

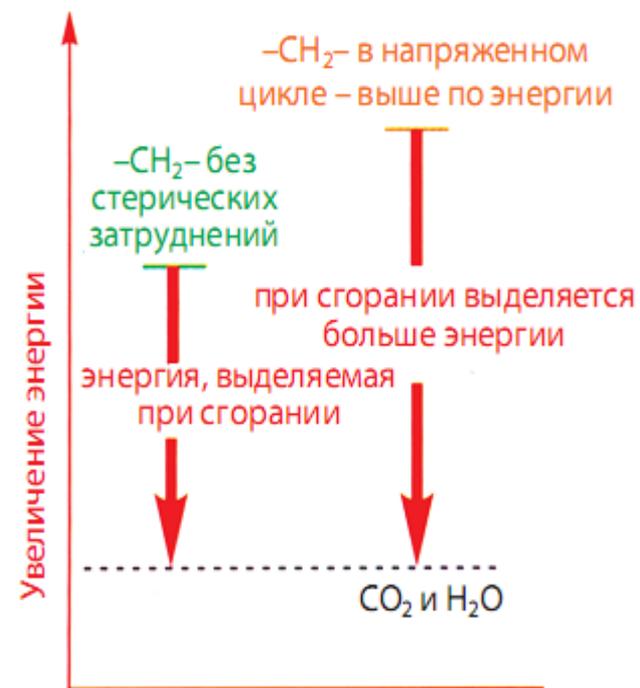
7. Образование циклов

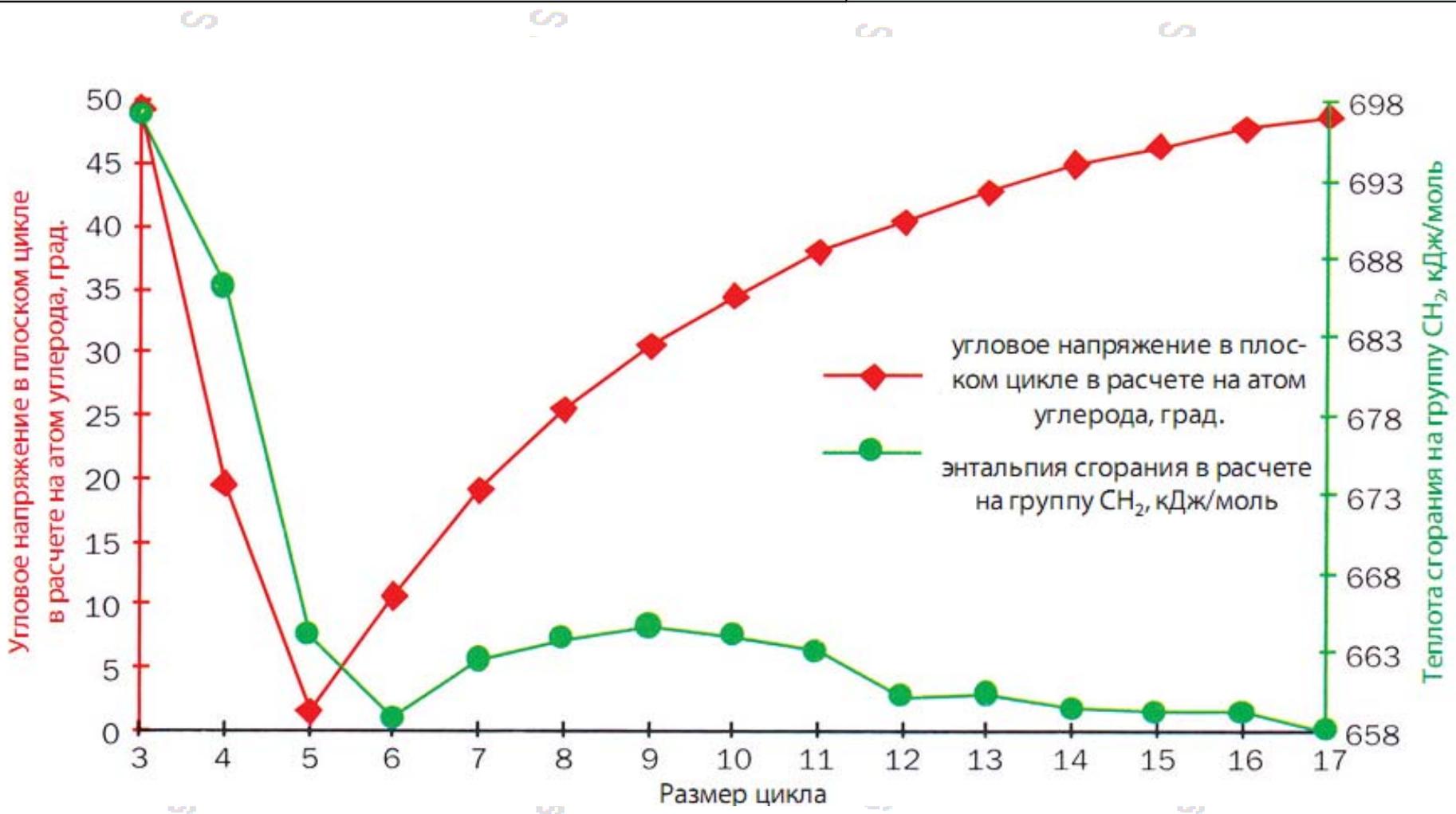
- Общие подходы к синтезу циклических молекул, понятие о циклизации и циклоприсоединении, способы сборки циклов
- Особенности реакций циклизации: правила Болдуина
- Трехчленные циклы: циклопропаны, азиридины, оксираны
- Четырехчленные циклы: циклобутаны, оксетаны
- Циклизации «анион-алкен/алкин», «радикал-алкен/алкин», «катион-алкен/алкин»
- Пятичленные циклы
- Шестичленные циклы

Общая энергия напряжения является суммой трех основных составляющих: углового напряжения, напряжения заслоненных, противостоящих С-Н связей и так называемого трансаннулярного напряжения. Угловое напряжение (синонимы напряжение углерод-углеродных связей или напряжение Байера) вызвано растяжением или сжатием валентных углов и отклонением их от тетраэдрического $109^{\circ}28'$. Напряжение заслоненных С-Н связей (синонимы торсионное напряжение или напряжение Питцера) по своей природе аналогично отталкиванию атомов водорода в заслоненной конформации этана и других предельных углеводородов для двух соседних заслоненных С-Н связей; энергии этого взаимодействия оценивается приблизительно в 1 ккал/моль. Трансаннулярное напряжение или напряжение Прелога обусловлено взаимодействием в пространстве двух или большего числа атомов водорода при атомах углерода на противоположных концах цикла. Его следует принимать во внимание, главным образом, для средних циклов C_8-C_{11} .

Таблица 18.2. Теплоты сгорания некоторых линейных алканов

Линейный алкан	n в $\text{CH}_3(\text{CH}_2)_n\text{CH}_3$	$-\Delta H_{\text{сгорания}}$, кДж/моль
Этан	0	1560
Пропан	1	2220
Бутан	2	2877
Пентан	3	3536
Гексан	4	4194
Гептан	5	4853
Октан	6	5511
Нонан	7	6171
Декан	8	6829
Ундекан	9	7487
Додекан	10	8148

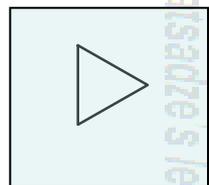




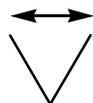
Напряжение по Питцеру и Прелогу

- Общие принципы образования циклов

Образование циклов

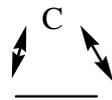


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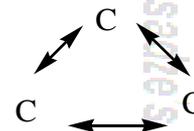


Циклизация

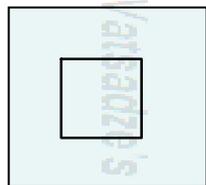
Циклоприсоединение



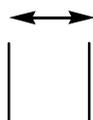
[1+2]



[1+1+1]

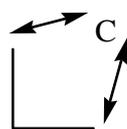


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Циклизация

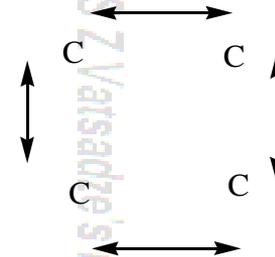
Циклоприсоединение



[1+3]

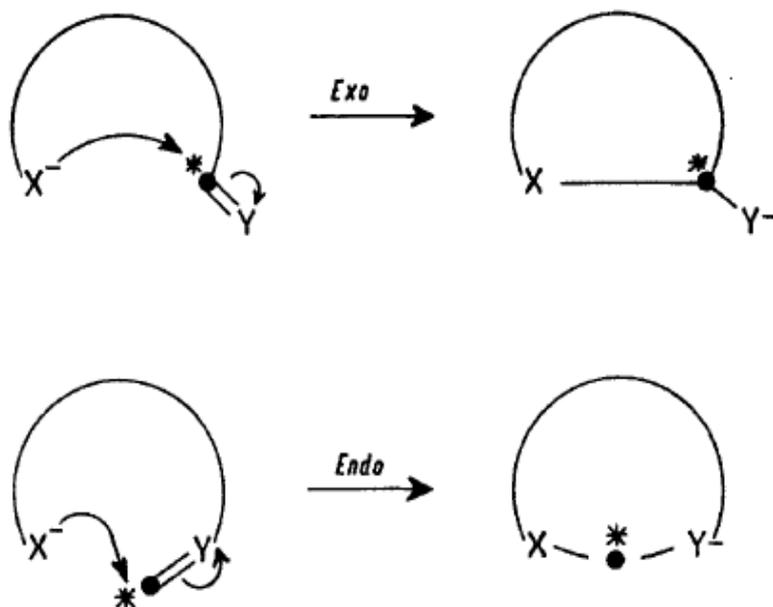


[2+2]



[1+1+1+1]

RING-FORMING reactions are important and common processes in organic chemistry. I now adumbrate a set of simple rules which I have found useful in predicting the relative facility of different ring closures. I believe these

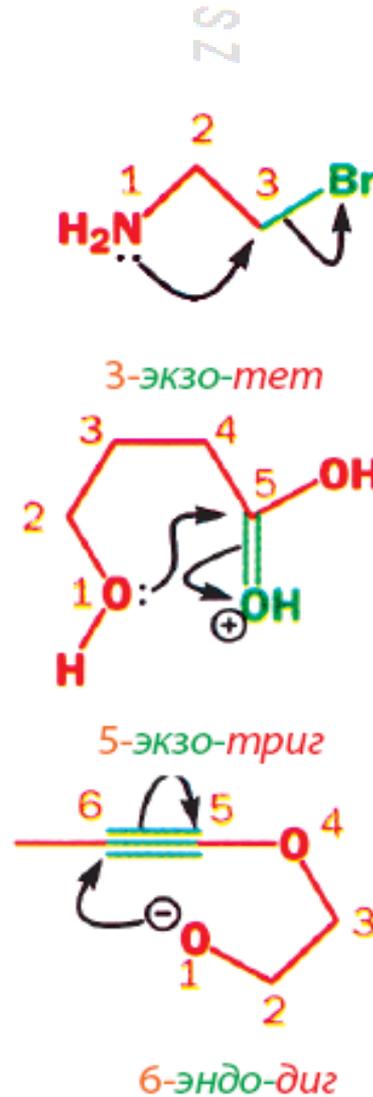


SCHEME 1

will be useful to organic chemists, especially in planning syntheses. Also these rules indicate certain experiments which may be helpful to define more precisely their limits. The rules are of a stereochemical nature and it is likely that unambiguous cases of all the possibilities I will discuss are as yet unknown.

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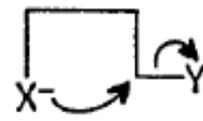


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3-Exo-Tet

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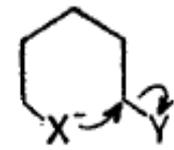
4-Exo-Tet

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5-Exo-Tet

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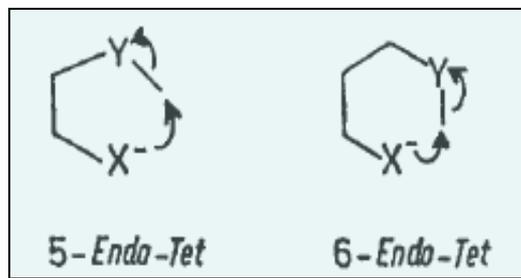


6-Exo-Tet

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7-Exo-Tet



5-Endo-Tet

6-Endo-Tet

SCHEME 2: Tetrahedral

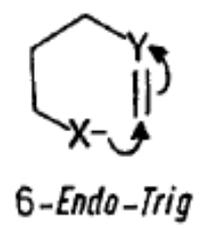
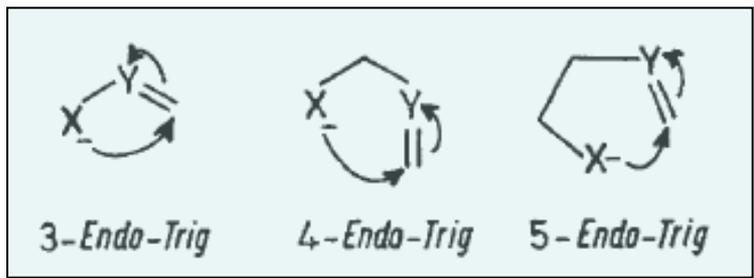
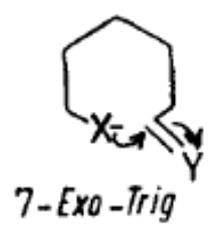
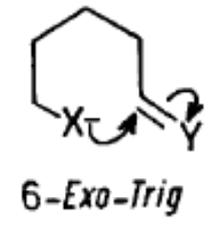
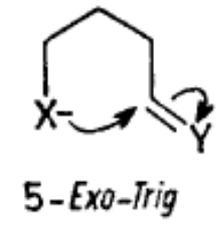
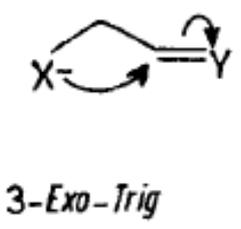
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SZ1

SZ

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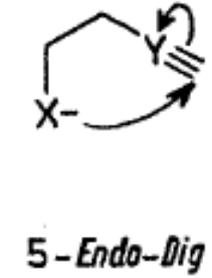
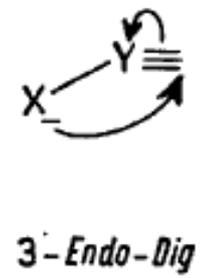
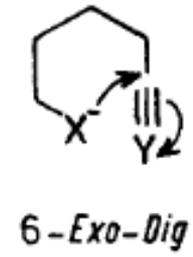
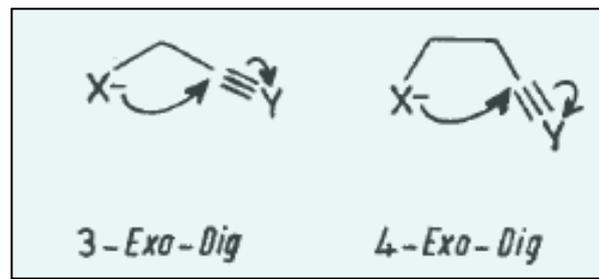
SCHEME 3: Trigonal

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SZVa

SZV

SZV



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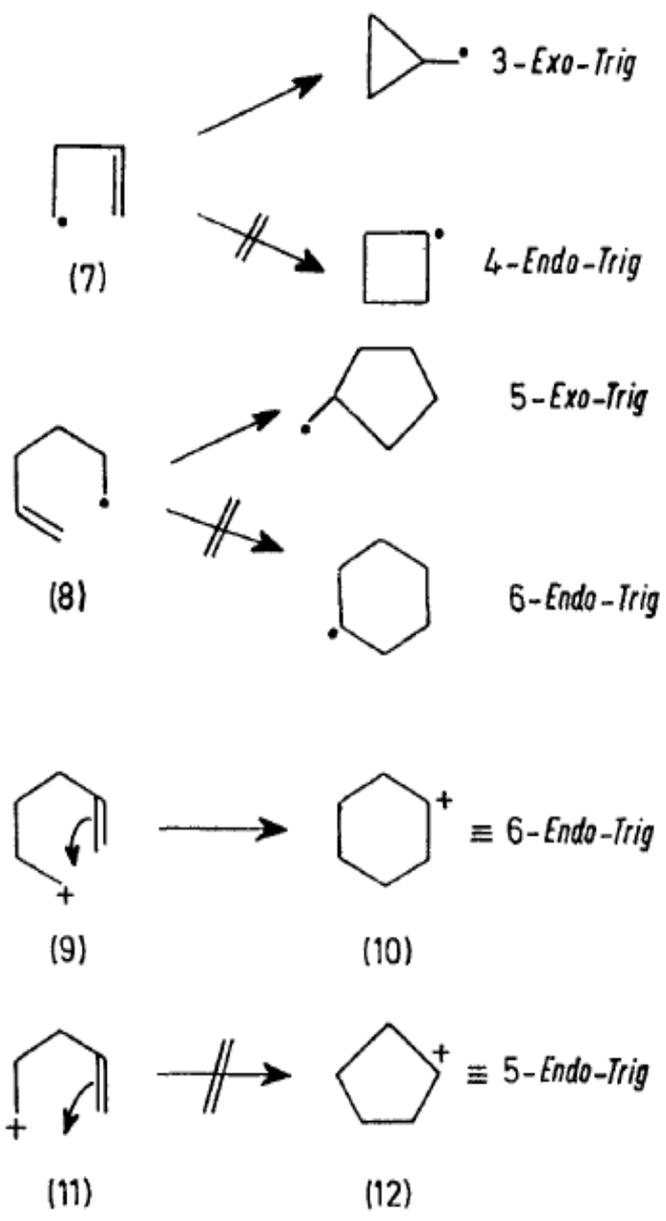
SCHEME 4: Digonal

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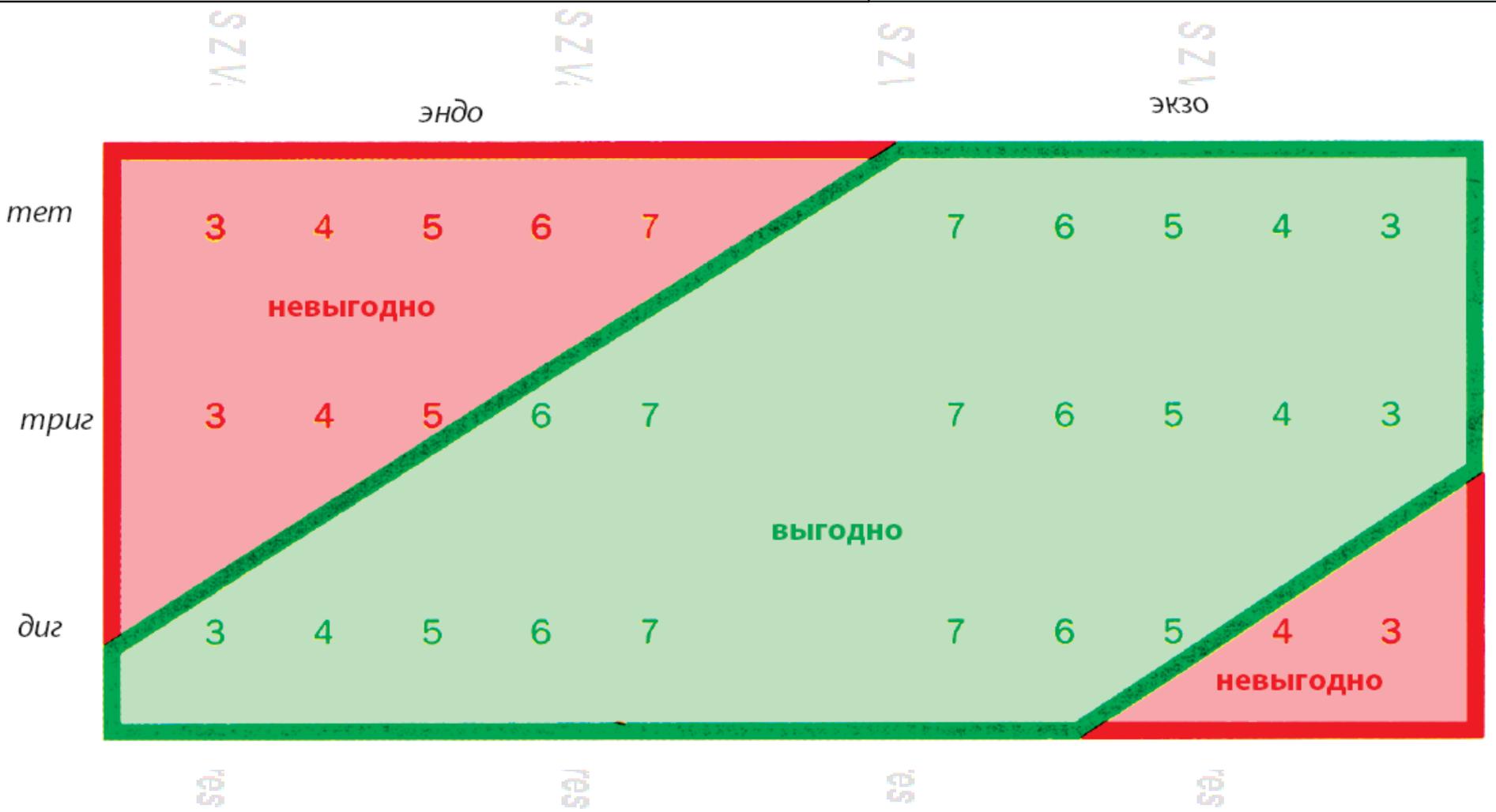
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S Z Vatsadze's lectures



• Правила Болдуина

Образование циклов





1. Циклопропаны:

- реакция Симмонса-Смита
- diaзосоединения
- илиды серы
- S_N2 -реакции

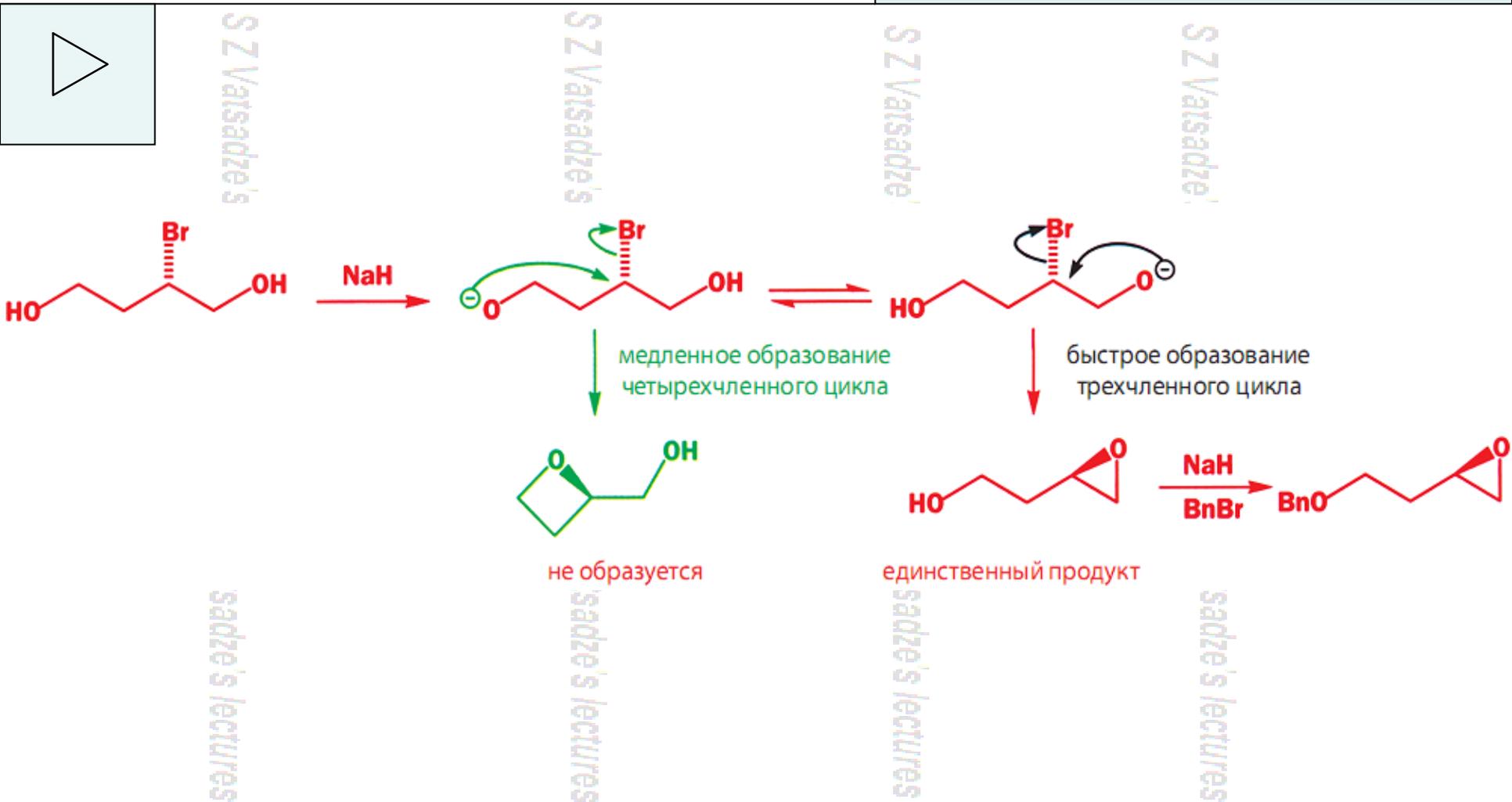
2. Эпоксиды:

- надкислоты, гидропероксиды, диоксираны
- галогенгидрины
- конденсация Дарзана
- илиды серы
- эпоксидирование, катализ. переходными металлами

3. Азиридины:

- нитрены
- S_N2 -реакции

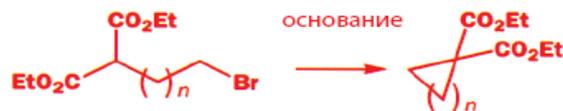
• S_N2 замыкание цикла



• S_N2 замыкание цикла



Таблица 42.3. Скорости реакций замыкания циклов



Размер цикла	Продукт	Относительная скорость ^a	Продукт реакции ^b	Относительная скорость ^a	Оценка реакции
3		0,07			С умеренной скоростью
4		0,001		0,58	Медленная
5		100		833	Очень быстрая
6		1		1	Быстрая
7		0,002		0,0087	Медленная
8				0,00015	Очень медленная

• Циклопропанирование



S Z Vaisadze

S Z Vaisadze

S Z Vaisadze

S Z Vaisadze

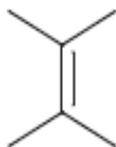
Parman, Schweizer *Org. React.* **1963**, 13, 55.

Moss *Acc. Chem. Res.* **1989**, 22, 15.

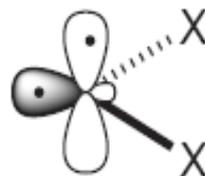
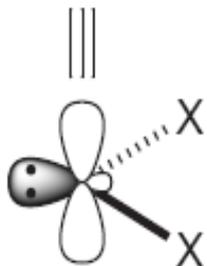
Acc. Chem. Res. **1980**, 13, 58.

Kostikov, Molchanov, Khlebnikov *Russ. Chem. Rev.* **1989**, 58, 654.

$2\pi^s + 2\omega^a$
cycloaddition



Addition of a singlet carbene proceeds by a concerted process in a *syn* fashion.



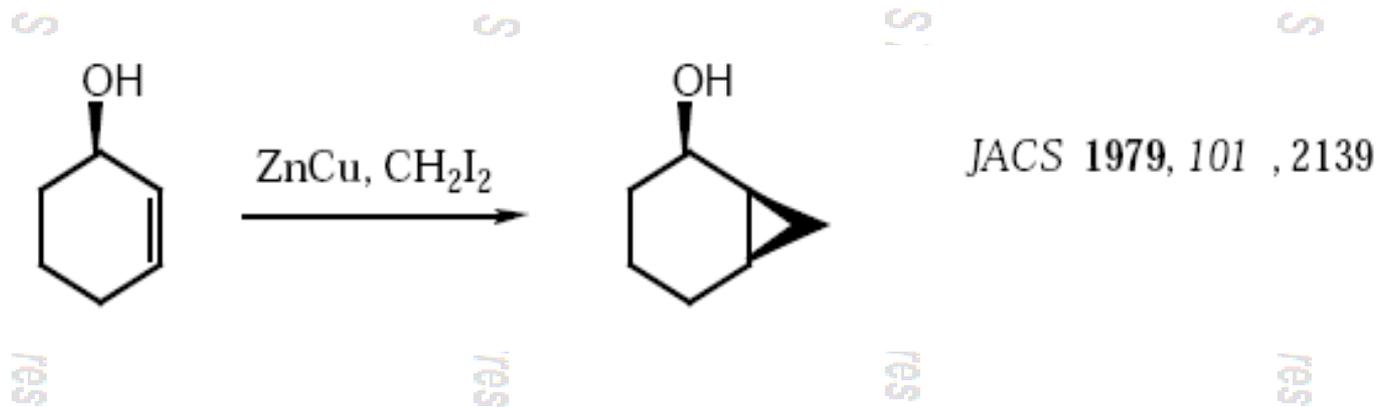
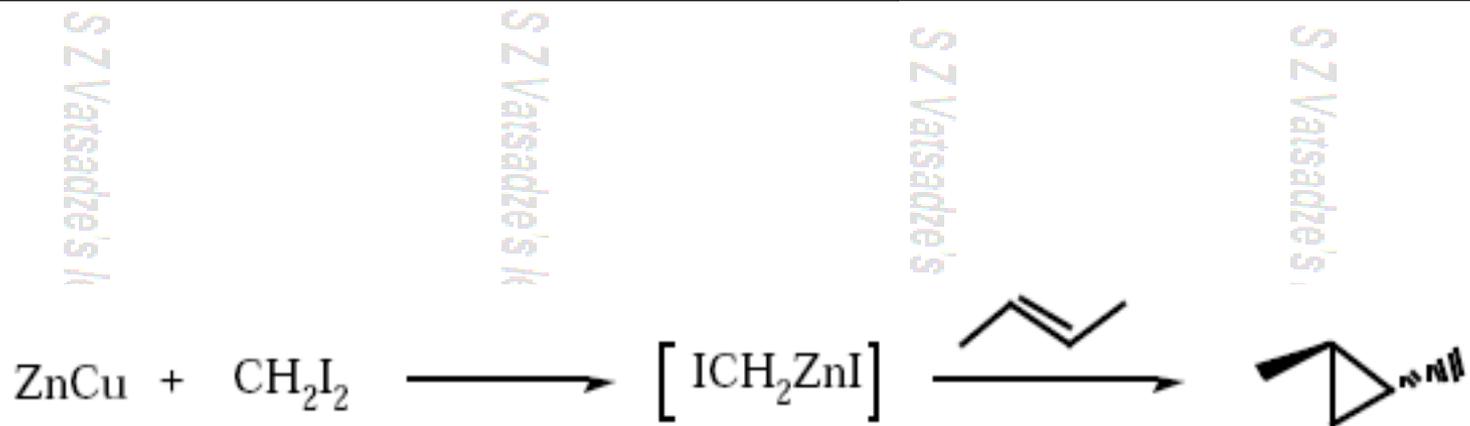
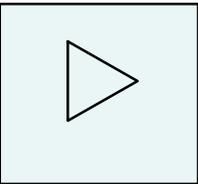
Triplet carbene behaves as a diradical.

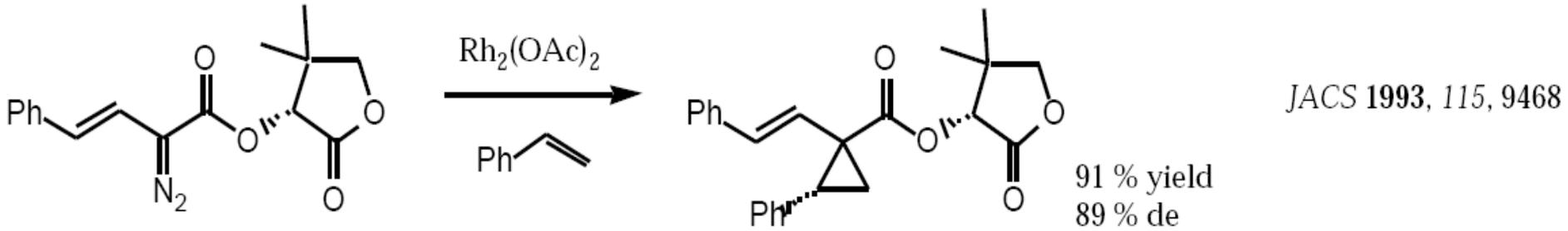
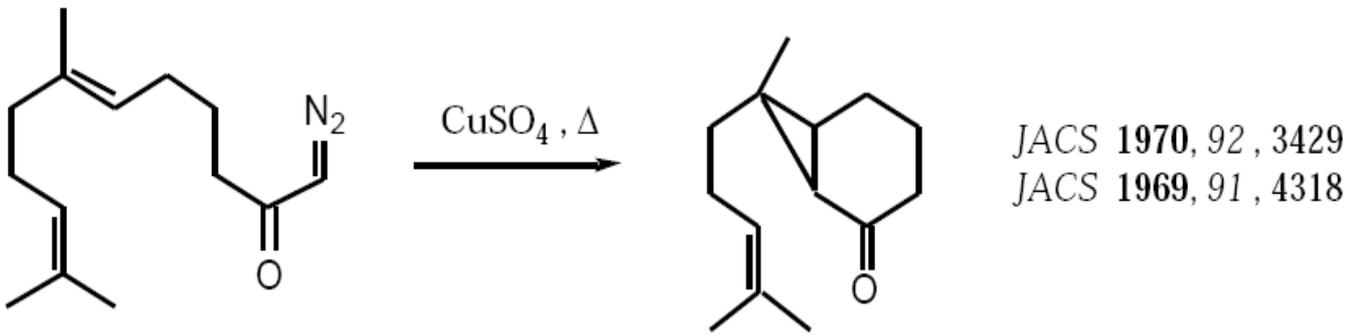
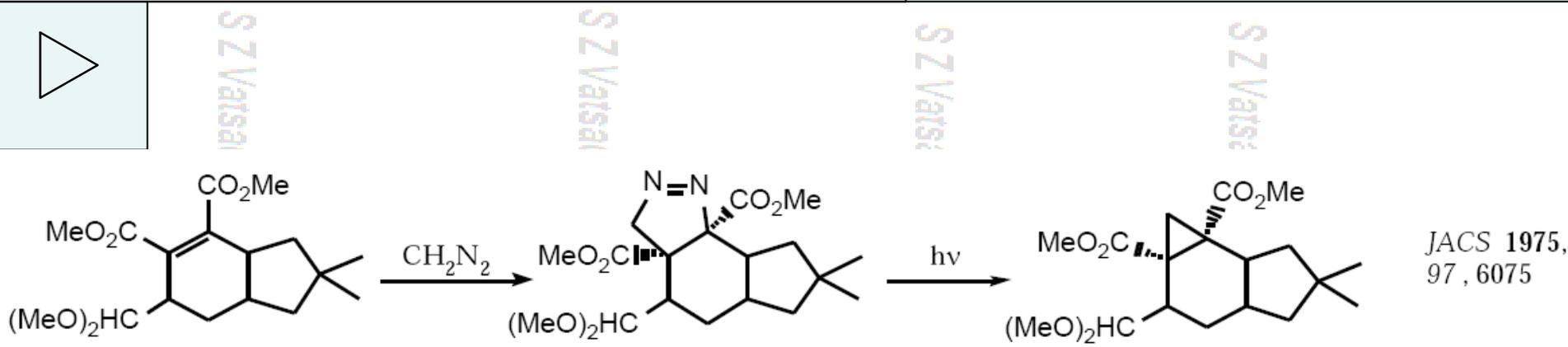
tuves

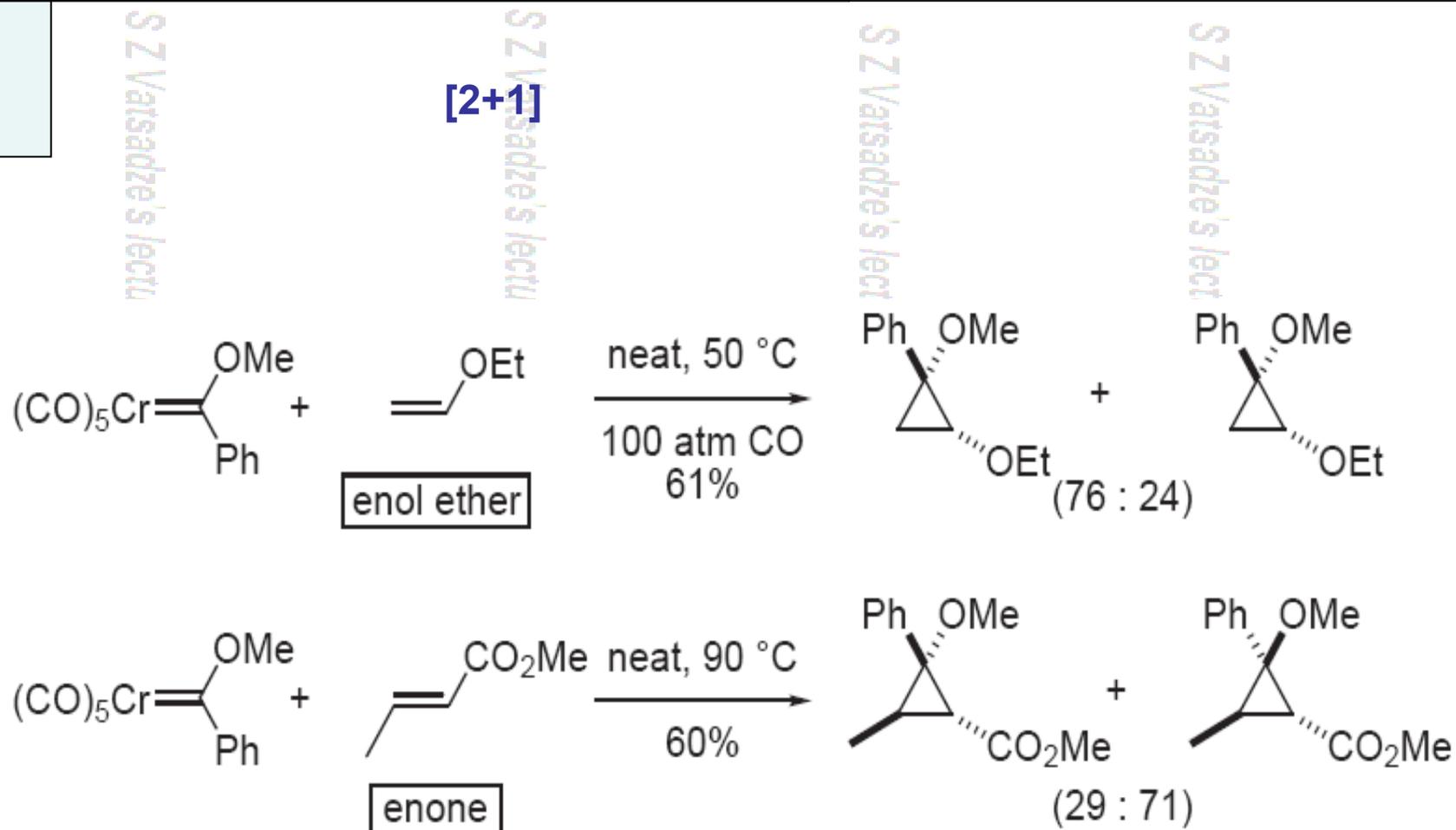
tuves

tuves

tuves



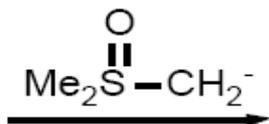
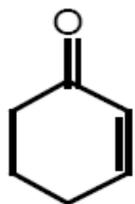




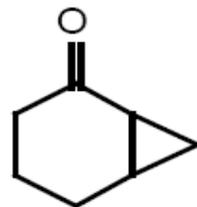
Fischer, Dötz *Chem. Ber.* **1972**, *105*, 3966.
Chem. Ber. **1972**, *105*, 1356.



S Z Vatsadze



S Z Vatsadze

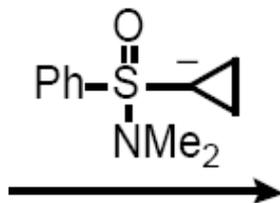
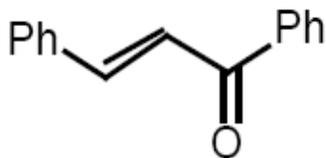


S Z Vatsadz

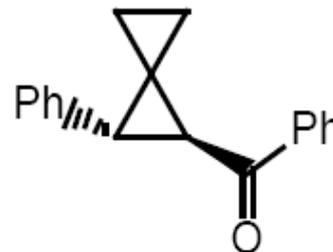
Tetrahedron
1987, 43, 2609

S Z Vatsadze's lectures

S Z Va



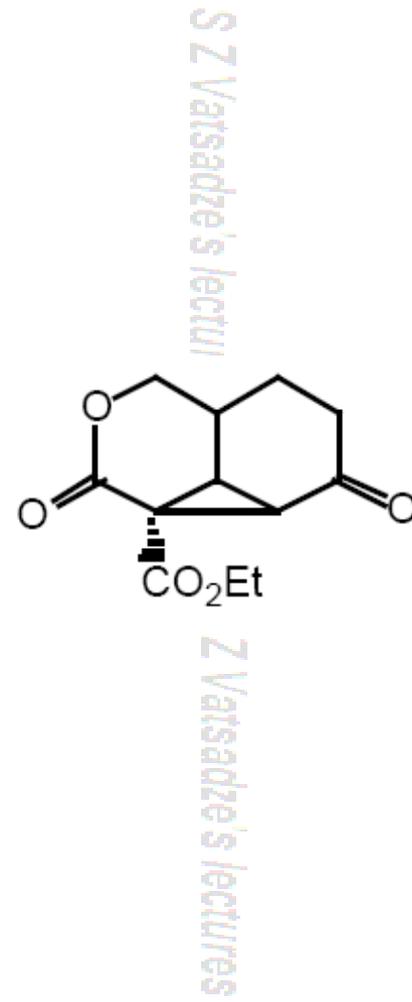
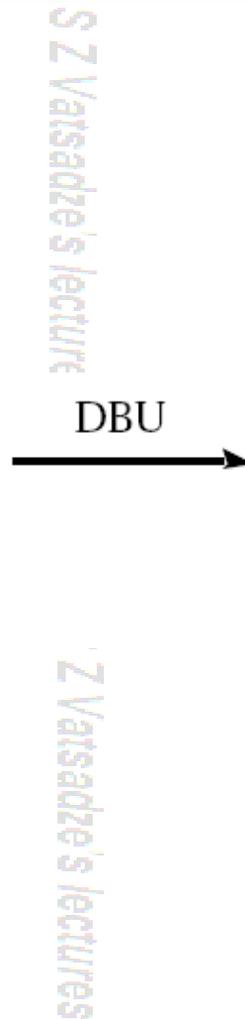
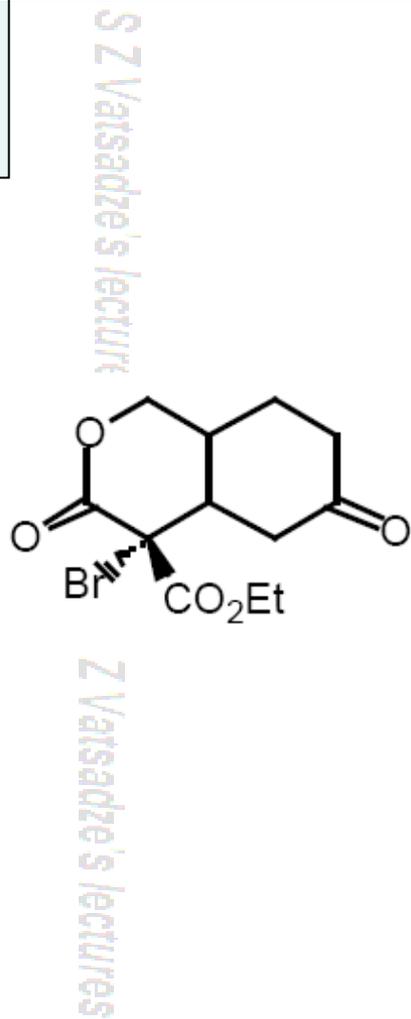
S Z Va



S Z Va

ACR 1973,
6, 341

S Z Va



S Z Vatsadze's lectu

JACS 1978,
99, 1940

Z Vatsadze's lectures

• Циклизации типа «радикал-радикал»

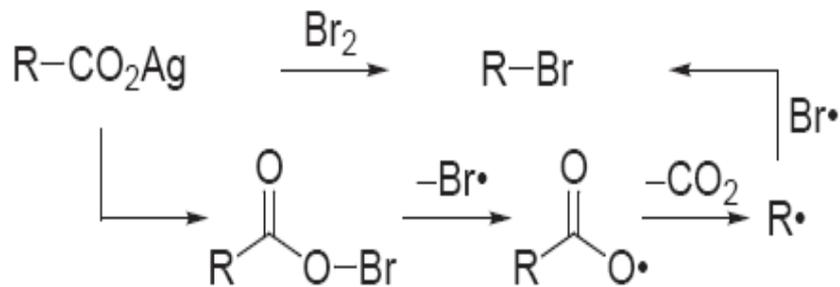


S Z Vaisadze's k

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Hunsdiecker, H.; Hunsdiecker, C. *Ber.* 1942, 75, 291.



Wiberg *Acc. Chem. Res.* 1984, 17, 379.

lectures

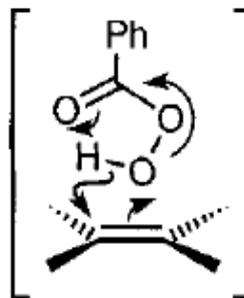
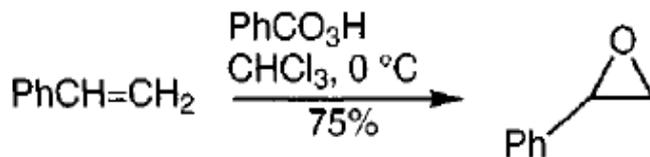
lectures

lectures

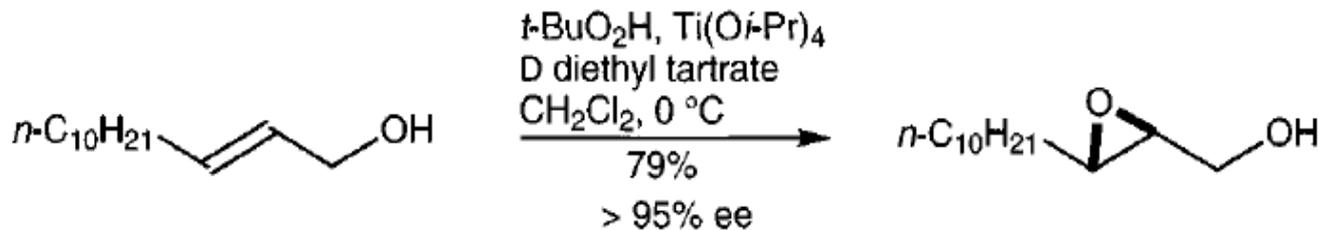
lectures



-надкислоты, гидропероксиды, диоксираны

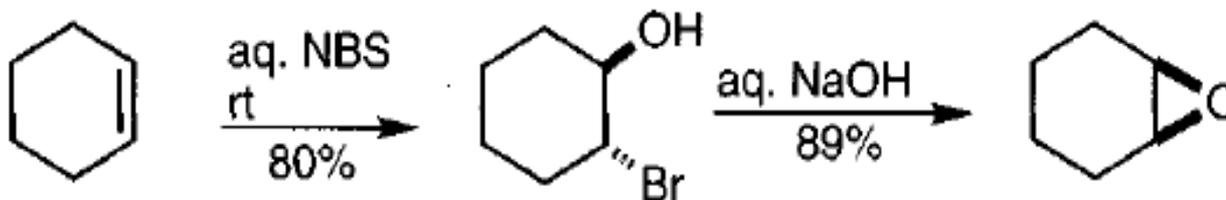


-эпокси́дирование, катализируемое переходными металлами
(Шарплесс, Якобсен)

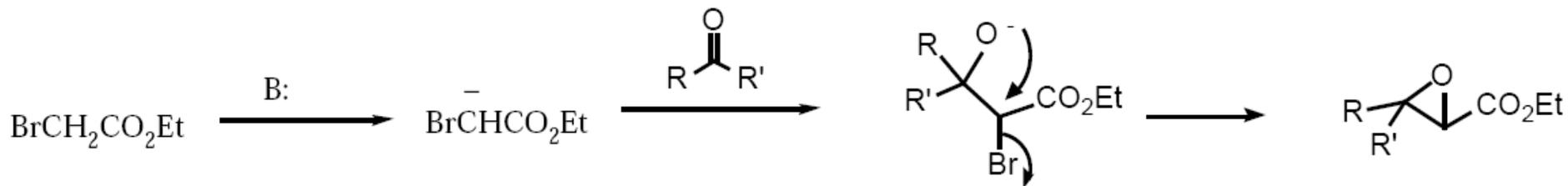




-галогенгидрины:

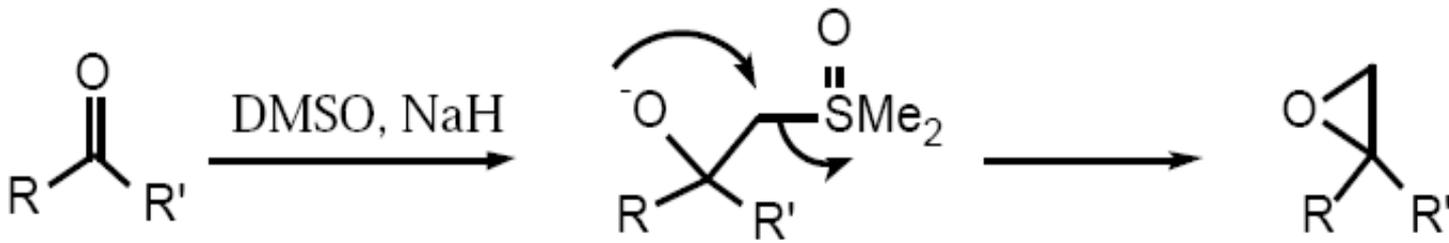


-конденсация Дарзана





-илиды серы



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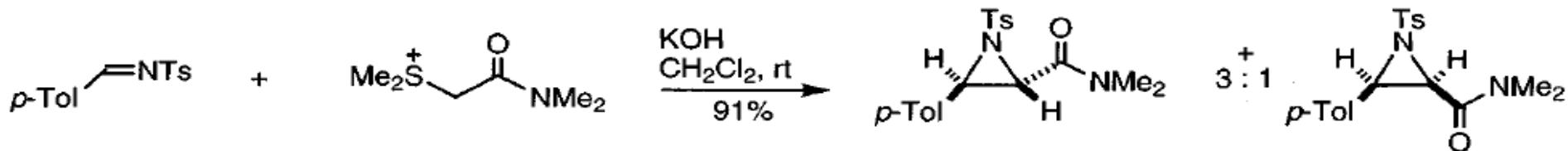
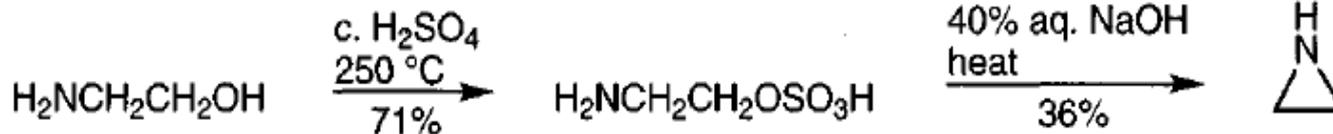
S Z Vatsadze's lectures

S Z Vatsadze's lecture

S Z Vatsadze's lecture

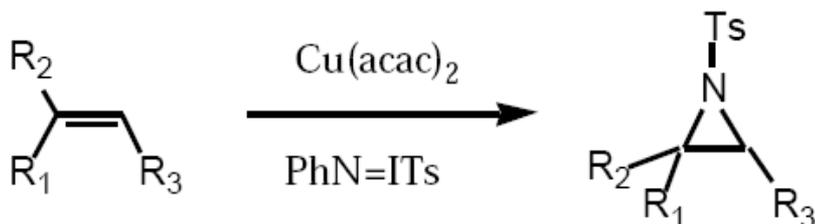


- S_N2-реакции



(не напоминает Кори-Чайковского?)

-нитрены:



J. Org. Chem. .1991, 56, 6744



1. Циклобутаны и циклобутены:

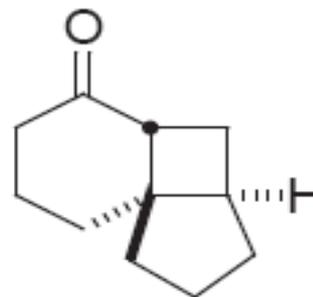
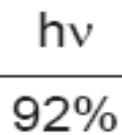
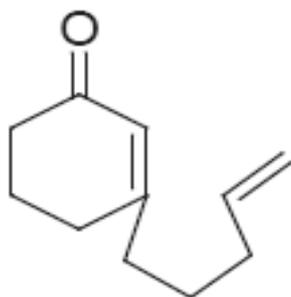
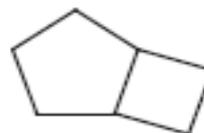
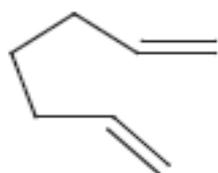
- [2+2]-циклоприсоединение кетены + алкены
- S_N2 -реакции
- ацилоиновая конденсация

- илиды серы

2. Оксетаны:

- [2+2]-циклоприсоединение (реакция Патерно-Бухи)
- S_N2 -реакции
- илиды серы
- β -лактоны

• [2+2]-циклоприсоединение



Cargill *Tetrahedron Lett.* 1978, 4465.

S Z Vatsadze's lectures

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S Z Vatsadze

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S Z Vatsadze's lectures



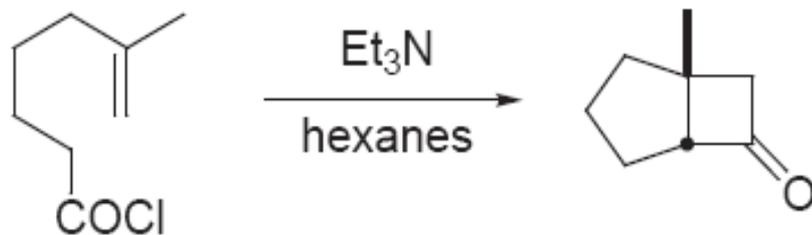
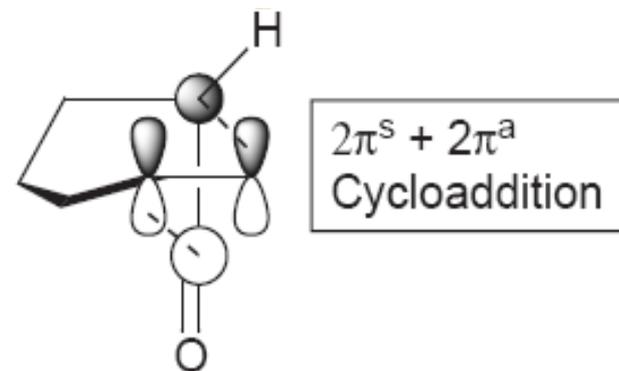
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S Z Vaisadze's

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Org. React. 1995, 45, 159.



Baldwin J. Chem. Soc., Chem. Commun. 1972, 1337.

• Фишеровские карбены



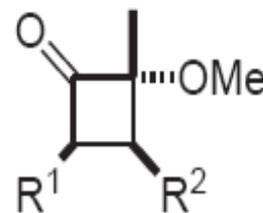
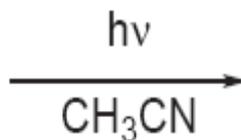
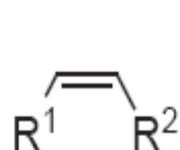
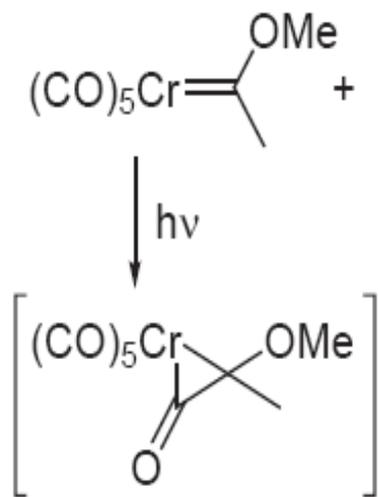
S Z Vatsadze's lectures

S Z Vatsadze's lectures

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S Z Vatsadze's lectures

[2+1+1]



$\text{R}^1 = \text{H}, \text{R}^2 = \text{OEt}, 85\%$
 $\text{R}^1 = \text{R}^2 = \text{Me}, 61\%$
 $\text{R}^1 = \text{H}, \text{R}^2 = \text{Ph}, 30\%$

Hegedus *J. Am. Chem. Soc.* **1989**, *111*, 2335.

es

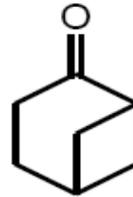
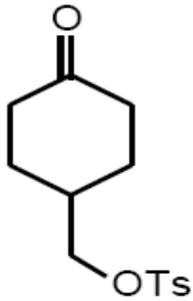
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SZ Vaisadze's lectures



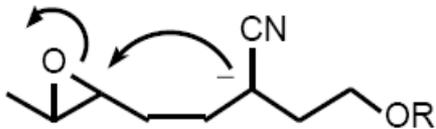
SZ Vaisadze's

SZ Vaisadze's

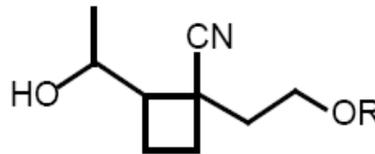
JACS 1980,
102, 1404

SZ Vaisadze's lectures

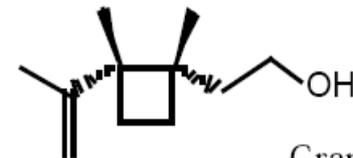
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JACS 1974, 96,
5268, 5272

SZV

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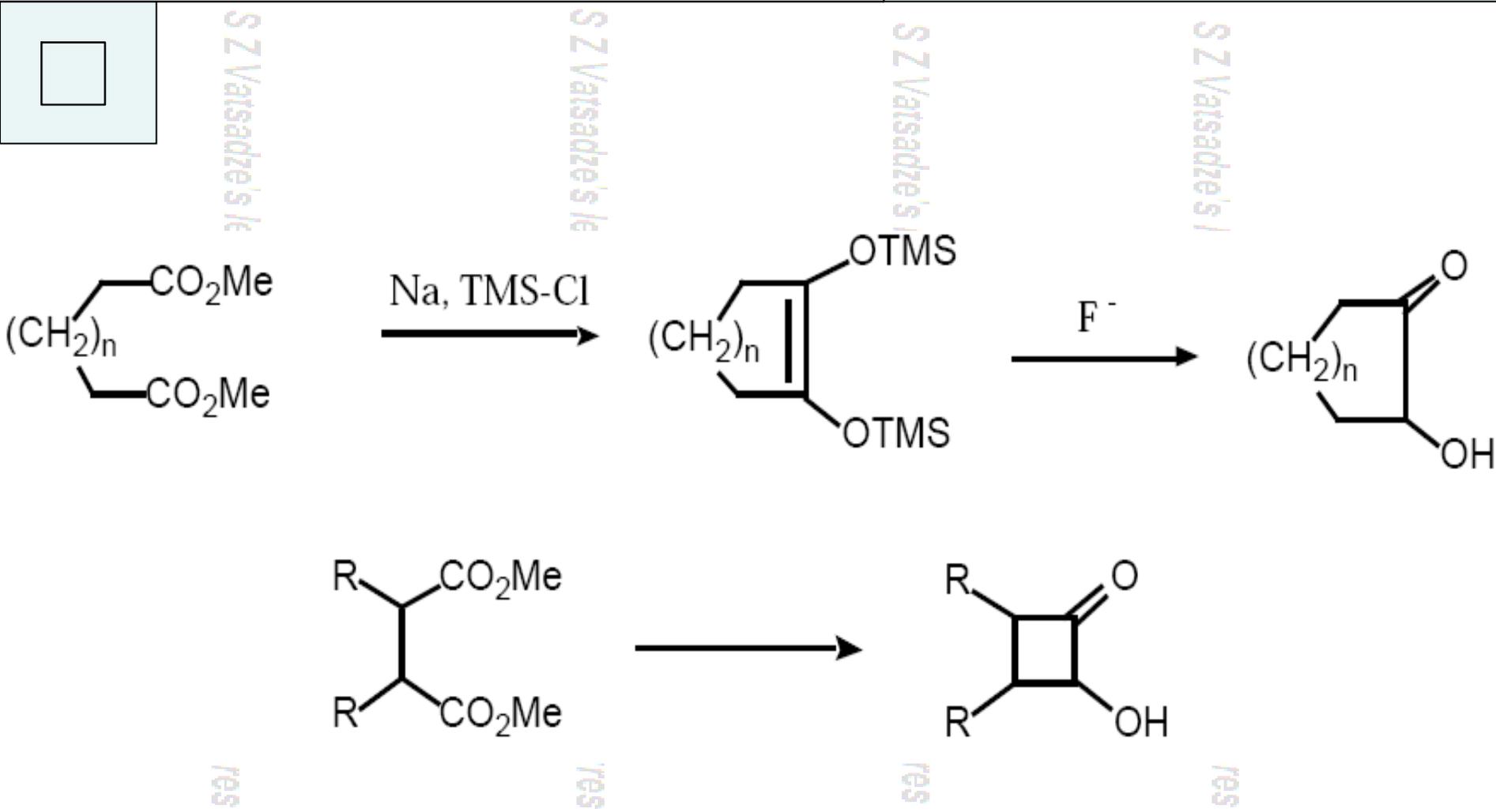
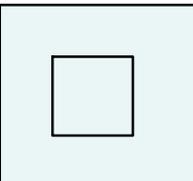
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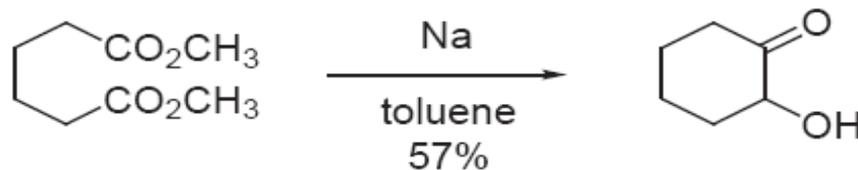
• Ацилоиновая конденсация

Образование циклов



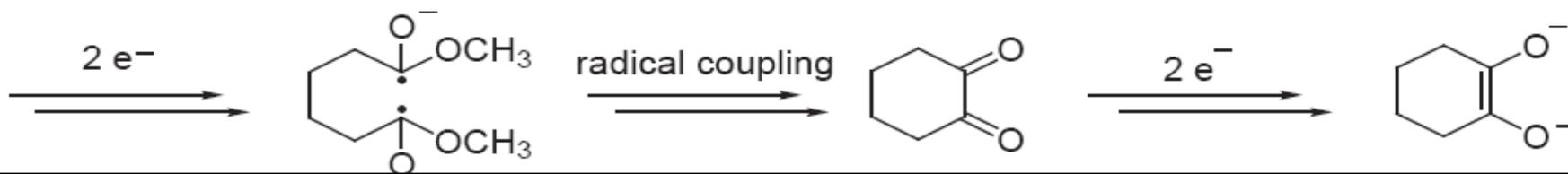
• Восстановительная димеризация карбонил

a. Acyloin Condensation

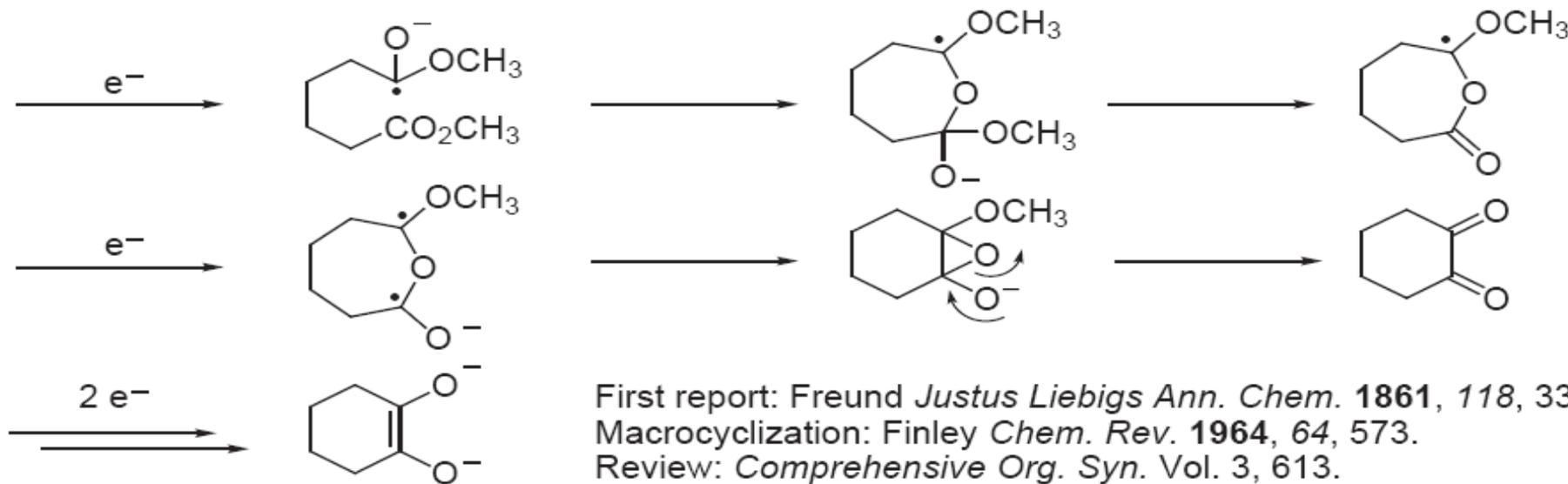


Sheehan *J. Am. Chem. Soc.* **1950**, *72*, 3376.

- Mechanism



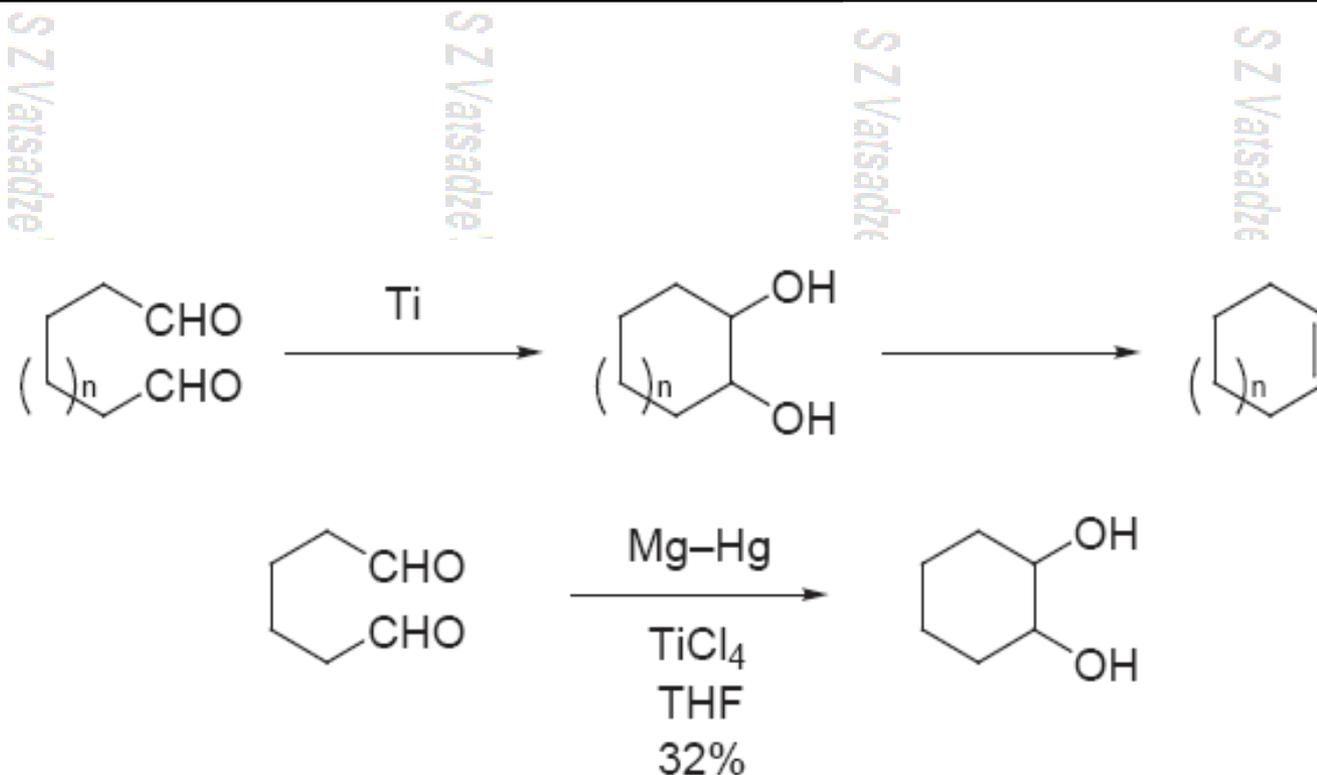
- Alternative



First report: Freund *Justus Liebigs Ann. Chem.* **1861**, *118*, 33.
 Macrocyclization: Finley *Chem. Rev.* **1964**, *64*, 573.
 Review: *Comprehensive Org. Syn.* Vol. 3, 613.

• Восстановительная димеризация карбонил

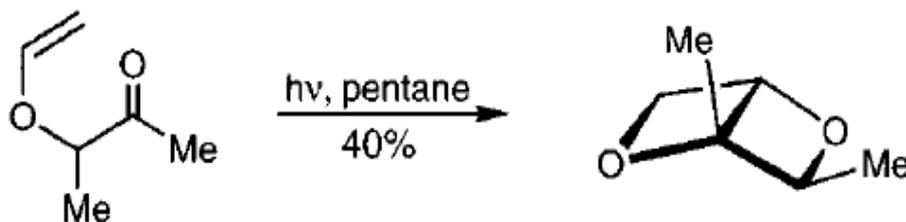
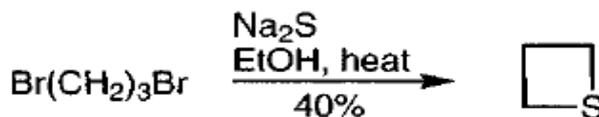
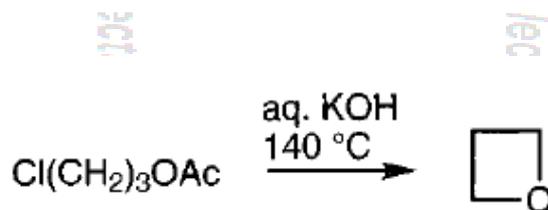
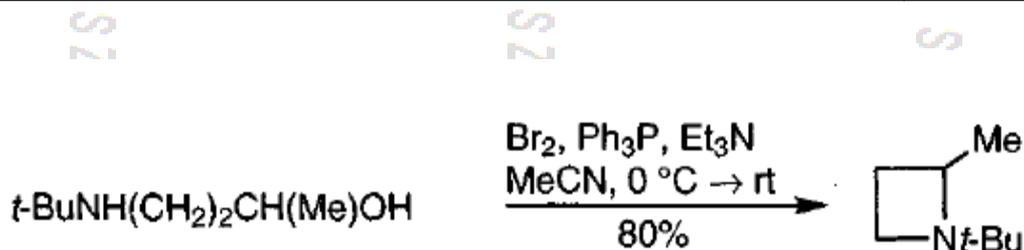
Образование циклов



Corey, Danheiser *J. Org. Chem.* 1976, 41, 260.

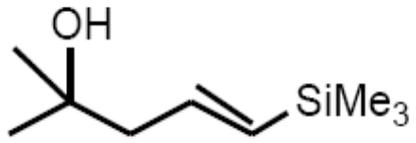
Вспомните реакцию МакМурри!

• Синтез 4-членных гетероциклов

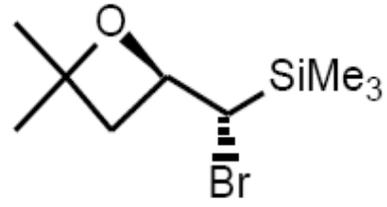




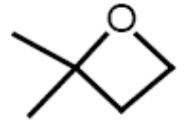
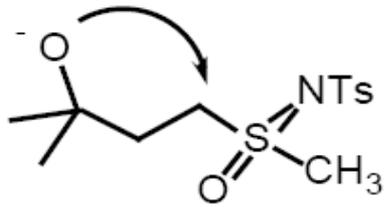
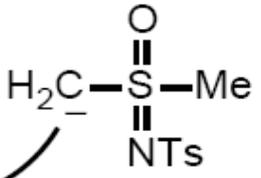
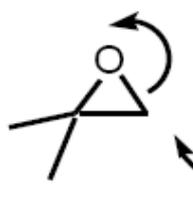
-S_N2-реакции



NBS



-илиды серы (вспомните Кори-Чайковский!)





SZVar

SZVar

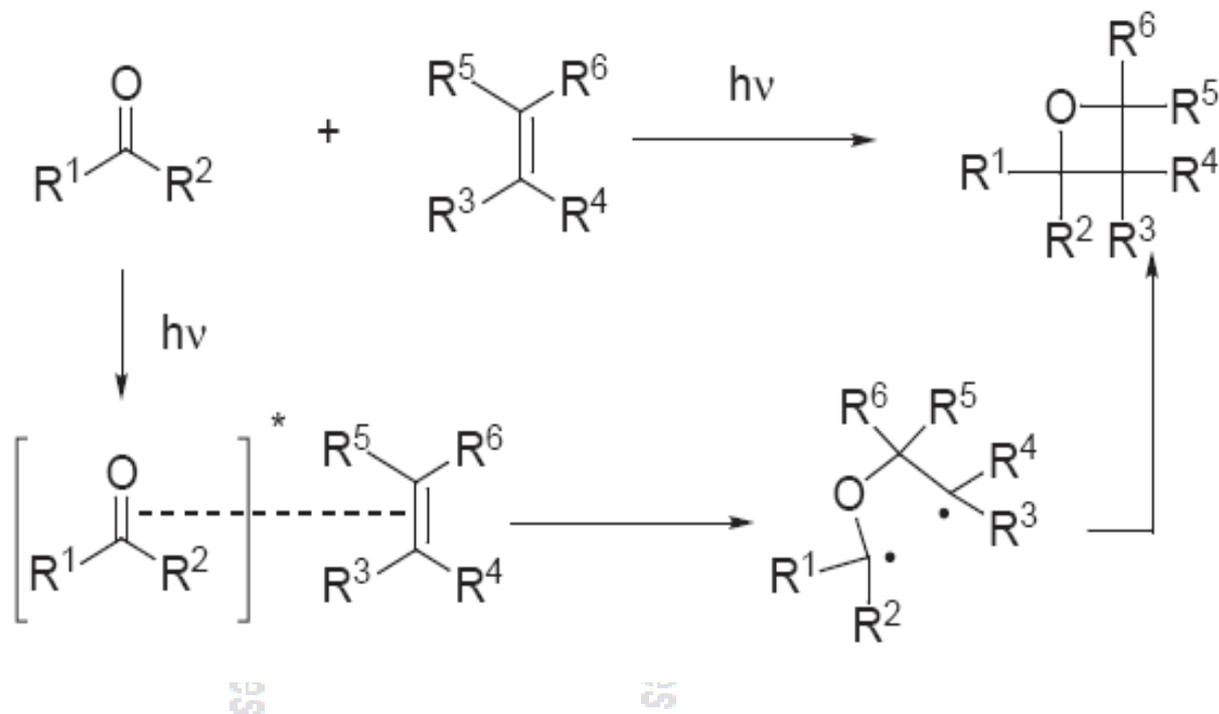
SZVa

SZVa

Comprehensive Org. Syn., Vol. 5, 151.

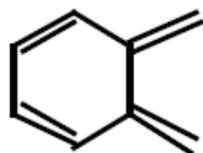
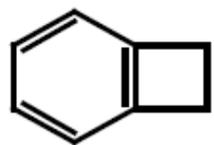
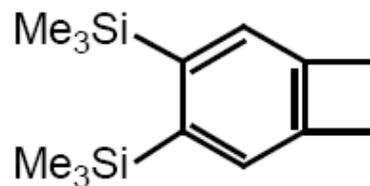
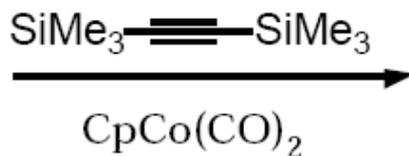
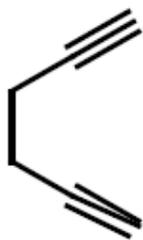
Dermuth Synthesis 1989, 152.

First studied in detail by Buchi *J. Am. Chem. Soc.* 1954, 76, 4327.



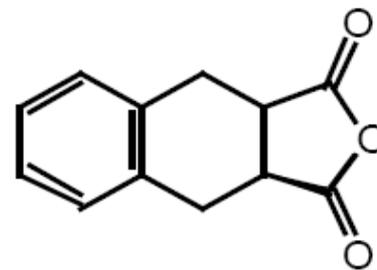
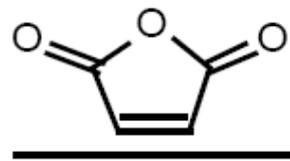


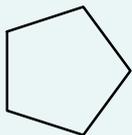
1. Циклобутаны и циклобутены:
- разное



benzocyclobutane

o-quinodimethane

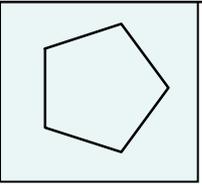




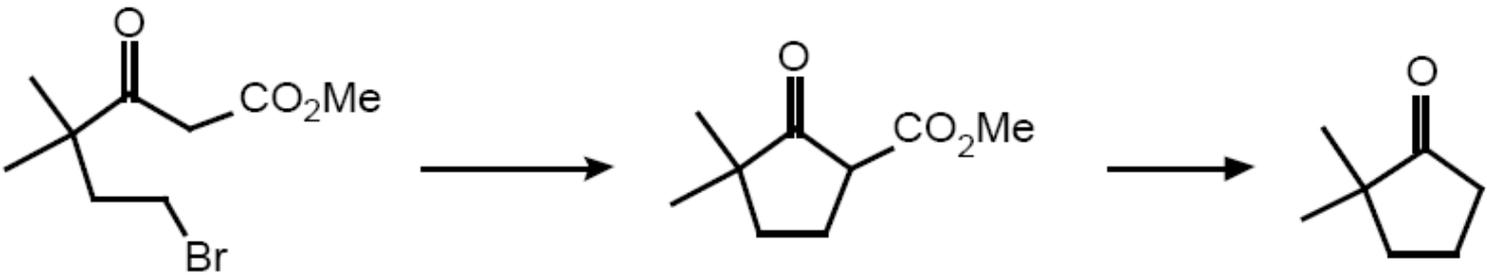
- **S_N2 -реакции**
- **ацилоиновая конденсация**
- **присоединение по Михаэлю**
- **альдольная конденсация**
- **внутримолекулярное олефинирование по Виттигу**
- **метатезис с замыканием цикла**
- **диазосоединения**
- **радикальные циклизации**
- **реакция Посона-Хэнда**
- 1,3-диполярное присоединение ([3+2]-циклоприсоединение)
- реакции расширения и сужения циклов:
 - a. 3 @ 5
 - b. 4 @ 5
 - c. 6 @ 5
- циклизация по Назарову
- циклизация по Нойори
- фотоциклизации арен+алкен

• S_N2 реакции

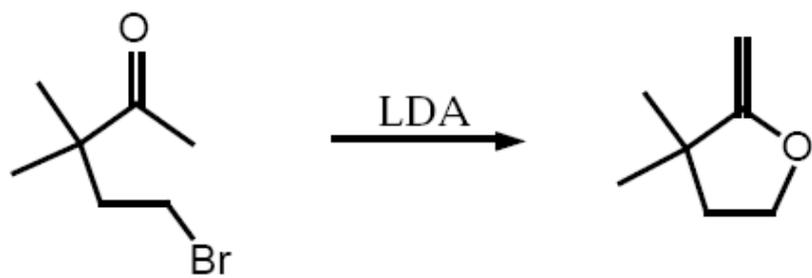
Образование циклов



5-ЭКЗО-ТЕТ

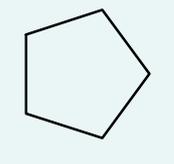


Но:



Почему:

Внутримолекулярное алкилирование енолятов

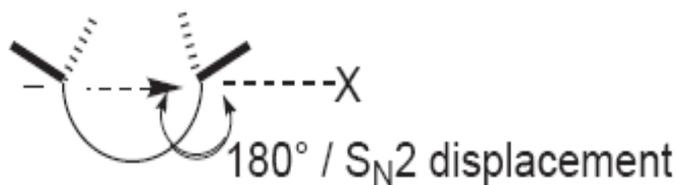


SZ Vaisa

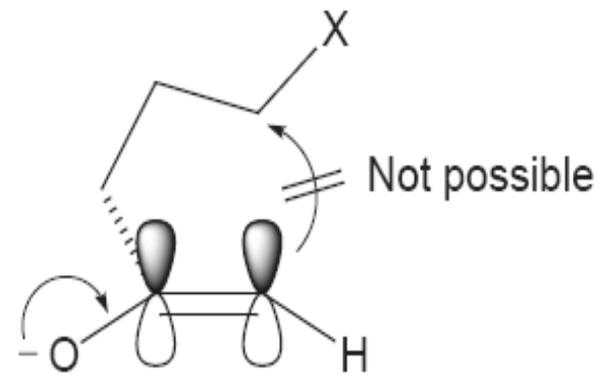
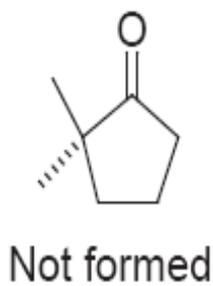
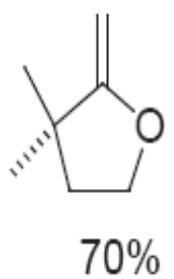
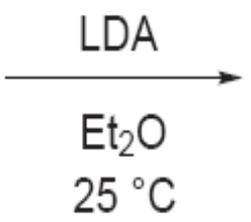
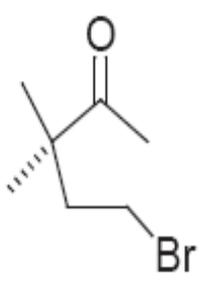
SZ Vaisa

SZ Vaisa

SZ Vaisa



- Note Baldwin's Rules
 Preceded by Eschenmoser
Helv. Chim. Acta 1970, 53, 2059.



tu/res

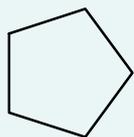
tu/res

tu/res

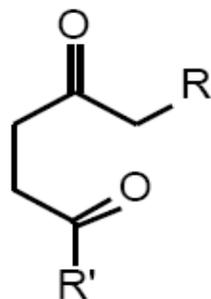
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• Альдольно-кратоновая конденсация

Образование циклов



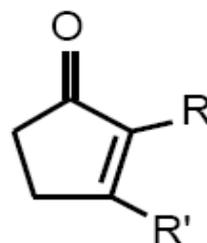
S Z Vaisadze's lectures



S Z Vaisadze's lectures

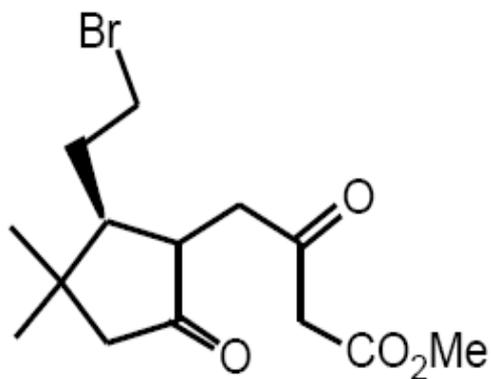


S Z Vaisadze's lectures

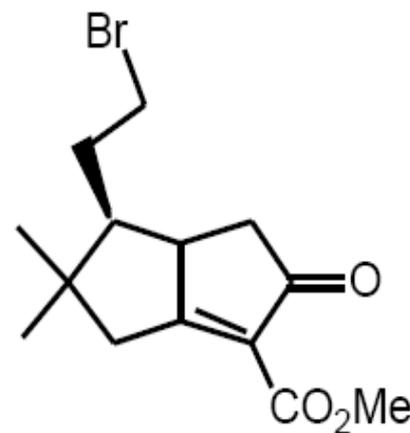


S Z Vaisadze's lectures

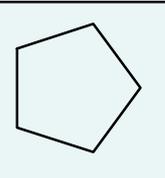
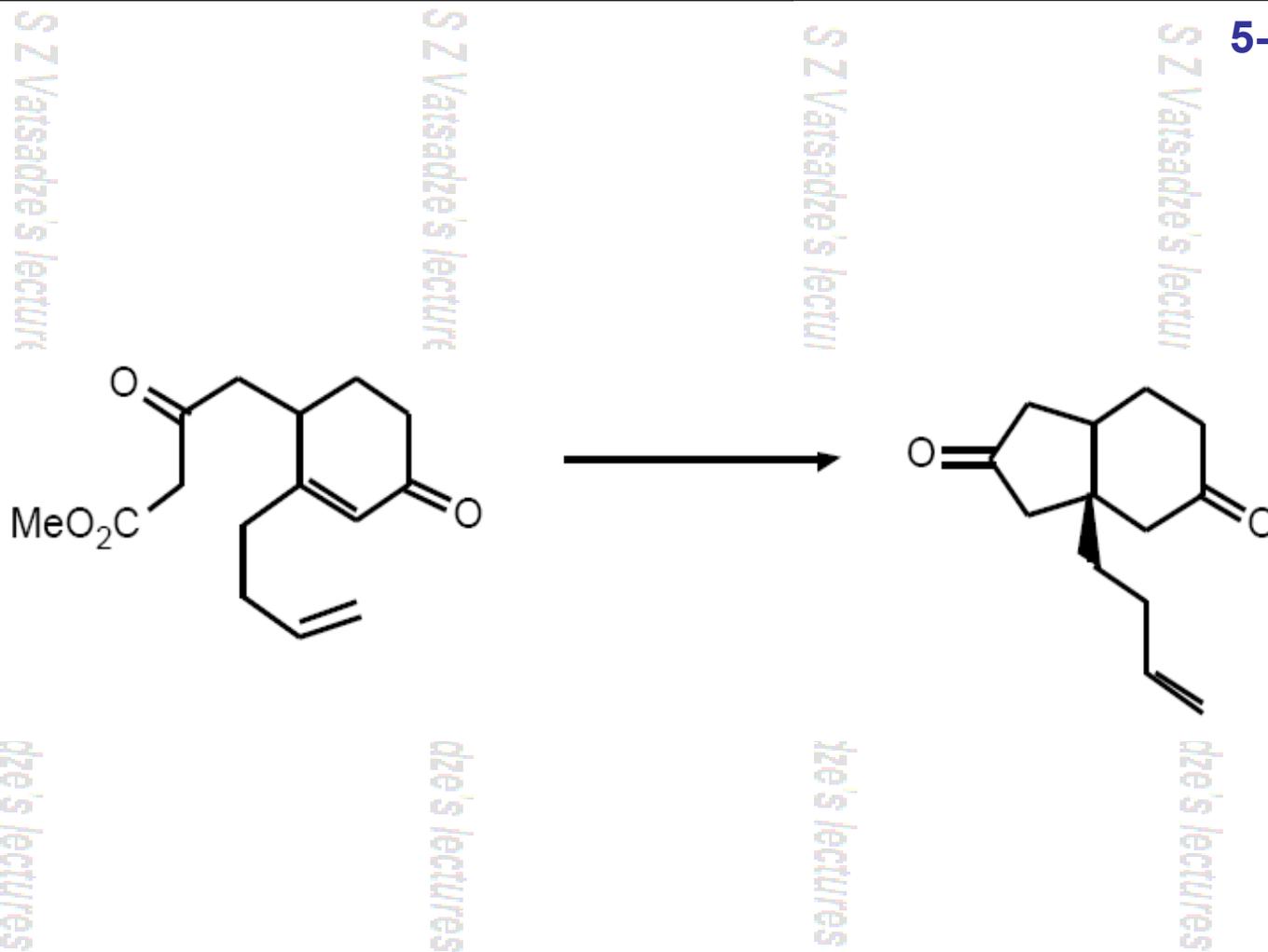
5-ЭКЗО-ТРИГ



NaOMe, MeOH



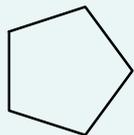
5-ЭКЗО-ТРИГ



- Присоединение по Михаэлю

• Внутримолекулярное олефинирование

Образование циклов

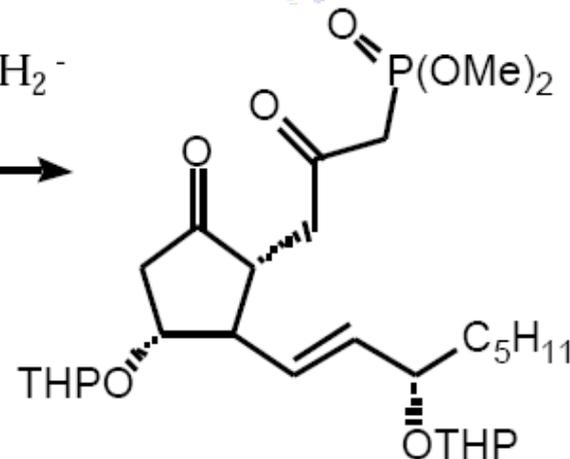
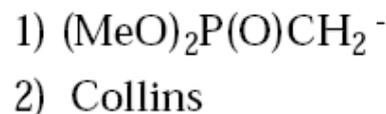
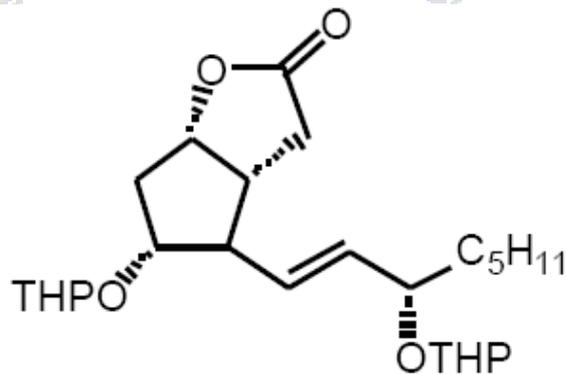


SZ Vats

SZ Vats

SZ Var

5-ЭКЗО-ТРИГ

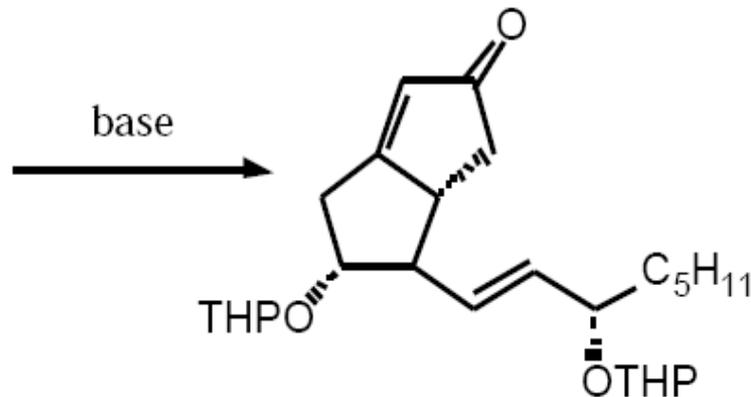


SZ Vatsadze's lectures

SZ I

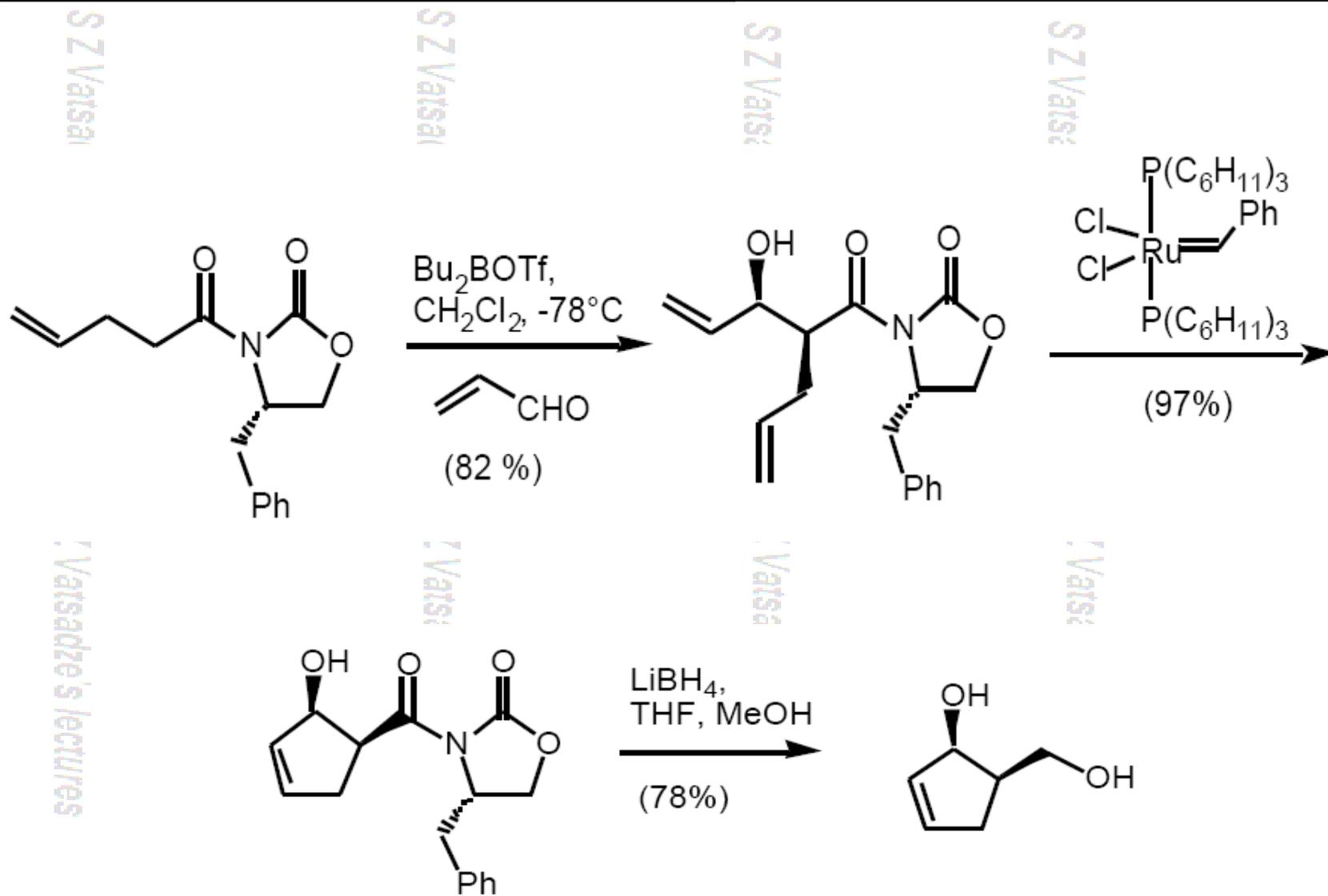
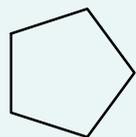
SZ V

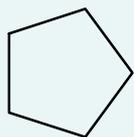
SZ Vatsadze's lectures



• Метатезис с замыканием цикла

Образование циклов



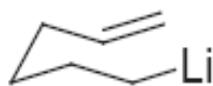


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S Z Vatsadze's lectu

S Z Vatsadze's lectu

S Z Vatsadze's lectu



stable at $-78\text{ }^{\circ}\text{C}$
 $t_{1/2} = 5.5\text{ min}$ at $25\text{ }^{\circ}\text{C}$

Bailey *J. Am. Chem. Soc.* **1992**, *114*, 8053.
J. Am. Chem. Soc. **1991**, *113*, 5720.
J. Am. Chem. Soc. **1987**, *109*, 2442.

Intramolecular carbometalation, review:
Comprehensive Org. Syn., Vol. 4, 871.

ie's lectures

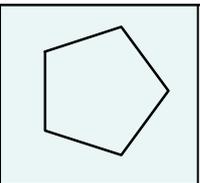
ie's lectures

ie's lectures

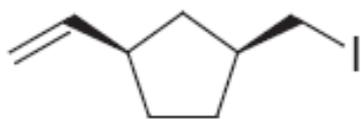
ie's lectures

• Анионные циклизации: примеры

Образование циклов



1ZV

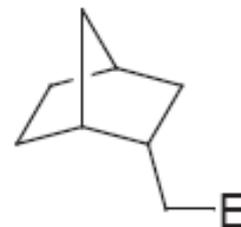


1ZS

1. ^tBuLi

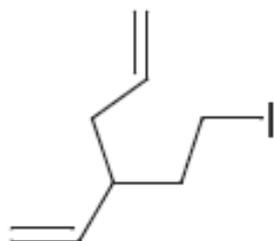
2. E⁺

1Z1



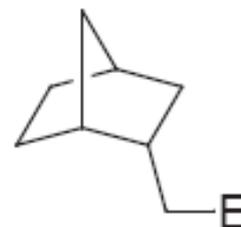
63–91%

1Z1

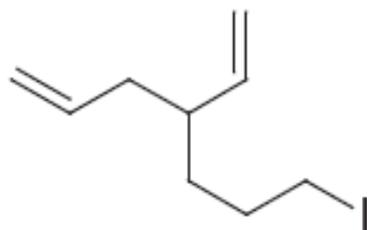


tandem

cyclizations

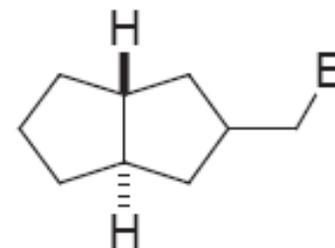


65–90%



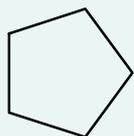
tandem

cyclizations



65–87%

• Анионные циклизации: примеры



S Z Va

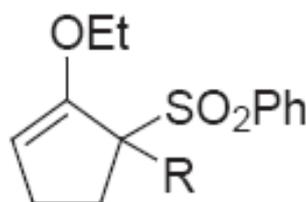
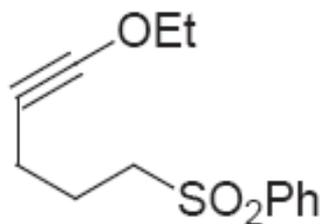
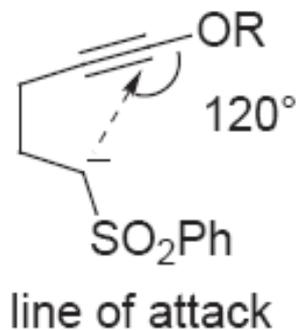
S Z Va

S Z Va

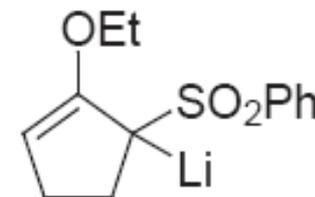
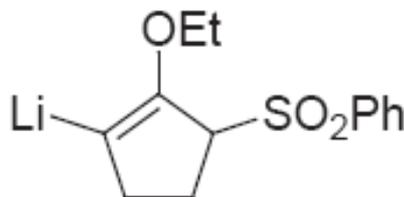
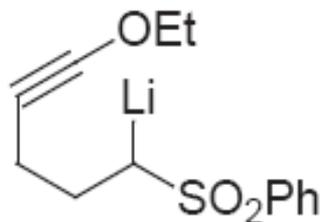
S Z Va

Funk J. Am. Chem. Soc. 1993, 115, 7023.

5-endo-dig cyclization



RX



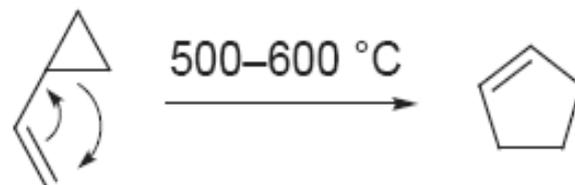
• 1,3-Сигматропные перегруппировки:
винилциклопропан

Образование циклов



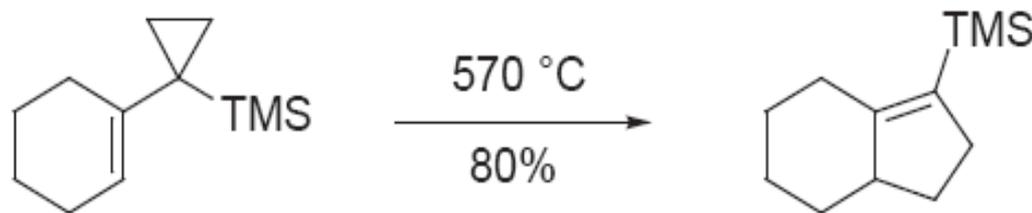
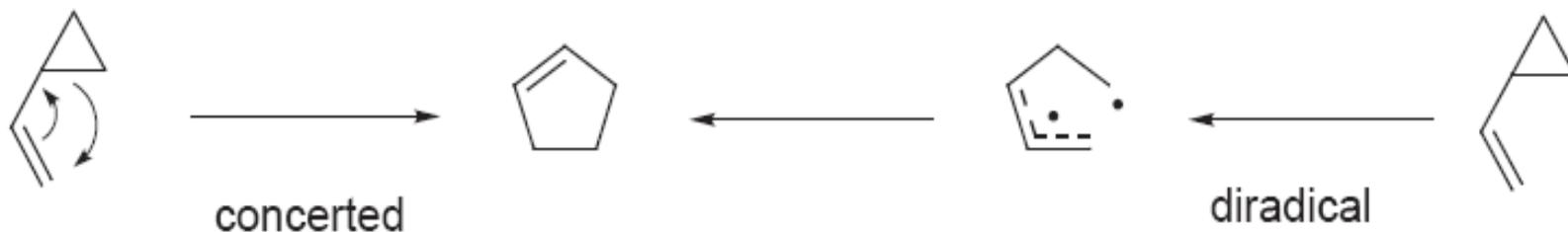
First report: Neureiter *J. Org. Chem.* **1959**, 24, 2044.

Review: Hudlicky *Chem. Rev.* **1989**, 89, 165.

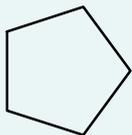


Org. React. **1985** 33, 247.

Mechanism:



Paquette *Tetrahedron Lett.* **1982**, 23, 263.



Что еще по циклопентанам:

• **радикальные циклизации**

-реакции расширения и сужения циклов:

a. 3 ® 5

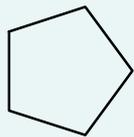
b. 4 ® 5

c. 6 ® 5

-ацетиловая конденсация

-1,3-диполярное присоединение

-фотоциклизации арен+алкен



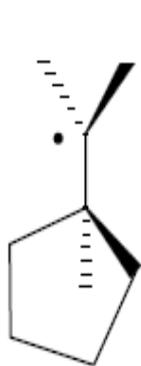
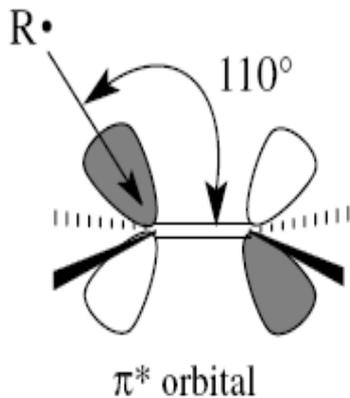
SZ Vats

Вспоминаем правила Болдуина!!!

SZ Vats

SZ Vats

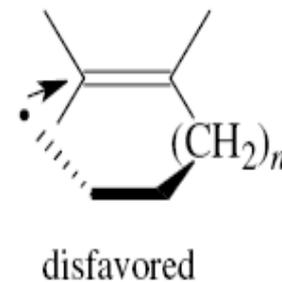
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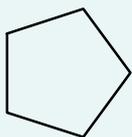
and



but



- Циклизации типа «радикал-олефин»



Радикалы бывают **электрофильные** и **нуклеофильные**.

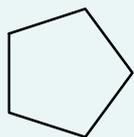
Алкильные радикалы относятся к **нуклеофильным** радикалам:

- Relative rates of addition to $\text{CH}_2=\text{CHPO}(\text{OEt})_2$: typical electron-deficient olefin.

$k_{\text{rel}} =$	$\text{CH}_3\cdot$	$\text{CH}_3\text{CH}_2\cdot$	$\text{CH}_3\text{OCH}_2\cdot$	$(\text{CH}_3)_2\text{CH}\cdot$	$(\text{CH}_3)_3\text{C}\cdot$
	1	1	2.7	4.8	24

• Циклизации типа «радикал-олефин»

Образование циклов

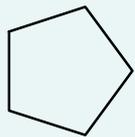


Стереoeлектронные требования

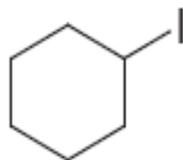
olefin	% addition to:		k_{rel}
	Ca	Cb	
	>95	<5	1.16
	>95	<5	18.4
	>95	<5	2×136
	50	50	2×0.50
	50	50	2×0.63
	>95	5	15
	<5	>95	13.9

- Циклизации типа «радикал-олефин»

Образование циклов

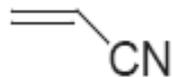


SZ Vais?



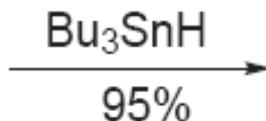
Nucleophilic radical

SZ Vais?

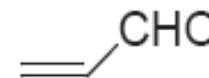
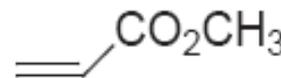
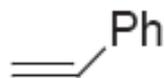
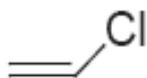
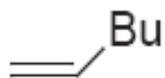
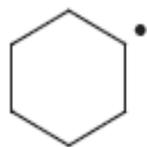
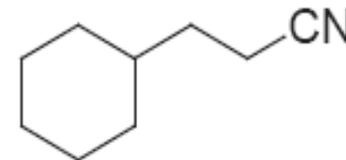


Electrophilic acceptor alkene

SZ Vais



SZ Vais



k_{rel}

1.0

8.4

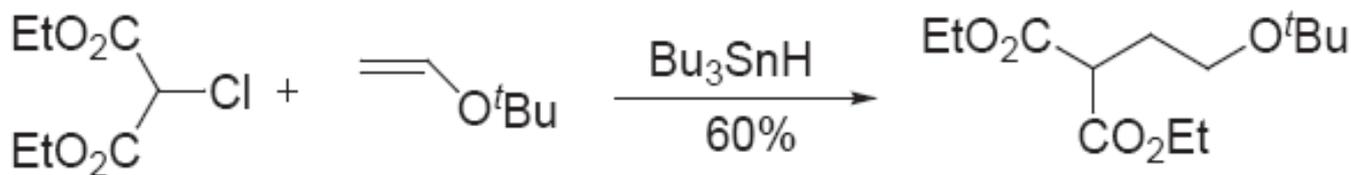
84

3000

8500

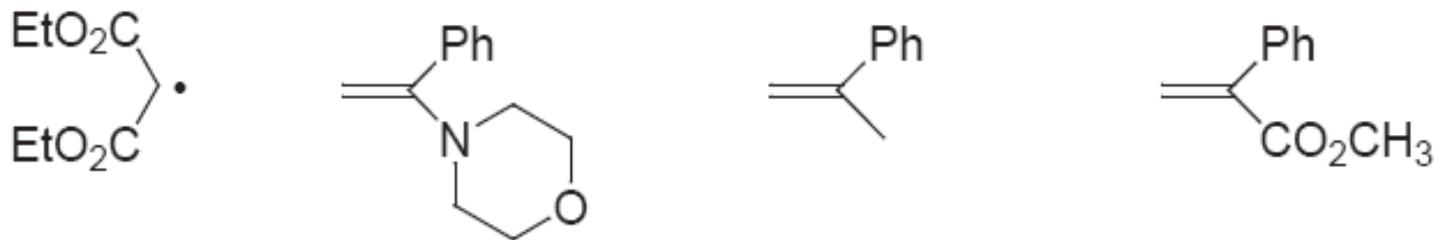
• Циклизации типа «радикал-олефин»

Электрофильные радикалы – стабилизированы электроноакцепторными группами:



Electrophilic radical

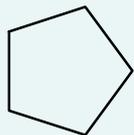
Nucleophilic acceptor alkene



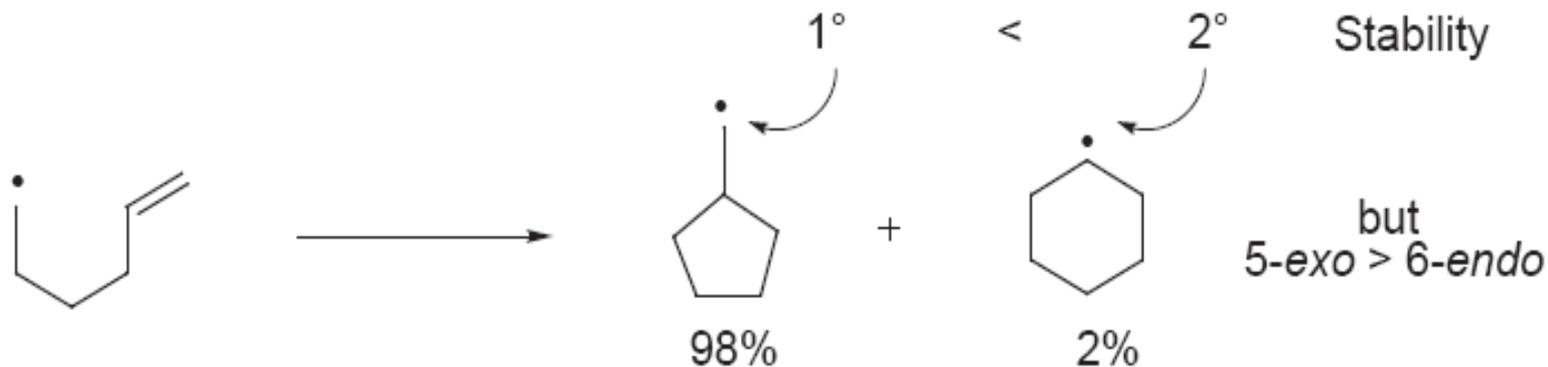
k_{rel}	23	3.5	1
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• Циклизации типа «радикал-олефин»

Образование циклов

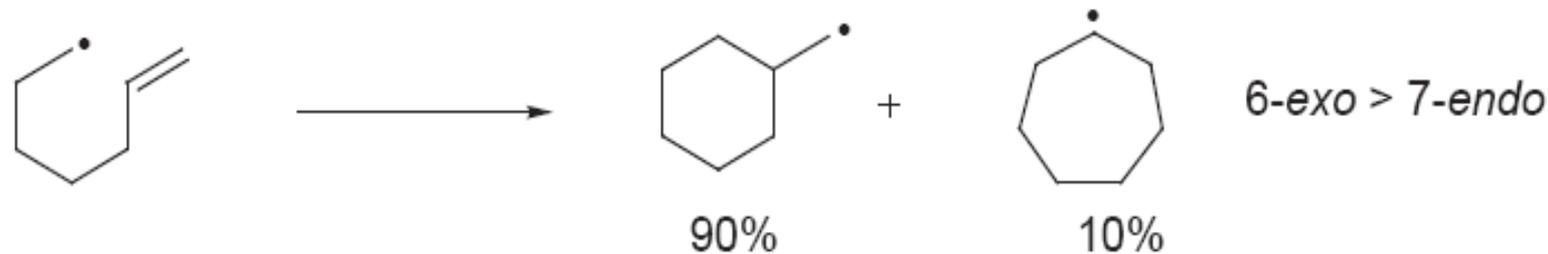


Скорости, регио- и диастереоселективность



Beckwith *J. Chem. Soc., Chem. Commun.* **1974**, 472.

Beckwith *J. Chem. Soc., Chem. Commun.* **1980**, 484.



- Циклизации типа «радикал-олефин»

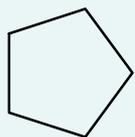
Образование циклов

Скорости, регио- и диастереоселективность

Radical	$k_{rel} \text{ } exo$	$k_{rel} \text{ } endo$	Product ratio
	1.0	0.02	(98 : 2)
	1.4	0.02	(99 : 1)
	2.4	<0.01	(>200:1)
	0.022	0.04	(36 : 64) <i>endo</i> predominates
	0.16	<0.002	<i>exo</i> >> <i>endo</i> (>80:1)

• Циклизации типа «радикал-олефин»

Образование циклов



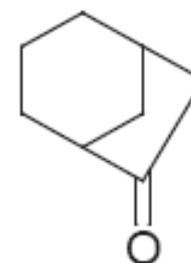
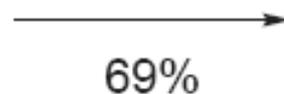
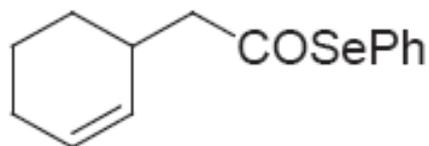
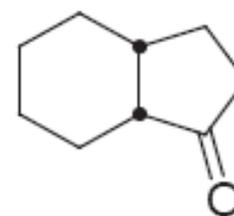
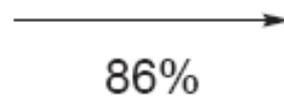
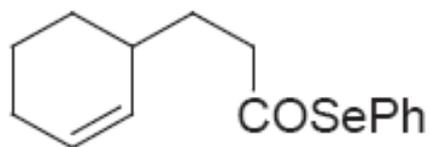
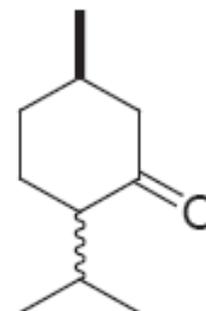
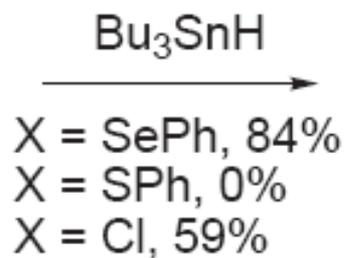
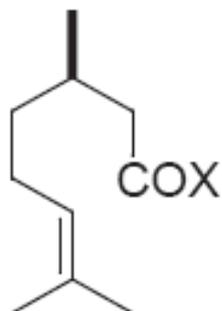
S Z Vatsa

Карбонилрадикалы

S Z Vatsa

S Z Vatsa

S Z Vatsa



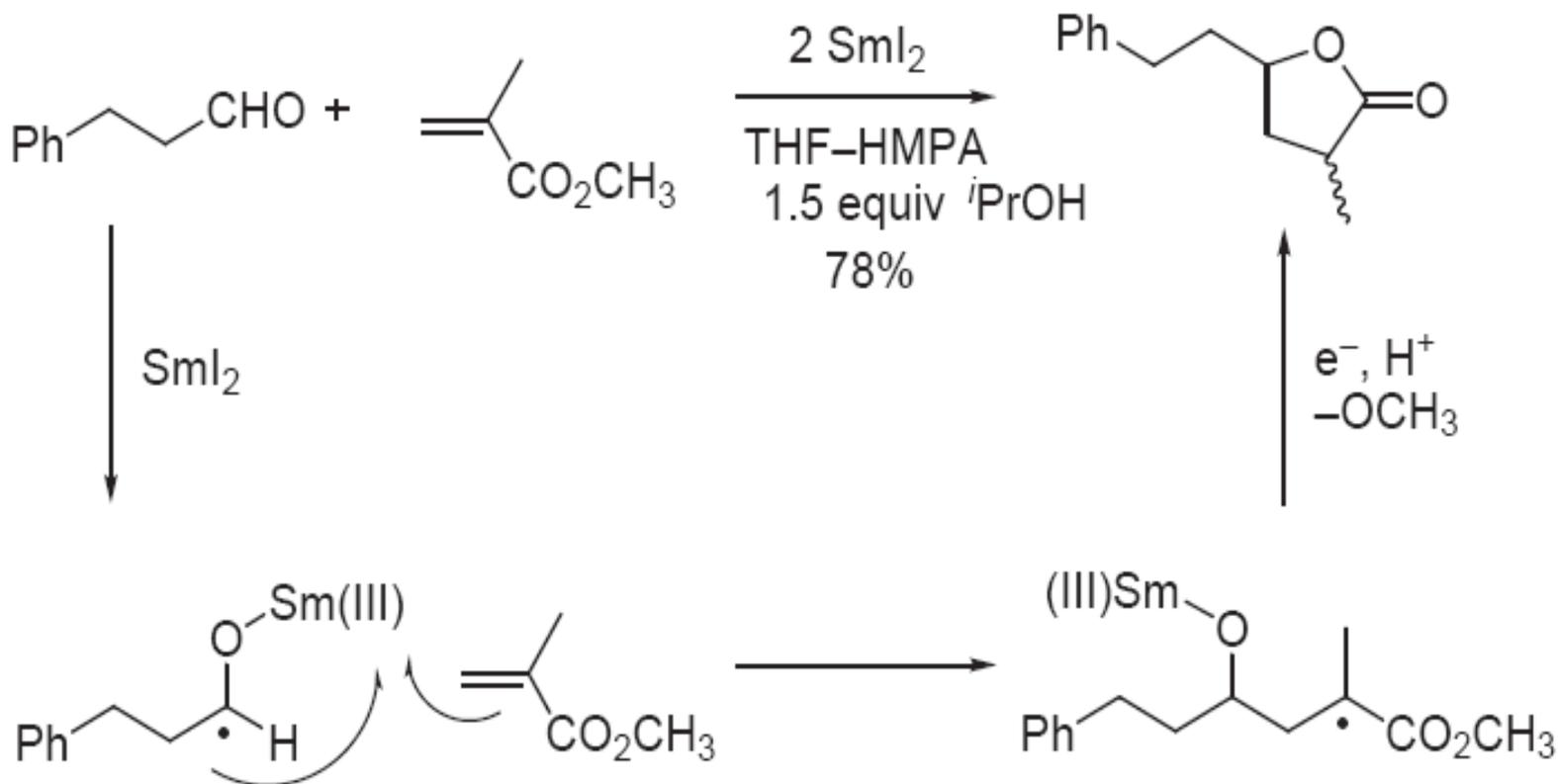
• Восстановительное сочетание с участием SmI_2



Molander *Chem. Rev.* **1992**, *92*, 29.

Molander in *Chemistry of the Carbon Metal Bond*, Hartley, F. R.; Patai, S., Eds.; Wiley: NY, 1989, Vol. 5

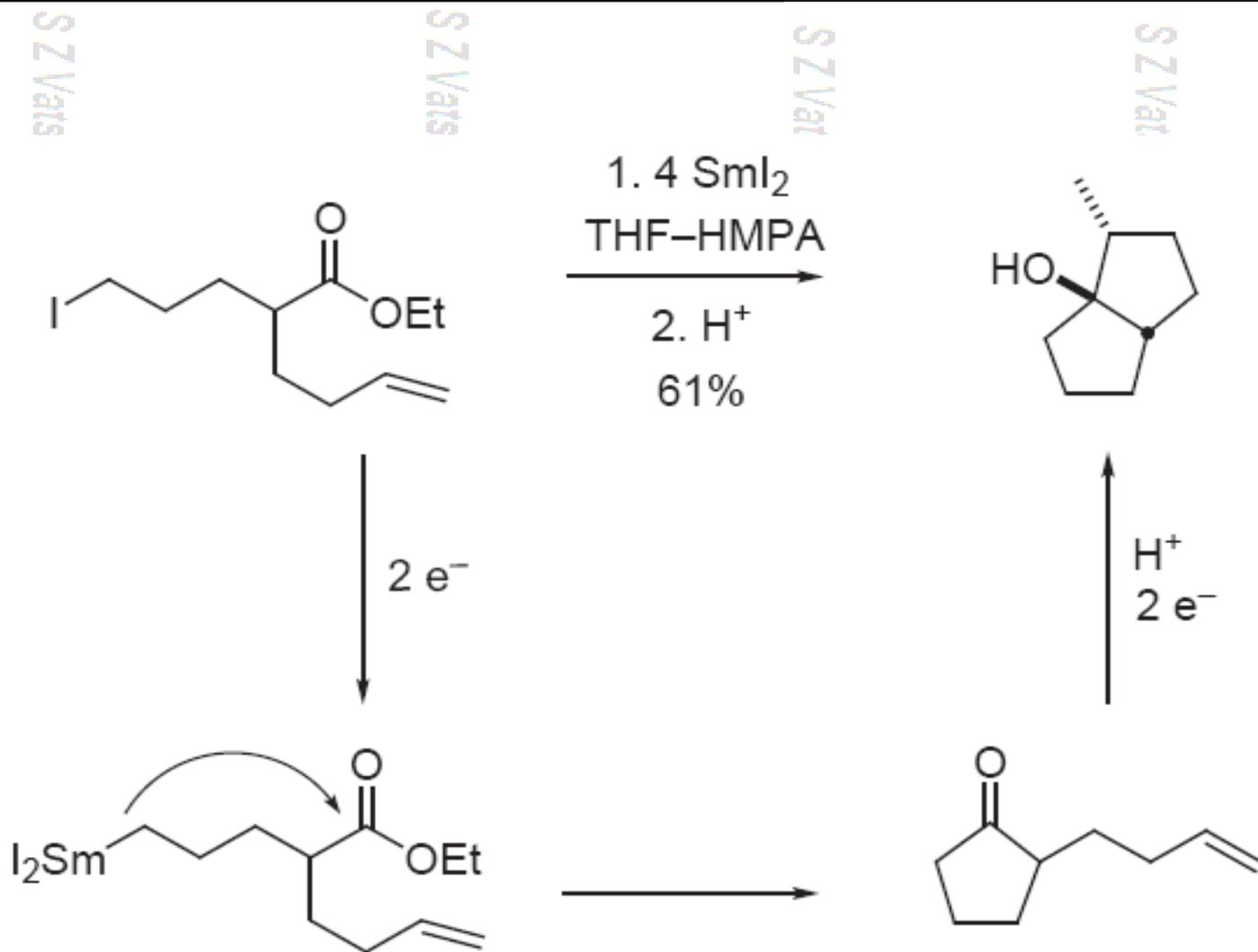
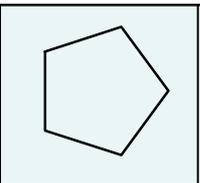
Molander in *Comprehensive Org. Syn.*, Vol. 1, 262.



Inanaga *Tetrahedron Lett.* **1986**, *27*, 5763.

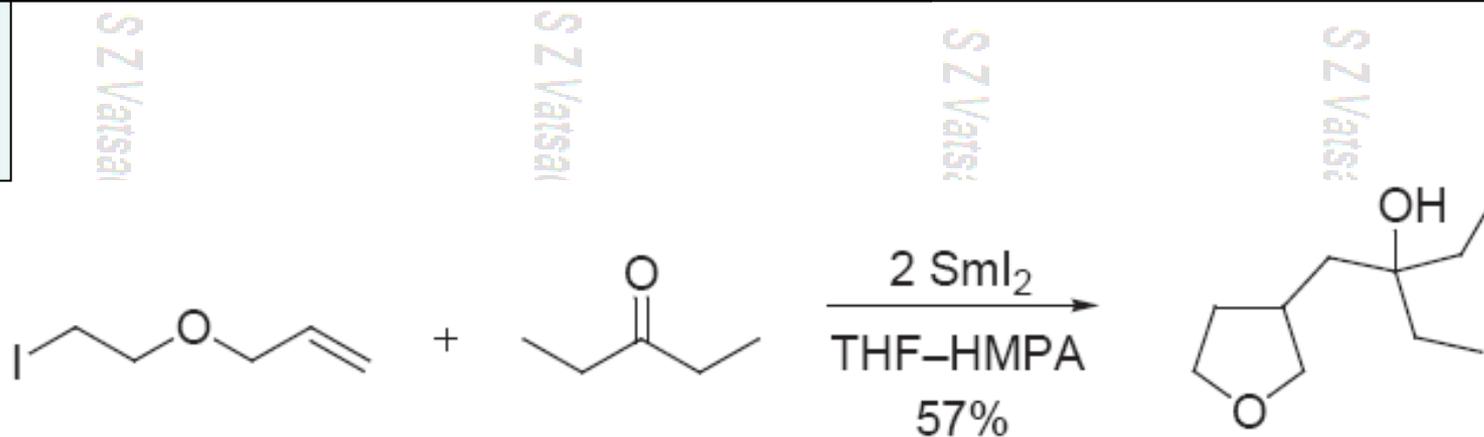
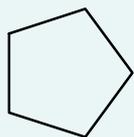
Tetrahedron Lett. **1989**, *30*, 2837.

- Восстановительное сочетание с участием SmI_2

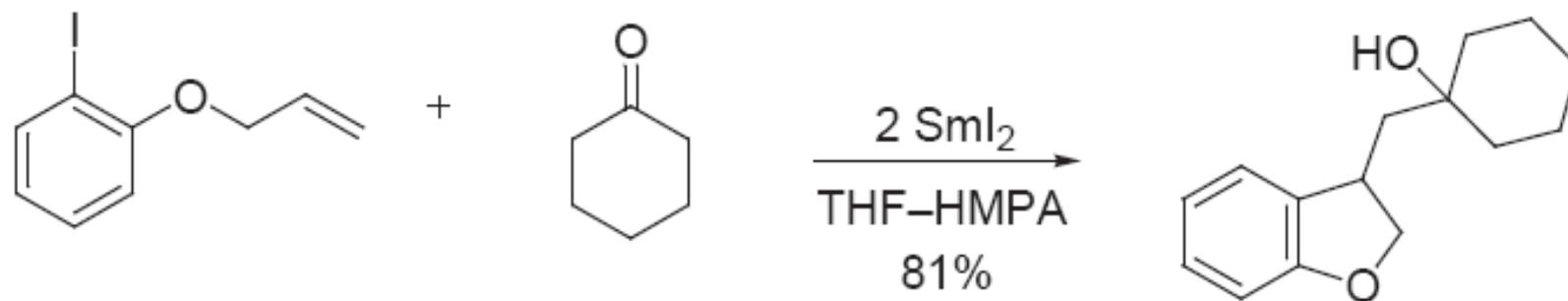


- Восстановительное сочетание с участием SmI_2

Образование циклов



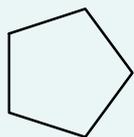
Molander *J. Org. Chem.* **1990**, *55*, 6171.



Curran *Synlett* **1990**, 773.

• Реакция Посона-Хэнда

Образование циклов



SZV

SZV

SZV

SZV

[2 + 2 + 1]

Comprehensive Org. Syn., Vol. 5, pp 1037–1064.

Org. React. **1991**, 40, 1.

Pauson *Tetrahedron* **1978**, 41, 5855.

Schore *Chem. Rev.* **1988**, 88, 1081.

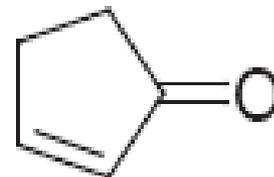
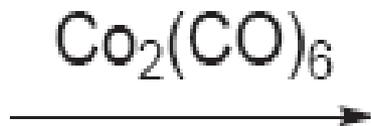
Brummond *Tetrahedron* **2000**, 56, 3263.

First detailed study: Khand *J. Chem. Soc., Perkin Trans. 1* **1973**, 977.

SZVatsadze's lectures



CO



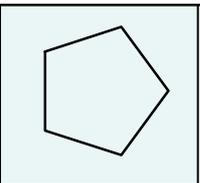
SZV

SZV

SZV

• Реакция Посона-Хэнда

Образование циклов



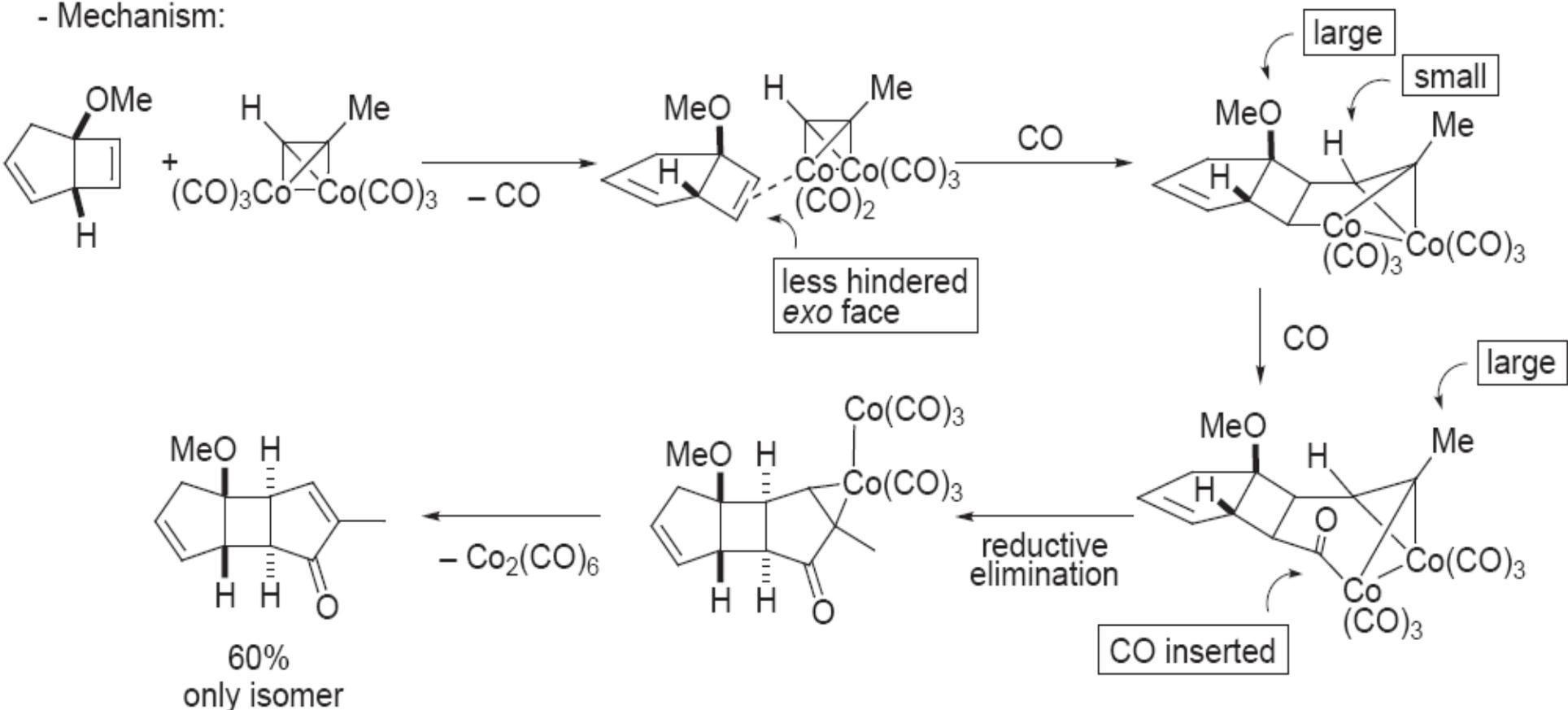
S Z Varsadz

S Z Varsadz

S Z Varsadz

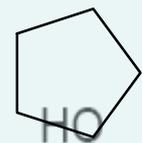
S Z Varsadz

- Mechanism:

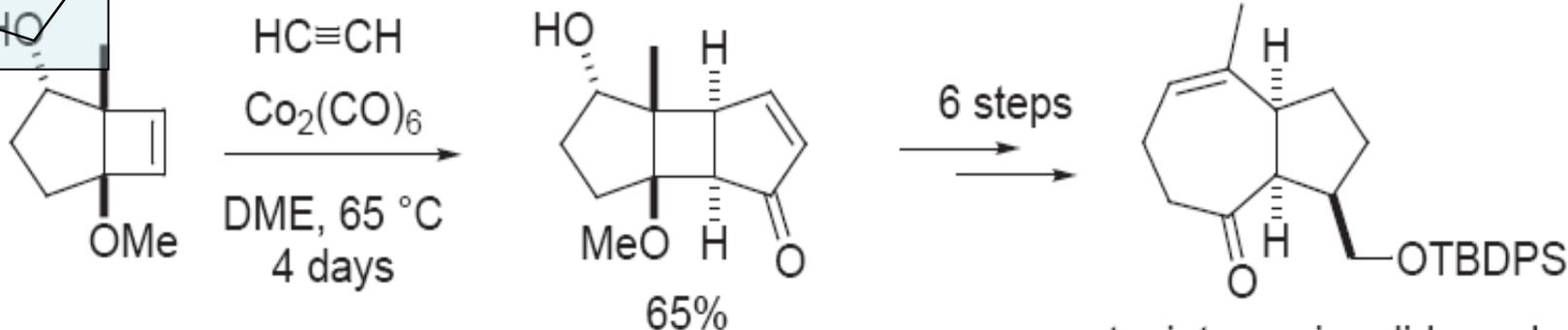


• Реакция Посона-Хэнда: примеры

Образование циклов



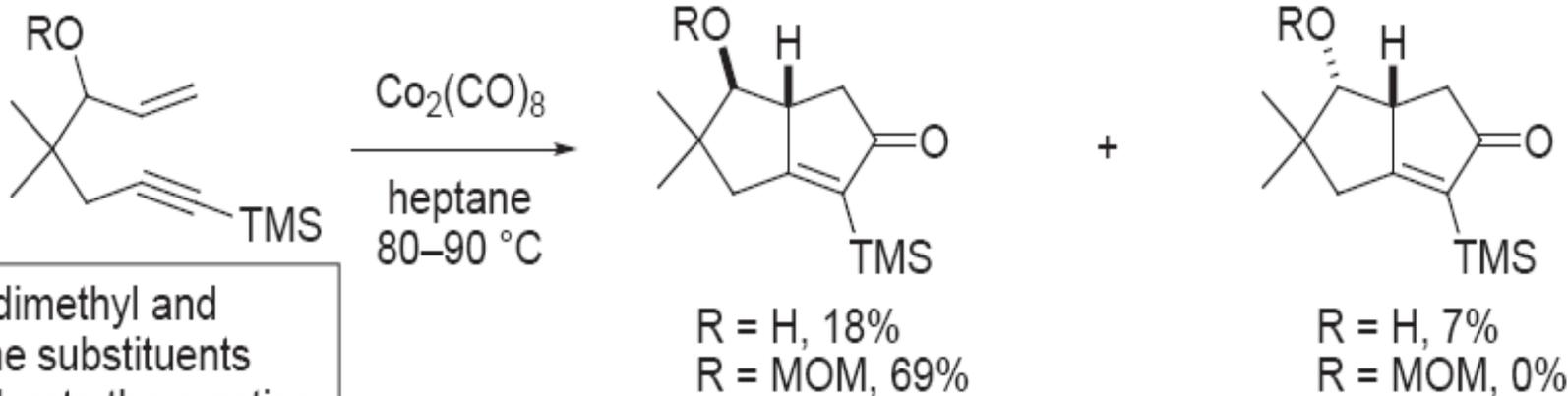
Межмолекулярная



Schore *J. Org. Chem.* **1987**, 52, 3595.

entry into guaianolide and pseudo-guaianolide natural products

Внутримолекулярная

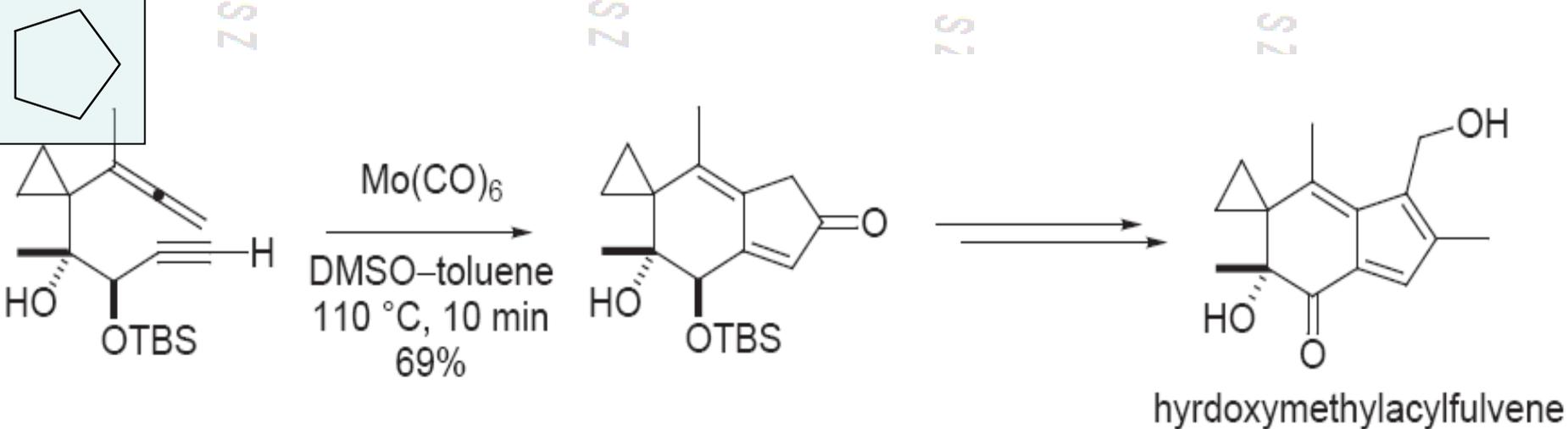


The dimethyl and alkyne substituents accelerate the reaction.

Magnus *J. Am. Chem. Soc.* **1983**, 105, 2477.

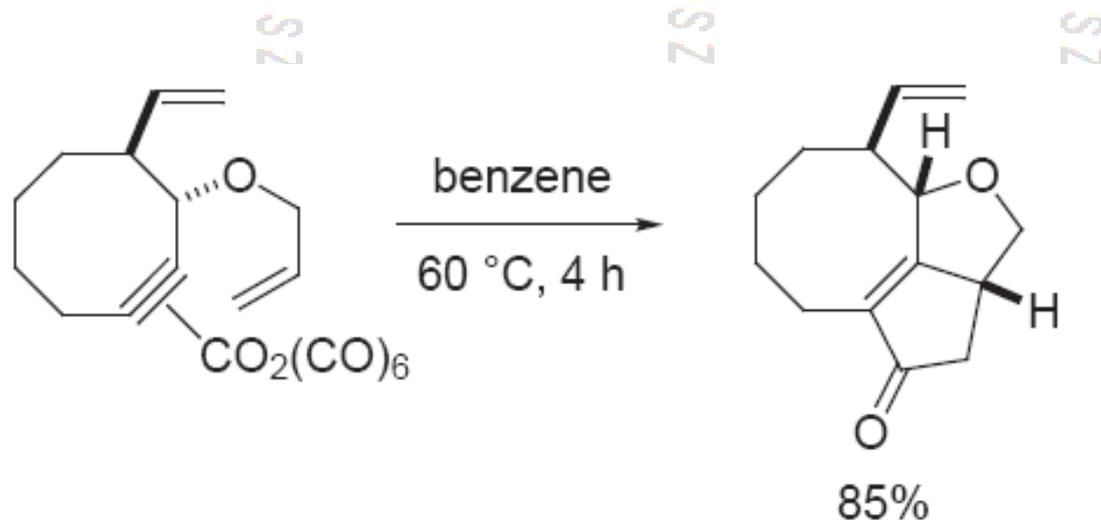
• Реакция Посона-Хэнда: примеры

Образование циклов

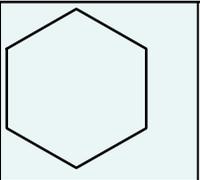


Brummond *J. Am. Chem. Soc.* 2000, 122, 4915.

SZ Varsadze's lectures



Schreiber *J. Am. Chem. Soc.* 1986, 108, 3128.

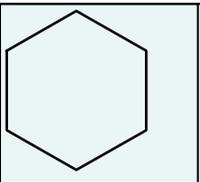


- сложноеэфирная и альдольная конденсация
- присоединение по Михаэлю
- аннелирование по Робинсону
- метатезис с замыканием цикла
- циклизации катион+алкен
- Фишеровские карбены (р-я Дётса)
- тримеризация ацетиленов
- циклизация по Бергману и родственные процессы

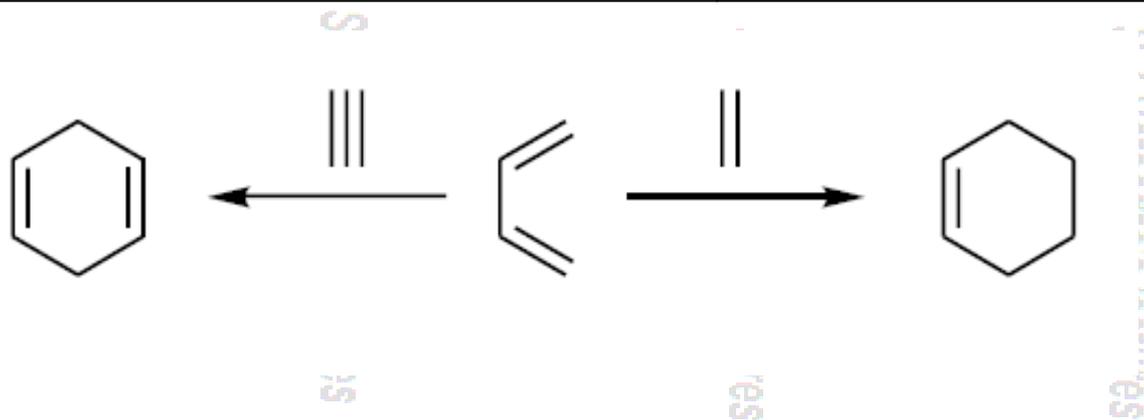
- внутримолекулярное олефинирование по Виттигу
- S_N2 -реакции
- ацилоиновая конденсация
- реакция Дильса-Альдера, *o*-хинодимеры
- расширение и сужение циклов
- ароматическое замещение с послед. гидрированием или восстановлением по Берчу
- радикальные циклизации
- сигматропные перегруппировки
- внутримолекулярная еновая реакция

• [4+2]-циклоприсоединение

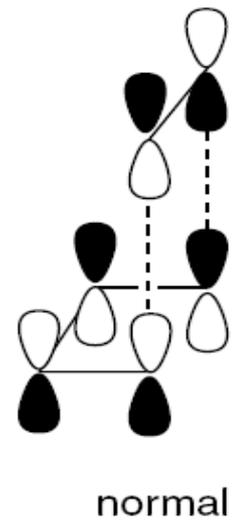
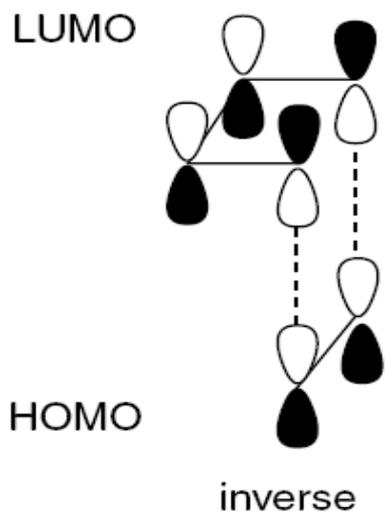
Образование циклов



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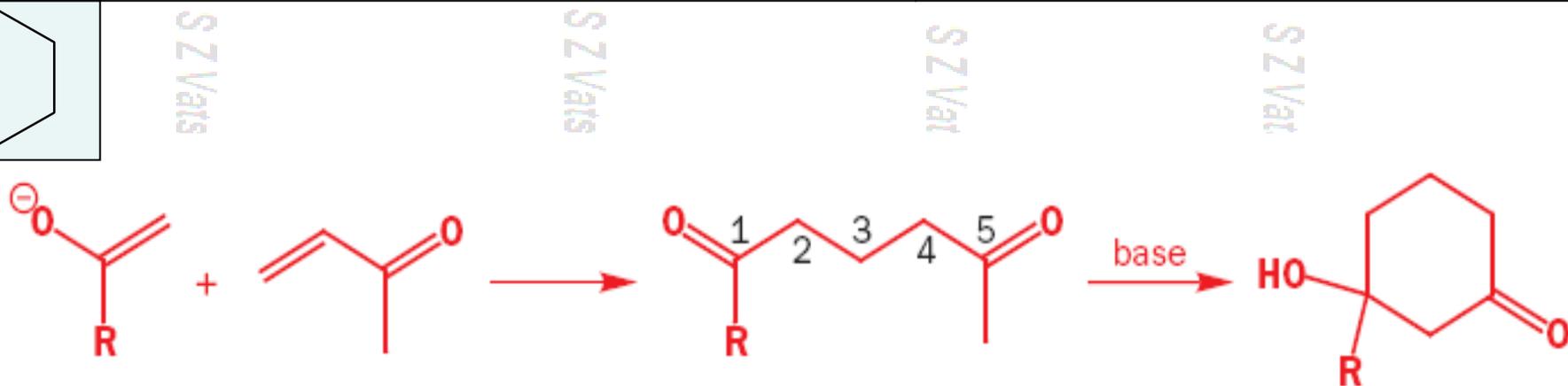
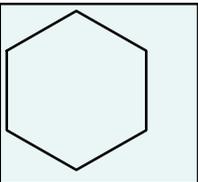


S Z Vatsadze's lectures



• Реакция Робинсона

Образование циклов

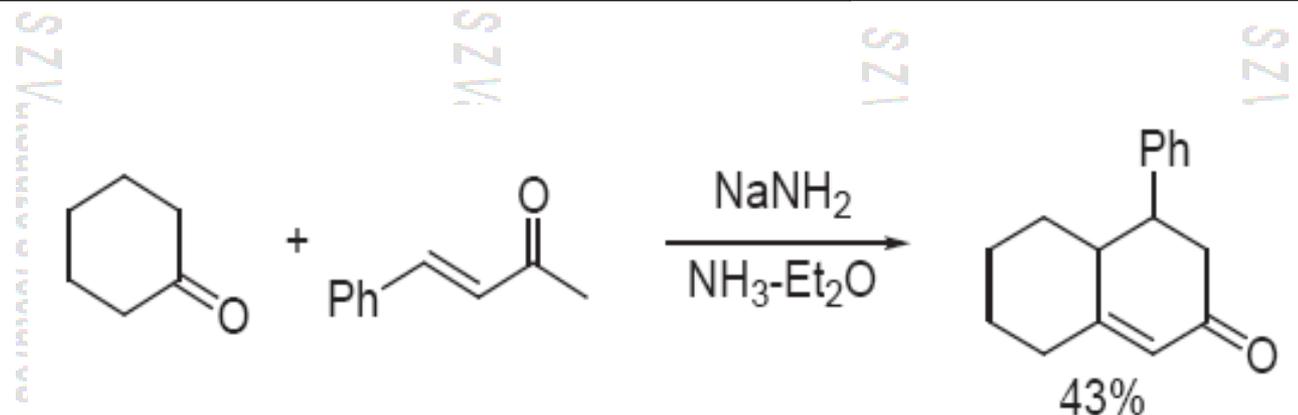
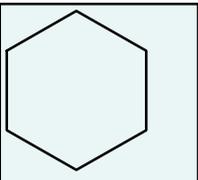


1947

SIR ROBERT ROBINSON for his investigations on plant products of biological importance, especially the alkaloids.



1886-1975



Robinson *J. Chem. Soc.* 1935, 1285.

Reviews

M. Jung, *Tetrahedron* 1976, 32, 3.
Org. React. 1959, 10, 179.
Org. React. 1968, 16, 3.
Synthesis 1976, 777.
Synthesis 1969, 49.

SZ Vatsadze's lectures

se.

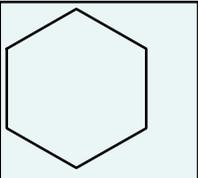
es

SZ Vatsadze's lectures

Robinson *J. Chem. Soc.* 1917, 762. (tropinone)

• Реакция Робинсона

Образование циклов



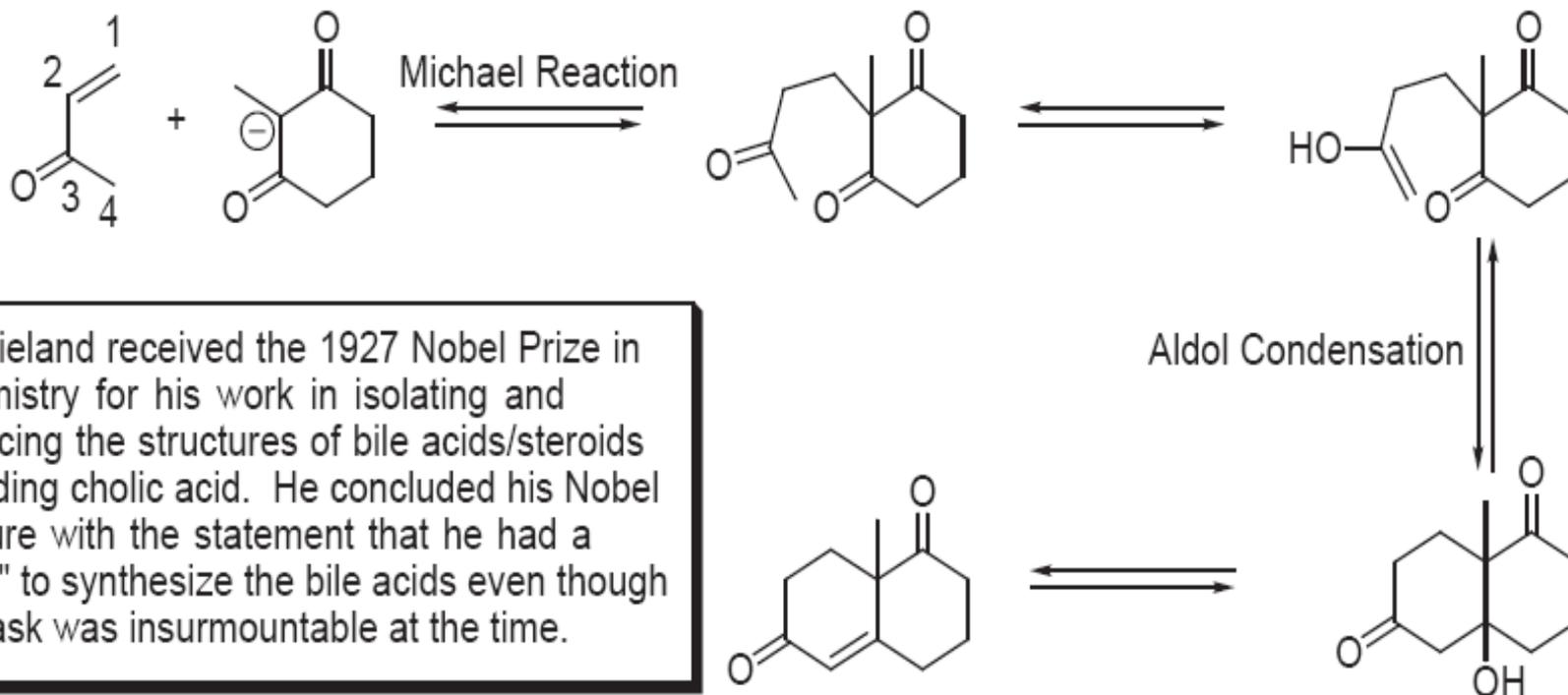
S Z Vatsad

S Z Vatsad

S Z Vatsai

S Z Vatsai

- Formally, a [4 + 2] condensation approach



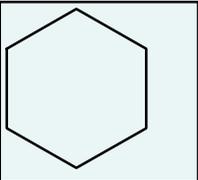
H. Wieland received the 1927 Nobel Prize in Chemistry for his work in isolating and deducing the structures of bile acids/steroids including cholic acid. He concluded his Nobel Lecture with the statement that he had a "duty" to synthesize the bile acids even though the task was insurmountable at the time.

Wieland-Miescher ketone

Wieland and Miescher *Helv. Chim. Acta* 1950, 33, 2215.

Реакция Робинсона

Образование циклов



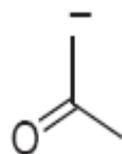
S Z Vatsadze's lectu

S Z Vatsadze's lectu

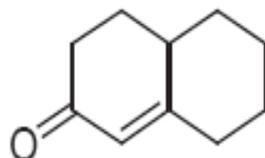
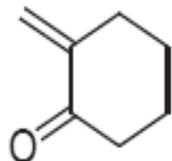
S Z Vatsadze's lectu

S Z Vatsadze's lectu

- Alternative "[3 + 3] Robinson Annulation"



+



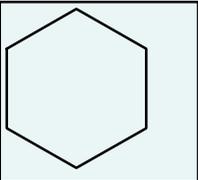
Both the [4 + 2] and [3 + 3] approaches were first generalized by Robinson *J. Chem. Soc.* 1937, 53.

sadze's lectures

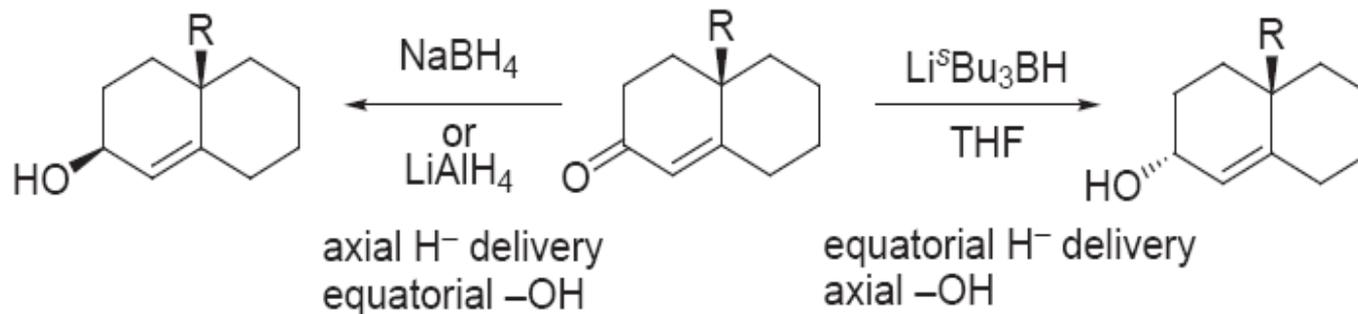
sadze's lectures

sadze's lectures

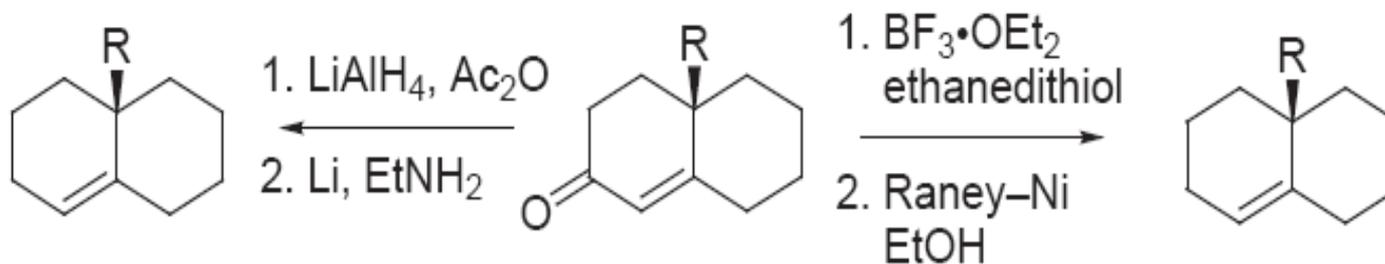
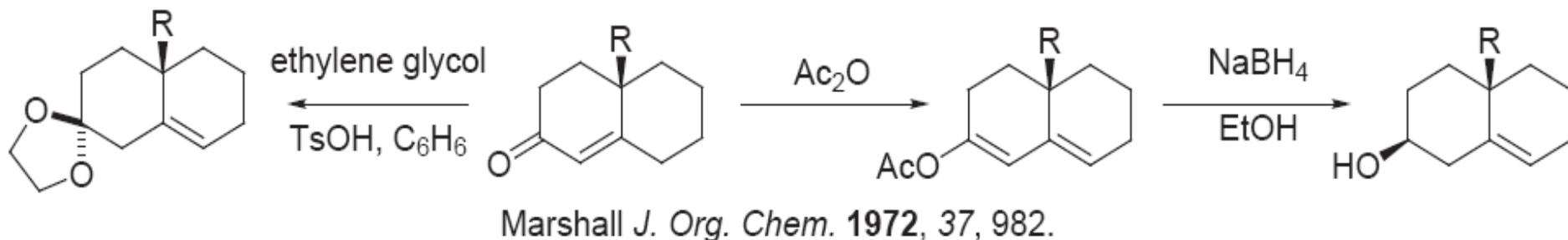
sadze's lectures



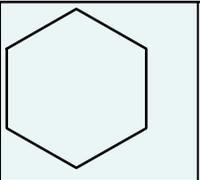
Некоторые синтетические превращения Робинсоновских продуктов



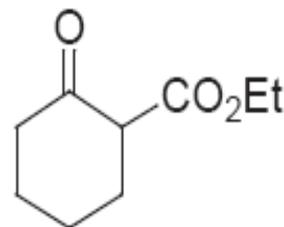
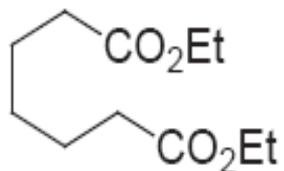
- Deconjugation with ketalization or reduction



- Внутримолекулярное ацилирование енолятов

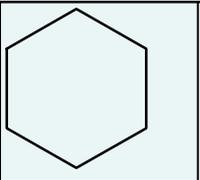


Конденсация Дикмана



• Внутримолекулярное алкилирование енолятов

Образование циклов



S Z Vatsadze's lectures



House *J. Org. Chem.* 1978, 43, 700.

S Z Vatsadze's lectures

S Z Vatsadze's le

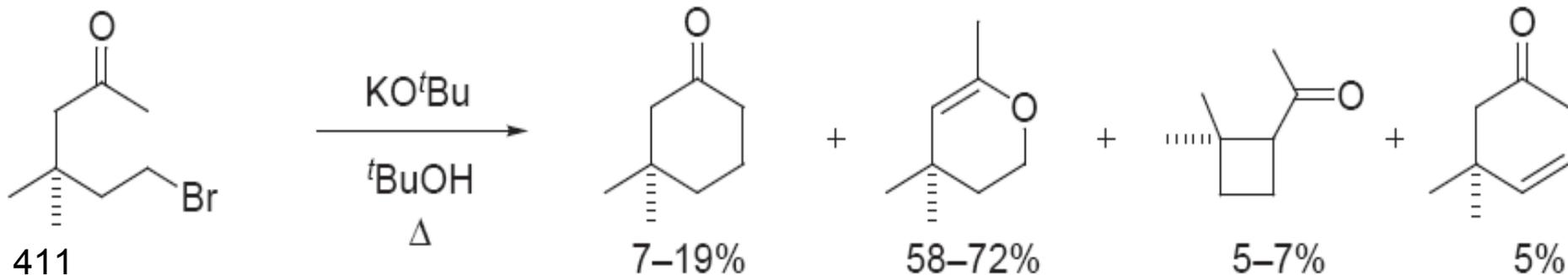


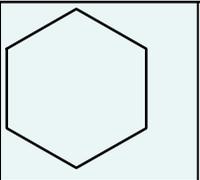
Z Vatsadze's le

Z Vatsadze's le

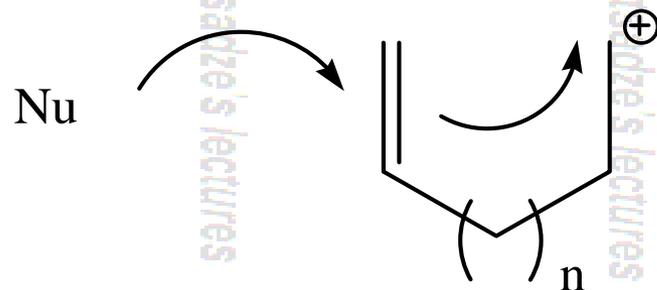
S Z Vatsadze's le

Сравните с т/д условиями:





S Z Vaisadze's lectures



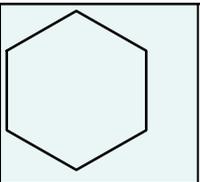
S Z Vaisadze's lectures

S Z Vaisadze's lectures

S Z Vaisadze's lectures

S Z Vaisadze's lectures

Johnson	<i>Acc. Chem. Res.</i> 1968 , 1, 1. <i>Angew. Chem., Int. Ed. Eng.</i> 1976 , 15, 9. <i>Bioorg. Chem.</i> 1976 , 5, 51.
van Tamelen	<i>Acc. Chem. Res.</i> 1968 , 1, 111.
Harding	<i>Bioorg. Chem.</i> 1973 , 2, 248.
Goldsmith	<i>Fortschr. Chem. Org. Nat.</i> 1972 , 29, 363.
Lansbury	<i>Acc. Chem. Res.</i> 1972 , 5, 311.
Speckamp	<i>Recl. Trav. Chim. Pays-Bas.</i> 1981 , 100, 345.
Sutherland	<i>Comprehensive Org. Syn.</i> Vol 3, pp 341–377.



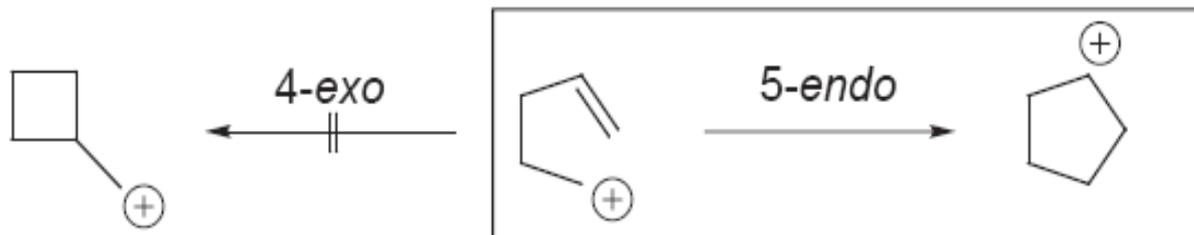
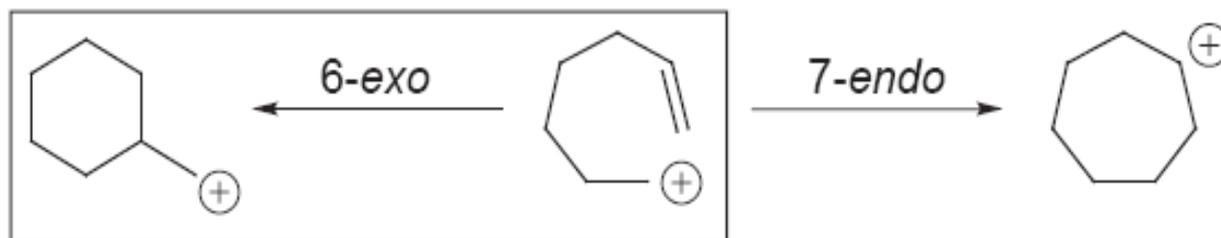
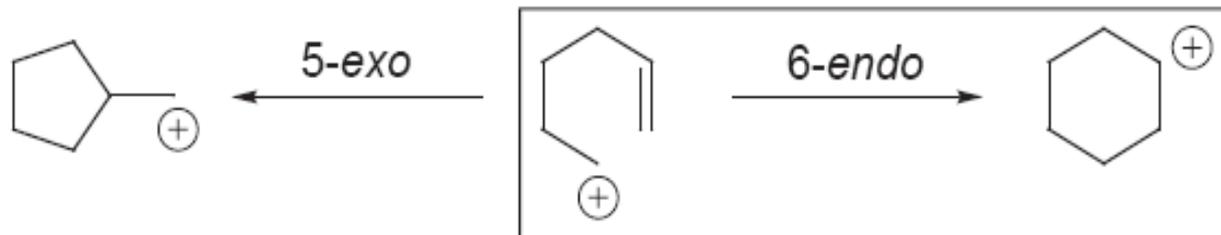
S Z Varsadze

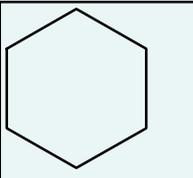
Алкены

S Z Varsadze

S Z Varsadze

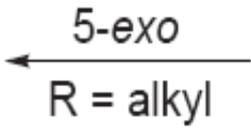
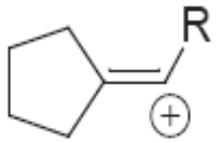
S Z Varsadze



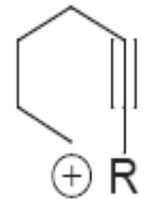


SZ Vats

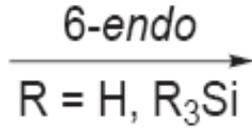
Алкины



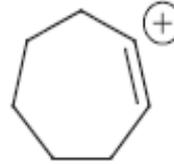
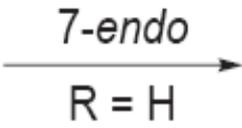
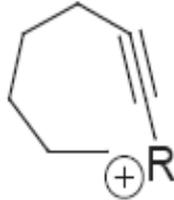
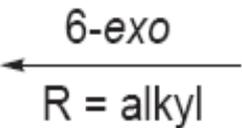
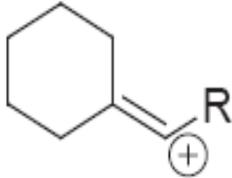
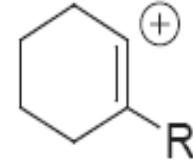
SZ Vats



SZ Var

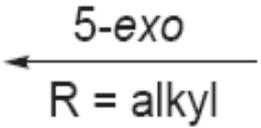
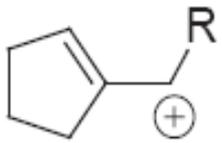


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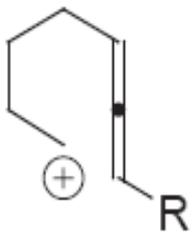


tsadze's lectur

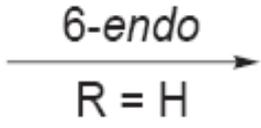
Аллены



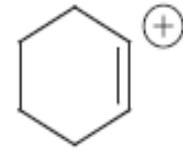
tsadze's lectur

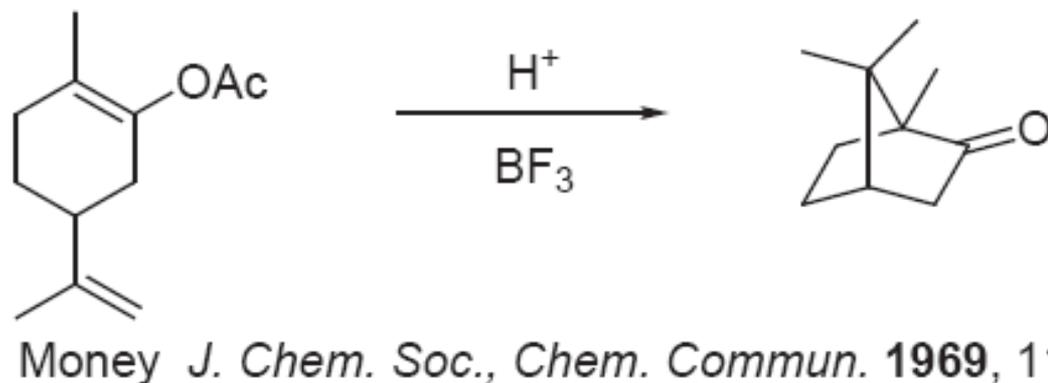
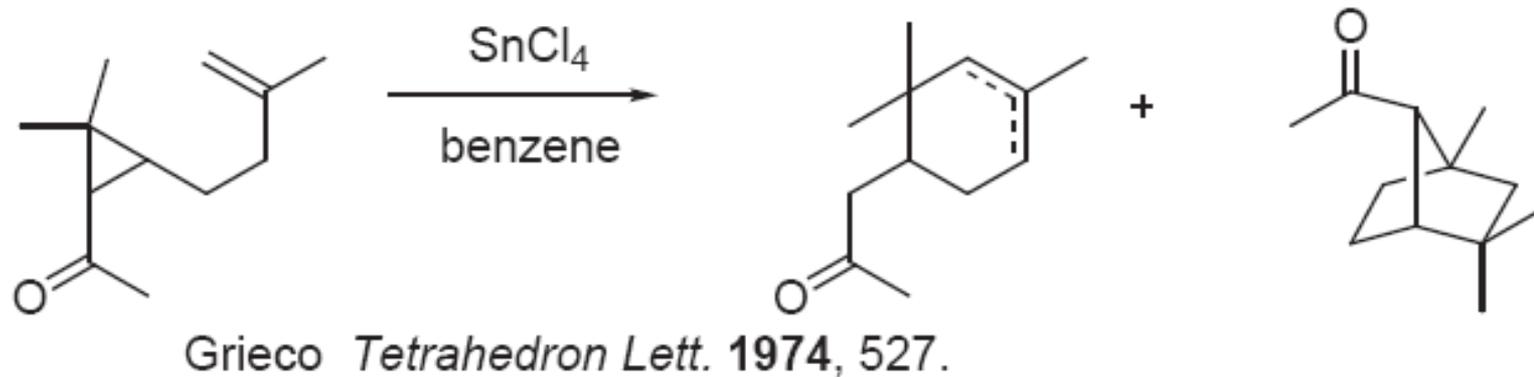
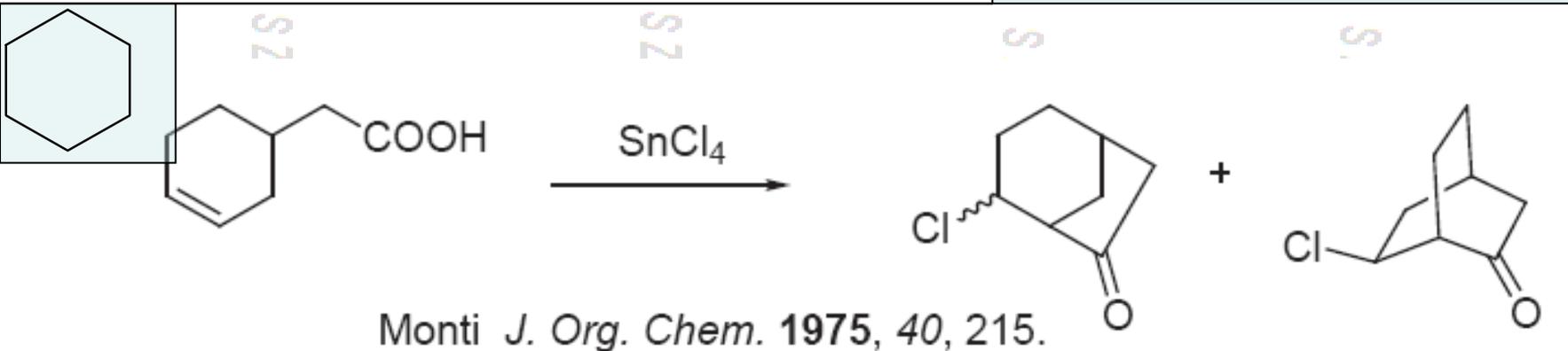


tsadze's lectur



tsadze's lectur

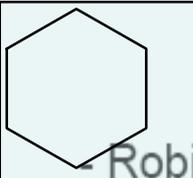




Goldsmith *J. Org. Chem.* 1970, 35, 3573.

• Катион-алкен/алкин

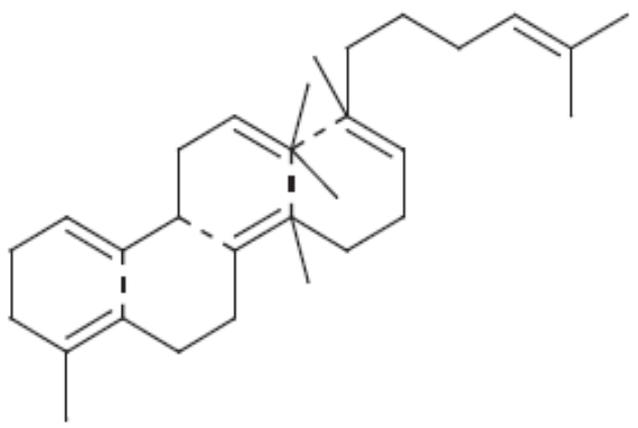
Образование циклов



SZVa

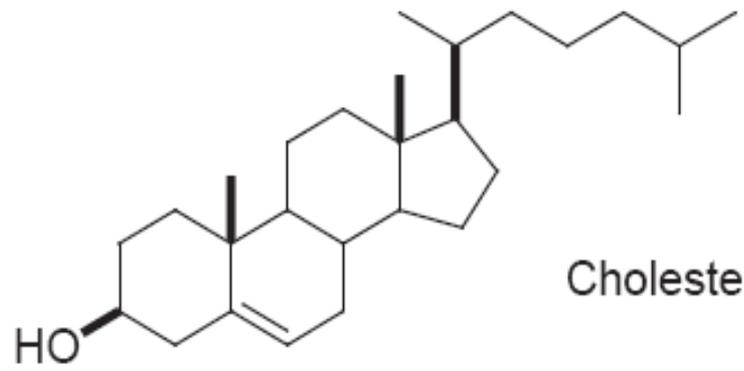
SZVa

- Robinson's proposal



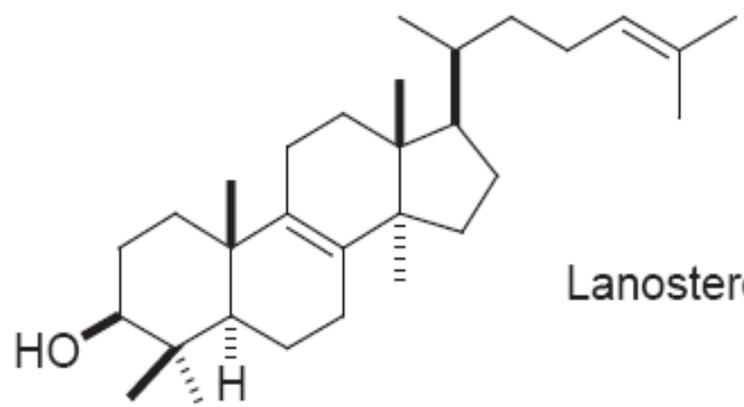
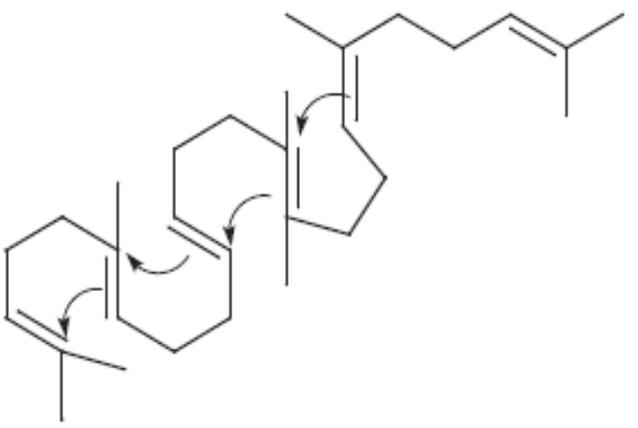
SZV_i

SZV_i



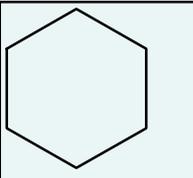
Cholesterol

- Correct cyclization scheme



Lanosterol

416 Lanosterol was proposed in 1953 by Woodward and Bloch.



Гипотеза Сторка-Эшенмозера: транс-анти-транс-стереохимия является результатом согласованной циклизации полиенов

S Z Vaisanze's lectures

S Z Vaisanze's lectures

S Z Vaisanze's lecture

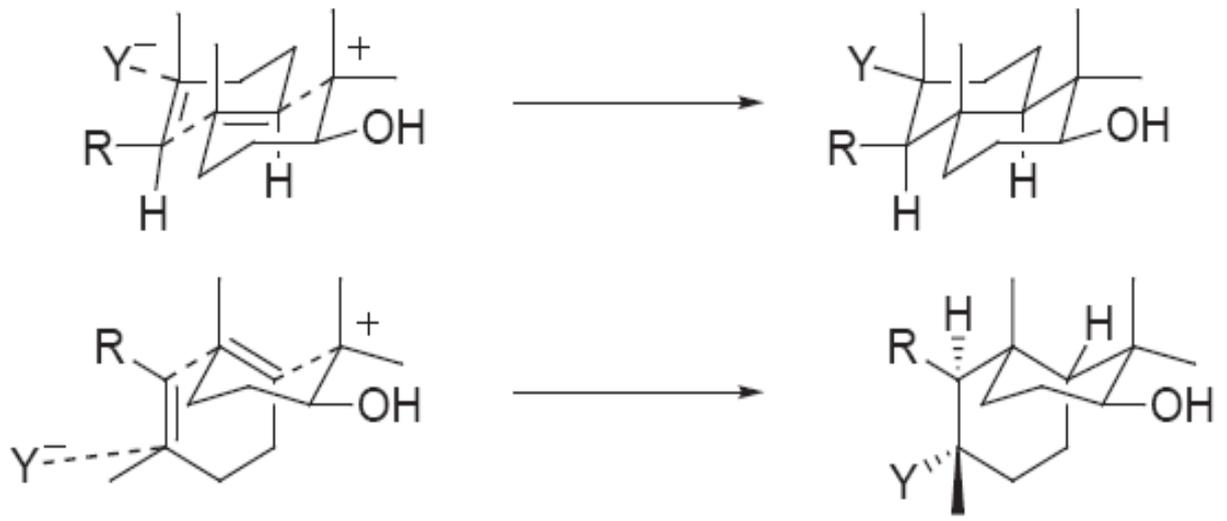
S Z Vaisanze's lecture

S Z Vaisanze's lectures

es

es

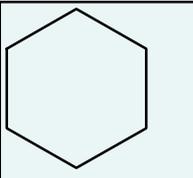
es



Сравните с AdE стереохимией!!!

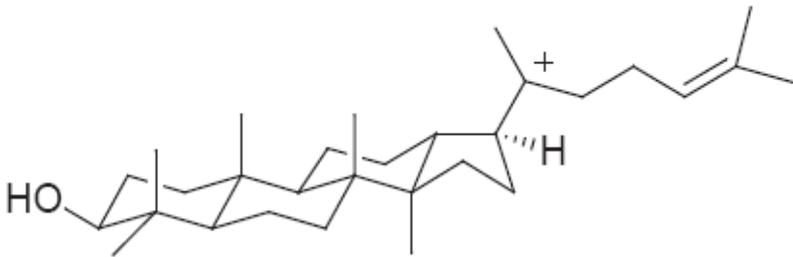
• Катион-алкен/алкин

Образование циклов



S Z Vaisadze's

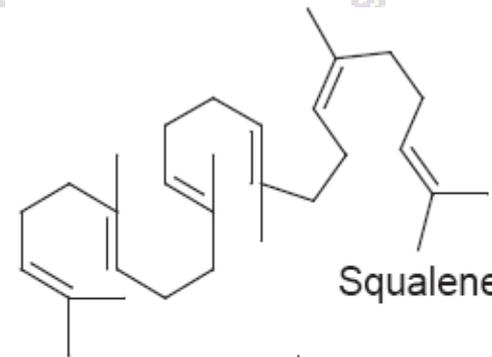
S Z Vaisadze's



418

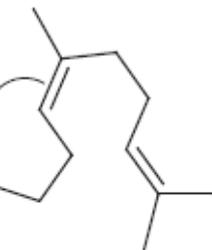
S Z Vaisadze's

S Z Vaisadze's



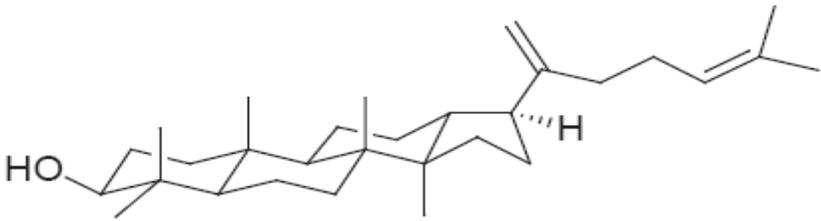
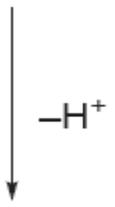
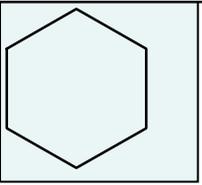
Squalene

Squalene monooxygenase



Squalene-2,3-oxide

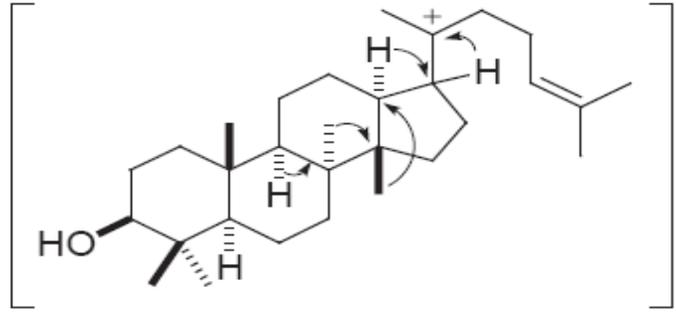
H⁺



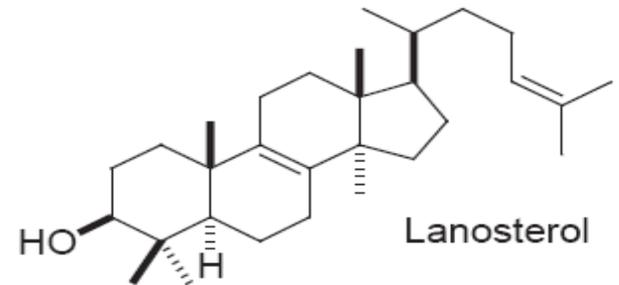
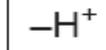
Dammaradienol

8 chiral centers with 256 possible stereoisomers

2,3-Oxidosqualene lanosterol cyclase



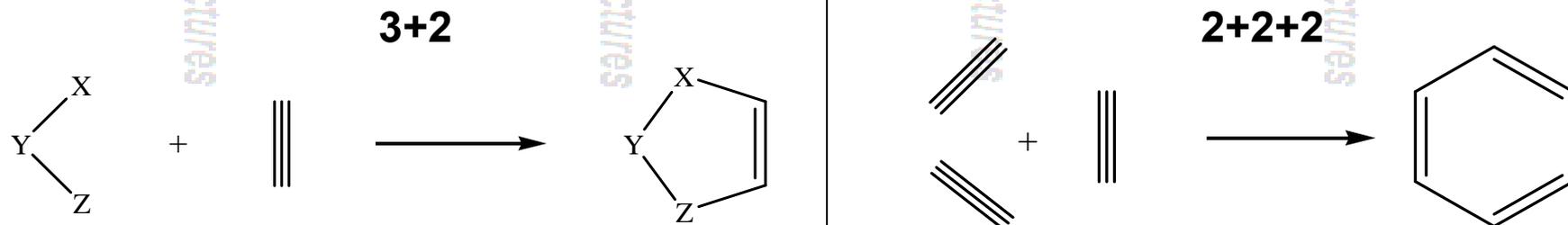
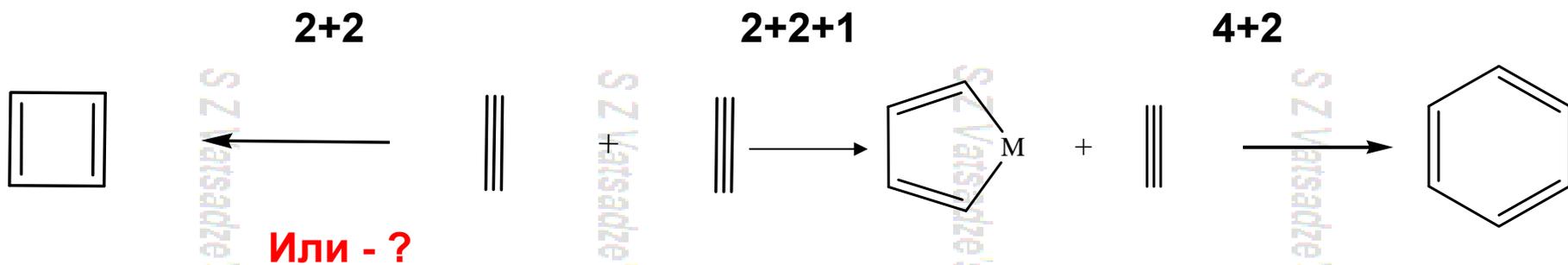
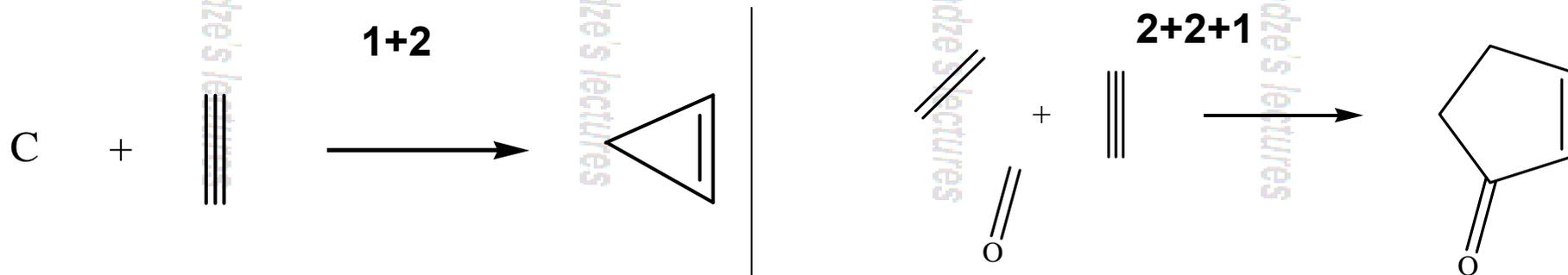
- Two methyl migrations and two hydride transfers



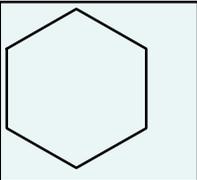
Lanosterol

Общие принципы

Алкин – поставщик C2-фрагмента. Следовательно, можно предложить несколько схем реакций циклизации с участием алкинов



- Фишеровские карбены



С алкинами обычно [3+2+1]

- Six-membered rings [3 + 2 + 1] (Fischer carbene addition to alkynes)

Dötz, *Fischer Transition Metal Carbene Complexes*, VCH: Deerfield Beach, FL, 1983.

Dötz *Angew. Chem., Int. Ed. Eng.* **1984**, 23, 587.

Casey in *Transition Metal Organometallics in Organic Synthesis*, Academic Press: New York, 1976, Vol. 1.

Dötz *Pure Appl. Chem.* **1983**, 55, 1689.

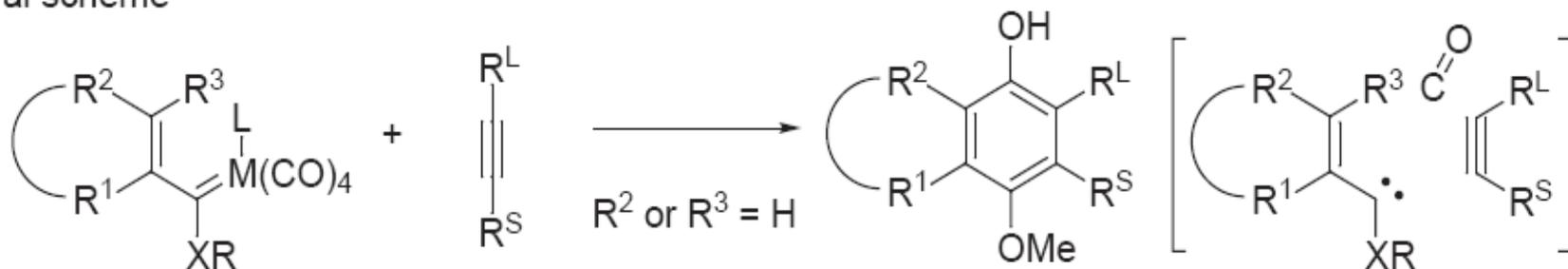
Casey in *Reactive Intermediates*, Wiley Interscience: New York, 1982, Vol. 2, and 1985, Vol. 3.

Hegedus *Principles and Applications of Organotransition Metal Chemistry*, University Science Books: Mill Valley, CA, 1987, 783.

Brown *Prog. Inorg. Chem.* **1980**, 27, 1.

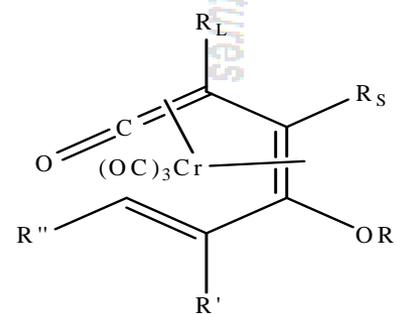
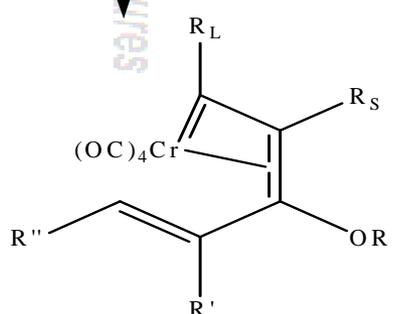
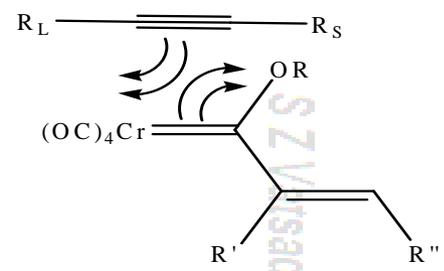
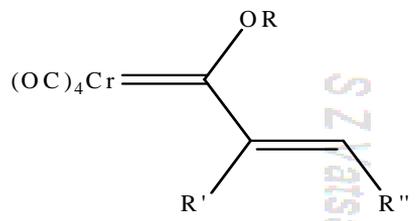
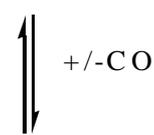
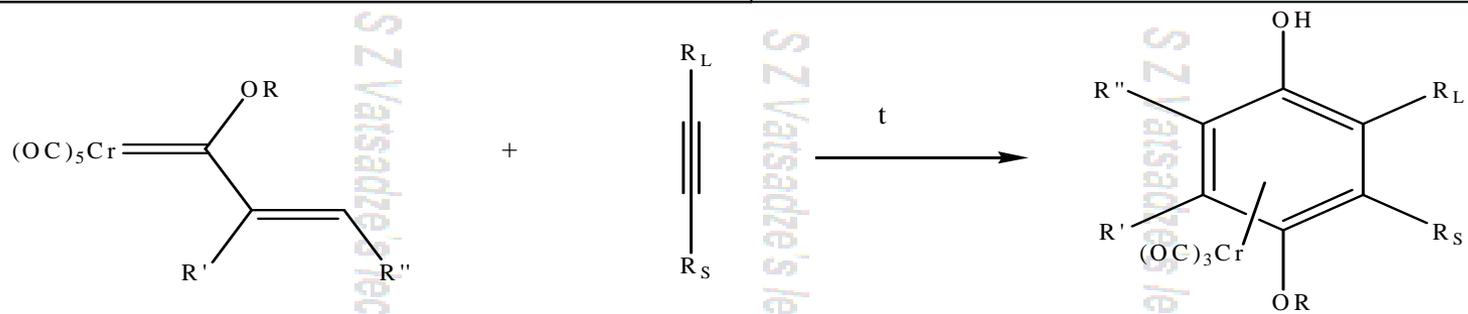
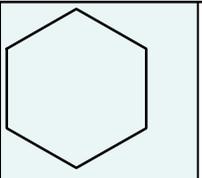
Wulff in *Advances in Metal-Organic Chemistry*, JAI Press: Greenwich, CT, 1989, Vol. 1.

- General scheme



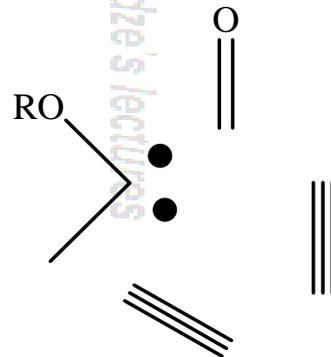
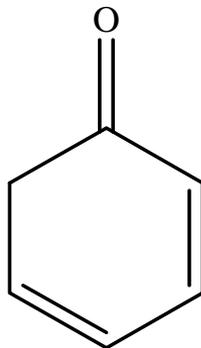
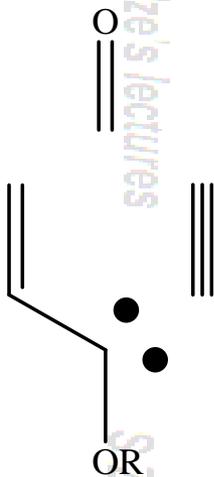
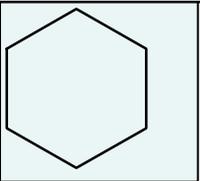
• 2+2: применение - реакция Дётса

Образование циклов

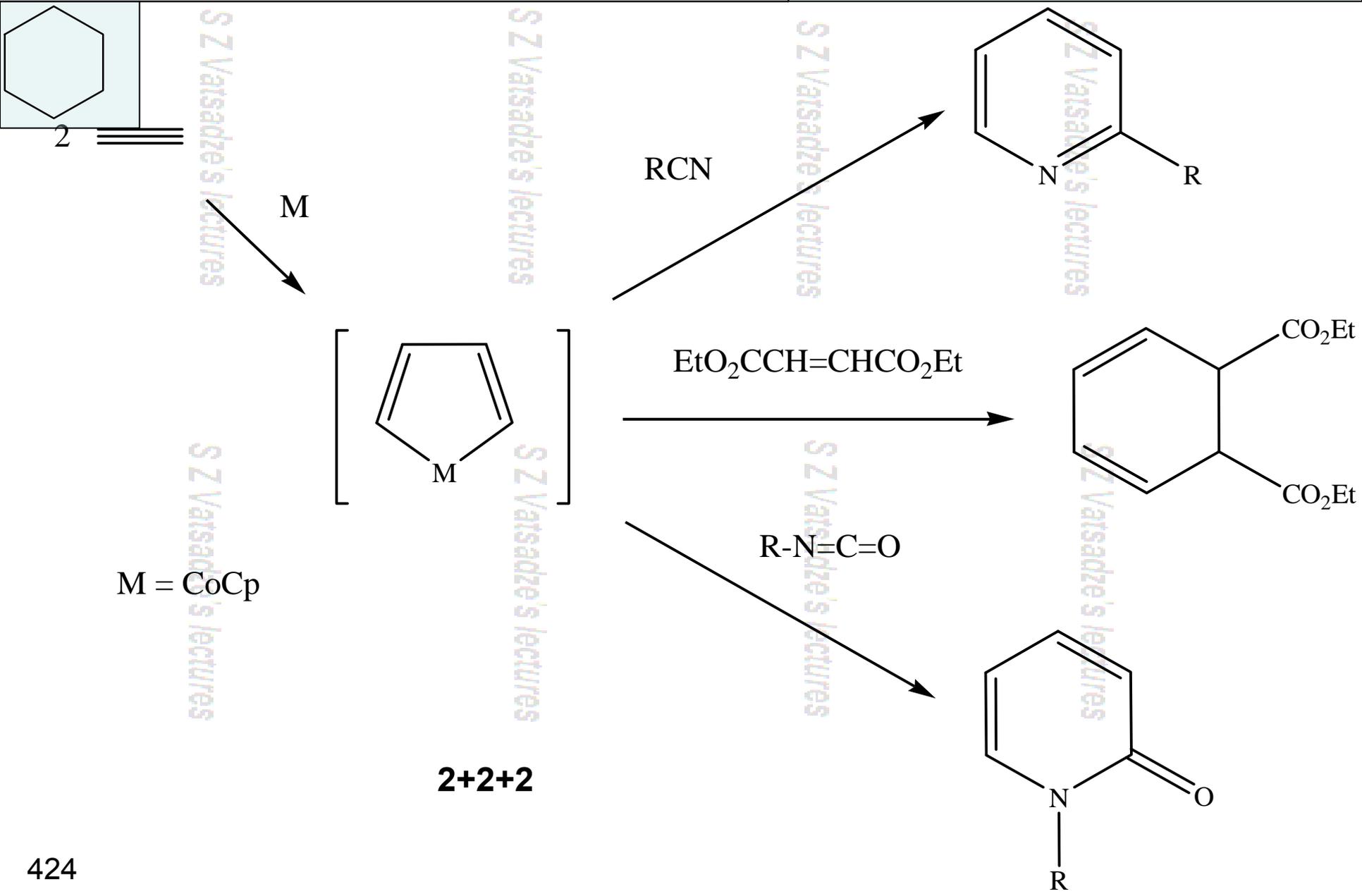


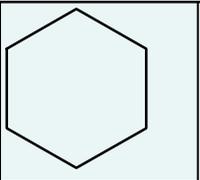
Закончить!

- 2+2: применение - реакция Дётса - обобщение



Каскадные реакции с металлами $(2+2)^n$

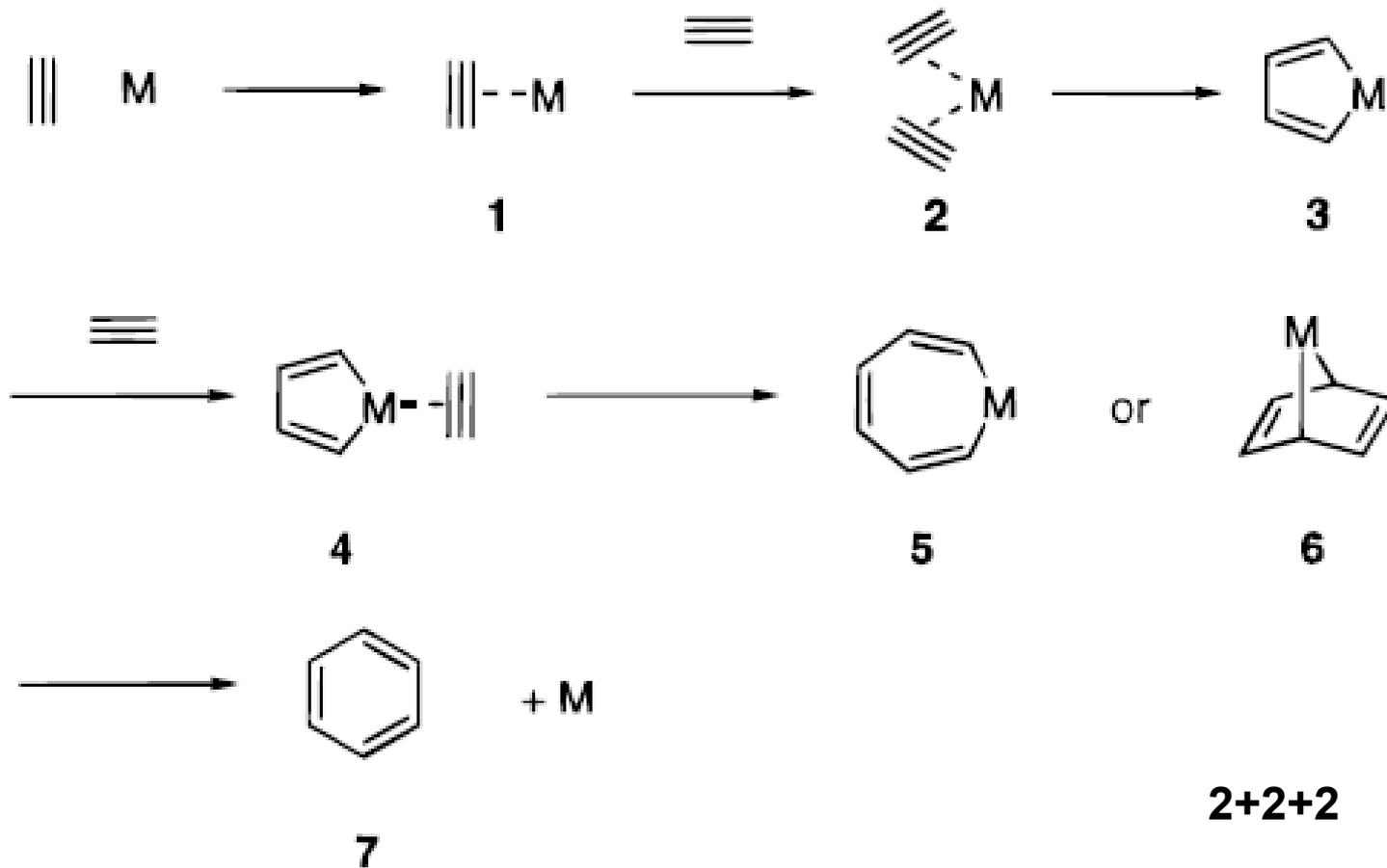


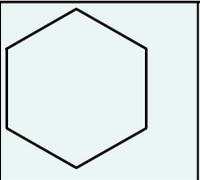


S Z Vaisadze's lect

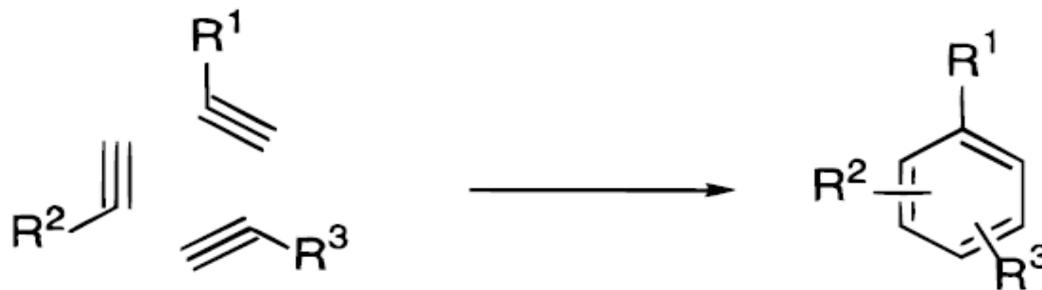
Recent Advances in the Transition-Metal-Catalyzed Regioselective Approaches to Polysubstituted Benzene Derivatives

Shinichi Saito* and Yoshinori Yamamoto*

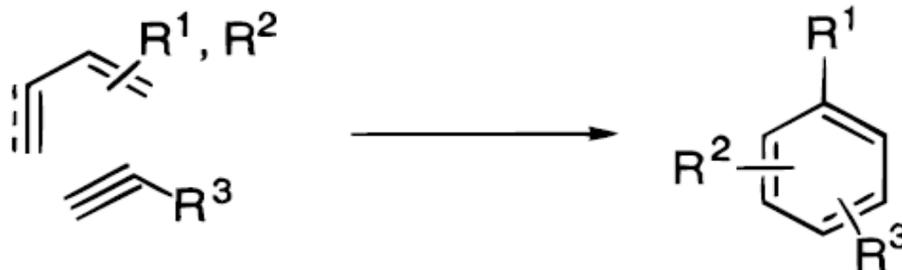




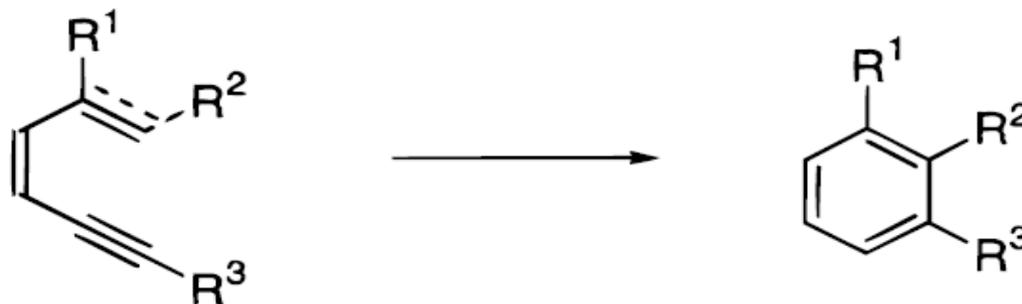
S Z Vatsadze's lectures



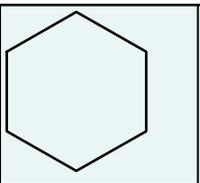
S Z Vatsadze's lectures



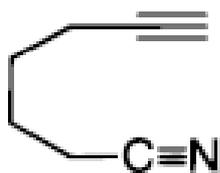
S Z Vatsadze's



2+2+2



S Z Vaisadz

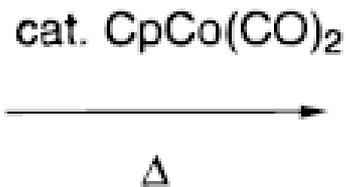


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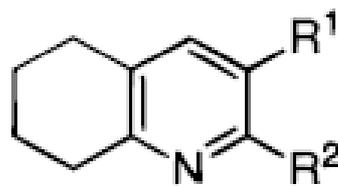


29



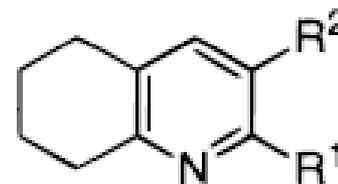
S Z Vaisadz

S Z Vaisadz



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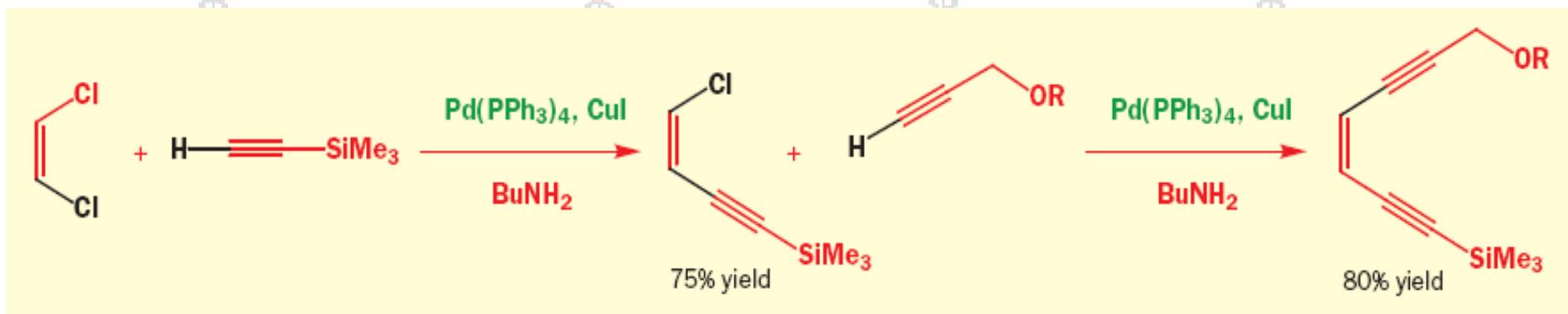
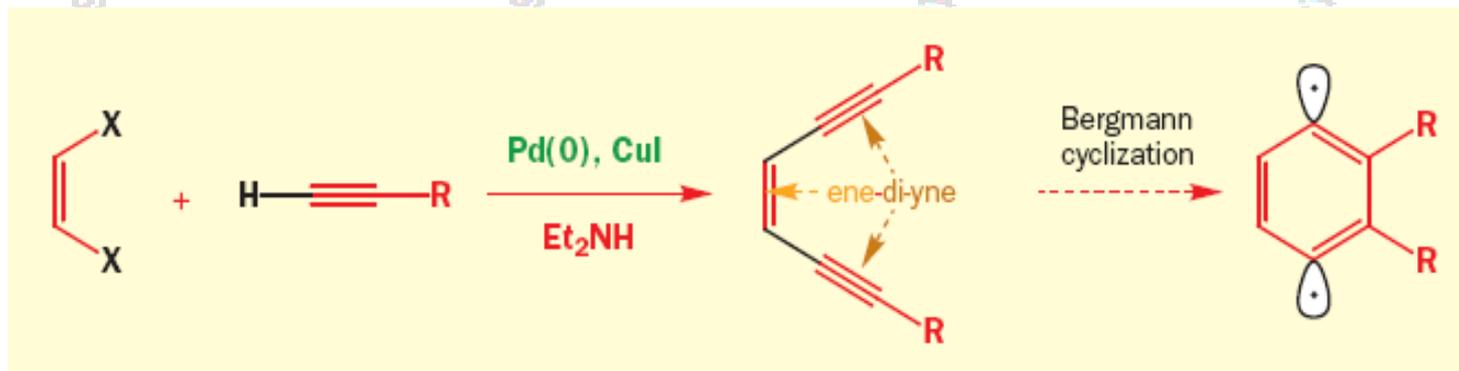
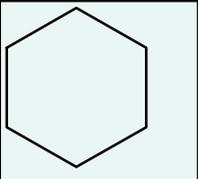
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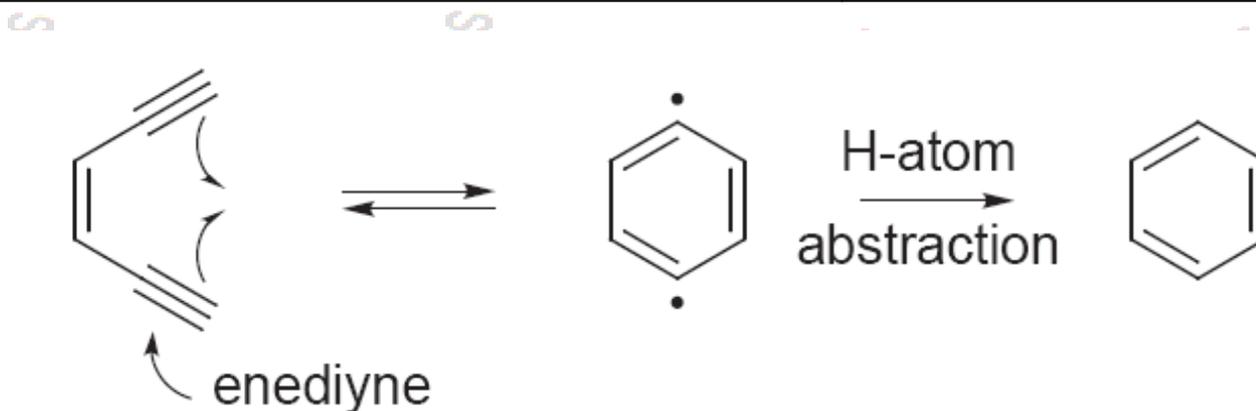
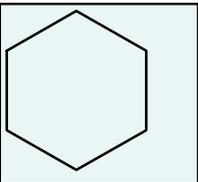
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• Циклизация по Бергману

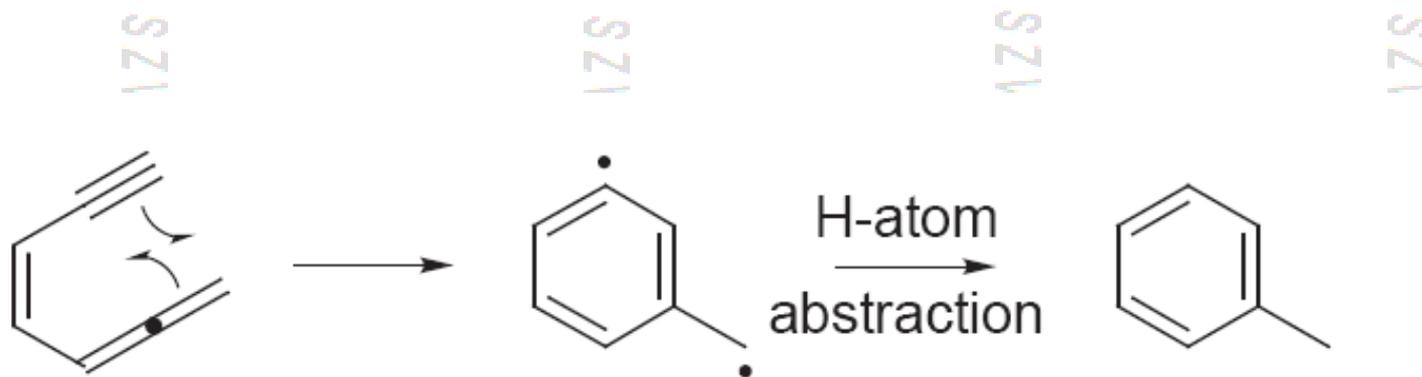


• Циклизация по Бергману и Майерсу-Сайто

Образование циклов

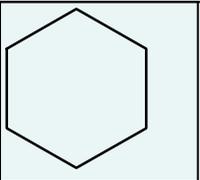


Bergman *J. Am. Chem. Soc.* **1972**, 94, 660.
Acc. Chem. Res. **1973**, 6, 25.



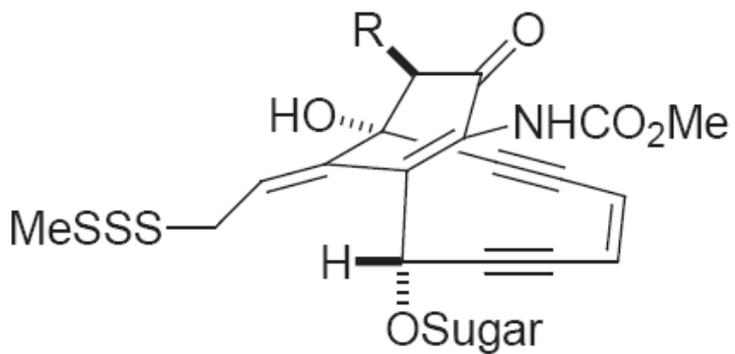
Myers *J. Am. Chem. Soc.* **1988**, 110, 7212; **1992**, 114, 9369.

- Циклизация по Бергману: эндиновые антибиотики



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S Z Vaisadze's k

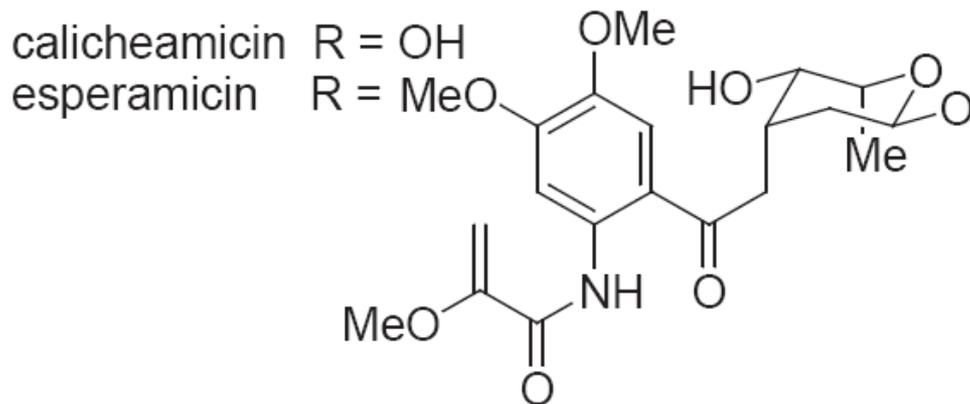


idze's lectures

idze's lectures

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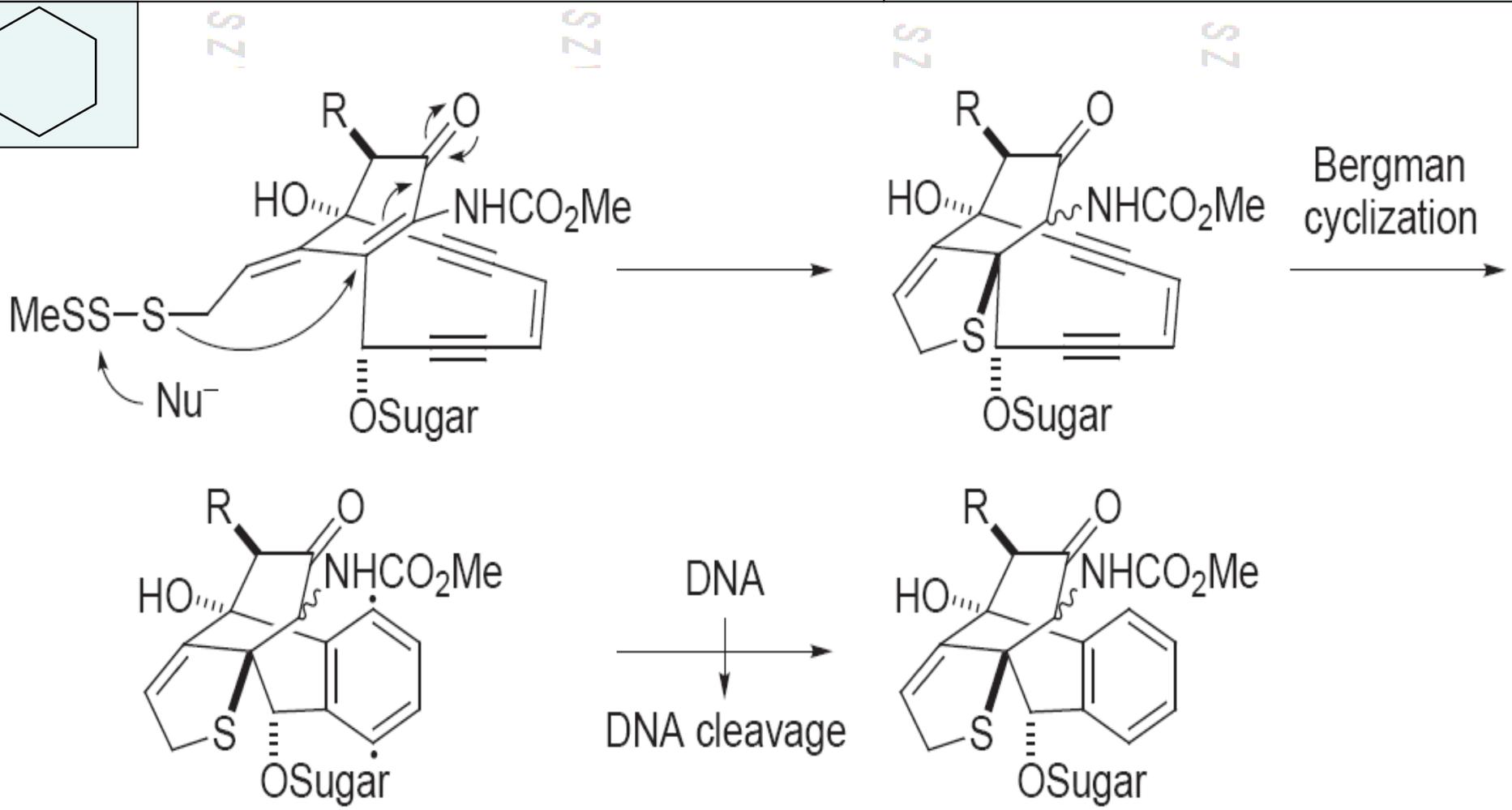
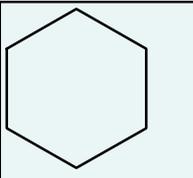


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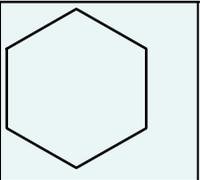
idze's lectures

• Циклизация по Бергману: ендиновые антибиотики

Образование циклов



- Циклизация по Майерсу-Сайто: ендиновые антибиотики

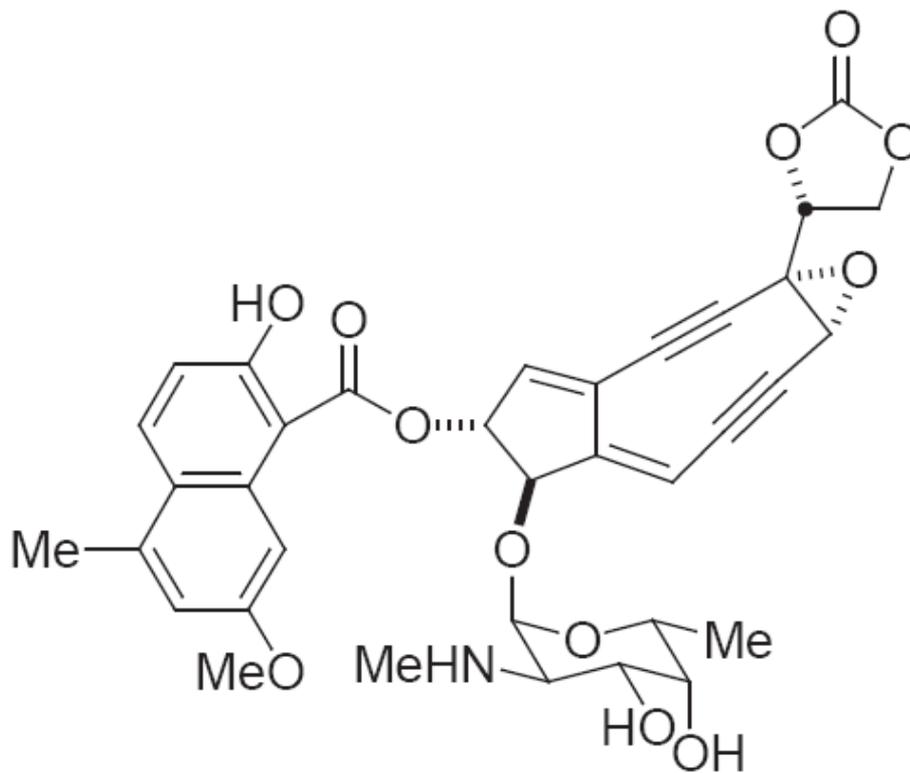


S Z Vatsadze's lectures

S Z Var

S Z Va

S Z Vatsadze's lectures



neocarzinostatin chromophore

S Z Vatsadze's lectures

S Z Vatsadze's lectures

• Циклизация по Майерсу-Сайто: эндиновые антибиотики

Образование циклов

