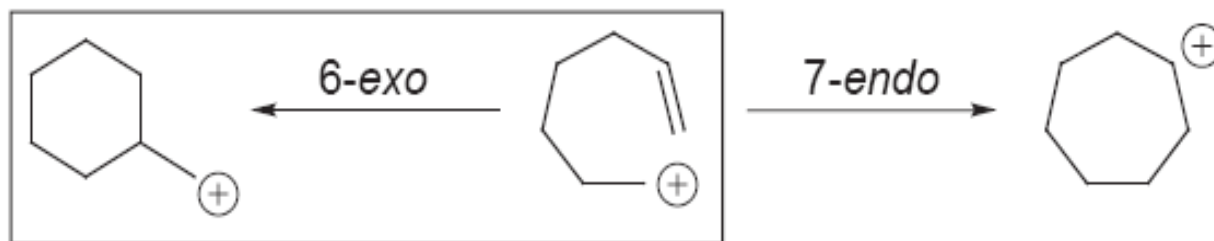
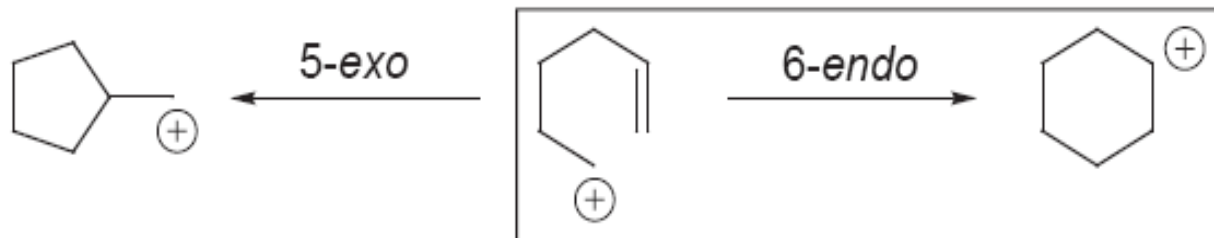
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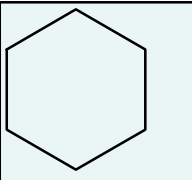
Алкены

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S Z Varsadze

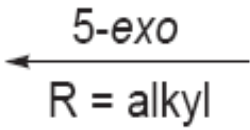
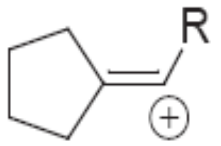
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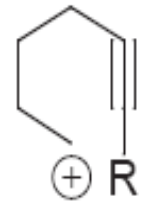


SZ Vats

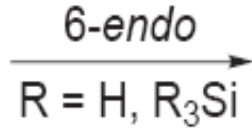
Алкины



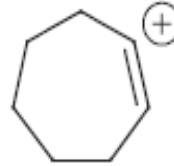
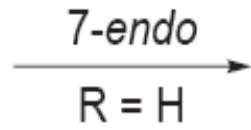
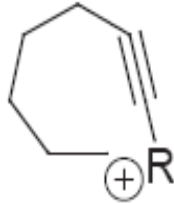
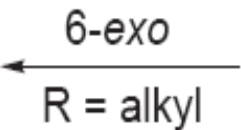
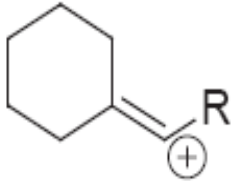
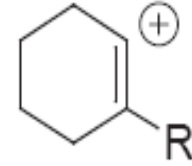
SZ Vats



SZ Var

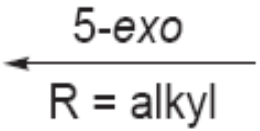
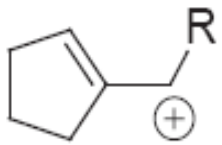


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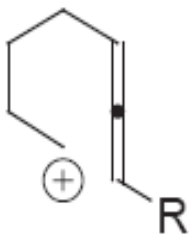


tsadze's lectur

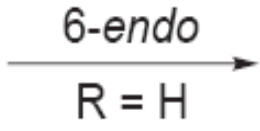
Аллены



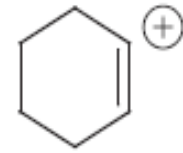
tsadze's lectur

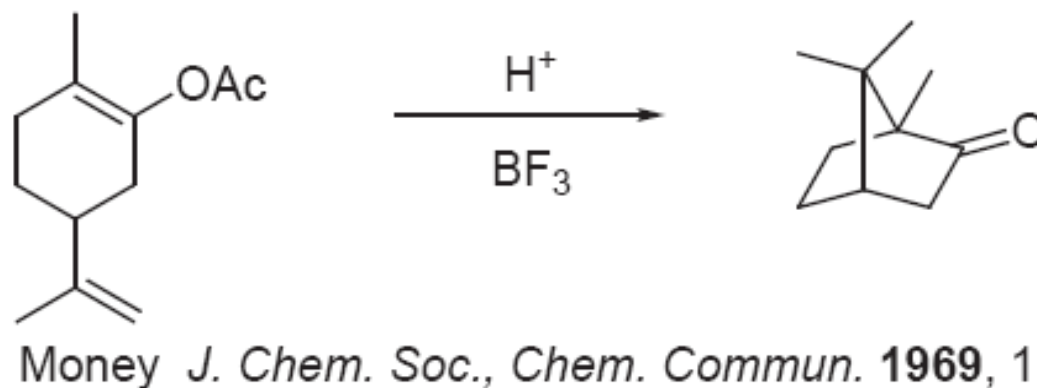
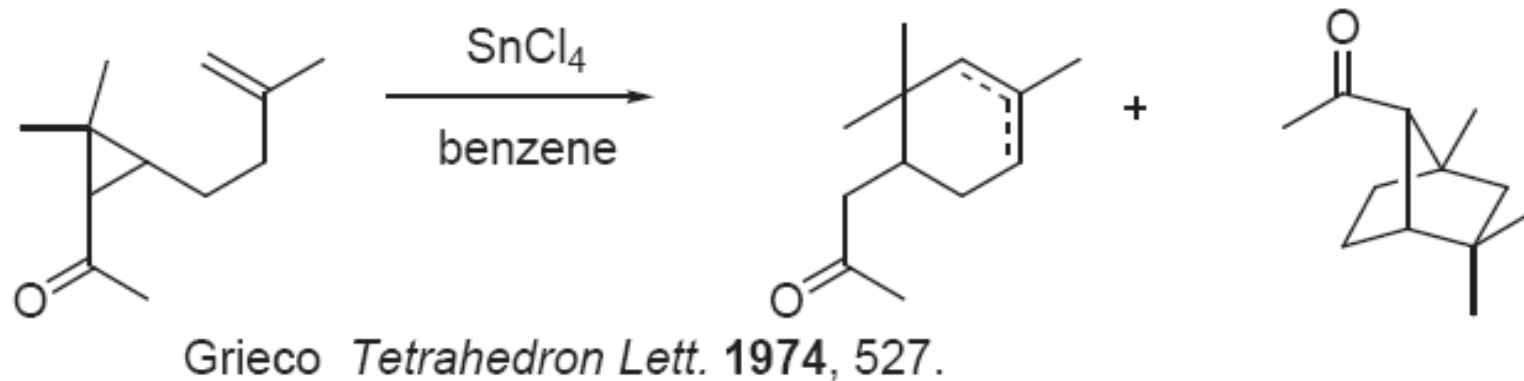
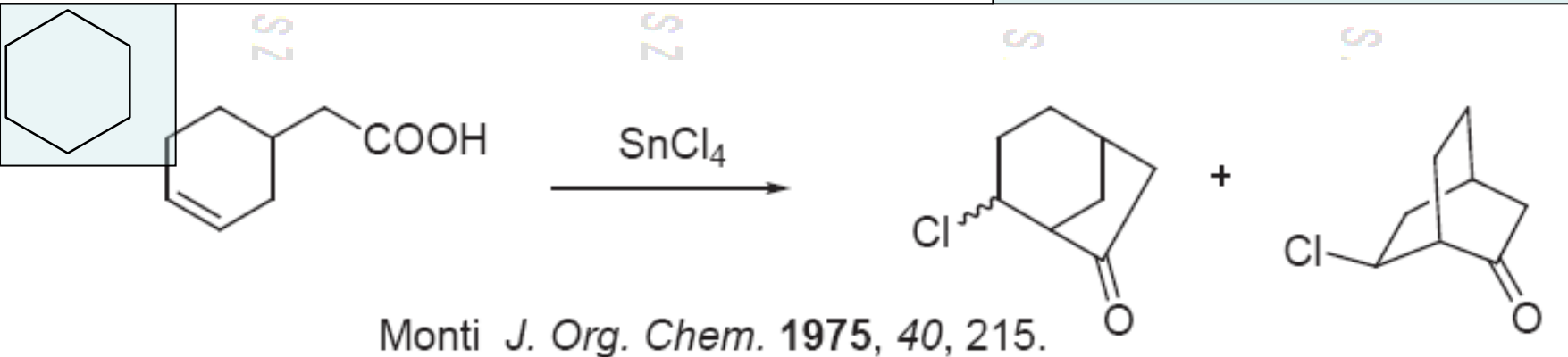


tsadze's lectur



tsadze's lectur

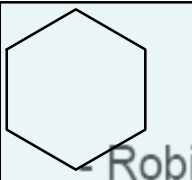




Goldsmith *J. Org. Chem.* 1970, 35, 3573.

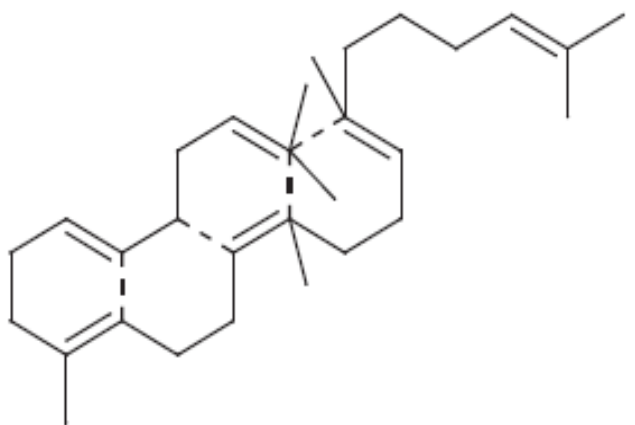
• Катион-алкен/алкин

Образование циклов



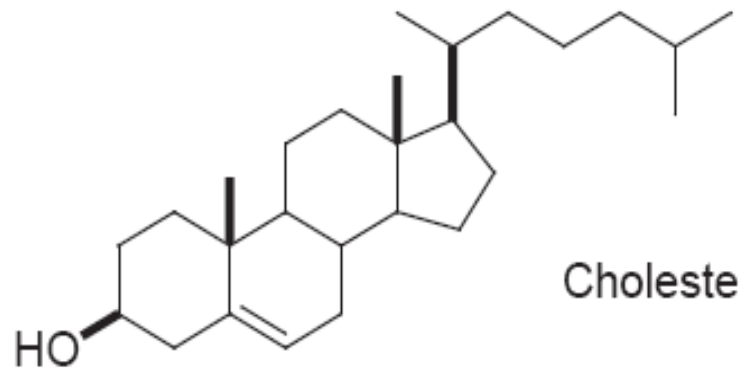
SZVa SZVa

- Robinson's proposal



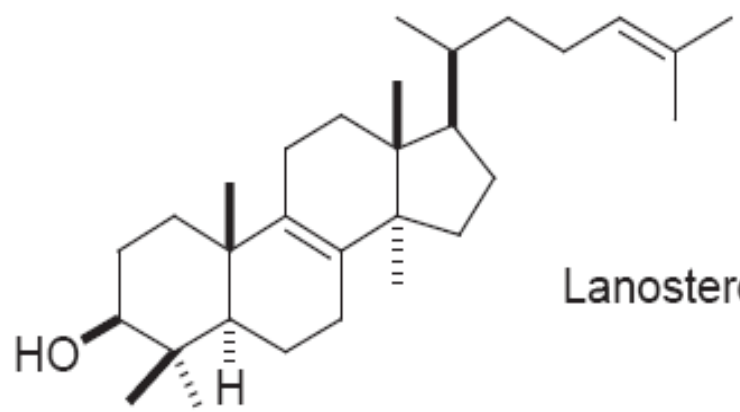
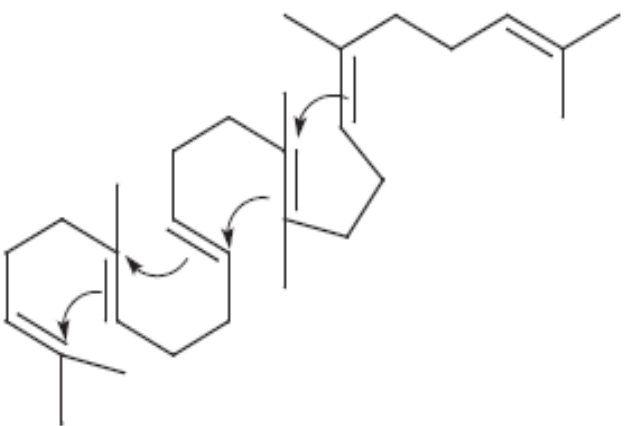
SZV_i

SZV_i



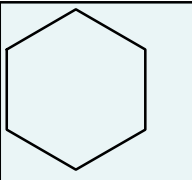
Cholesterol

- Correct cyclization scheme



Lanosterol

416 Lanosterol was proposed in 1953 by Woodward and Bloch.



Гипотеза Сторка-Эшенмозера: транс-анти-транс-стереохимия является результатом согласованной циклизации полиенов

S Z Vaisanze's lectures

S Z Vaisanze's lectures

S Z Vaisanze's lecture

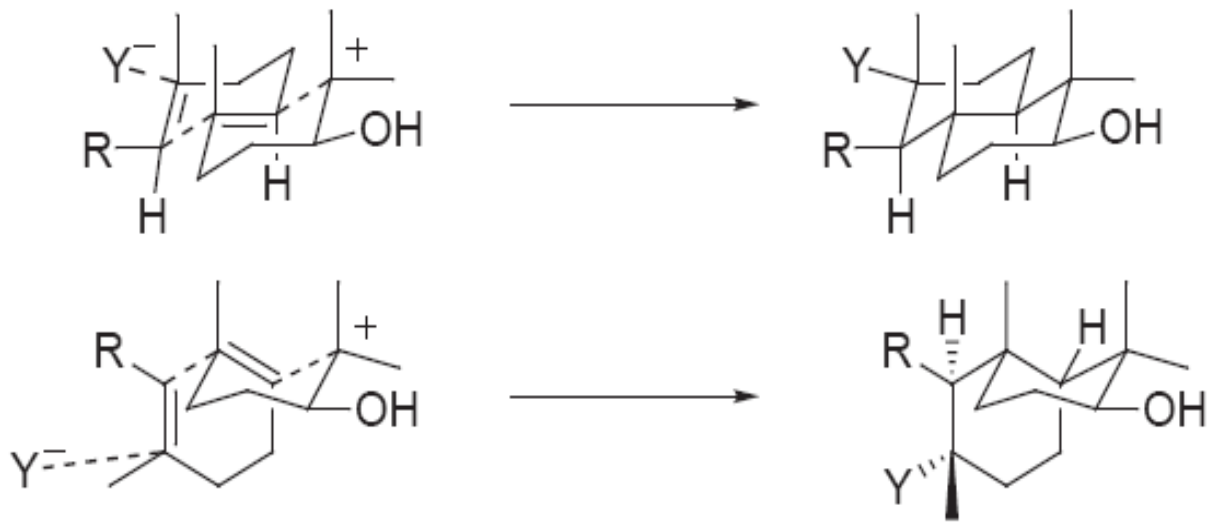
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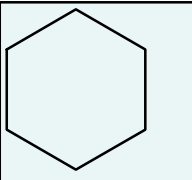
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Сравните с AdE стереохимией!!!

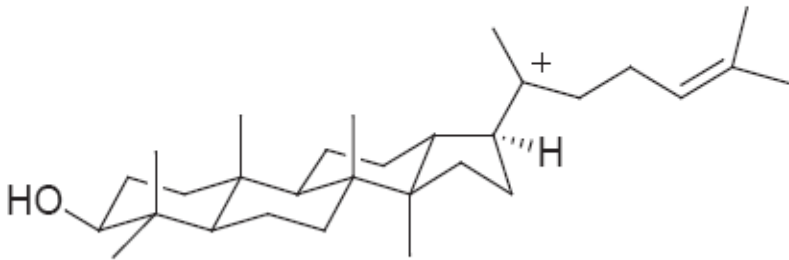
• Катион-алкен/алкин

Образование циклов



S Z Vaisadze's

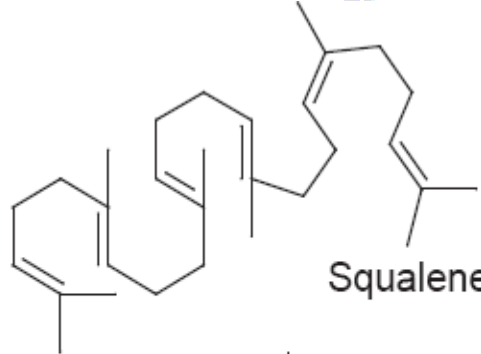
S Z Vaisadze's



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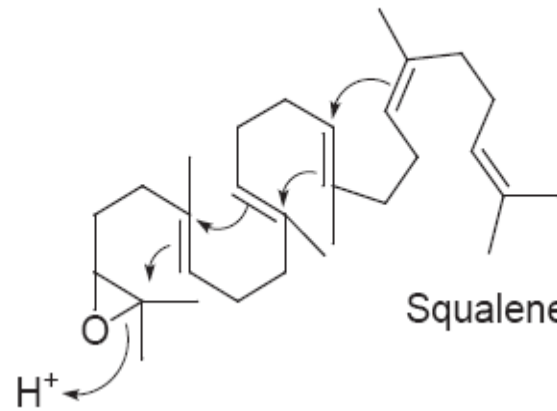
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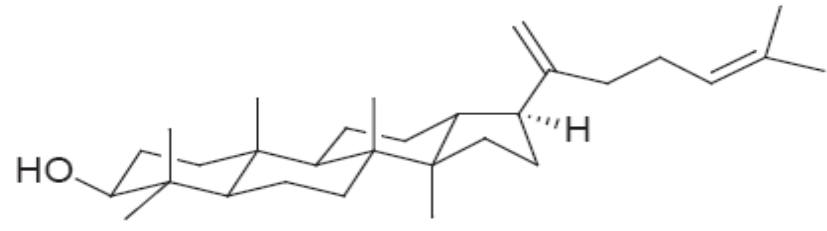
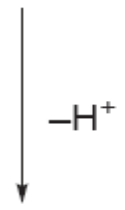
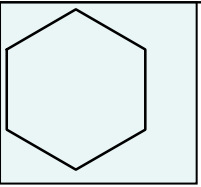


Squalene

Squalene monooxygenase

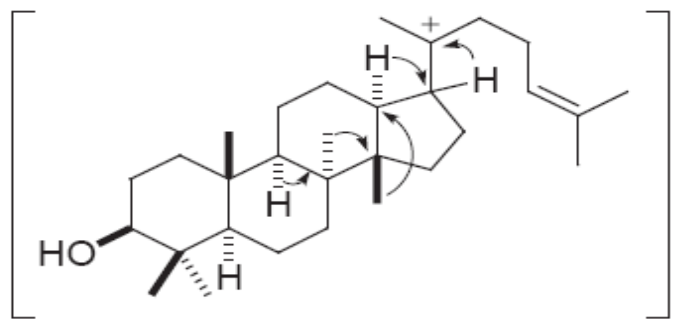
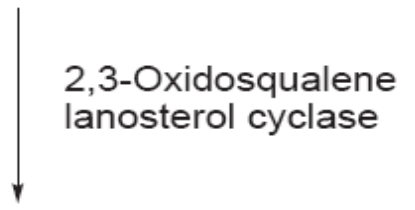


Squalene-2,3-oxide

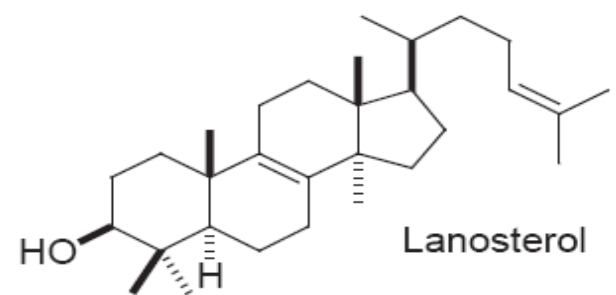
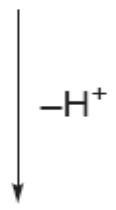


Dammaradienol

8 chiral centers with 256 possible stereoisomers



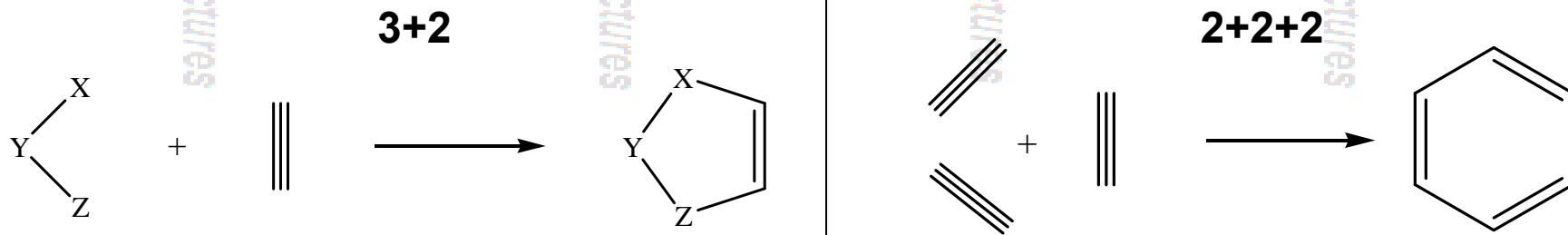
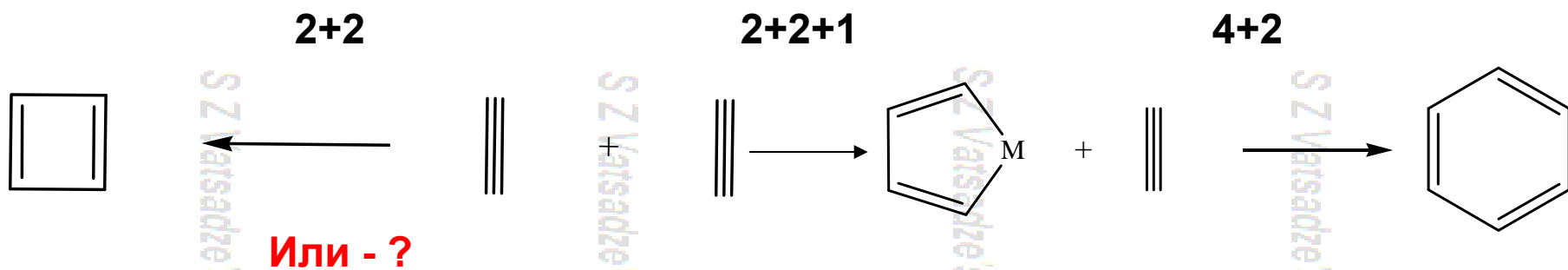
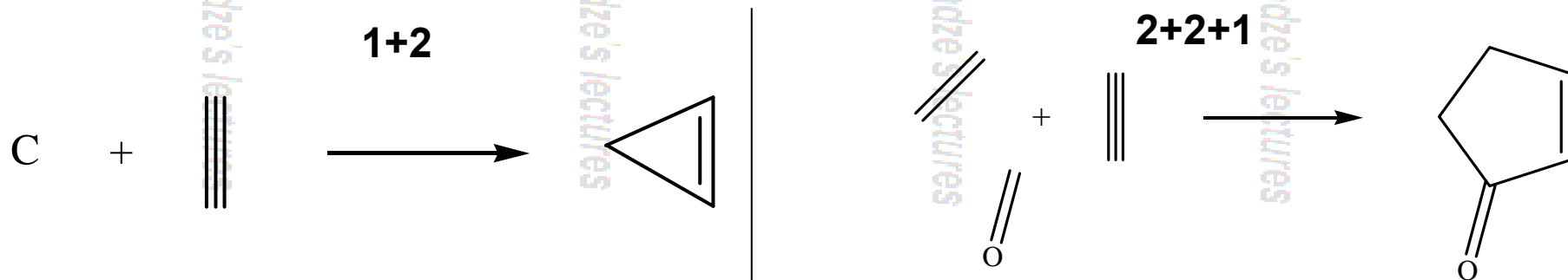
- Two methyl migrations and two hydride transfers



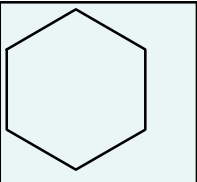
Lanosterol

Общие принципы

Алкин – поставщик C2-фрагмента. Следовательно, можно предложить несколько схем реакций циклизации с участием алкинов



- Фишеровские карбены



С алкинами обычно [3+2+1]

- Six-membered rings [3 + 2 + 1] (Fischer carbene addition to alkynes)

Dötz, *Fischer Transition Metal Carbene Complexes*, VCH: Deerfield Beach, FL, 1983.

Dötz *Angew. Chem., Int. Ed. Eng.* **1984**, 23, 587.

Casey in *Transition Metal Organometallics in Organic Synthesis*, Academic Press: New York, 1976, Vol. 1.

Dötz *Pure Appl. Chem.* **1983**, 55, 1689.

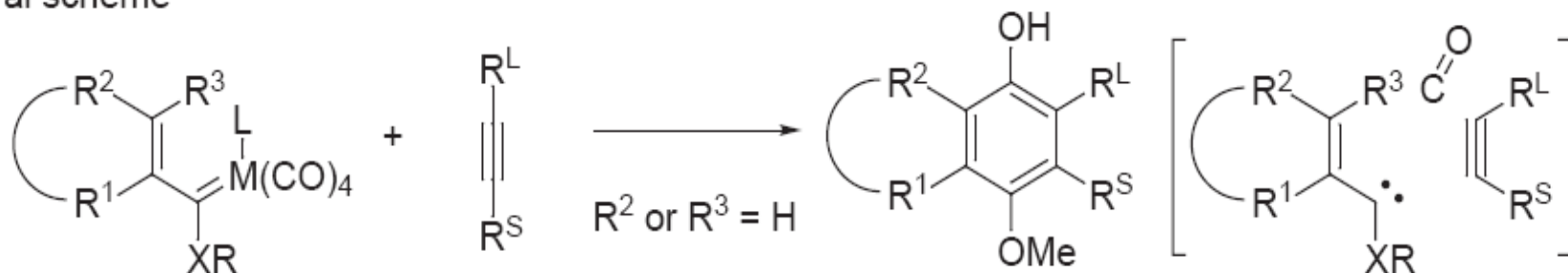
Casey in *Reactive Intermediates*, Wiley Interscience: New York, 1982, Vol. 2, and 1985, Vol. 3.

Hegedus *Principles and Applications of Organotransition Metal Chemistry*, University Science Books: Mill Valley, CA, 1987, 783.

Brown *Prog. Inorg. Chem.* **1980**, 27, 1.

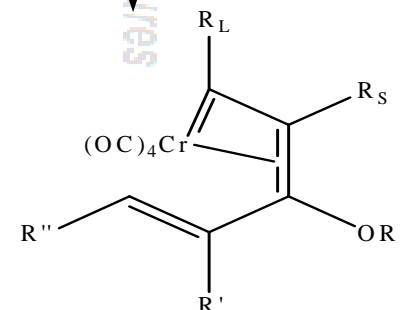
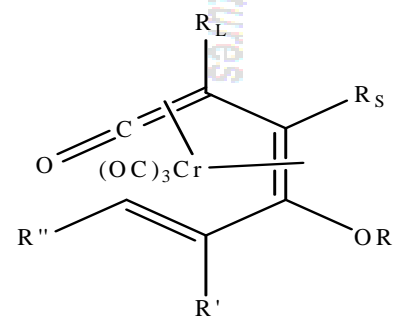
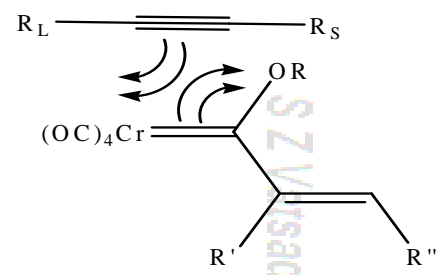
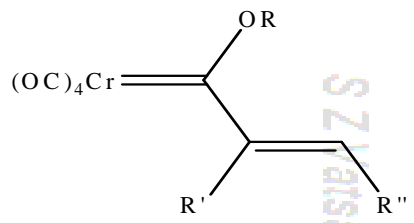
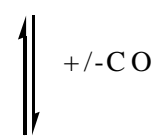
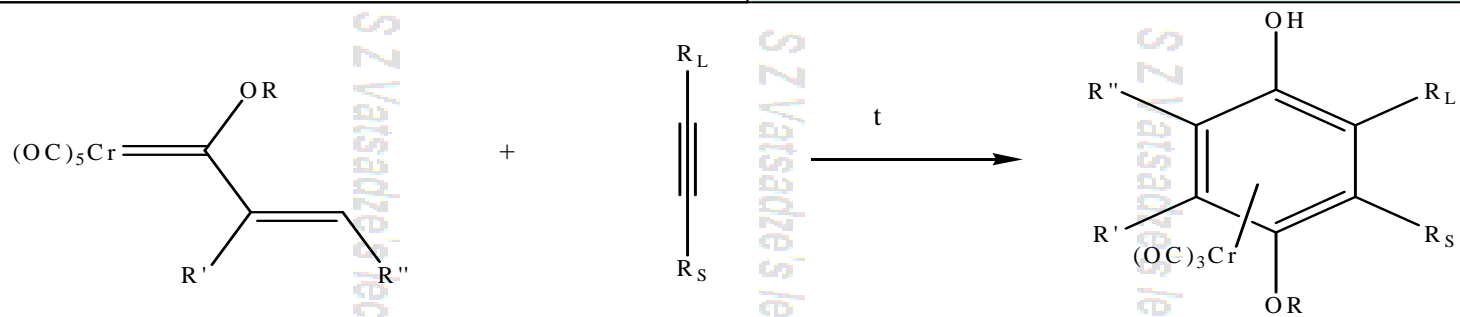
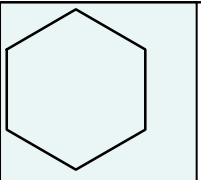
Wulff in *Advances in Metal-Organic Chemistry*, JAI Press: Greenwich, CT, 1989, Vol. 1.

- General scheme



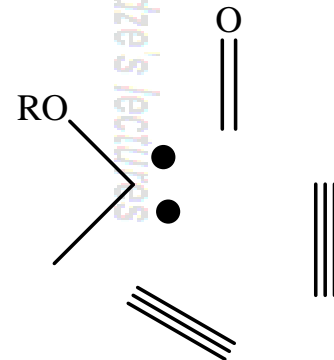
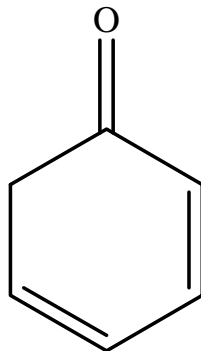
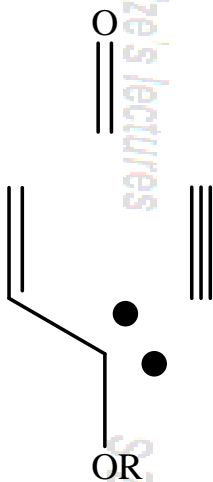
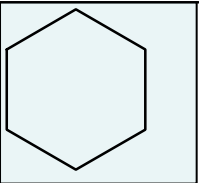
• 2+2: применение - реакция Дётса

Образование циклов



Закончить!

- 2+2: применение - реакция Дётса - обобщение



S Z Vatsadze's lectures

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S Z Vatsadze's lectures

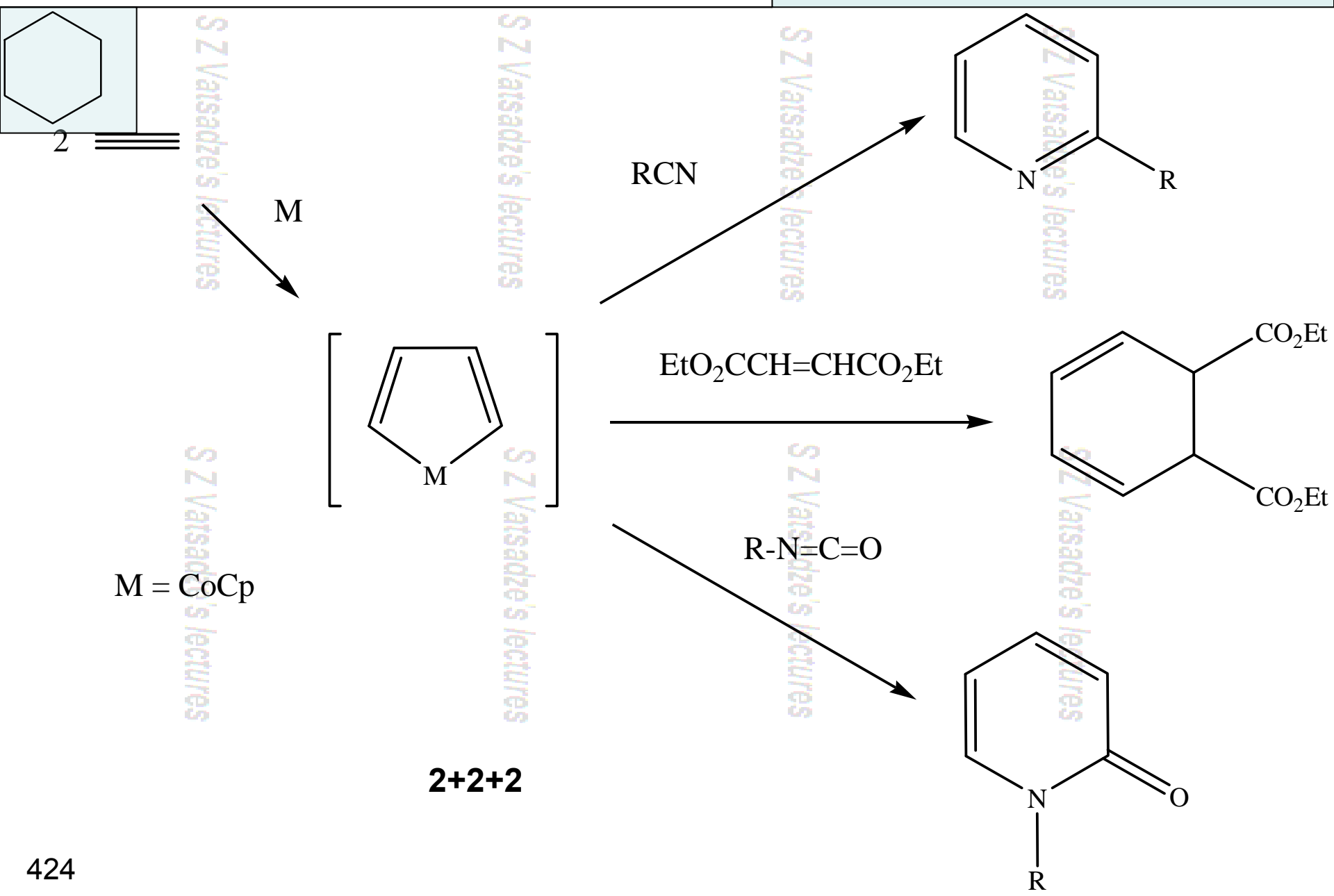
S Z Vatsadze's lectures

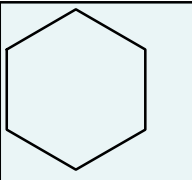
S Z Vatsadze's lectures

S Z Vatsadze's lectures

S Z Vatsadze's lectures

Каскадные реакции с металлами $(2+2)^n$

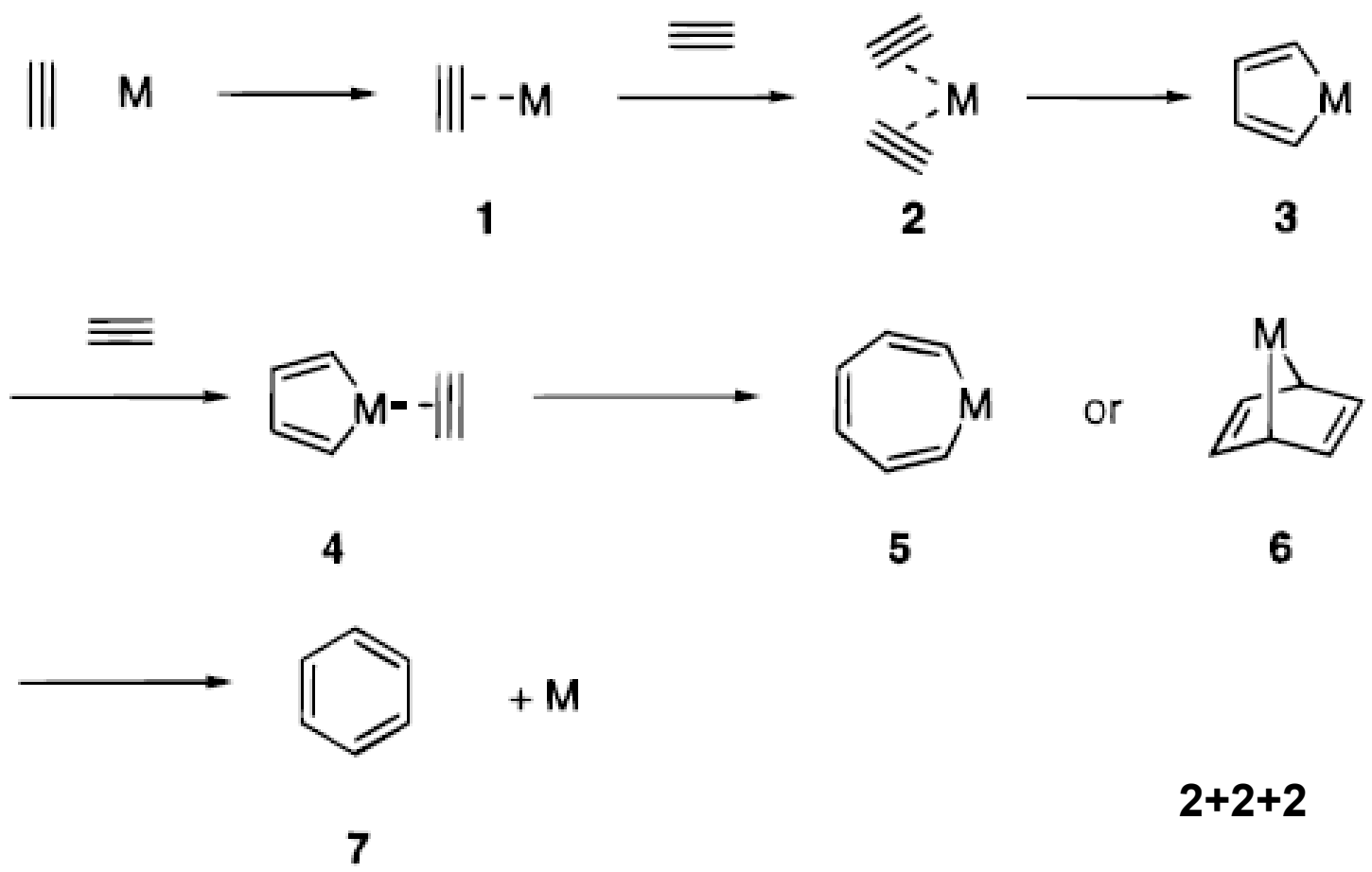


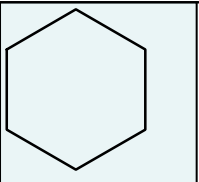


S Z Vaisadze's lect

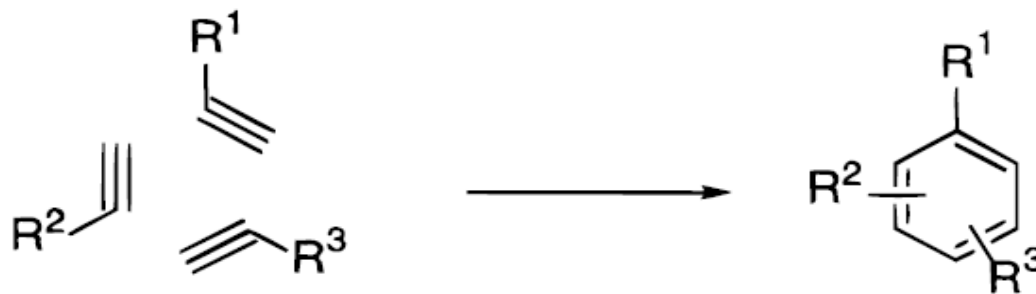
Recent Advances in the Transition-Metal-Catalyzed Regioselective Approaches to Polysubstituted Benzene Derivatives

Shinichi Saito* and Yoshinori Yamamoto*

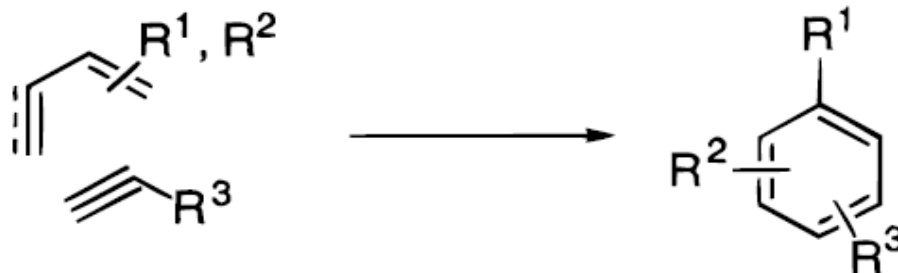




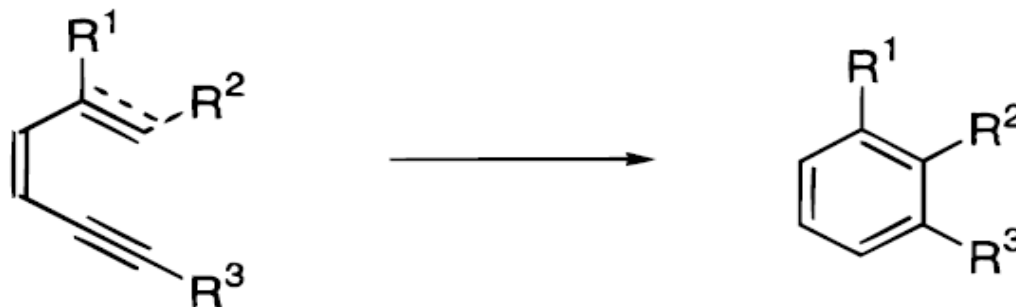
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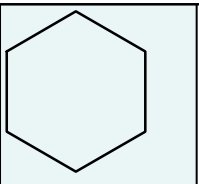
S Z Vatsadze's lectures



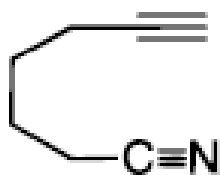
S Z Vatsadze's



2+2+2

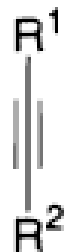


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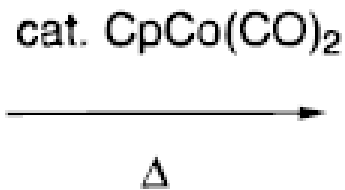


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S Z Vaisadz

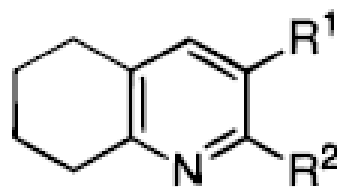


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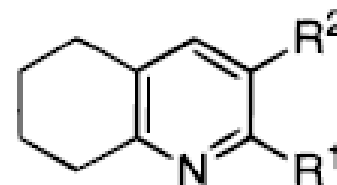
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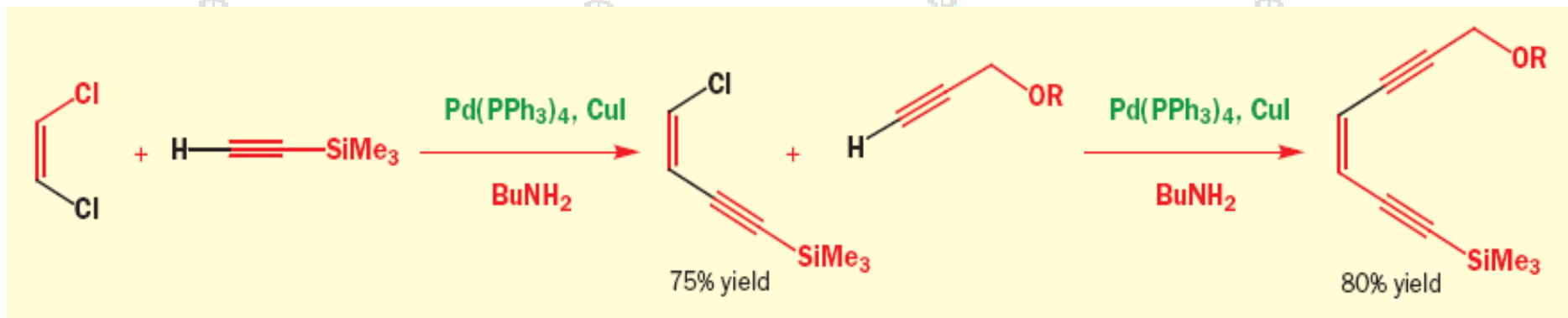
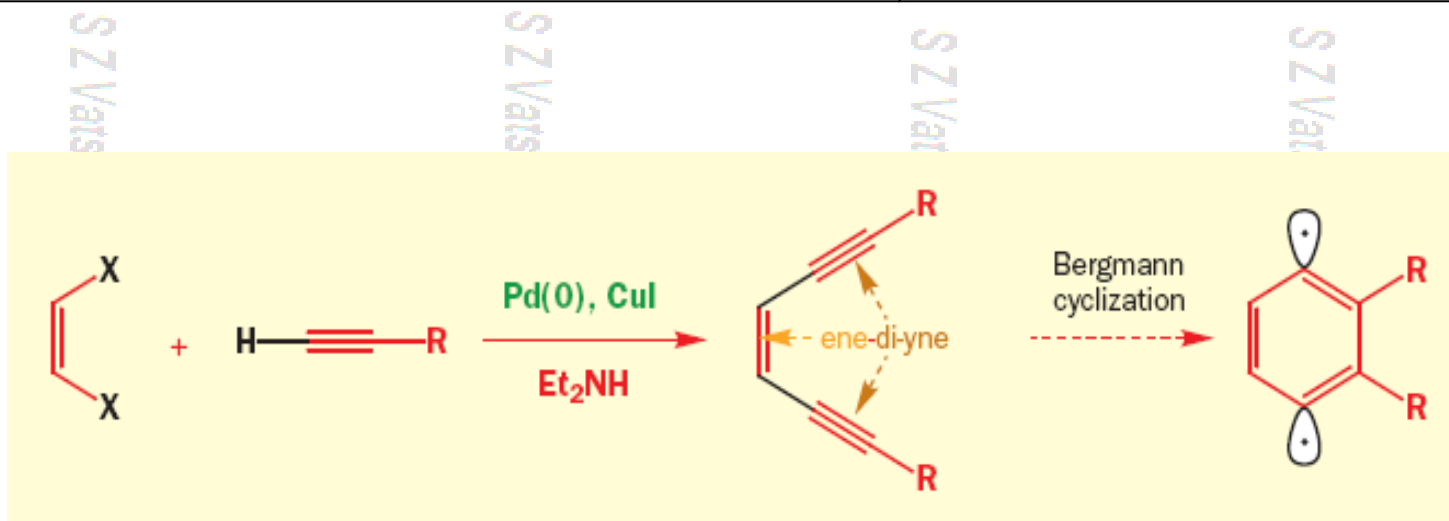
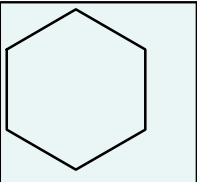
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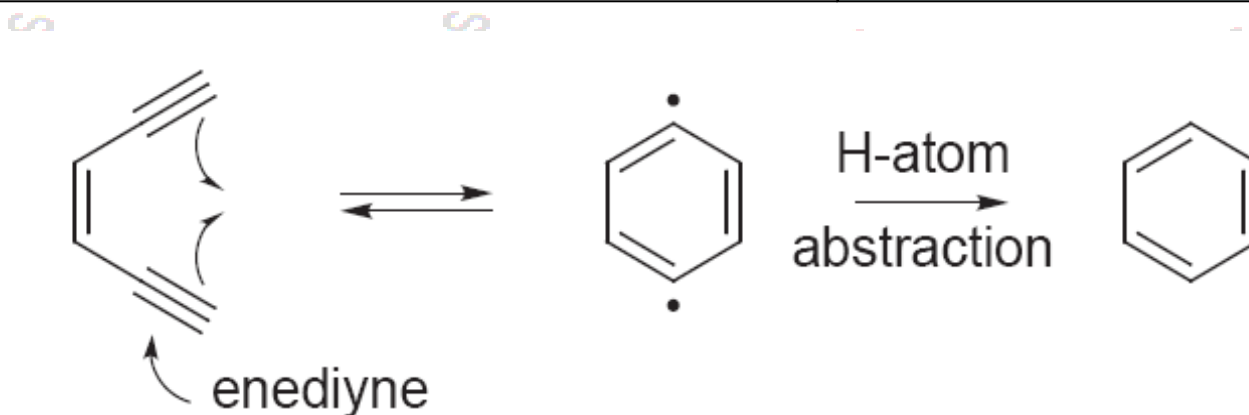
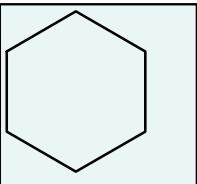
2+2+2

• Циклизация по Бергману

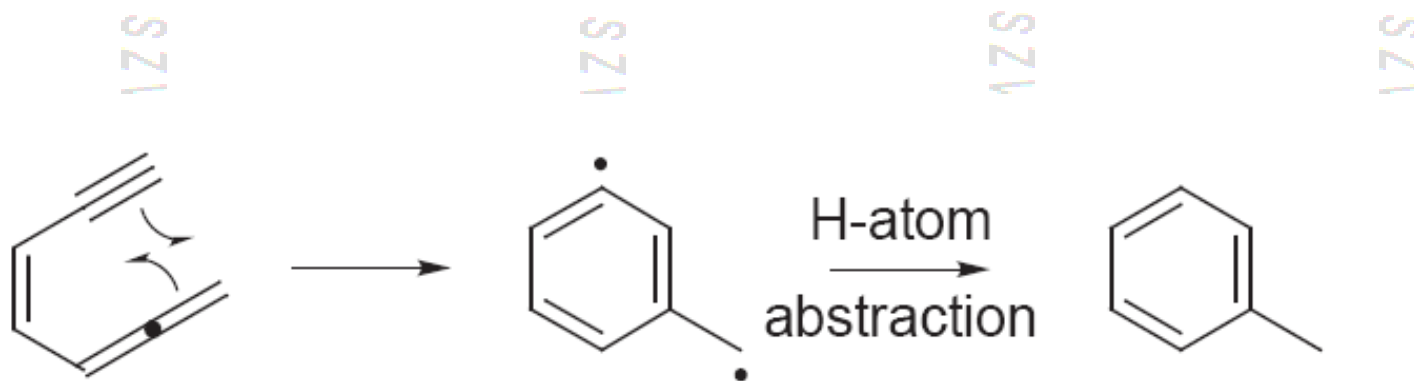


- Циклизация по Бергману и Майерсу-Сайто

Образование циклов

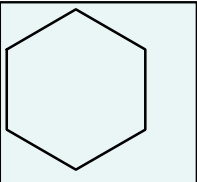


Bergman *J. Am. Chem. Soc.* **1972**, 94, 660.
Acc. Chem. Res. **1973**, 6, 25.



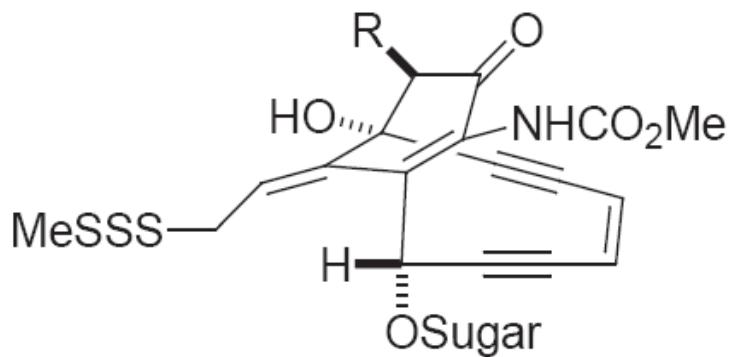
Myers *J. Am. Chem. Soc.* **1988**, 110, 7212; **1992**, 114, 9369.

- Циклизация по Бергману: эндиновые антибиотики



S Z Vaisadze's k

S Z Vaisadze's k

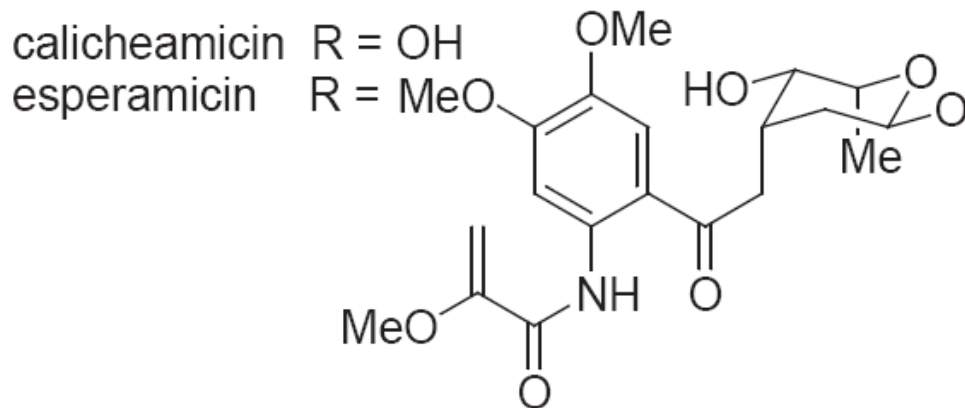


idze's lectures

idze's lectures

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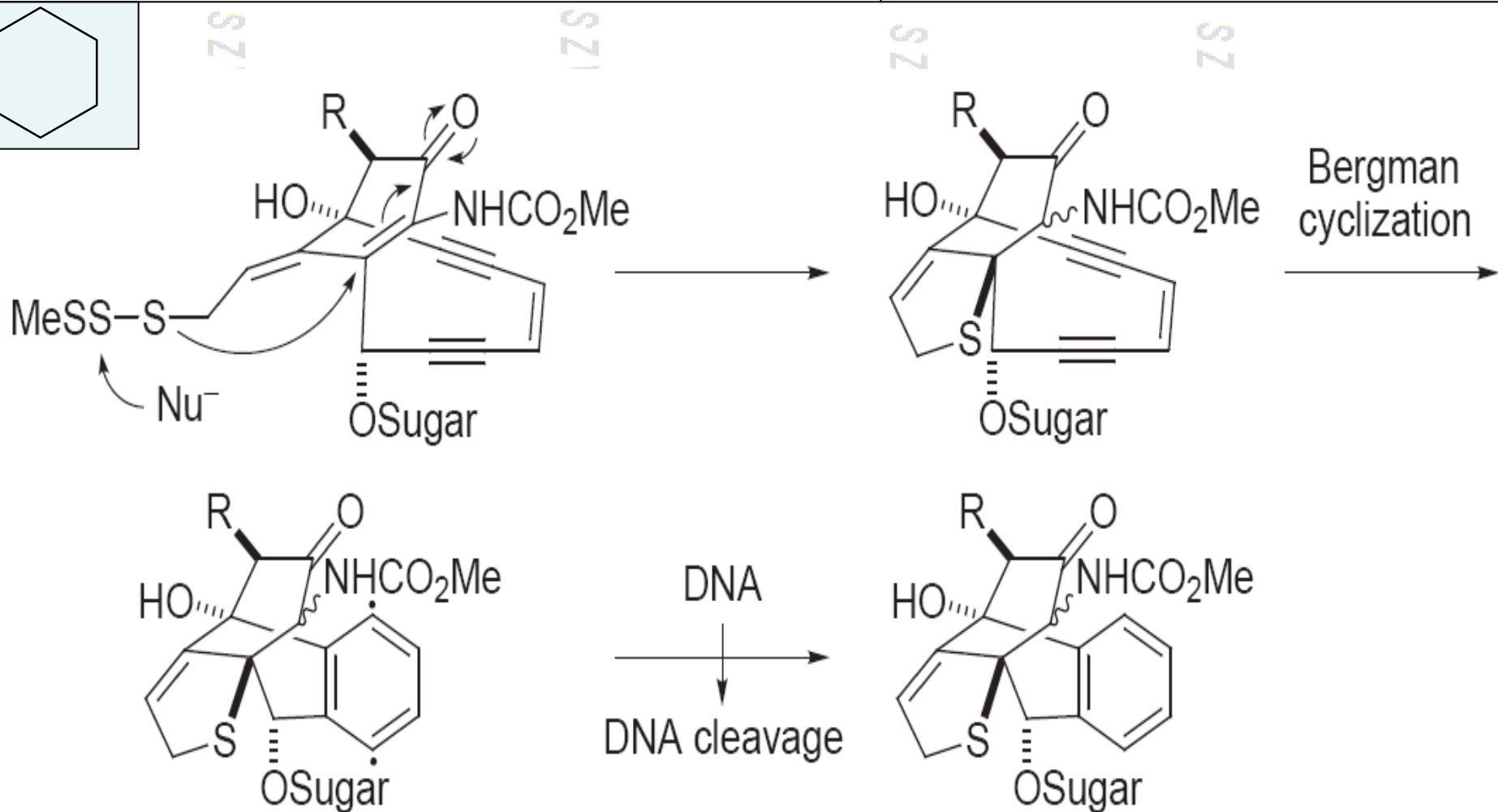
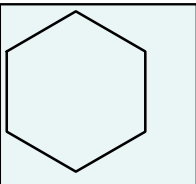


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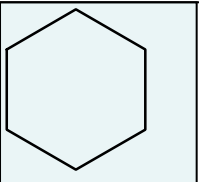
idze's lectures

- Циклизация по Бергману: эндиновые антибиотики

Образование циклов



- Циклизация по Майерсу-Сайто: ендиновые антибиотики

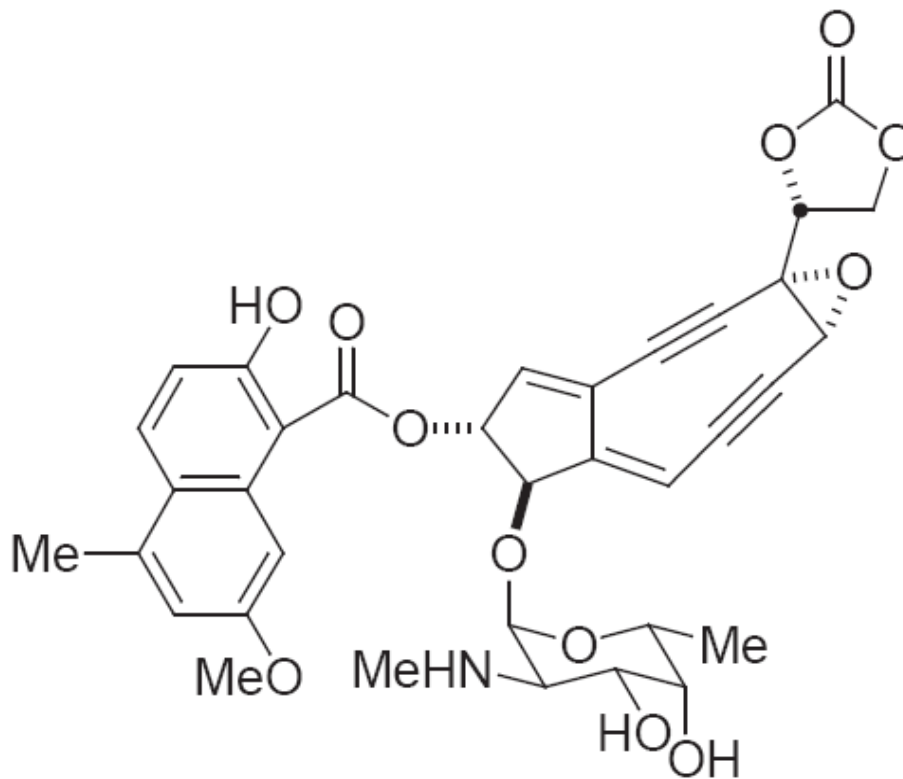


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• Циклизация по Майерсу-Сайто: ендиновые антибиотики

Образование циклов

